
Flood inputs in a Mediterranean coastal zone impacted by a large urban area: Dynamic and fate of trace metals

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Abstract:

Trace elements and organic carbon inputs to the Mediterranean sea from an urbanized area (Marseille city) were studied and characterized during flood events. Inputs were brought to the sea by two small coastal rivers whose waters were mixed together and also with treated wastewaters (TWW) just before discharge. The monitoring of the rivers during flood events showed the high temporal dynamics of water flow, suspended particulate matter (SPM), organic carbon and trace metals concentrations, typical of small coastal Mediterranean rivers and requiring an appropriate sampling strategy. Dissolved/particulate partition coefficient ($\log K_d$) in rivers during floods remained quasi-constant for a given trace element, but differed from one element to another according to their affinity toward particles. Because of high SPM concentrations, trace elements were mainly brought to the sea during floods as particles, despite a weaker affinity for particles when compared to baseflow conditions for all studied elements but Pb. If the contribution of TWW dominated the elements baseflow discharge to the coastal zone, rivers outweighed during floods. When discharged to the sea, most trace elements underwent partial desorption in the salinity gradient, especially at highest salinity. Laboratory desorption experiments results were consistent with field data and showed slower desorption kinetics than in baseflow conditions, suggesting that trace elements desorption rates from particles are slower than sedimentation rates. With regard to heavy particles, it results in a potential impact of the sediment on benthic organisms and a possible further desorption after sediment resuspension events. With regard to light particles, it results possible additional desorption during offshore transport.

Highlights

► Fast concentration changes during flood events require appropriate monitoring ► Desorption kinetics from particles in plume slower during floods than baseflow ► Untreated wastewaters input resulted in higher contamination by dissolved species ► Differences between flood and baseflow particles strongly control metals fate

Keywords : WWTP ; metals contamination ; coastal zone ; flood events ; Mediterranean Sea ; small rivers

1. Introduction

Estuaries constitute a transition zone between continent and ocean, and represent a major source of material to the coastal zone. The presence of trace elements in rivers relates to the regional geochemistry and inputs from various anthropogenic sources, especially where population density is high. These anthropogenic inputs are mainly due to leaching of impervious urban surfaces (roads, carparks, roofs) and soil surfaces by runoff, and to sewage overflow from sanitary sewer system during high rainfall events; locally treated wastewaters (TWW) discharge and industrial effluents can also be of importance (Bay et al., 2003, Bothner et al., 2002, Gonzalez et al., 1999, Matthai et al., 2002, Nicolau et al., 2012, Oursel et al., 2013 and Wei et al., 2010). When discharged to the aquatic environment, trace elements can raise toxic levels and cause adverse effects to organisms or even human health (Gupta et al., 2009).

In freshwater/seawater mixing zones such as estuarine systems, the biogeochemistry of trace metals is controlled by a complex interplay of hydrodynamic factors, industrial and

municipal wastewater discharges and biogeochemical processes. Differences of physico-chemical conditions between river and seawater can cause a redistribution of metals between solution and suspended particulate matter (SPM). The geochemical reactivity of trace metals in estuaries is commonly ascribed to changes in metal adsorption-desorption equilibrium and flocculation processes along the salinity gradient (e.g. Boyle et al., 1977; Zwolsman et al., 1997; Thill et al., 2001). Mobilization of dissolved metals from solid phase is indeed the subject of considerable interest and numerous field data confirm the significance of metal release from river SPM along the salinity gradient (Elbaz-Poulichet et al., 1996; Kraepiel et al., 1997; Waeles et al., 2005; Audry et al., 2007). The morphological characteristics of the river watershed as climate, catchment area hydrology, geology and land use, and the physico-chemical characteristics of the water column as mixing time of waters or metal speciation can be decisive (Nicolau et al., 2012).

The Mediterranean environment is known for its high-intensity rains, promoting high rates of terrestrial runoff that can occur after long dry periods which allow contaminants accumulation in the watershed surfaces. The spatial and seasonal variability of rainfall follows a complex pattern, with wide and unpredictable rainfall fluctuations from one year to another (Martínez-Casasnovas et al., 2002; Nicolau et al., 2012). Numerous studies focused on Rhône River, the main river of the western Mediterranean basin and the largest contributor with regard to water and SPM discharge to the Mediterranean Sea (Ollivier et al., 2011; Radakovitch et al., 2008; Raimbault and Durrieu de Madron, 2003). Small coastal rivers, despite their low flows, can also have a high ecological impact because they rapidly bring to the sea the contamination occurring in the coastal area, especially where population density is high, however only few studies dealt with small Mediterranean rivers impacted by urbanization (Dassenakis et al., 1997; Elbaz-Poulichet, 2005; López-Flores et al., 2003; Nicolau et al., 2012; Oursel et al., 2013). In Marseille agglomeration (1.7 million inhabitants), Huveaune and Jarret rivers constitute a typical example of such rivers. Their watersheds are representative of modern anthropization in most of the Mediterranean coastal strip, mixing urban, semi-urban, small industrial and some preserved natural areas. Their waters are mixed together and with city wastewater treatment plant (WWTP) effluents before their outlet to the sea (Fig.1) at the Calanque of Cortiou, situated in the French National Park of "Calanques". Studies on the dynamics of elements in transitional coastal zone (e.g. estuaries) are well documented (Elbaz-Poulichet et al., 1996; Ollivier et al., 2011; Shiller, 1997; Waeles et al., 2008), but a system such as Marseille is quite specific. Oursel et al. (2013) have demonstrated that the WWTP effluents represent the main source of inorganic contaminants brought to the

coastal zone during baseflow periods. During flood events, the relative contributions of rivers vs. WWTP effluents are likely different. Indeed, studying experimentally the fate of Marseille watershed particles and associated pollutants when discharged to the coastal zone, Oursel et al. (2014) have observed that characteristics of particles from flood events were dominated by that of rivers waters, except during WWTP by-pass.

In this context, the aims of the present study were (1) to quantify and characterize carbon and trace metals dynamics and inputs to the sea during flood events in comparison to baseflow periods and (2) to understand the mechanisms governing elements behaviour and fate in the mixing zone during these contrasted conditions.

2. Material and methods

2.1. Study site

Huveaune River extends over 48.4 km long and runs through a watershed with an area of 523 km² which consists in karstic formation (60%) and detrital sediments. Land-use in the downstream part of the watershed is urban and industrial. Jarret River extends over 21 km with a 102 km² watershed mainly urban and industrial. For the period 1961-2009, the mean annual precipitation in Marseille was 544 mm and the mean number of days with total rainfall of 1 mm or more was 59 (Infoclimat, 2014).

In baseflow period, these two rivers merge in Marseille; the resulting water is mixed with the Marseille City WWTP effluents, and then channelled and discharged at sea at the Calanque of Cortiou (Fig. 1). This treatment plant, one of the largest in Europe (1.7 million inhabitant eq.), uses both physical and biological treatment processes. During baseflow periods (~ 300 days), around 100 and 80 Mm³ y⁻¹ of TWW and river water, respectively (average flows of 1.9 and 0.9 m³ s⁻¹ for Huveaune and Jarret, respectively), are discharged to the sea (Le Masson, 1997). The corresponding annual SPM discharge is around 3950 t, of which 850 t are coming from rivers, highlighting the main contribution of TWW in such conditions (Jany et al., 2012; Oursel et al., 2014). During flood events, Huveaune and Jarret Rivers flows can overpass 60 and 16 m³ s⁻¹, respectively. During such extreme events, as the maximal outlet discharge capacity cannot exceed 30 m³ s⁻¹, the exceeding part of rivers waters is channelled through the Huveaune former bed to the Prado's beach (Fig. 1). During rainy periods, around 4.1 and 11 Mm³ y⁻¹ of TWW and rivers waters, respectively, are discharged to the sea. Le Masson (1997) has evaluated the corresponding SPM discharge to be 6500 t, 94%

coming from rivers. On average, during flood events, 90% of Huveaune and Jarret waters (and SPM) are channelled through the outlet and only 10% are deviated through the Huveaune former bed. Additionally, discharge of untreated wastewater (due to WWTP bypass) brought to the coastal zone ranged between 456 and 1450 t y⁻¹ of SPM during the period 2001-2007 (Jany et al., 2012).

2.2. Sampling

All bottles (FEP and HDPE, from Nalgene) used throughout the whole sampling and conditioning were previously cleaned with 10% HNO₃ (HNO₃ 70% Analytical reagent grade, Fisher Scientific), rinsed with milliQ water (Millipore 18.2 MΩ), filled with milliQ water and acidified at 0.1% with suprapur (s.p.) HNO₃ (Merck) until use. Main physico-chemical parameters (T, S, pH, O₂, conductivity) were measured on-field using a HACH LANGE multi-probe, calibrated before each campaign.

2.2.1. Freshwater sampling

In order to characterize rivers inputs during a flood event, a monitoring of Huveaune and Jarret Rivers in rainy conditions was performed on 18/09/2009. The sampling stations were immediately located upstream the river confluence. In order to have representative samples, nine composite samples were collected from 8:00 am to 8:20 pm by sampling 100 mL of the river water using an HDPE bucket each 10 min during 1 h or each 20 min during 2 h, depending on the intensity of the rain and the rivers flow. Rivers were also sampled one day before and one day after the flood to characterize river baseflow. Then, 6 additional rain samples were collected at various dates after this monitoring to better take into account the inputs variability and to more accurately evaluate their contribution to the average flux of elements to the coastal zone.

A monitoring of WWTP effluent and rivers waters mixing at the outlet was also performed from 18/09/2009 7:00 am to 20/09/2009 7:00 am, using an automatic sampler (mobile refrigerated automatic sampler, Hydreka) installed in the outlet, thanks to the logistic of the SERAM company and IFREMER. During these 48 hours, 24 composite samples of 900 mL were collected by sampling 150 mL of water every 20 min during 2 h. Rainfall data were provided by Meteo-France, river flows by SERAM and DEA.

2.2.2. Seawater sampling

In rainy conditions, the accessibility to the study site was difficult as rain was often accompanied with strong wind and big waves, so only 3 sampling campaigns (25/10/2011, 21/05/2012 and 28/11/2012) were conducted. Two of them were performed at the Calanque of Cortiou and the 25/10/2011 campaign was performed at the Prado beach.

During each campaign, sampling was performed along a transect (Fig. 1, inset) situated in the plume. Depending on weather conditions and plume length, 5 to 11 surface samples were taken from the outlet to the defined marine end-member (2 km offshore and 4 m depth). Considering the high stratification of the plume, sampling was performed using a 2.2 horizontal sampler (Wildco) which allows sampling within a ten-centimeter thick layer. Samples were partitioned in 1 L FEP bottles for metal analysis and 1 L HDPE bottles for organic matter (OM) analysis. After sampling, bottles were stored in cooler, brought back to the laboratory and immediately filtered (in-lab filtration). Additionally, for these 3 campaigns, at each sampling site an aliquot was also immediately filtered on board (on-field filtration) to avoid trace elements dissolved/particulate repartitioning which can alter the assessment of their behaviour along the salinity gradient (Oursel et al., 2013).

The 3 flood campaigns will hereafter be compared to 2 contrasted baseflow conditions campaigns already described in Oursel et al. (2013): a 22/03/2011 campaign with typical baseflow conditions, and a 18/04/2012 campaign characterized by a planned by-pass of the WWTP (due to servicing).

