

## Long-term records of cadmium and silver contamination in sediments and oysters from the Gironde fluvial–estuarine continuum – Evidence of changing silver sources

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### Abstract:

The Gironde fluvial estuarine system is impacted by historic metal pollution (e.g. Cd, Zn, Hg) and oysters (*Crassostrea gigas*) from the estuary mouth have shown extremely high Cd concentrations for decades. Based on recent work (Chiffoleau et al., 2005) revealing anomalously high Ag concentrations (up to 65 mg kg<sup>-1</sup>; dry weight) in Gironde oysters, we compared long-term (~1955–2001) records of Ag and Cd concentrations in reservoir sediment with the respective concentrations in oysters collected between 1979 and 2010 to identify the origin and historical trend of the recently discovered Ag anomaly. Sediment cores from two reservoirs upstream and downstream from the main metal pollution source provided information on (i) geochemical background (upstream; Ag: ~0.3 mg kg<sup>-1</sup>; Cd: ~0.8 mg kg<sup>-1</sup>) and (ii) historical trends in Ag and Cd pollution. The results showed parallel concentration-depth profiles of Ag and Cd supporting a common source and transport. Decreasing concentrations since 1986 (Cd: from 300 to 11 mg kg<sup>-1</sup>; Ag: from 6.7 to 0.43 mg kg<sup>-1</sup>) reflected the termination of Zn ore treatment in the Decazeville basin followed by remediation actions. Accordingly, Cd concentrations in oysters decreased after 1988 (from 109 to 26 mg kg<sup>-1</sup>, dry weight (dw)), while Ag bioaccumulation increased from 38 up to 116 mg kg<sup>-1</sup>, dw after 1993. Based on the Cd/Ag ratio (Cd/Ag ~2) in oysters sampled before the termination of zinc ore treatment (1981–1985) and assuming that nearly all Cd in oysters originated from the metal point source, we estimated the respective contribution of Ag from this source to Ag concentrations in oysters. The evolution over the past 30 years clearly suggested that the recent, unexplained Ag concentrations in oysters are due to increasing contributions (>70% after 1999) by other sources, such as photography, electronics and emerging Ag applications/materials.

### Highlights

► Ag and Cd records in river sediment support common release by former ore-treatment. ► Cd levels in oysters reflect Cd decrease in sedimentary records after 1988. ► Ag levels in oysters reach maximum values between 1993 and 2003. ► Ag accumulation in oysters was attributed to additional sources.

**Keywords:** Silver ; Cadmium ; Estuary ; Oyster ; Biomonitoring

### 1. Introduction

The Gironde Estuary (Fig. 1) is known for its metal pollution since high Cd accumulation (> 100 µg g<sup>-1</sup>, dry weight) in oysters (*Crassostrea gigas*) was discovered in the end of 1970s (e.g. Claisse, 1989) resulting in prohibition of consumption, production and purification of

50 oysters since 1996 (e.g. Audry et al., 2004a). The main source of metal pollution was coal  
51 mining and Zn-ore treatment in the Decazeville watershed. Since the cessation of this  
52 metallurgical industry (1986), remediation and pollution management have clearly reduced  
53 Cd emission (Audry et al., 2004a). In the Gironde watershed, suspended particles are major  
54 vectors for Cd, Ag and other metals (e.g. Audry et al., 2004a; Lanceleur et al. 2011,). Part of  
55 these particles accumulate in hydroelectric reservoirs, where the pollutants may be recycled  
56 (e.g. released to porewaters; Roulier et al., 2010) or stored over decades, providing  
57 sedimentary records of past pollutions (Audry et al., 2004b; Castelle et al., 2007; Valette-  
58 Silver, 1993). The hydroelectric reservoirs on the Lot River downstream from the Decazeville  
59 watershed contain huge sediment stocks with high metal loads ( $9.47 \pm 0.9 \cdot 10^6 \text{ m}^3$  of sediment  
60 containing approximately 200 t of Cd), that may be remobilized by floods, dredging and  
61 construction work (Audry et al., 2004a; Coynel et al., 2007). Although Cd has long been  
62 considered as the major pollutant, many other metals derived from the same source also show  
63 important anomalies in both sediment and water (e.g. Hg, Cu, Zn; Castelle et al., 2007;  
64 Schäfer et al., 2006). Furthermore, the regional geology and agricultural activities also  
65 contribute (e.g. As, Cu, Zn) to the metal/metalloid loads in the Gironde watershed (Masson et  
66 al., 2006, 2007). Recent work has revealed remarkably high Ag concentrations (up to 60  
67  $\text{mg kg}^{-1}$  in 2003/2004) in wild oysters at the mouth of the Gironde Estuary (Fig. 1A; i.e. the  
68 highest measured in oysters and mussels along the French coast; Chiffolleau et al., 2005),  
69 indicating the fluvial-estuarine system is also contaminated by Ag. Both Ag and Cd are highly  
70 toxic to aquatic organisms and may be easily bioaccumulated (Eisler, 1996; US EPA, 2001;  
71 Vijayavel et al., 2007), especially by oysters (e.g. Gunther et al., 1999; Ivanina et al., 2011;  
72 Reinfelder et al., 1997). The measured concentrations in the Gironde oysters clearly exceed  
73 safety limits for human consumption for Cd ( $5 \text{ mg kg}^{-1}$ ; dw; EC No 466/2001) and although  
74 there is no legal threshold for Ag, consumption of oysters from several sites along the French

