

# Silicon pool dynamics and biogenic silica export in the Southern Ocean inferred from Si-isotopes

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Abstract. Silicon isotopic signatures ( $\delta^{30}$ Si) of water column silicic acid (Si(OH)<sub>4</sub>) were measured in the Southern Ocean, along a meridional transect from South Africa (Subtropical Zone) down to 57° S (northern Weddell Gyre). This provides the first reported data of a summer transect across the whole Antarctic Circumpolar Current (ACC).  $\delta^{30}$ Si variations are large in the upper 1000 m, reflecting the effect of the silica pump superimposed upon meridional water transfer across the ACC: the transport of Antarctic surface waters northward by a net Ekman drift and their convergence and mixing with warmer upper-ocean Si-depleted waters to the north. Using Si isotopic signatures, we determine different mixing interfaces: the Antarctic Surface Water (AASW), the Antarctic Intermediate Water (AAIW), and thermoclines in the low latitude areas. The residual silicic acid concentrations of end-members control the  $\delta^{30}$ Si alteration of the mixing products and with the exception of AASW, all mixing interfaces have a highly Si-depleted mixed layer end-member. These processes deplete the silicic acid AASW concentration northward, across the different interfaces, without significantly changing the AASW  $\delta^{30}$ Si composition. By comparing our new results with a previous study in the Australian sector we show that during the circumpolar transport of the ACC eastward, the  $\delta^{30}$ Si composition of the silicic acid pools is getting slightly, but significantly lighter from the Atlantic to the Australian sectors. This results either from the dissolution of biogenic silica in the deeper layers and/or from an isopycnal mixing with the deep water masses in the different



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oceanic basins: North Atlantic Deep Water in the Atlantic, and Indian Ocean deep water in the Indo-Australian sector. This isotopic trend is further transmitted to the subsurface waters, representing mixing interfaces between the surface and deeper layers.

Through the use of  $\delta^{30}$ Si constraints, net biogenic silica production (representative of annual export), at the Greenwich Meridian is estimated to be  $5.2 \pm 1.3$  and  $1.1 \pm 0.3$  mol Si m<sup>-2</sup> for the Antarctic Zone and Polar Front Zone, respectively. This is in good agreement with previous estimations. Furthermore, summertime Si-supply into the mixed layer of both zones, via vertical mixing, is estimated to be  $1.6 \pm 0.4$  and  $0.1 \pm 0.5$  mol Si m<sup>-2</sup>, respectively.

## 1 Introduction

In the Southern Ocean, deep nutrient-rich waters ascend into the surface layer and are returned equatorward as subsurface waters, before the available nitrogen pool is fully used by phytoplankton. This contrasts with silicon (in the form of silicic acid, Si(OH)<sub>4</sub>), which is much more depleted by diatom growth and exported along the same pathway (Sarmiento et al., 2004). The Southern Ocean redistributes nutrients at a global scale, thus representing the largest unused nutrient reservoir of the ocean interior and bearing most of the potential to increase the efficiency of the marine biological carbon pump (Sigman et al., 2010). The strong decoupling between nitrogen (N) and silicon (Si) impacts marine productivity at lower latitudes and is due to a complex interplay between, (i) different diatom morphological types presenting varying cellular silicification (Baines et al., 2010), (ii) iron and light limitation favouring heavily silicified diatoms (Takeda, 1998; Franck et al., 2003), (iii) the "silicate pump" reflecting the deeper remineralization of Si compared to N (Dugdale et al., 1995; Brzezinski et al., 2003a) and (iv) grazing-induced changes in cell wall silicification as an inducible defence against herbivory (Smetacek et al., 2004; Pondaven et al., 2007; Schultes et al., 2010). The zonal silicic acid gradient is thus controlled by surface water Si uptake by phytoplankton (e.g. diatoms) to form opaline cell walls (referred to as biogenic silica, bSiO<sub>2</sub>; Pondaven et al., 2000; Brzezinski et al., 2001). At the same time, it is affected by remineralization and physical processes such as vertical mixing, lateral advection and subsequent intermediate water mass formation (Sarmiento et al., 2004, 2007). Deciphering the relative importance of these different processes, which control the nutrient distribution in the Southern Ocean, and understanding how they affect nutrient export to the lower latitudes is necessary to better constrain the role of the Southern Ocean in global biogeochemical cycles of the past, present, and future (Sarmiento et al., 2004; Sigman et al., 2010).