2.3. Sample filtration, conditioning and treatment

Samples for organic carbon analyses were filtered through 25 mm glass fiber filters (Whatman GFF, 0.7 μm). Filters were dried (40°C) until constant mass and stored in aluminium paper until particulate organic carbon (POC) quantification. The dissolved fraction was preserved with 25 μL 1 M NaN_3 ($\text{NaN}_3 > 99\%$, Aldrich) and stored at 4°C in 24 mL glass tubes (Wheaton, equipped with Teflon/silicone septum) prior to analysis of dissolved organic carbon (DOC). All tubes, filters and glass filtering systems (Wheaton) were previously cleaned with 10% HNO_3 , rinsed with MilliQ water and calcinated during 4 h at 450 °C.

Samples for dissolved/particulate metal analyses were filtered through 47 mm pre-cleaned (rinsed with 100 mL of MilliQ water) cellulose nitrate filters (Sartorius, 0.45 μm). Filters were then dissolved by microwave digestion (AntonPaar Multiwave 3000) in aqua

regia (trace metal grade acids, Fisher Scientific) for further analysis of metals in the particulate fraction, following the procedure previously validated (Tessier et al., 2011). The dissolved fraction was stored in 60 mL FEP bottles, acidified with 60 μL HNO_3 (Merck Suprapur) and digested 2 h under UV-irradiation (Metrohm 705, 500 W), a necessary step for stripping voltammetry analysis of total dissolved metal concentrations.

On-field filtration was achieved using 0.2 μm in-line syringe cellulose nitrate filters (Sartorius). Filtrates were stored in 60 mL FEP bottles for dissolved metals analysis and in 24 mL glass tubes for DOC analysis. The filtrates were then processed as described for in-lab filtrations.

2.4. Remobilization experiment

For a better understanding of the mechanisms governing trace metals behaviour in the salinity gradient, a remobilization experiment designed to simulate mixing of outlet effluent with seawater was performed in the laboratory. During baseflow period, a 10 L composite outlet water (Outlet baseflow) was sampled by collecting ten times 1 L during ~ 15 min, in order to obtain a more representative outlet sample. Flood deposit collected in the outlet (FDOut2, Oursel et al., 2014) was added in this composite sample in order to obtain an “artificial flood” outlet sample (SPM close to 800 mg L^{-1}) and equilibrated during 24 h. Filtered marine “end-member” (Oursel et al., 2013) was mixed in 500 mL FEP bottles with the artificial flood outlet water to cover the following range of salinity: 1, 2.5, 5, 10, 20 and 30. The 500 mL FEP bottles were installed on an head-over-head agitation system (Reax 20, Heildolph) and aliquots were sampled at 15 min, 1, 6 and 24 h of mixing time using pre-cleaned syringe, then filtered through pre-cleaned in-line filters (Sartorius, 0.2 μm). Filtrates were stored and analyzed for DOC and dissolved trace metals as explained in section 2.5.

2.5. Sample chemical analysis

2.5.1. Dissolved and particulate carbon analysis

The organic carbon content was determined using the high-temperature (900°C) catalytic oxidation method with CO_2 IR detection (Ammann et al., 2000; Callahan et al., 2004), calibrated using glucose (Analytical reagent grade, Fisher Scientific) with an accuracy of $8.3 \mu\text{M C}$. POC contents were determined on GFF filters using a TOC- V_{CSH} analyzer (Shimadzu),

coupled with a SSM-5000A module. GFF filters were dried to constant weight at 60°C, and then exposed to HCl fumes for 4 h to remove all the inorganic carbon (Lorrain et al., 2003).

DOC and DIC concentrations were determined using the same TOC-V_{CSH} analyzer, calibrated using potassium hydrogenophthalate (Shimadzu) and NaHCO₃/Na₂CO₃ (Shimadzu) standard solutions, respectively, with an accuracy of 1.7 µM C (Louis et al., 2009). DOC and DIC analysis were validated using certified reference material (MISSIPPI-03, Canada); the obtained values were within the certified limits.

2.5.2. Minor and trace elements analysis

In the following, concentration and content terms refer to mass/mol per volume unit (e.g. nM) and mass/mol per mass unit (e.g. µmol g⁻¹), respectively

2.5.2.1. Dissolved trace metals in saline samples

Total dissolved metal concentrations were determined by Differential Pulse Anodic Stripping Voltammetry (Cd, Cu, Pb and Zn) and Differential Pulse Adsorptive Cathodic Stripping Voltammetry (Co and Ni) using fully automated analytical procedures previously described (Louis et al., 2009; Omanović et al., 2006; Oursel et al., 2013). Total dissolved As was determined using a hydride generation atomic fluorescence spectrometry (HG-AFS) technique (ThermoScientific – PS Analytical) (Dang et al., 2014).

Analysis were validated using certified “Nearshore Seawater Reference Material for Trace Metals” - CASS5 (NRC CNRC). All metals determinations were within the certified limits.

2.5.2.2. Particulate minor and trace elements and dissolved trace elements in freshwater samples

Particulate elements (Ag, Al, As, Cd, Co, Cr, Cu, Fe, Li, Mn, Ni, Pb, V and Zn) were quantified from the acid-digested filters by High Resolution Inductively Coupled Plasma Mass Spectrometry (HR ICP-MS Element 2, Thermo Finnigan). Blanks didn't overpassed 10% of the elements concentrations measured on samples and consequently were not subtracted from these values. The instrument was calibrated using standard solutions and indium was used as an internal standard to correct for changes in peak intensities due to instrumental drift (Lenoble et al., 2013). Quality control of HR ICP-MS measurements was checked by the determination of elements concentration on Certified Reference Material

(SLRS-4 river water, PACS-2 sediment, National Research Council Canada). All results fell in the range of the certified data.

3. Results and Discussion

3.1. Dynamic of elements in rivers during flood event

Table 1 gives the average of the measured parameters in Huveaune and Jarret Rivers, at the outlet and in the marine end-member during the flood, compared to those observed during floods in Eygoutier River, a small coastal river of the same geographic area (Nicolau et al., 2012) and to those observed in floods of Rhône River (discharge over $1500 \text{ m}^3 \text{ s}^{-1}$), the main river of the French Mediterranean area (Ollivier et al., 2011).

Figure 2A represents the discharge in rivers and the rainfall at Marseille ("MARSEILLE-OBS" Météo-France rain gauge) as a function of time during the 18/09/2009 flood. A first 14 mm-rainfall between 7:00 and 9:00 am resulted in a fast increase of the river discharge with a maximum of $28 \text{ m}^3 \text{ s}^{-1}$ in Huveaune River and $10 \text{ m}^3 \text{ s}^{-1}$ in Jarret River. A second 11 mm-rainfall between 3:00 and 6:00 pm resulted in an increase of the river flows with a maximum of $22 \text{ m}^3 \text{ s}^{-1}$ in Huveaune River and $6.6 \text{ m}^3 \text{ s}^{-1}$ in Jarret River. Contrarily to Huveaune, the flow of Jarret returned at its baseflow (around $1.6 \text{ m}^3 \text{ s}^{-1}$) less than two hours after the rainfall. These results attested the high reactivity and dynamic of this urbanized system, a specificity common to most of the small Mediterranean rivers submitted to extreme events.

3.1.1. Suspended particulate matter

Figure 2B shows the suspended particulate matter (SPM) and the turbidity in the rivers as a function of time during the same day. For both rivers, SPM and turbidity were highly correlated (data not shown, $R^2 = 0.94$), increasing during the rain and decreasing after, with a maximum (1-h composite value) for the first rain at 256 mg L^{-1} and 248 mg L^{-1} for Huveaune and Jarret Rivers, respectively. Both rivers presented similar SPM time variations and maximal values despite much lower flow in the Jarret River. However, between the 2 rain events, SPM concentrations in Jarret River return to baseflow values when Huveaune River ones remained 4-fold higher. The flood carried SPM concentrations 20 times higher, on average, than in baseflow conditions (Oursel et al., 2013), but were in the same range of

concentrations than in Eygoutier River floods and even lower than in Rhône River floods (Table 1).

3.1.2. DOC, POC, DIC and conductivity

In both rivers, DOC was on average 3-fold higher during flood than during baseflow periods (Table 1). Figure 2C represents DOC and POC concentrations for both rivers as a function of time during the same flood. DOC and POC concentrations showed similar variations ($R^2 = 0.85$), increasing at the beginning of each rain as already observed in other systems (Nicolau et al., 2012; Shafer et al., 1997; Tao, 1998; Wu et al., 2013), with a maximum for DOC at 480 and 720 $\mu\text{M C}$ and for POC at 1700 and 1250 $\mu\text{M C}$ for Huveaune and Jarret Rivers, respectively. However, Nicolau et al. (2012) have shown that very different patterns between DOC and POC can be observed from one flood event to another. The variations of DOC concentrations as a function of POC concentrations for baseflow and floods in rivers and at the outlet are plotted in Figure 3A. For baseflow conditions, in Huveaune River, DOC fraction is dominant whereas in Jarret River, it is POC (Oursel et al., 2013). At the outlet, DOC and POC are in the same range, both strongly increasing in comparison to river values (Oursel et al., 2013). During floods, DOC and POC increased in both rivers, POC being the major fraction. There is no significant difference between outlet and rivers values, indicating that the WWTP signature was overridden by the rivers water, on contrary to baseflow conditions (Oursel et al., 2013).

POC expressed in content (i.e. $\mu\text{molC g}_{\text{SPM}}^{-1}$) didn't shown significant variation during the flood (data not shown) but was significantly lower than in baseflow conditions, likely because the enrichment of SPM by mineral particles due to erosion processes resulted in a dilution of POC.

In both rivers, DIC and conductivity decreased with increasing discharge, reaching minimal values of 1250 and 1060 $\mu\text{M C}$ for DIC and 0.21 and 0.17 mS cm^{-1} for conductivity for Huveaune and Jarret Rivers, respectively (data not shown), due to the dilution of river water by rain water (baseflow levels: 4080 and 3080 $\mu\text{M C}$, 0.71 and 0.61 mS cm^{-1} for Huveaune and Jarret Rivers, respectively). The good linear correlation ($R^2 = 0.93$) between conductivity and DIC was expected as conductivity was mostly controlled by DIC (carbonates and Ca) and corresponded to the calcareous signature of the rivers watersheds.

3.1.3. Minor and trace elements

3.1.3.1. Particulate and dissolved metal concentrations

Particulate concentrations (μM) of Ag, Al, Cd, Co, Cr, Cu, Fe, Li, Mn, Ni, Pb, V and Zn were well correlated to SPM ($0.53 \leq R^2 \leq 0.78$ for Huveaune River, $0.68 \leq R^2 \leq 0.93$ for Jarret River ($p < 0.05$)). Their average concentrations were similar in the two rivers (Table 1). Some differences, however, were observed between metals. In Huveaune River, particulate Cu, Pb, Cd, Zn and Ni were much more concentrated during the second rain despite a lower SPM concentration (Fig. 2B and 2D). Oppositely, particulate Co was evenly distributed between the 2 rains. In Jarret River, particulate Cu, Pb, Cd and Zn were more concentrated during the first rain and particulate Co and Ni concentrations were evenly distributed between the 2 rains. Between the 2 rain events, particulate organic carbon and metals concentrations rapidly returned to baseflow values in Jarret River, confirming the high reactivity of this river, due to the predominance of impervious surface areas in the watershed.

The behaviour of dissolved metals in rivers during the flood was also different between each rain and rivers. Dissolved Cu is given as an example in Figure 2D. In Huveaune River, dissolved Cu increased at the beginning of each increasing flood, decreased during the flow peak and returned close to the baseflow concentration between the 2 rains, whereas in Jarret River, it decreased with increasing flow and increased between the two rains. Dissolved Pb, Cd and Zn showed the same trend as dissolved Cu during the flood but dissolved Co and Ni had a different behaviour. Well-correlated one to the other ($R^2 = 0.94$, $p < 0.001$), they increased at the beginning of the first flood in both Huveaune and Jarret Rivers, decreased at the flow peak, but increased only in Jarret River at the beginning of the second flood.