75 coast (Chiffoleau et al., 2005) may be incompatible with human maximum Ag daily intake  
76 recommended by the EPA ( $0.005 \text{ mg kg}^{-1} \text{ day}^{-1}$ ; [www.epa.gov/iris/subst/0099.htm](http://www.epa.gov/iris/subst/0099.htm)).  
77 The present work aimed at documenting for the first time the historical Ag records in  
78 reservoir sediments upstream and downstream from the Decazeville watershed to (i) establish  
79 geochemical background for Ag in the upper Lot River watershed and (ii) quantify the extent  
80 of historical Ag contamination by the metal point source (Fig. 1B). Furthermore, we compare  
81 and discuss the chronologies of Cd and Ag accumulation in reservoir sediments and oysters  
82 from the Gironde watershed to understand (i) their relationships and (ii) changes over time in  
83 both concentrations and potential sources.

84

## 85 **2. Materials and Methods**

86

### 87 **2.1. Study area**

88 The Gironde Estuary (~170 km length; Fig. 1), draining a 80,000 km<sup>2</sup> watershed, is a major  
89 European estuary. Its mean annual freshwater discharge varies between 800 and 1000 m<sup>3</sup> s<sup>-1</sup>,  
90 with typical water and particle residence times of ~20–90 days and ~1–2 years, respectively  
91 (Sottolichio and Castaing, 1999). Suspended Particulate Matter (SPM) concentrations in its  
92 well developed Maximum Turbidity Zone (MTZ) frequently exceed  $1 \text{ g L}^{-1}$ , making the  
93 Gironde Estuary one of the most turbid estuaries in Europe. A high number of studies on  
94 physical, hydrological and geochemical characteristics of the Gironde Estuary and its  
95 watershed (e.g. metal pollution) make it a model estuary, especially for understanding trace  
96 element transport and reactivity in impacted watersheds and macrotidal estuaries (e.g. Schäfer  
97 et al., 2002; Sottolichio and Castaing, 1999).

98 The main source of the historical metal pollution is located in the upper Lot River watershed,  
99 where a small tributary (i.e. the Riou Mort River) drains a waste area of a former zinc ore

100 manufacturing facility (1842 to 1987) in the Decazeville basin (Fig. 1B). The metallurgical  
101 activity ended after a major pollution accident in 1986 followed by ongoing remediation  
102 efforts such as the treatment of drainage water and the confinement of a part of the waste  
103 deposits in storage basins (1987). However, the major part of the metallurgical waste is still  
104 exposed to atmospheric agents, i.e. leaching and erosion continue (e.g. Audry et al., 2004b).

105

## 106 **2.2. Core retrieval and sampling**

107 Reservoir sediments usually accumulate at high rates (typically  $> 2 \text{ cm y}^{-1}$ ; e.g. Müller et al.,  
108 2000, Audry et al., 2004b) and consequently are considered as being little affected by early  
109 diagenesis processes, i.e. providing preserved records of historical inputs (Callender, 2000).  
110 In June 2001, sediment cores (0–140 cm depth) were collected in two hydroelectric reservoirs  
111 on the Lot River, upstream (Marcenac) and downstream (Cajarc) from the pollution source  
112 (Fig. 1B) as described elsewhere (Audry et al., 2004b). Briefly, horizontal sections of the  
113 cores were sliced, dried at  $50^{\circ}\text{C}$  to constant weight, powdered and homogenized with an agate  
114 mortar and stored in the dark awaiting analyses.

115

## 116 **2.3. Bioindicator sampling**

117 As suspension-feeding bivalves, oysters are exposed to both dissolved metals and metals  
118 bound to mineral particles, phytoplankton and organic matter. Their capacity to accumulate  
119 both Ag and Cd from their environment (e.g. Baudrimont et al., 2005; Gunther et al., 1999)  
120 makes them good bioaccumulators for monitoring studies. Within the National Network for  
121 the Observation of Marine Environment Quality (RNO/ROCCH; i.e. the French Mussel-Watch;  
122 <http://www.ifremer.fr/deltn/pages/rno.htm>), 2-year old wild-growing Japanese oysters  
123 (*Crassostrea gigas*) were collected according to the guidelines for monitoring contaminants in  
124 biota defined by the OSPAR commission (<http://www.ospar.org>). Three sites at the mouth of

125 the Gironde Estuary (Fig. 1A) have been sampled four times a year during 1979-2002, twice a  
126 year from 2003 to 2007 (February and November) and annually (February) since 2008. The  
127 sampling sites are located in the downstream part of the Gironde Estuary dominated by  
128 marine sand and gravel with some shell fragments (Larrose et al., 2010). Wild oyster banks at  
129 the Bonne Anse and La Fosse sites are surrounded by silty/sandy sediment, whereas at  
130 Pontailiac they are close to rocky outcrops (calcareous cliffs). The three sites are  
131 representative of different salinity ranges with typical salinity ranges being 25 to 30 at the  
132 Bonne Anse site, 20 to 25 at the Pontailiac site and 15 to 20 at the La Fosse site.

133

#### 134 **2.4. Analyses**

135 All the labware used was acid cleaned (HNO<sub>3</sub> analytical grade; 10%; 72 h), thoroughly rinsed  
136 with Milli-Q water, dried under a laminar flow hood and stored in double sealed polyethylene  
137 bags prior to use. For the determination of total trace element concentrations in reservoir  
138 sediments, representative sub-samples (30 mg of dry, powdered and homogenized material)  
139 were digested in acid-cleaned PP-tubes (DigiTUBEs<sup>®</sup>, SCP SCIENCE) using 1.5 mL HCl  
140 (12 M, Plasmapur), 0.5 mL HNO<sub>3</sub> (14 M, Plasmapur) and 2 mL HF (26 M, Plasmapur) at  
141 110°C for 2 h in a temperature-controlled digestion system (DigiPREP MS<sup>®</sup>, SCP  
142 SCIENCE). After complete cooling, the digestate was evaporated to dryness, then the residue  
143 was dissolved using 150 µL HNO<sub>3</sub> (12 M, Plasmapur), brought to 10 mL with double  
144 deionized Milli-Q water and stored in the dark at 4°C awaiting analysis (adapted from Audry  
145 et al., 2004a and Schäfer and Blanc, 2002). The Cajarc core was <sup>137</sup>Cs-dated using a low  
146 background noise  $\gamma$ -ray spectrometer with Ge detector (Intertechnique EGSP 2200–25; Audry  
147 et al., 2004b). Cadmium and Ag bioaccumulation in oysters was measured from aliquots  
148 (200 mg) of freeze-dried samples digested at 90°C with HNO<sub>3</sub> for Cd and a mixture of HNO<sub>3</sub>  
149 and HCl for Ag as described elsewhere (Chiffolleau et al., 2005).

150 Cadmium and Ag concentrations in the sediment digestates were measured by ICP-MS (X7,  
151 Thermo-Fisher) at EPOC, Bordeaux 1 University. Precision, evaluated from parallel analyses  
152 of international certified reference materials (NCS-DC-70317, NCS-DC-70311), was < 5%  
153 (rsd) and results were consistently within the certified ranges (Table 1; Audry et al., 2004b;  
154 Lanceleur et al., 2011).

155 Cadmium and Ag concentrations in oyster digestates were determined at Ifremer, Centre de  
156 Nantes by graphite furnace atomic absorption spectrometry (AA800, Varian) using the  
157 Zeeman background correction for non-specific absorption. Quality control was satisfactorily  
158 performed by analyses of CRM (CRM TORT-2, CRM DOLT-2, SRM 1566b; Table 1;  
159 Chiffoleau et al., 2005) and through regularly successful participations in QUASIMEME  
160 (Quality Assurance of Information for Marine Environmental Monitoring in Europe)  
161 intercalibration exercises.

162 Flux estimates of particulate Cd and Ag gross fluxes (fluvial fluxes into the estuary) at the La  
163 Réole site, i.e. the main fluvial entry of the Gironde Estuary were performed from daily  
164 discharge and SPM (suspended particulate matter) data and monthly measured metal  
165 concentrations using commonly applied methods described in detail elsewhere (e.g. Audry et  
166 al., 2004a; Masson et al., 2007; Schäfer et al. 2002). The particulate Ag fluxes for the 1990-  
167 1998 period (no Ag data available) were estimated from the SPM fluxes (Schäfer et al., 2002)  
168 and using a particulate Ag concentration of  $1.0 \text{ mg kg}^{-1}$ , i.e. the mean particulate Ag  
169 concentration measured during the last decade at the La Réole site (Lanceleur et al., 2011).