The dissolved metal concentration as a function of the particulate metal concentration for baseflow and floods in rivers and at the outlet is given in Figure 3. In rivers, during baseflow conditions, Cu, Cd and Zn (Figure 3B, 3D and 3E, respectively) were on average evenly distributed between the dissolved and particulate fractions. Lead was mainly contained in the particulate fraction (Figure 3C) and Co and Ni mainly contained in the dissolved fraction (Figure 3F and 3G, respectively). At the outlet, in baseflow conditions, dissolved concentrations were similar to those in rivers for all metals except Zn exhibiting higher values than in rivers. Oppositely, particulate concentrations strongly increased for all metals, due to TWW inputs (Oursel et al., 2013). During floods, concentrations of Cu, Cd, Zn and Pb increased in both dissolved and particulate fractions, but with an increase of the

particulate/dissolved ratio. Concentrations of Co and Ni increased only in the particulate fraction, concentrations in the dissolved fraction remaining constant or slightly diminishing. In such flood conditions, no significant differences were observed between rivers and outlet values, indicating that the WWTP signature was overridden by the rivers water, on contrary to baseflow conditions (Oursel et al., 2013). The absence of any significant relationships between dissolved and particulate trace metals concentrations can result from repartitioning during the flood events due to the strong variation of SPM concentrations, which will further be studied using partition coefficients (see section 3.2.3).

3.1.3.2. Minor and trace elements content in the particulate fraction

No significant trend was observed for metal contents as a function of time during the flood (data not shown). Oursel et al. (2014) have shown that metals contents in the particulate fraction of Huveaune and Jarret Rivers were close to those observed in Rhône River, but when normalizing the values to Al content, particles from Huveaune and Jarret Rivers appeared to be more contaminated. Metal contents in rivers during floods (Table SI-1) were in the same order of magnitude than those observed during baseflow periods (Oursel et al., 2013) indicating the same origin of particles during baseflow and flood periods. Nevertheless, in Huveaune River, when the river flow increased, Cd and Cr contents decreased (data not shown) as it was observed in Rhône River in floods (Radakovitch et al., 2008; Ollivier et al., 2011). In Jarret River, some differences were observed, when the river flow increased, Al, Co, Fe, Li, Mn, Ni, and V contents decreased until, more or less, constant values giving an indication of the background value. On the contrary, Cd, Cu, Pb and Zn contents increased with increasing flow, indicating additional anthropogenic contamination. Some metal contents had good correlations one to the other, testifying similar origin and/or behaviour. In both rivers, Li, V and Fe contents presented a good correlation to Al content (R^2 of 0.75, 0.69 and 0.50, respectively ($p < 0.01$)), an example is given with Li in Figure 4A. This is consistent to the fact that these elements in rivers are known to be mainly related with rock weathering (Milot et al., 2010; Schiller and Mao, 2000). Regarding non-crustal elements, the absence of correlations with Al suggests that anthropogenic sources are dominant compared to natural ones. Cu was correlated with Zn ($R^2 = 0.81$ ($p < 0.001$), Fig. 4B) and Cd with Pb ($R^2 = 0.74$ ($p < 0.001$)) which informed about their similar source or reactivity.

3.2. Dynamics of elements in the mixing zone

Sampling inside the outlet during rain events was impossible for safety reasons. As the rivers were largely the main contributors at the outlets, we supposed that metals concentrations at the outlets were close to rivers values for the 3 flood campaigns.

3.2.1. Suspended particulate matter

The distribution of SPM along the salinity gradient decreased with increasing salinity, showing for each flood campaign a conservative behaviour (data not shown), indicating that that freshwater SPM (min 135, max 195 mg L⁻¹) were just diluted in seawater without neither significant flocculation processes, nor sediments resuspension due to the absence of tidal currents. Differences between the flood campaigns were due to the quantity of SPM discharged to the sea at the outlets. The baseflow with WWTP bypass campaign (18/04/2012) also showed a conservative behaviour, with SPM values in the range of flood events (outlet: 202 mg L⁻¹), 9-fold higher than normal baseflow conditions (outlet: 23 mg L⁻¹ in average (N=50); Oursel et al., 2013).

3.2.2. DOC and POC

DOC variations as a function of salinity for the flood and the baseflow campaigns are given in Figure 5A. Average DOC in rivers during the 3 flood campaigns was 330 µM C (min 190, max 400) for Huveaune River and 480 µM C (min 470, max 500) for Jarret River, so in the range of values recorded during the flood monitoring (18/09/2009). No differences were observed between on-field and in-lab filtrations neither in rivers nor in plume samples. During floods, as well as in baseflow conditions, the DOC values decreased in the plume with increasing salinity, following a conservative behaviour (Fig. 5A), as it was already observed by Pujo-Pay et al. (2006) in the Rhône River Plume. This contrasts with the non-conservative behaviour observed in the Krka River estuary (Louis et al., 2009; Sempere and Cauwet, 1995), a peculiarity due to the low DOC level in the freshwater end-member enhancing possible effects from biological activities. Oursel et al. (2013) have shown that for the by-pass campaign, DOC concentrations were higher in outlet, especially when samples were directly filtered on field. In that case, DOC showed non conservative trend along the salinity gradient for both filtrations.

During floods, the average POC contents were 5700 and 7600 $\mu\text{molC g}^{-1}$ for Huveaune and Jarret, respectively, values much lower than the ones encountered during baseflow (9200, 19000 and 21000 $\mu\text{molC g}^{-1}$ on average in Huveaune, Jarret and outlet, respectively (Oursel et al., 2013)). Such differences probably reflect the contrasted source/nature of particles: terrigenous and mineral during floods, organic-rich and anthropogenic with TWW dominance during baseflow (Oursel et al., 2014). Along the salinity gradient, POC content decreased with increasing salinity but no clear trend was observed (data not shown). During the by-passed campaign, although SPM concentrations were similar to those recorded during floods, much higher POC content ($\sim 33000 \text{ mg g}^{-1}$) were measured at low salinity due to the predominance of organic-rich particles from untreated wastewaters.

3.2.3. Trace elements

For all studied trace elements, no differences were observed between on-field and in-lab filtration (data not shown) during flood campaigns, contrarily to what was observed during baseflow campaigns (Oursel et al., 2013). This means that metals were at the equilibrium between dissolved and particulate fractions when entering in the coastal waters and that desorption processes along the salinity gradient were slow with regard to mixing times. Dissolved Cu, Pb, Cd, Zn, Co, Ni and As as a function of salinity for the flood and the baseflow campaigns are given in Fig. 5. Variations of element dissolved/particulate fractionation in rivers and in the salinity gradient and between campaigns were studied by plotting particulate trace elements (expressed in percent of total concentration) as a function of SPM (log scale), in comparison to the theoretical lines for given $\log Kd$ values (Fig. 6 and Fig. SI-1). In the following sections, the $\log Kd$ values were optimized from measured data using Excel Solver subroutine, based on the following equation determined from Kd and mass balance equations:

$$\frac{M_p}{M_D + M_p} = \frac{K_d \text{ SPM}}{1 + K_d \text{ SPM}}$$

M_p and M_D (nM) are respectively the particulate and dissolved concentrations of a given element, Kd is the optimized partition coefficient (L kg^{-1}) and SPM the measured suspended particulate matter concentration (kg L^{-1}).

3.2.3.1. Cu

During floods, dissolved Cu decreased with increasing salinity following a quasi-conservative trend (Fig. 5B), as already observed in other estuarine systems (Elbaz-Poulichet et al., 1996; Guieu et al., 1998; Shim et al., 2012). Oppositely, Cu behaviour when WWTP was by-passed and during baseflow conditions was different, showing points well above the conservative mixing line at low salinity, indicating a fast metal release from particles. During the WWTP by-pass, dissolved Cu at the beginning of the plume reached 350 nM, a level significantly increased when compared to the concentrations observed in the outlet during this by-pass event (170 nM) or baseflow conditions (33nM, Oursel et al. (2013)) and more than 100-fold higher than marine end-member value (3.3 nM). Such high levels, greatly overpassing acute water quality criteria (WQC) for the protection of coastal ecosystems (WQC_{Cu} 22 nM; Durán and Beiras (2013)), could cause significant effects to living organisms. For instance, impact of untreated urban discharge on cage mussels was recently evidenced by De los Rios et al. (2013). During floods, Cu was mainly associated to particulate fraction (more than 90%) at the outlet but its fractionation progressively evolved in the plume towards a more evenly distribution between dissolved and particulate fractions in the marine end-member (Fig. 3B). The Cu dissolved/particulate fractionation during the 18/09/2009 floods followed a quasi-constant log Kd value in Huveaune and Jarret Rivers (4.7, $R^2 = 0.90$ (N=21), Fig. 6B), which suggests that Cu dynamic was more controlled by the strong changes of SPM concentrations during flood than by variations in particles affinity. This value is slightly higher than the average one observed in Rhone River (4.3, Ollivier et al., 2011). Otherwise, the Cu log Kd in the plume gradually evolved from 4.5 (SPM > 100 mg L⁻¹; $R^2 = 0.73$ (N=13)) to 5.3 (SPM < 20 mg L⁻¹; $R^2 = 0.53$ (N=10)) with increasing salinity, a behaviour already observed by Turner et al. (2002), which could imply a repartitioning onto particles of higher affinity despite an apparently conservative behaviour (Fig. 5B). Although more dispersed, data points corresponding to baseflow conditions (22/03/2011) showed similar trend (Fig. 6B), exhibiting higher log Kd values than during floods and reaching the highest values (~5.7) at the lowest SPM concentrations. Ollivier et al. (2011) also observed higher log Kd in baseflow than in floods in Rhone River. Similar phenomena was attributed by Cobelo-Garcia et al. (2004) to different origin of SPM: (1) detrital material from soil weathering, (2) organic material from biota and urban/industrial effluents, the former being predominant during flood events together with an increase of sand material. In the present

case, the significant correlation between main detrital components (e.g. Al-Fe, $R^2 = 0.91$ (N=29)) observed for flood particles, and the higher POC (see part 3.2.2) content in baseflow particles confirmed such hypothesis. However, during the 18/04/2012 by-pass event, a lower log Kd related to a significant desorption (Fig. 5B) was observed at low salinity (~4.1), which suggests that sorption properties of untreated wastewater particles are more sensitive to salinity, despite a very high organic carbon content.

3.2.3.2. *Pb*

During floods, the decrease of dissolved Pb with increasing salinity was non-conservative with points slightly above the conservative line at salinity between 15 and 35 (Fig. 5C), when in baseflow conditions points were above the conservative mixing line at low salinity, especially during the WWTP by-pass event. Both conditions indicate desorption of Pb from particles, but the observed differences between flood and baseflow conditions could result from different desorption kinetic, particles nature/reactivity and/or mixing time. This contrasts with conservative behaviour of dissolved Pb observed in Rhône River plume (Elbaz-Poulichet et al., 1996) or in the Gironde estuary (Kraepiel et al., 1997). At any period, Pb was mainly in the particulate fraction (more than 95%) at the outlet and progressively evolved in the plume towards a lower particulate/dissolved fractionation (Fig. 3C). As for Cu, the Pb dissolved/particulate fractionation during the 18/09/20109 floods followed a quasi-constant log Kd value in Huveaune and Jarret Rivers (6.1, $R^2 = 0.89$ (N=20), Fig. 6F). The value was higher than observed for Cu, which highlights a stronger affinity for particles, as already observed in previous studies (e.g. Cobelo-Garcia, 2004; Ollivier et al., 2011). In the salinity gradient, the log Kd values slightly decreased (5.7, $R^2 = 0.88$ (N=28)); some points had low values (~5.1) likely corresponding to marine-born particles. Similarly to Cu, during the 18/04/2012 by-pass event, the significant Pb desorption observed at low salinity (Fig. 5C) resulted in much lower log Kd values (down to ~4.5). In baseflow conditions (22/03/2011), the log Kd values in the salinity gradient were also lower (5.5, $R^2 = 0.78$ (N=9)) than those observed during floods.

3.2.3.3. *Cd*

During the 25/10/2011 flood and in baseflow conditions, dissolved Cd remained quite constant with increasing salinity showing a quasi-conservative behaviour as it was observed

in other study (Michel et al., 2000). During the 2 other flood campaigns and the WWTP by-pass event, points far above the theoretical conservative mixing line indicated a release of Cd from particles (Fig. 5D). Such desorption is well documented and is related to the formation of stable and soluble chlorocomplexes in the dissolved fraction (Dai and Martin, 1995; Elbaz-Poulichet et al., 1987; Audry et al., 2007; Waeles et al., 2008). This process could partly explain that Cd was mainly found in the particulate fraction (more than 94%) during flood events at the outlet and mainly switched as dissolved at the end of the plume (Fig. 3D). Accordingly, if the Cd dissolved/particulate log Kd during the 18/09/20109 floods remained quasi-constant in Huveaune and Jarret Rivers (~ 4.7 , $R^2 = 0.72$ (N=22), Fig. 6D), it decreased in the plume down to ~ 4.5 (25/10/2011 and 21/05/2012 campaigns) and ~ 4.1 (28/11/2012 campaign), in relation to a significant Cd desorption (Fig. 5D), as already evidenced in other environment (e.g. Turner et al., 2004). In baseflow conditions (22/03/2011), higher log Kd values (5.2, $R^2 = 0.75$ (N=12)) could be due to a different affinity for particles mostly originating from TWW (Oursel et al., 2013).

3.2.3.4. Zn

During two floods (25/10/2011 and 21/05/2012) and in baseflow conditions, dissolved Zn decreased with increasing salinity with a quasi-conservative behaviour (Fig. 5E) as it was observed in the Gironde estuary (Michel et al., 2000). During the 28/11/2012 flood, points above the mixing line indicated a release of Zn from particles at low salinity as previously reported for the Scheldt estuary (Zwolsman et al., 1997) and Rhône River plume (Elbaz-Poulichet et al., 1996). During the WWTP by-pass event, points situated far below the mixing line at salinity lower than 15 indicated an adsorption on particles. The particulate/dissolved fractionation of Zn remained quite constant along the salinity gradient during flood events, but increased in baseflow conditions (Fig. 3E). However, during floods, even if Zn was mainly found in the particulate fraction (more than 90%) at the outlet, dissolved Zn concentrations can be 3-fold higher than WQC_{Zn} value (126 nM, Durán and Beiras (2013)) and even more during WWTP by-pass (4-fold higher). Zn dissolved/particulate log Kd (Fig. SI-1A) followed no clear pattern, neither from one campaign to another nor within the plume.

3.2.3.5. Co

During floods, dissolved Co exhibited a non-conservative behaviour with points above the mixing line (Fig. 5F), indicating Co release from particles at low salinity as it was observed in the Sagami and Wasaka estuaries (Takata et al., 2010) and in the Hudson River estuary (Tovar-Sánchez et al., 2004). In baseflow conditions and during the WWTP by-pass event, dissolved Co showed a quasi-conservative behaviour. Particulate Co was the dominant fraction during floods in outlet (more than 90%), on contrary to baseflow conditions (Fig. 3F). The Co dissolved/particulate log Kd during the 18/09/2009 floods remained stable (4.8, $R^2 = 0.96$ (N=22), Fig. SI-1B) in Huveaune and Jarret Rivers. It decreased at low salinity down to ~ 4.4 for the 18/09/2009 floods and to ~ 3.5 for the 18/04/2012 by-pass event. As noticed for Cu and Cd, higher log Kd (~ 5) were observed during baseflow.

3.2.3.6. Ni

In any conditions, dissolved Ni decreased with increasing salinity with a non-conservative behaviour, with points located above the mixing line indicating Ni release from particles (Fig. 5G). However, a conservative behaviour was commonly observed in the Ob and Yenisey estuaries (Dai and Martin, 1995), in Rhône River plume (Elbaz-Poulichet et al., 1996) and in the Gironde estuary (Michel et al., 2000). As Ni was mainly in the dissolved fraction in baseflow conditions and in the particulate fraction in floods (more than 80%), the particulate/dissolved fractionation of Ni increased in the plume in baseflow conditions, but decreased during flood events (Fig. 3G). The variations of Ni partitioning (Fig. SI-1C) followed similar trends than Co: stable value of log Kd in rivers during the 18/09/2009 floods (4.4, $R^2 = 0.98$ (N=22)), slight decrease at low salinity down to ~ 4.1 for floods and to ~ 3.8 for the by-pass event, higher values up to ~ 6 during baseflow.

3.2.3.7. As

Whatever the conditions, dissolved As increased with increasing salinity showing a conservative behaviour (Fig. 5H) as it was reported in the Gironde estuary (Michel et al., 2000) or in the Krka River estuary (Seyler and Martin., 1991). This conservative behaviour contrasted with a non-conservative behaviour observed for instance in the Penzé estuary (Vandenhecke et al., 2010). In this last study, the non-conservative behaviour was ascribed to inputs of dissolved As, probably from sediments (Vandenhecke et al., 2010), a mechanism not expected in our case considering the surface extension of the plume. In the Krka River

estuary, dissolved As concentration was lower in Krka River than in the Mediterranean Sea due to a low contamination of the Krka River basin (Seyler and Martin., 1991). Total dissolved As was observed between 12.4 and 18.9 nM in the Adriatic Sea (Pettine et al., 1997), slightly lower than in this study (21 nM, Table 1). In baseflow period, in rivers and in outlet, dissolved As form was dominant (in higher proportion in outlet) whereas during floods, particulate fraction dominated (70-80%). Finally in the marine end-member, dissolved As was in higher proportions compared to particulate one. Such variations can reasonably be explained considering the dissolved/particulate fractionation (Fig. SI-1D): during floods, in both rivers and salinity gradient, $\log Kd$ exhibited constant values (3.9, $R^2 = 0.94$ (N=31)), significantly lower than the ones observed for other studied metals; low values for the by-pass event (3.5, $R^2 = 0.99$ (N=7)) (18/04/2012); higher values up to ~4.1 during baseflow.

3.3. Remobilization experiment

The differences between the outlet baseflow sample and the artificial flood outlet sample were the following (Fig. 6A, 6C, 6E and SI-2A to SI-2D, open vs. close symbols at $S=0$). For all studied metals but As, dissolved concentrations were higher in the artificial flood sample than in the outlet baseflow sample, indicating a partial desorption from particles: from 20% (Ni) to 550-600% (Pb and Cu). However, the particulate fraction remained dominant (83 to 98%) for all the studied elements, a similar situation as encountered in field samples from rivers and outlet during flood events. Copper, Cd and Pb exhibited three contrasted behaviors (Fig. 6), according to the range of salinity where the maximal desorption occurred. The DOC (data not shown) showed a conservative behavior with salinity, in agreement with field data (Fig. 5A). Conversely, all the studied trace elements were desorbed from SPM. As observed in many other estuarine systems, such non-conservative behavior resulted from a complex interplay of various parameters and processes such as: intrinsic affinity of trace metals vs. flood particles (Kd), SPM concentration, kinetic of association/dissociation reactions, ionic strength, pH changes, competition of major ions (cations and anions) toward trace metals and particles binding site, Considering the deviation from the conservative line, Cd was the most intensively released element (Fig. 6B), whereas As was the less released (Fig. SI-2D). The shape of desorption curves showed that the maximum absolute release occurred at low salinity for Cu, Zn and Ni, at low to mid-salinity for As, mid-salinity for Cd and Co, and at high salinity for Pb. However, when expressed in term of percent of released metal (i.e. the fraction of metal released from SPM), the maximum release was in all cases obtained for the

longest contact time (24 h) at the highest salinity ($S=30$); it varied from 4% (Pb) to 25% (Cd and Cu) of initial particulate metal content. These discrepancies between elements resulted from different affinities for particles ($\log K_d$) and effect of salinity. No element but Co has reached its desorption isotherm plateau after 24 h of contact time (data not shown), which suggested slow kinetics. The restricted number of data, however, prevents modelling the kinetics of metal/particles affinity. In order to compare this experiment with field conditions, experiment results corresponding to 15 min and 24 h of contact time were plotted in terms of metals dissolved/particulate fractionation vs. SPM concentration (Fig. 6 and SI-1). Consistently with field patterns, Pb and As depicted the highest ($\log K_d > 5$) and lowest ($\log K_d < 4$) values, but $\log K_d$ experimental values were slightly lower than flood events field values. That may be due to a higher proportion of large particles having weak affinity for trace element in the flood deposit sample compared to SPM in floods (Benoit and Rozan, 1999).

Such results, gathering field data and lab experiments, allowed a better assessment of metals behaviour and fate in the studied environment. Compared to the results from another experiments performed by Oursel et al. (2013) to described baseflow conditions, they suggested that during floods, sorption/desorption mechanisms in the salinity gradient are slower than in baseflow conditions. This is likely due to differences in SPM: particles are mostly terrigenous during flood and anthropogenic with a TWW signature during baseflow (Oursel et al., 2014). Slower kinetics in the plume during flood events, however, may result in subsequent desorption after sediment deposition, amplifying a potential impact on benthic organisms.

4. Concluding remarks

During rain periods, the dissolved and particulate trace metal temporal dynamics in rivers were flood, river and element dependent. The studied system was highly dynamic and rapidly answered from the beginning of rain, which is typical of small Mediterranean rivers, especially when the water catchment is urbanized, requiring an adapted monitoring strategy. Metals contamination was mainly brought by the particulate fraction that originated from the leaching of impervious surfaces and soil erosion. The dissolved/particulate fractionation of a given trace element expressed in $\log K_d$ followed in rivers quasi-constant values during floods; Pb and As had respectively the highest (~ 6.1) and the lowest (~ 3.9) $\log K_d$ values. For all studied elements but Pb, $\log K_d$ values were lower in floods than during baseflow,

indicating lower affinities for particles; the high values of particulate trace element concentration during floods mostly resulted therefore from high SPM concentrations.

During flood conditions, the average concentrations of dissolved and particulate metals at the outlet were in the same range than in rivers, indicating that the WWTP signature, predominant during baseflow, was diluted and masked during floods. As floods brought to the sea around 94% of the annual SPM discharge, differences in total trace element contents between floods and baseflow had little weight on the annual particulate trace element fluxes that were mostly driven by rain events, as already observed in similar Mediterranean rivers. Oppositely, annual dissolved metals fluxes were likely mainly brought during baseflow periods, due to the strong influence of treated/untreated wastewaters inputs. An accurate evaluation of the global pollutants discharge to the coastal zone, however, would require a continuous monitoring of inputs.

When discharged to the sea, most of the studied elements presented a non-conservative behaviour along the salinity gradient due to desorption from particles, especially at highest salinity. The apparent inconsistency between floods was likely due to temporal variations of element-bearing phases at the outlet, depending on element dissolved/particulate partitioning in freshwater before discharge.

The laboratory remobilisation results were consistent with field data and suggested that element desorption from particles is slower than sedimentation. With regard to heavy particles, it results in a potential impact of the sediment on benthic organisms and a possible further desorption after sediment resuspension events. With regard to light particles, it results in a possible desorption during offshore transport.