170

### 171 **3. Results**

172

#### 173 **3.1. Ag and Cd concentrations in sediment cores**

174 At the Marcenac site, which served as a reference site for the geochemical background of the  
175 upper Lot River watershed (Audry et al., 2004b), concentrations (mean  $\pm$  SD) were low and  
176 rather constant, i.e.  $0.31 \pm 0.06 \text{ mg kg}^{-1}$  for Ag and  $0.77 \pm 0.22 \text{ mg kg}^{-1}$  for Cd (Fig. 2). In the  
177 Cajarc sediment, directly affected by the former industrial activity in the Decazeville  
178 watershed (e.g. Audry et al., 2004b; Castelle et al., 2007), Ag concentrations ranged from  
179  $0.43$  to  $6.96 \text{ mg kg}^{-1}$  and were correlated with Cd concentrations ( $R_{\text{Spearman}} = 0.90$ ;  $n = 43$ ;  
180  $p < 0.01$ ). Concentrations of both Ag and Cd decreased from the 42 cm depth to the surface  
181 (Cd: from 300 to  $11 \text{ mg kg}^{-1}$ ; Ag: from 6.7 to  $0.43 \text{ mg kg}^{-1}$ ), corresponding to the 1986-2001  
182 period according to  $^{137}\text{Cs}$  records (Fig. 2). Normalization of metal concentrations by Sc  
183 presented in previous work showed that grain size variations had a negligible impact on the  
184 historical metal record (Audry et al., 2004b).

185

### 186 **3.2. Ag and Cd concentrations in oysters**

187 Total metal concentrations in oysters, expressed as  $\text{mg kg}^{-1}$  dry weight (dw) of oyster tissue,  
188 undergo seasonal variations due to weight gains and losses during the reproduction cycle  
189 (Figs. 3A and 3B; Páez-Osuna et al., 1995). In order to (i) limit scattering induced by  
190 seasonal environmental parameters on oyster life cycles and (ii) account for different  
191 sampling frequencies (see sampling section) we used only the February data obtained during  
192 the whole observation period for the comparison of historical data with recent values  
193 (Figs. 3C to 3G).

194 Oysters from the La Fosse site (the most upstream site) showed systematically higher  
195 concentrations in both Cd and Ag, while the lowest concentrations occurred at the Bonne  
196 Anse site (Figs. 3C and 3D). Cadmium concentrations in oysters at the La Fosse site  
197 decreased after 1988 from 109 to  $26 \text{ mg kg}^{-1}$ , dw, while Ag bioaccumulation increased from  
198  $38$  to  $116 \text{ mg kg}^{-1}$ , dw, after 1993 (Figs. 3C and 3D). Although being clearly lower, the Cd

199 and Ag concentrations at the other sites showed similar trends. Compared to the La Fosse site,  
200 the Cd and Ag concentrations at the Pontailac and the Bonne Anse sites were respectively  
201 equivalent to  $54 \pm 11\%$  for Cd and  $78 \pm 18\%$  for Ag and  $31 \pm 10\%$  for Cd and  $38 \pm 16\%$  for  
202 Ag.

203

## 204 **4. Discussion**

205

### 206 **4.1. Geochemical background and historical record of Cd and Ag in river sediment**

207 The historical metal records in the Cajarc sediments closely reflect the evolution of the  
208 industrial activity in the source area from the 1950's to present, including termination of the  
209 Zn extraction in 1986 and subsequent remediation (Audry et al., 2004b; Castelle et al. 2007).  
210 The geochemical background values of the upstream Lot River watershed estimated from the  
211 Marcenac values (Cd:  $0.77 \text{ mg kg}^{-1}$ ; and Ag:  $0.30 \text{ mg kg}^{-1}$ ) were 15 to 384 and 1.4 to 23 times  
212 lower, respectively, than Cd and Ag concentrations at the contaminated Cajarc site. Co-  
213 evolution of Cd and Ag concentration-depth profiles supports common source and transport,  
214 suggesting that the Decazeville basin formerly was an important Ag source to the Lot-  
215 Garonne-Gironde fluvial estuarine continuum (Fig. 2). The observed Ag concentrations in the  
216 Lot River sediments were of similar magnitude as those reported for polluted English  
217 estuaries (up to  $5 \text{ mg kg}^{-1}$  with  $0.1 \text{ mg kg}^{-1}$  as the background concentration; Bryan, 1984). As  
218 in many industrially affected aquatic systems at the global scale, metal stocks in the  
219 contaminated Lot River sediments are of great concern for the Gironde fluvio-estuarine  
220 system due to their possible remobilisation during floods (e.g. Coynel et al., 2007), dredging  
221 (e.g. Audry et al., 2004a), passive diffusion at the water-sediment interface, (e.g. Rivera-  
222 Duarte and Flegal, 1997; Van Den Berg et al., 1999) or bioturbation (e.g. Zoumis et al.,  
223 2001).

## 224 **4.2. Bioavailability of Cd and Ag to oysters**

225 The concentration gradient of Cd and Ag in oysters from the La Fosse site to the Bonne Anse  
226 site, i.e. decreasing concentrations from upstream to downstream, could be explained by  
227 differences in exposure. Exposure of oysters to dissolved metals results in direct uptake  
228 (direct pathway) and is considered the major contamination pathway for both Cd (e.g. Ettajani  
229 et al., 2001) and Ag (e.g. Abbe and Sanders, 1990). In estuarine salinity gradients, both  
230 dissolved Cd and Ag typically show highest concentrations in the 15-20 salinity range (e.g.  
231 Dabrin et al., 2009; Tappin et al., 2010) due to desorption from the particulate phase by  
232 chloride-complexation (e.g. Luoma et al., 1995; Turner et al., 1993). Fluvial Cd and Ag  
233 transport into the Gironde Estuary mainly occurs in the particulate phase (respectively ~80%  
234 and ~96%; Lancelour et al., 2011; Schäfer et al., 2002). Their addition to the dissolved phase  
235 results in the typical ‘bell-shaped’ distribution of dissolved Cd (e.g. Dabrin et al., 2009) and  
236 Ag (Lancelour et al., unpublished; Tappin et al., 2010) along the estuarine salinity gradient. In  
237 the Gironde Estuary, dissolved Cd and Ag concentrations decrease from respectively 150-  
238 170 ng L<sup>-1</sup> and 6-8 ng L<sup>-1</sup> in the 15-20 salinity range towards the seaward end due to dilution  
239 by oceanic water (Atlantic Ocean: Cd<sub>D</sub> = 4 ± 3 ng L<sup>-1</sup>; Kremling, 1985; North Atlantic Ocean:  
240 Ag<sub>D</sub> = 0.39 ng L<sup>-1</sup>; Rivera-Duarte et al., 1999). This may induce systematic spatial differences  
241 in oyster exposure with decreasing concentrations from the La Fosse to the Bonne Anse site  
242 (Figs. 1 and 3). During the whole observed period, concentrations in oysters from the  
243 Pontailiac site were lower than those in La Fosse oysters, i.e. by 46 ± 11% for Cd and  
244 22 ± 18% for Ag (with metal concentrations at La Fosse being set equal to 100%).  
245 Accordingly, relative (compared to the La Fosse site) metal accumulation at the Pontailiac site  
246 (i.e. at higher salinity) appeared significantly higher for Ag than for Cd (p < 0.001; Log  
247 transformation of data followed by a one-way ANOVA). Similarly, but to a lesser extent, the  
248 relative metal accumulation at Bonne Anse was significantly higher (p < 0.001) for Ag

249 (38 ± 16%) than for Cd (31 ± 10%). Further work is necessary to understand if these  
250 differences are due to (i) differences in the slopes of dissolved Cd and Ag distribution in the  
251 high salinity range, i.e. the fact that maximum Ag release from particles occurs at higher  
252 salinities than for Cd as it was shown experimentally (Ciffroy et al., 2003) and/or (ii) different  
253 assimilation efficiencies for Cd and Ag species. In fact, compared to Cd, data on dissolved Ag  
254 distribution and speciation along estuarine salinity gradients are extremely scarce, i.e.  
255 inexistent for most estuaries. From recent maximum dissolved Cd (170 ng L<sup>-1</sup>) and Ag  
256 (8 ng L<sup>-1</sup>) concentrations in the Gironde Estuary salinity gradient (S=15-20; Dabrin et al.,  
257 2009; Lancelleur et al., unpublished) and the respective concentrations in oysters from the La  
258 Fosse site (Cd: ~30 mg kg<sup>-1</sup>, dw; Ag: ~80 mg kg<sup>-1</sup>, dw), we estimated minimum  
259 bioaccumulation factors (concentration in oysters/concentration in water) at 0.18 10<sup>6</sup> for Cd  
260 and 10 10<sup>6</sup> for Ag. This suggests that in the Gironde Estuary oysters bioaccumulation factors  
261 are ~55 times higher for Ag than for Cd. These results are consistent with previous  
262 experimental work showing that even if assimilation efficiency by trophic transfer in oysters  
263 is higher for Cd (*C. virginica*) than for Ag (Reinfelder et al., 1997), Ag-chlorocomplexes play  
264 an important role in Ag accumulation via the direct pathway by oysters compared to Cd  
265 (Engel et al., 1981), making Ag the strongest accumulated trace metal for certain marine  
266 bivalves (e.g. Luoma et al., 1995).