The WWTP by-pass in baseflow conditions resulted in higher contamination of the dissolved phase (e.g. close to 350 nM of dissolved Cu), due to weaker affinity of untreated wastewater particles for the studied trace elements. Discharge of these untreated effluents leads to trace metals levels significantly overpassing acute water quality criteria for the protection of coastal ecosystems (e.g. WQC_{Cu} 22 nM). Consequently, the potential impact of such strongly-polluted events on coastal organisms may result from (1) direct exposure to desorbed metals due to salinity effects and (2) ingestion of organic-rich wastewater particles and associated pollutants by zooplankton and further transfer to food web (Bănar et al., 2014). Further studies, however, are needed to investigate the impact of such large agglomeration discharge on coastal micro- and macro-organisms in comparison to other sources (large rivers, aerosols).

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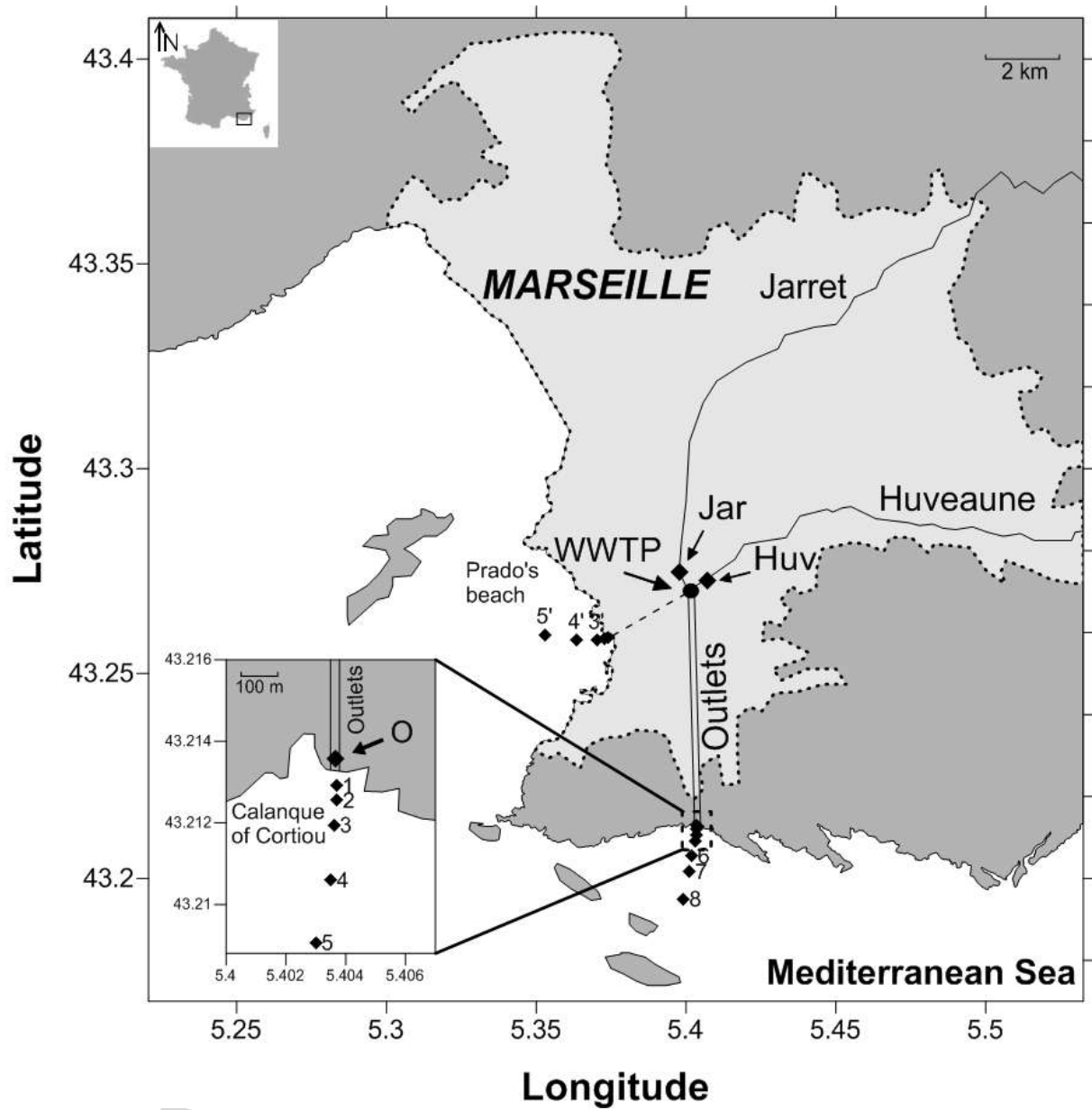
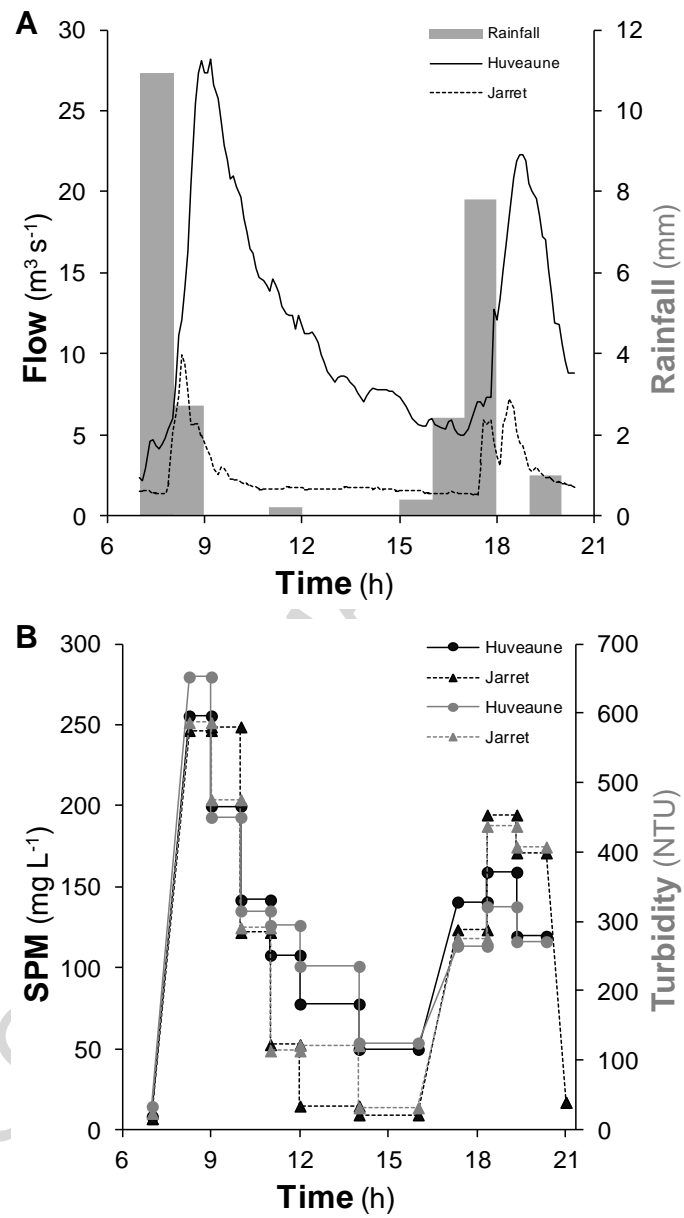
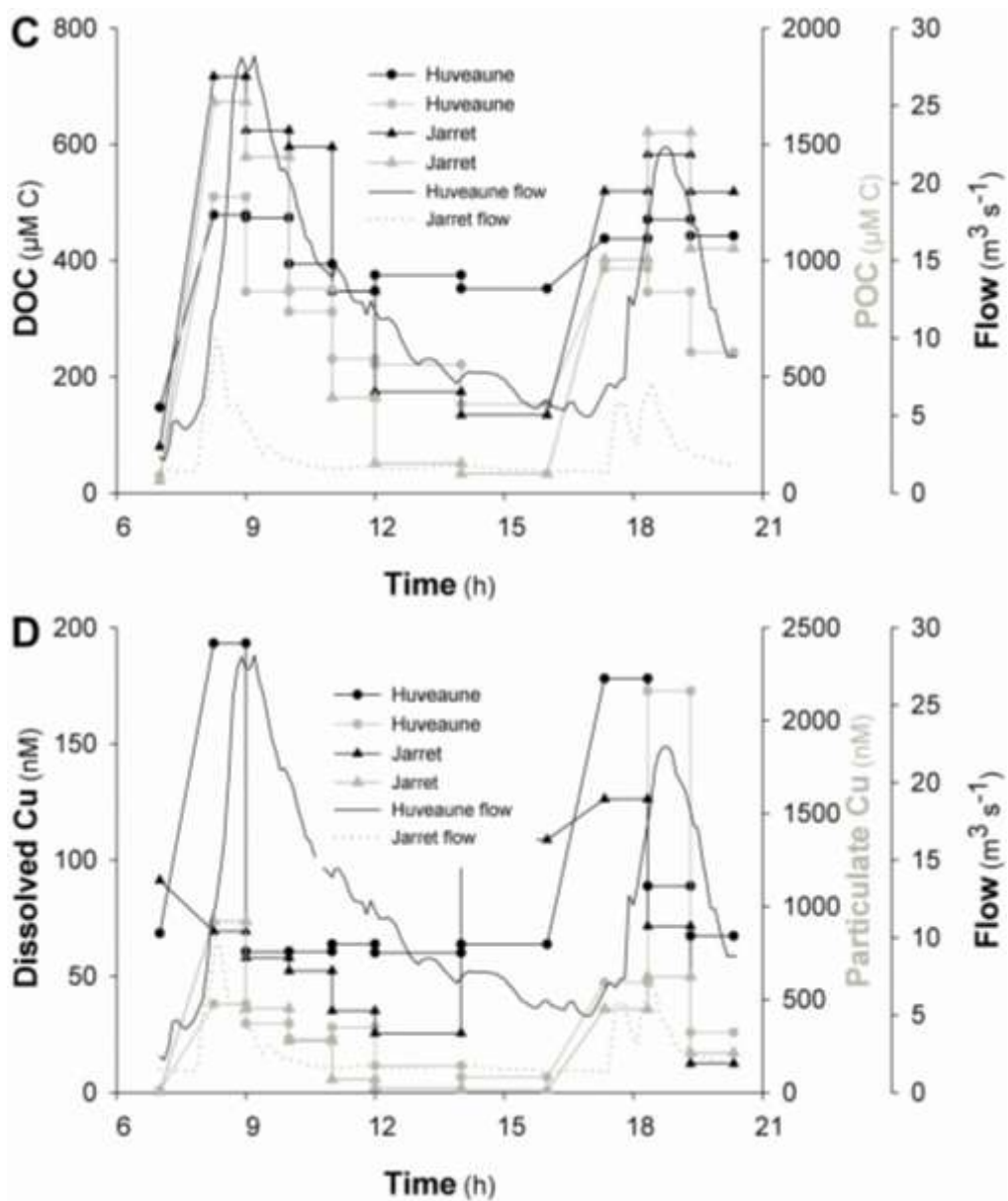