267

#### 268 **4.3. Comparison of particulate Cd and Ag fluvial gross fluxes to the Gironde Estuary** 269 **with Cd and Ag accumulation in oysters**

270 The Garonne River (the La Réole site; Fig. 1) is the main fluvial entry of particles and  
271 associated metals transported to the Gironde Estuary with > 80% (Cd; Schäfer et al., 2002)  
272 and ~96% (Ag; Lancelleur et al., 2011) of the fluvial gross fluxes (i.e. fluvial fluxes to the  
273 estuary) occurring in the particulate phase. The evolution over time of Cd and Ag annual

274 gross fluxes to the Gironde Estuary reflects (i) decreasing emissions in the historic source area  
275 (i.e. the Decazeville watershed) and (ii) the hydrological variability (e.g. Schäfer et al., 2002).  
276 Accordingly, Cd particulate gross fluxes at the La Réole site globally decreased by a factor of  
277 ~4 over the last two decades and may explain the parallel decrease in Cd concentrations in La  
278 Fosse oysters (from 95 to 25 mg kg<sup>-1</sup>, dw; Fig. 3E). Similar to Cd, the estimated particulate  
279 Ag gross fluxes also tended to decrease during 1992-2009 by a factor of ~4 (Fig. 3F). In  
280 contrast to Cd, Ag concentrations in oysters were at maximum during 1998-2003 and their  
281 evolution over time did not reflect that of the estimated Ag gross fluxes (Fig. 3F). This shows  
282 that after 1993 the fluvial Ag gross fluxes cannot explain the evolution of Ag bioaccumulation  
283 in oysters.

284

#### 285 **4.4. Historical trend of Ag and Cd pollution**

286 Based on the observation that the Cd/Ag ratio in oysters from the La Fosse site was constant  
287 before the termination of Zn ore treatment in the Decazeville basin (Cd/Ag=2 ± 0.4; 1981-  
288 1985) and assuming that during the whole observation period nearly all Cd in oysters was  
289 derived from the metal point source, we estimated the respective (theoretical) Ag  
290 concentrations in oysters over time (Fig. 3G). The latter assumption is based on mass balances  
291 at the watershed scale, showing that even 13 to 16 years after the termination of Zinc ore  
292 treatment (i.e. 1999 to 2002), the Decazeville basin contributed more than 61 to 92% of total  
293 Cd fluxes into the Gironde Estuary (Masson et al., 2006). This is further supported by (i)  
294 historical records of Cd in estuarine freshwater sediments, reflecting the industrial activity in  
295 the Decazeville watershed (Grousset et al. 1999) and (ii) mass balances at the estuary scale,  
296 suggesting that intra-estuarine sources (e.g. the Bordeaux agglomeration) are negligible,  
297 compared to the main source (e.g. Audry et al., 2007; Dabrin et al., 2009; Schäfer et al.,  
298 2002).

299 Comparing the estimated theoretical Ag concentrations to the measured data suggests that  
300 before ~1993, Ag in the oysters from La Fosse was mostly derived from the Decazeville basin  
301 (Fig. 3G). In contrast, increasing Ag concentrations after 1993 were attributed to increasing  
302 contributions (> 70% after 1999) by other sources. While soil erosion is the main natural Ag  
303 source to rivers, there are various anthropogenic Ag sources such as coal combustion,  
304 smelting, photography manufacturing and development, urban runoff, sewage treatment  
305 plants, cloud seeding and electrical supplies, etc. (e.g. Lanceleur et al., 2011; Purcell and  
306 Peters, 1998; US PHS, 1990). In 1986, photography accounted for 45% of Ag demand in the  
307 United States (US PHS 1990) and greatly contributed to the Ag contamination of San  
308 Francisco Bay sediment, water and biota until the mid 1980s (Flegal et al., 2007). In France,  
309 photographic film sales increased from 103 to 136 million in 1995-2001 and decreased to 24  
310 million in 2007 and 9 million in 2009, due to the development of numeric photography  
311 (Fig. 3G; data from the French Observatory for Imaging Professions OPI, 2010). Even if Ag  
312 load in oysters from the La Fosse site apparently followed this trend during 1995-2005,  
313 concentrations tended to increase after 2005 (Fig. 3G). Future work should aim at quantifying  
314 the contribution of recent increases in silver-containing paper sales (+21% between 2006 and  
315 2007; OPI, 2010), electronic products and emerging sources, e.g. growing use of Ag  
316 nanoparticles (e.g. Blaser et al., 2008).

317

## 318 **Conclusion**

319

320 Anomalously high Cd and Ag concentrations, up to 384 and 23 times higher than the  
321 respective regional background concentrations of 0.77 mg kg<sup>-1</sup> (Cd) and 0.30 mg kg<sup>-1</sup> (Ag) in  
322 the Lot River sediments downstream from the Decazeville watershed showed that Ag is a  
323 historical pollutant in the Gironde fluvial-estuarine system. The decreasing Cd concentrations

324 in sediments since the end of the metallurgical activity in 1986 and the evolution of fluvial  
325 particulate Cd gross fluxes, fully explain the decreasing Cd concentrations in oysters from the  
326 estuary mouth. In contrast, neither the historical Ag record in river sediments nor the fluvial  
327 Ag gross fluxes can explain the recent increase in Ag concentrations in oysters, clearly  
328 suggesting increasing contributions (> 70% after 1999) by other, probably intra-estuarine  
329 sources. The observed gradients in Cd and Ag concentrations in oysters from the three sites  
330 near the estuary mouth probably reflect exposure to dissolved Cd and Ag concentrations, i.e.  
331 their typical distribution along the estuarine salinity gradient. Further work is necessary to  
332 identify these recent Ag sources and to understand their role in the estuarine Ag cycle and in  
333 Ag contamination of aquatic organisms.

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340

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Table 1. Results obtained from analyses of Certified Reference Materials. *n*: number of independent analyses.

Reference material	Description	<i>n</i>	Ag (mg kg <sup>-1</sup> , dw)		Cd (mg kg <sup>-1</sup> , dw)	
			Certified value	Measured value	Certified value	Measured value
NCS-DC-70311	River sediment	5	6.73 ± 0.62	6.28 ± 0.38	3.76 ± 0.23	3.66 ± 0.10
NCS-DC-70317	River sediment	5	0.32 ± 0.02	0.34 ± 0.01	0.57 ± 0.04	0.55 ± 0.02
SRM-1566b	Oyster tissue	47	0.67 ± 0.01	0.61 ± 0.04	2.48 ± 0.08	2.41 ± 0.09
CRM-278	Mussel tissue	6	-	0.20 ± 0.04	0.35 ± 0.01	0.33 ± 0.02

520 Figure 1: The Gironde Estuary (SW France) with sampling locations of A) oysters and B)  
521 sediment cores. The La Réole site, located at the upstream limit of tidal dynamics, is a  
522 permanent observation site for trace element concentrations and fluxes at the main entry of  
523 the Gironde Estuary (for details see Schäfer et al., 2002).

524

525 Figure 2: Cadmium concentrations, Ag concentrations and  $^{137}\text{Cs}$  activity vs depth for the  
526 sediment cores from the Marcenac (geochemical background; empty symbols) and the Cajarc  
527 (historical record of metallurgical pollution; filled symbols) sites. Cadmium and  $^{137}\text{Cs}$  profiles  
528 are from Audry et al. (2004b).

529

530 Figure 3: Concentrations of Cd (A, C and E) and Ag (B, D, F and G) in oysters from the La  
531 Fosse (A, B, C, D, E and F), Pontailiac (C and D) and Bonne Anse (C and D) sites during  
532 1979-2010. Oysters have been sampled four times a year during 1979-2002, twice a year from  
533 2003 to 2007 (February and November) and annually (February) since 2008 (A and B).  
534 February data are represented for Cd (C and E) and Ag (D, F and G) concentrations in  
535 oysters. Particulate Cd (E; white bars) and Ag (F; white and grey bars) gross fluxes at the La  
536 Réole site over time. (E): Particulate Cd gross fluxes in 1990-1999 are from Schäfer et al.  
537 (2002), Audry et al. (2004a) for 2000, Masson et al. (2006) for 2001-2002, from Dabrin et al.  
538 (2009) for 2003-2007 and unpublished data for 2008-2009. (F): The grey bars represent  
539 estimated particulate Ag gross fluxes at the La Réole site in 1990-1998. Particulate Ag fluxes  
540 in 1999-2009 are from Lanceleur et al. (2011). (G): Ag concentrations in oysters and  
541 photographic film sales in France vs time. Ag levels were measured (full circles) and  
542 estimated (open squares) from pre-1986 Cd/Ag ratio and from Cd concentrations in oysters.