Figure 1

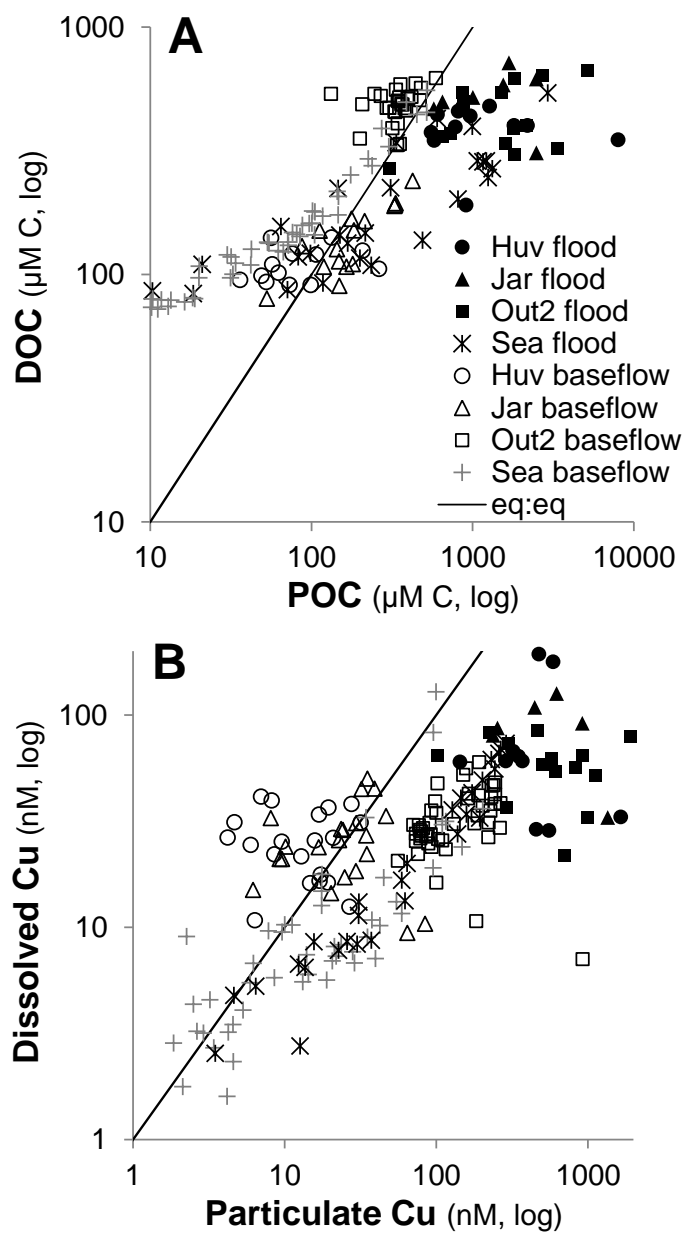
Oursel et al. Fig. 2

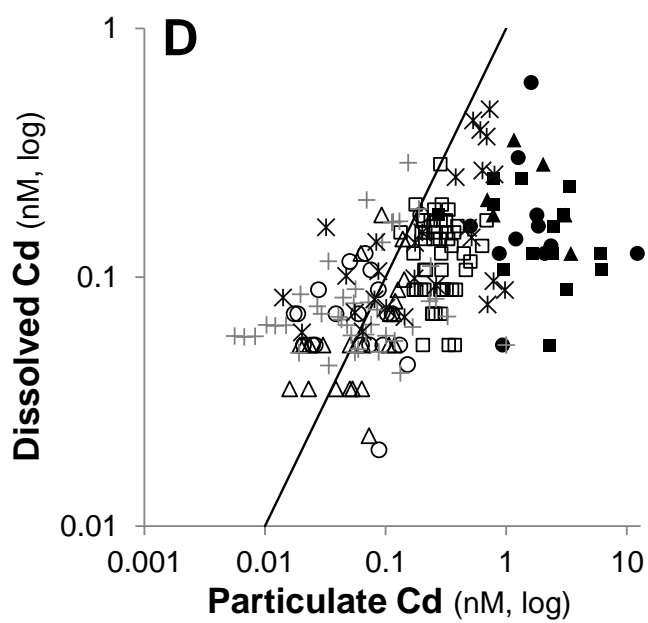
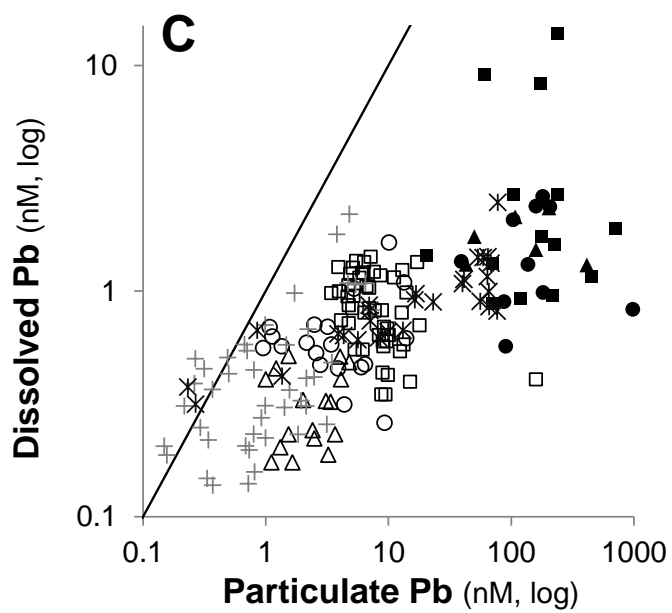


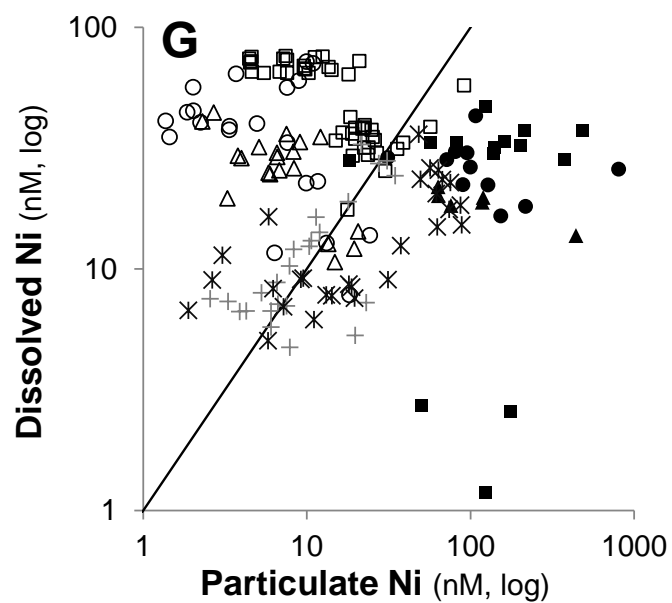
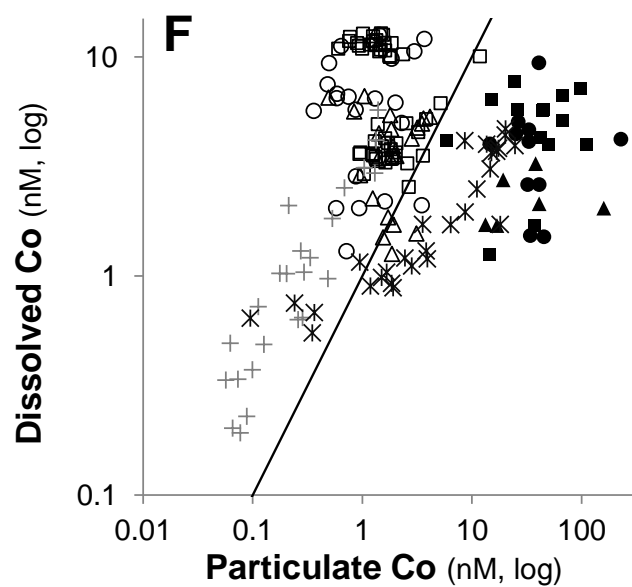
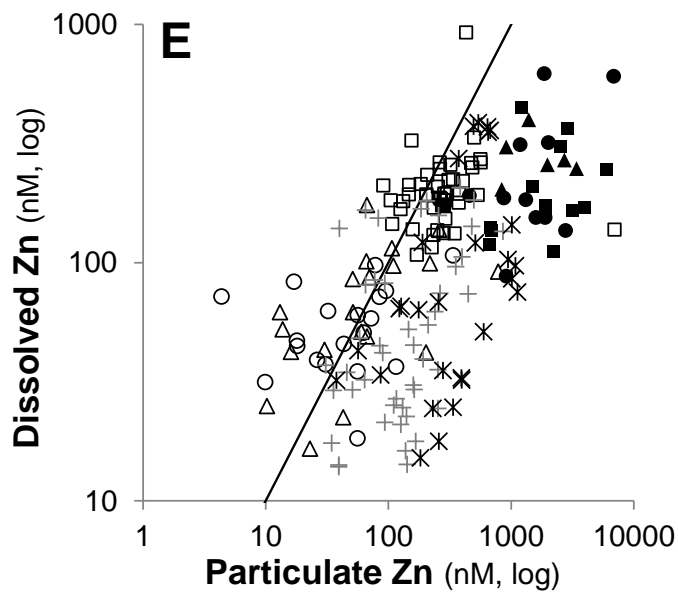


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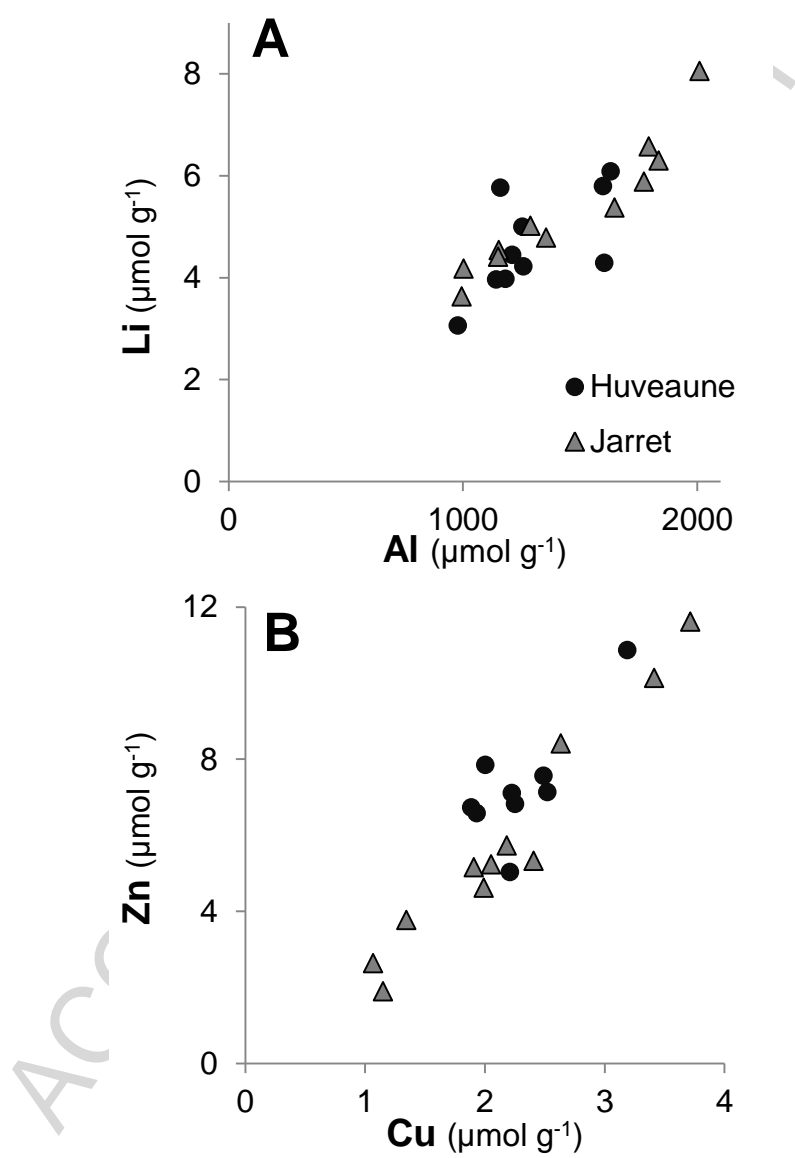
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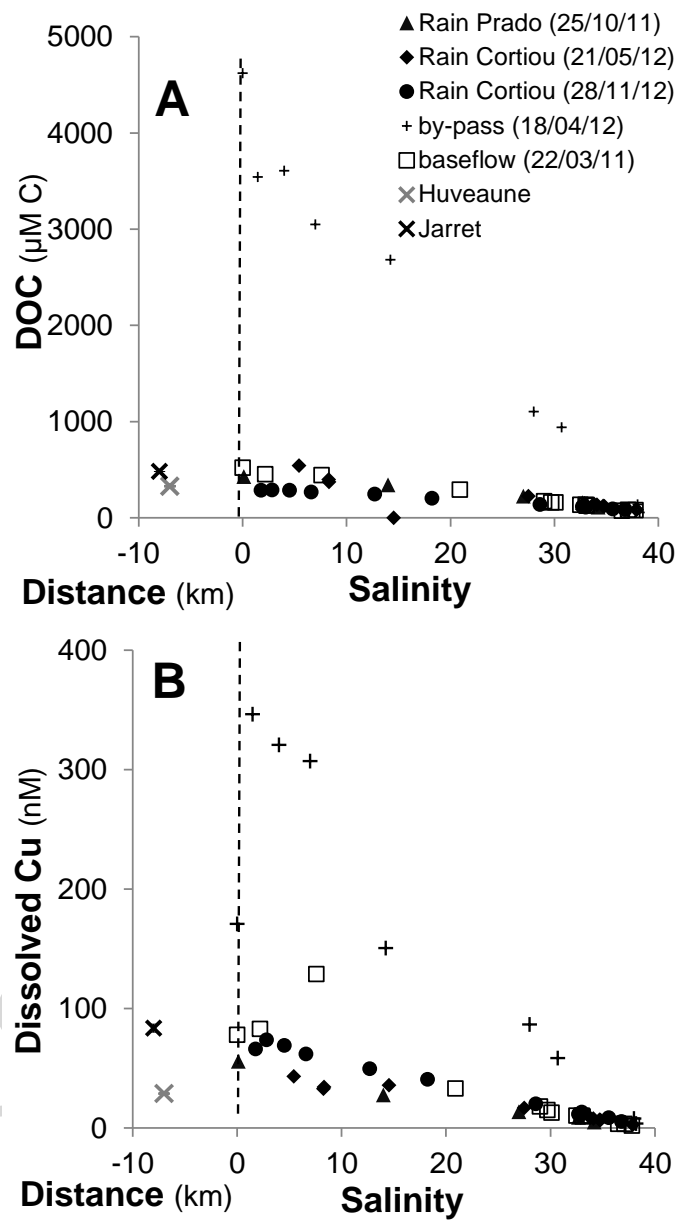


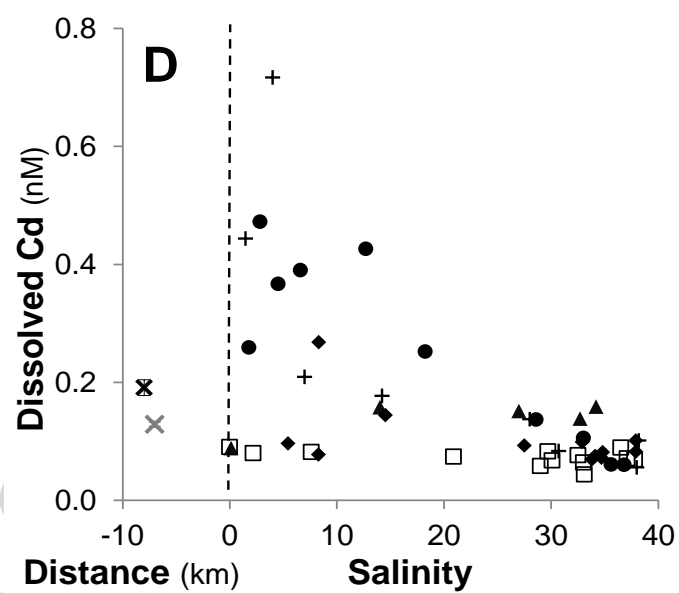
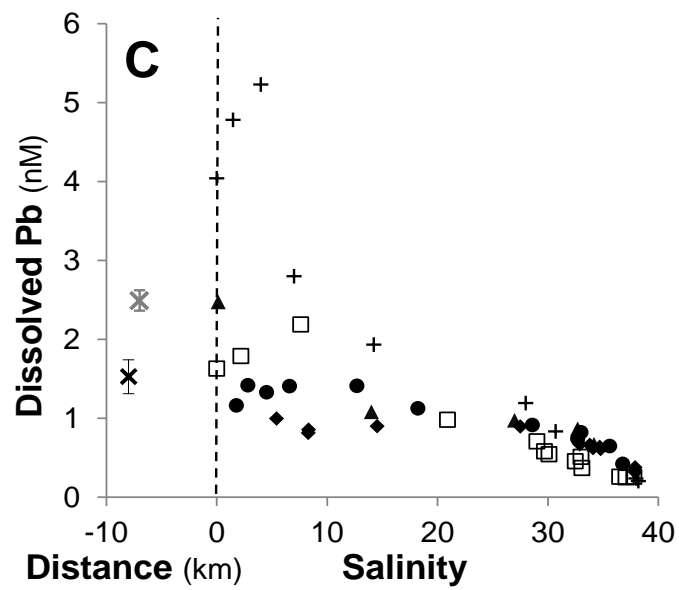


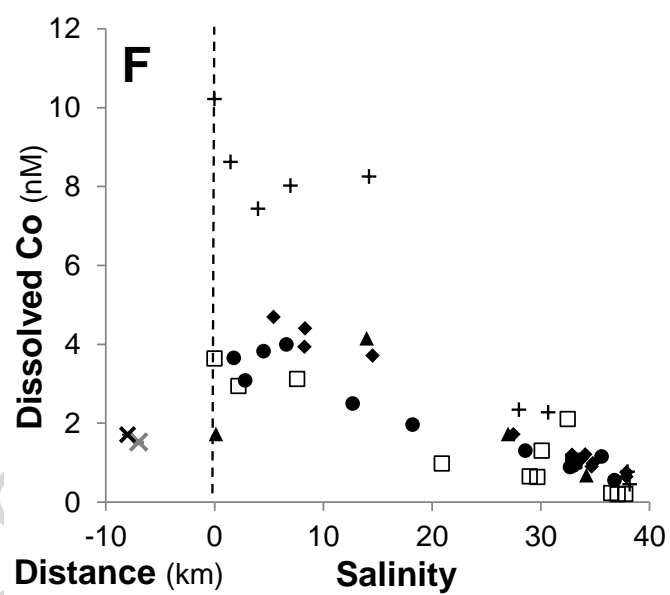
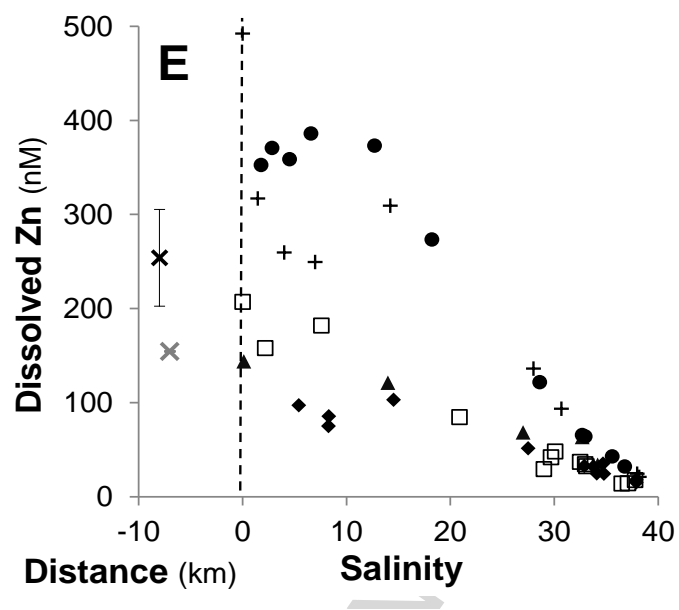
Oursel et al. Fig. 4

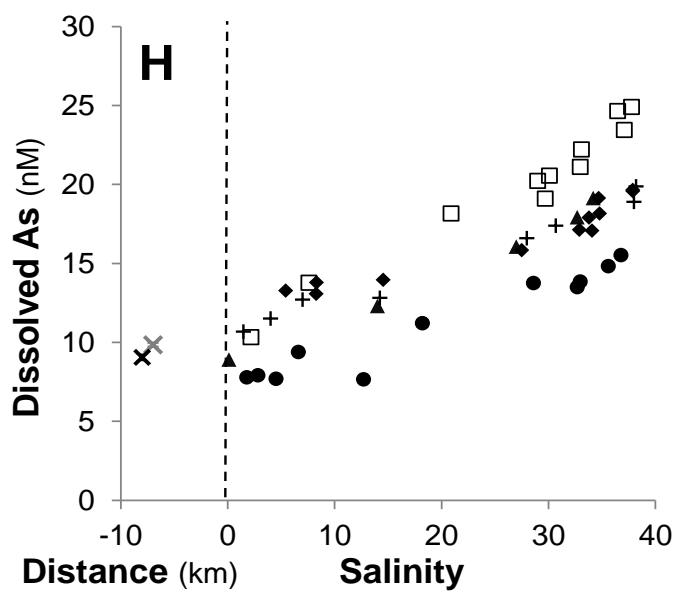
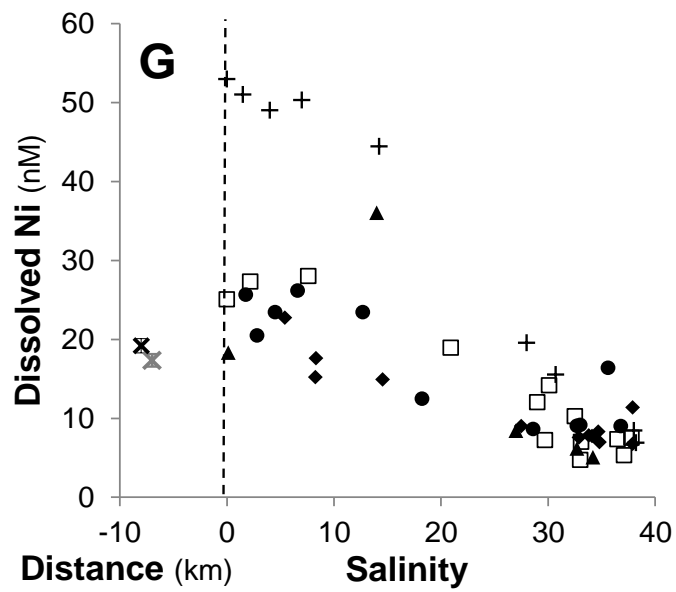