Figure 1

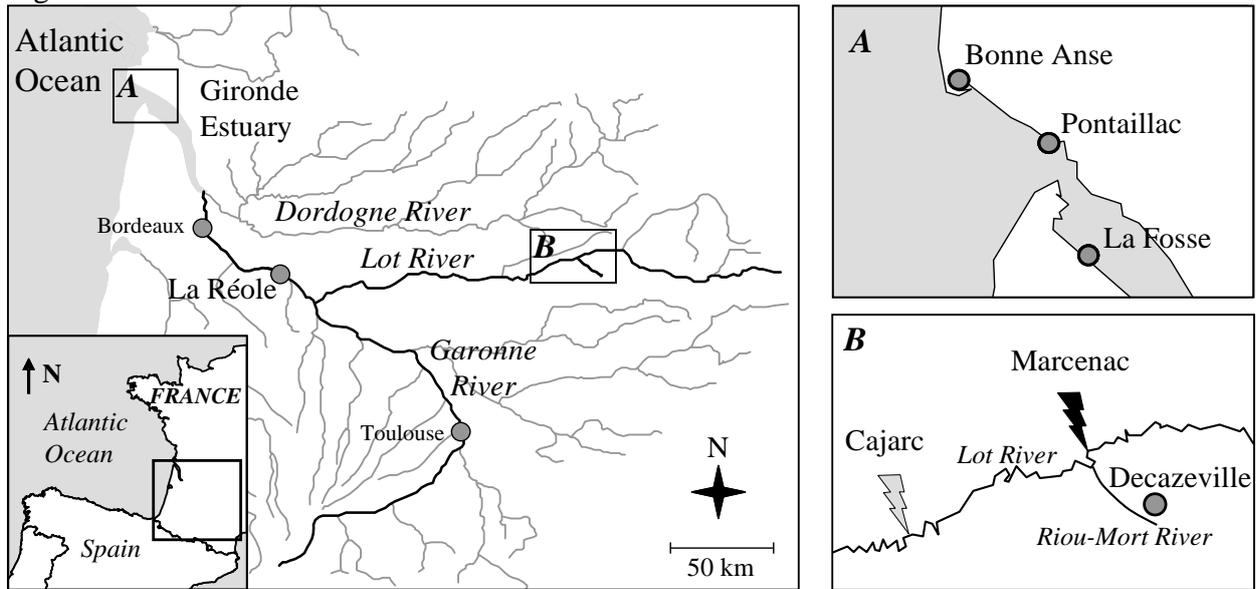


Figure 2

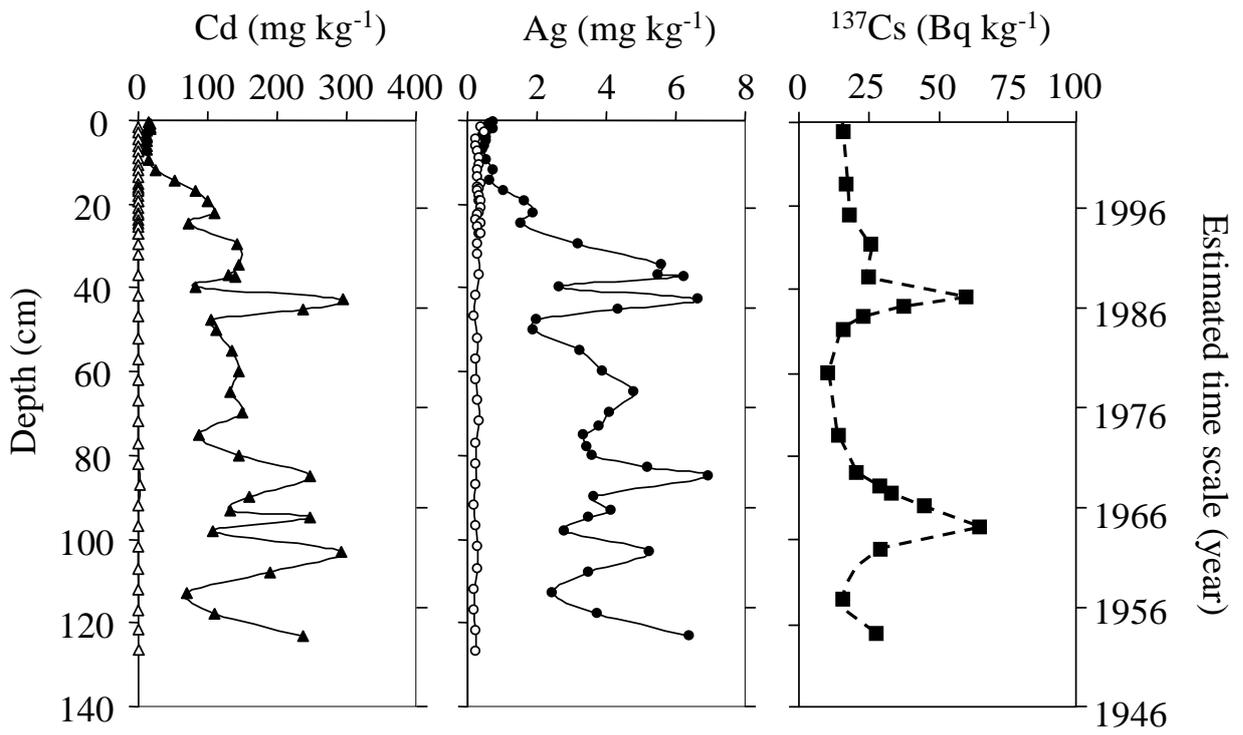


Figure 3