Oursel et al. Fig. 5

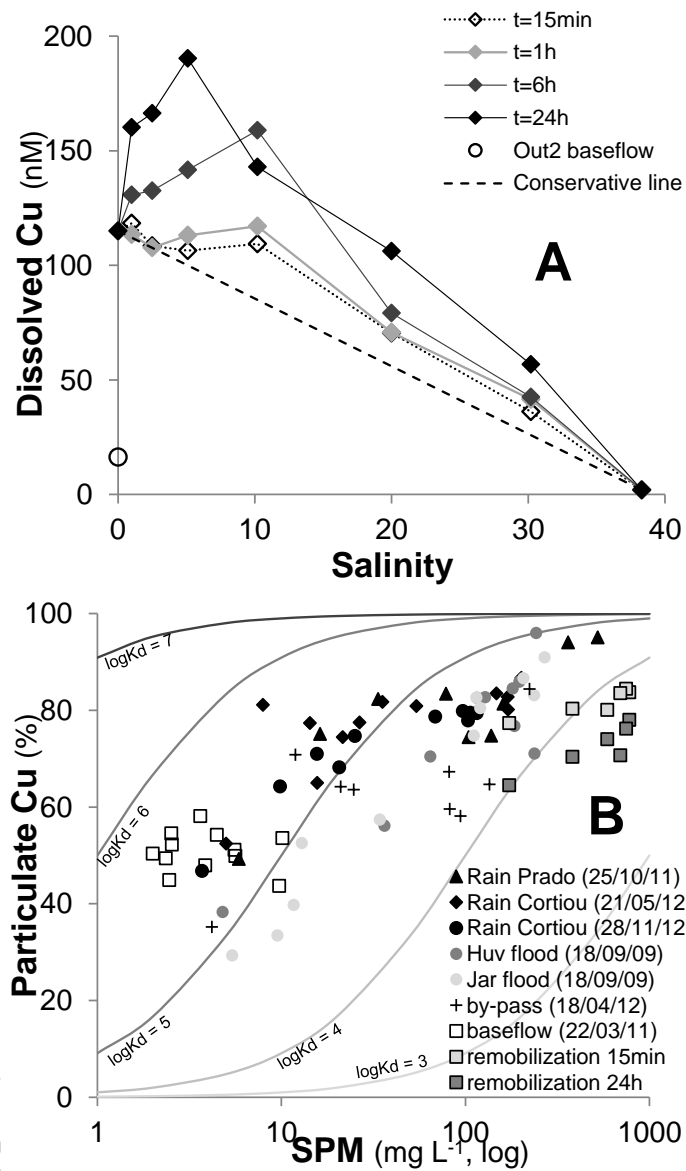


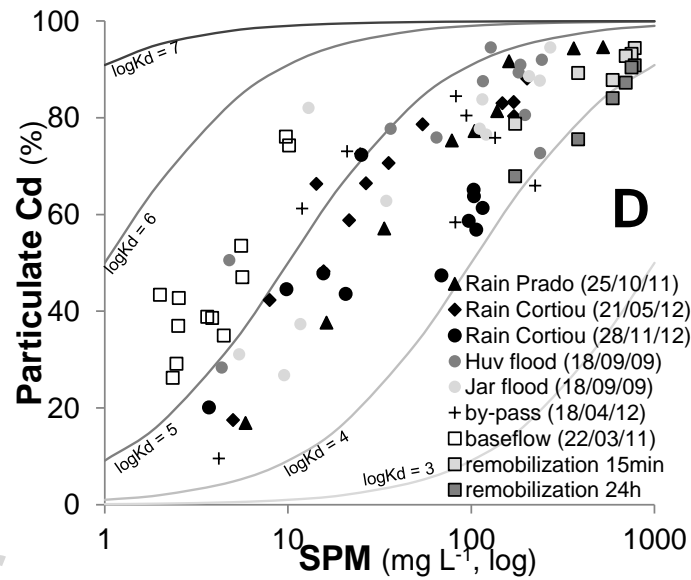
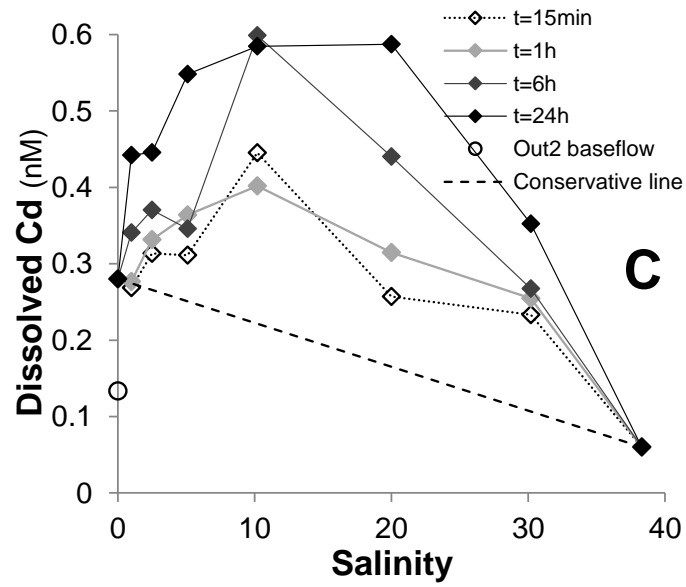


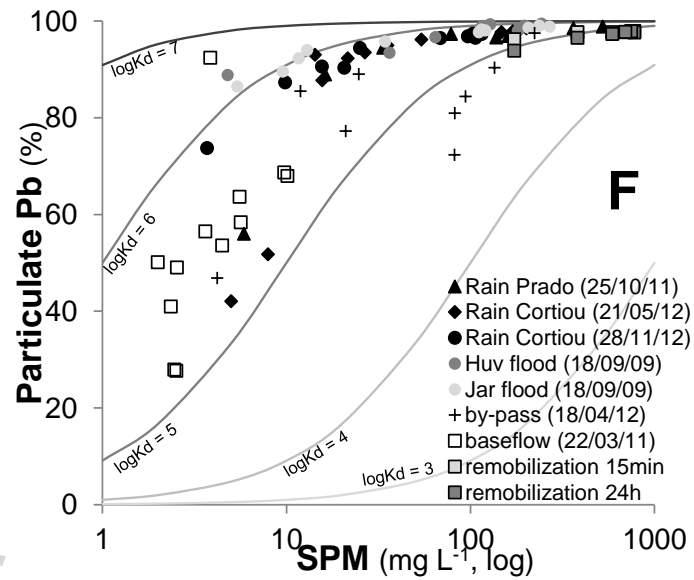
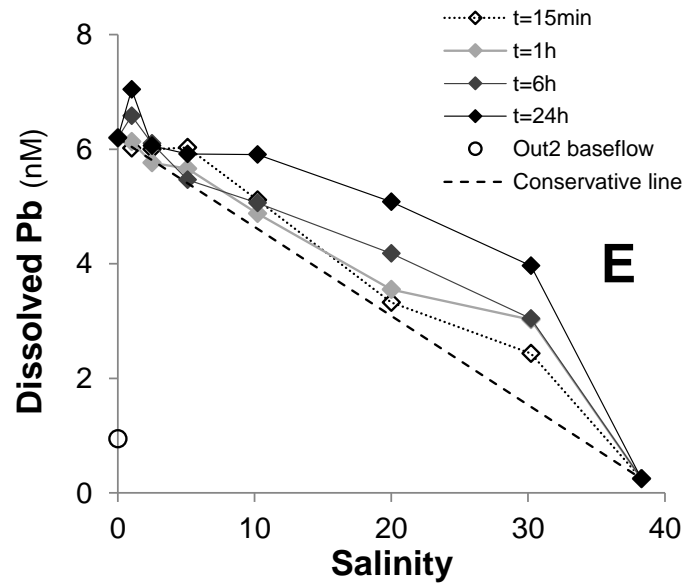




Oursel et al. Fig. 6







Oursel et al. Table 1

	SPM (mg L ⁻¹)	DOC (µM C)	POC (µmol C g ⁻¹)	Cu (nM)		Pb (nM)		Cd (nM)		Zn (nM)		Co (nM)		Ni (nM)		As (nM)	
				D	P	D	P	D	P	D	P	D	P	D	P	D	P
Huveaune-average	200	400	5800	79	640	1.5	210	0.19	2.3	270	1900	4.0	48	27	160	9.9	38
(min-max)	(78-490)	(190-480)	(4300-7700)	(28-193)	(140-2200)	(0.57-2.6)	(39-980)	(0.05-0.60)	(0.51-12)	(88-620)	(460-6900)	(1.5-9.4)	(15-230)	(17-43)	(31-800)	(9.6-10)	(24-50)
Jarret-average	230	530	7600	88	580	1.7	150	0.22	1.7	280	1700	2.3	44	19	140	9.1	17
(min-max)	(120-460)	(310-720)	(4600-10800)	(33-130)	(240-1400)	(1.3-2.3)	(43-420)	(0.12-0.36)	(0.7-3.4)	(200-400)	(630-3400)	(1.7-3.3)	(13-160)	(14-22)	(63-440)	(9.0-9.1)	(13-20)
Outlet-average	210	450	8300	59	680	3.5	210	0.16	2.5	210	2100	5.0	46	27	170	N.M.	N.M.
(min-max)	(35-590)	(270-670)	(4500-14200)	(22-85)	(100-1900)	(0.88-14)	(21-700)	(0.05-0.25)	(0.27-6.1)	(110-450)	(290-6000)	(1.3-7.7)	(5.9-110)	(1.2-47)	(18-490)		
Seawater-average	4.5	83	3600	3.3	5.2	0.25	0.26	0.07	0.04	21	120	0.52	0.24	8.4	14	21	0.44
(min-max)	(3.1-6.0)	(70-100)	(1800-5400)	(1.6-4.1)	(1.9-9.2)	(0.14-0.36)	(0.04-0.37)	(0.05-0.08)	(0.01-0.09)	(14-31)	(61-160)	(0.22-1.0)	(0.06-0.55)	(5.7-9.9)	(6.1-20)	(20-21)	(0.32-0.54)
Eygoutier-average	260	1400	12500	52	720	5.6	230	0.051	14	20	3300	N.M.	N.M.	N.M.	120	N.M.	N.M.
Rhône-average	660	N.M.	N.M.	31	260	0.34	98	N.M.	2.0	35	970	N.M.	87	18	310	27	N.M.
(min-max)	(95-2900)			(16-50)	(42-990)	(0.076-2.1)	(15-600)		(0.34-7.1)	(11-130)	(190-3600)		(17-300)	(14-28)	(63-1100)	(16-63)	

N.M.: not measured

Figure caption

Table 1: Variations of dissolved (D) and particulate (P) elements concentrations during flood conditions in the rivers, at the outlet and in the seawater end-member compared to the Eygoutier River (Nicolau et al., 2012) and to the Rhône River (Ollivier et al., 2011).

Figure 1: Map of the studied site: the dotted area represents the urbanized zone of Marseille, (inset) position of the sampling sites. Sampling points: O: outlet; Jar: Jarret River; Huv: Huveaune River.

Figure 2: Temporal variation of (A) the flow in both rivers and hour rainfall in Marseille; (B) the suspended particulate matter (SPM) and the turbidity; (C) the DOC and POC; (D) dissolved and particulate Cu during the 18/09/2009 flood events. In each figure, data points represent measurements on composite samples collected during one or two hours (see part 2.2.1.).

Figure 3: Variation of dissolved elements as a function of particulate elements expressed in $\mu\text{M C}$ for DOC/POC (A) and in nM for Cu (B), Pb (C), Cd (D), Zn (E), Co (F) and Ni (G) for all flood and baseflow period (Oursel et al., 2013) samples; the line (eq:eq) represents the same element concentration in the dissolved and particulate fractions.

Figure 4: Particulate contents variation of (A) Li vs. Al and (B) Zn vs. Cu during Huveaune and Jarret flood events.

Figure 5: Variation of DOC and dissolved metal in Huveaune and Jarret rivers, as a function of the distance until outlet, and from outlet to seawater, as a function of the salinity gradient for flood, baseflow (Oursel et al., 2013) and by-pass sampling campaigns for DOC (A), Cu (B), Pb (C), Cd (D), Zn (E), Co (F), Ni (G) and As (H); Error bar for the river points correspond to measured minimum-maximum range.

Figure 6: Dissolved Cu (A), Cd (C) and Pb (E) concentrations as a function of salinity after different contact time during the remobilization experiment; Particulate Cu (B), Cd (D) and Pb (F) fraction vs. suspended particulate matter (SPM) concentration for the various studied conditions compared to theoretical lines for given $\log K_d$ values.