The isotopic composition of silicon ( $\delta^{30}$ Si) is a valuable tool to help address this point: (i) it integrates longer timescales and larger spatial scales compared to direct flux measurements (e.g. isotope dilution experiments; sediment traps), (ii) it does not manipulate micro-organisms which can induce artefacts, and (iii) the different sensitivities of the mass and isotopic balances in marine systems mean sources can be identified as well as the discrimination between different processes, e.g. bSiO<sub>2</sub> production (De La Rocha et al., 1997), bSiO<sub>2</sub> dissolution (Demarest et al., 2009), and mixing (Fripiat et al., 2011a). Si(OH)<sub>4</sub> uptake by diatoms leaves a clear imprint on the isotopic compositions of both Si(OH)<sub>4</sub> and bSiO<sub>2</sub>. The fractionation factor ( ${}^{30}\varepsilon = -1.1 \pm 0.4$  ‰) reported by De La Rocha et al. (1997) seems independent of temperature, species (De La Rocha et al., 1997), cell size (Cardinal et al., 2007) and appears relatively constant across contrasted hydrological environments including fresh water systems (Alleman et al., 2005). By compiling existing data, Fripiat et al. (2011a) estimate an average  ${}^{30}\varepsilon$  for the Antarctic Circumpolar Current (ACC) of  $-1.2 \pm 0.2$  ‰, a value indistinguishable from in vitro estimates. Biogenic silica dissolution preferentially releases the light Si isotope, <sup>28</sup>Si, with a fractionation factor of -0.55 ‰ (Demarest et al., 2009). This fractionation is opposite to that occurring during Siuptake, by about half the absolute magnitude. By increasing the bSiO<sub>2</sub> dissolution over production (D:P) ratio, the overall fractionation factor would be dampened (Demarest et al., 2009). Currently, spatial and seasonal coverage of  $\delta^{30}$ Si measurements in the global ocean are limited and additional measurements, controlled experiments and modelling efforts are required to better constrain the origin and fate of the isotopic composition, as well as to better assess modern and past distributions. Indeed, an important application of natural Si-isotopic variability is to track past silicon supply: consumption ratios by measuring  $\delta^{30}$ Si of opal accumulating in sediments (De La Rocha et al., 1998; Brzezinski et al., 2002; Beucher et al., 2007).

In this study we present the first complete summer transect of  $\delta^{30}$ Si<sub>Si(OH)4</sub> (surface to seafloor) in the different fronts and zones of the Antarctic Circumpolar Current (ACC) and its adjacent subsystems (the SubTropical Zone and the northern part of the Weddell Gyre). The results are discussed in terms of source, pathway, and fate of silicon across the different Southern Ocean water masses and the interaction of the latter with the water masses from other oceanic basins. Results are compared with a similar transect sampled in spring and located in the Australian sector of the Southern Ocean (CLIVAR-SR3; Cardinal et al., 2005). The isotopic constraints allow us to quantify the annual net bSiO<sub>2</sub> production and the vertical summertime Si-supply into the mixed layer, both in the Antarctic Zone and Polar Front Zone, using the approach described in Fripiat et al. (2011a). A companion study, Fripiat et al. (2011c), is focused on the origin and fates of  $\delta^{30}$ Si<sub>bSiO2</sub> across the whole water column.

### 2 Materials and methods

Between 7 February and 24 March 2008, the International Polar Year (IPY) BONUS-GoodHope (BGH) cruise aboard the R/V *Marion Dufresne* covered a transect from Cape Town (South Africa) up to  $58^{\circ}$  S in the Southern Ocean roughly centred on the 0° meridian (Fig. 1a). The general aim of BGH, a GEOTRACES-IPY endorsed project, is to understand the interactions between the physics and biogeochemistry in the Atlantic sector of the Southern Ocean and its exchanges with the Indo-Atlantic connection in the wake of the Agulhas system. Eleven stations were sampled for Siisotopes which were distributed across the different subsystems (Fig. 1a); complete water column profiles (~13 depths per station) were obtained.

Seawater was collected using a CTD (conductivitytemperature-depth) rosette equipped with 121 Niskin bottles. Water samples (0.25 to 101) were immediately filtered on Nuclepore polycarbonate membranes (0.4 µm porosity) using Perspex filtration units under the pressure of filtered air (<2 bar). Filtered water samples for silicic acid analysis were stored in acid-cleaned polypropylene (PP) bottles at room temperature in the dark. Potential temperature ( $\Theta$ , sd = ± 0.0017 °C), salinity (S, sd = ± 0.0016) and dissolved oxygen (O<sub>2</sub>, sd = ± 2.1 µmol kg<sup>-1</sup>) were recorded using a SBE 911+ Seabird<sup>®</sup> 216 probe, with respectively, SBE3+, SBE4 and SBE43 sensors (Branellec et al., 2010). Data recording and validation of potential temperature, salinity and oxygen are described by Branellec et al. (2010).



**Fig. 1.** Map of the BONUS-GoodHope (left panel; February–March 2008) and CLIVAR-SR3 (right panel; October–December 2001) sections with bathymetry and main hydrographic features (following Sokolov and Rintoul, 2009, and Swart et al., 2008; mean path position): STF = SubTropical Front; SAF = SubAntarctic Front (middle branch); PF = Polar Front (middle branch); N-SACCF = northern branch of the Southern ACC front; N-SB = northern branch of the Southern Boundary. The area north of the STF is referred to as the SubTropical Zone (STZ), between STF and SAF the SubAntarctic Zone (SAZ), between the SAF and PF the Polar Front Zone (PFZ), between the PF and SB the Antarctic Zone (AZ), and south of the SB the Weddell Gyre (WG) on the left side and Seasonal Ice Zone on the right side. S refers to super stations and L to large stations of BGH. Mapping from Ocean Data View (R. Schlitzer, 2003, available at http://odv.awi.de).

### Si-analyses

Si(OH)<sub>4</sub> concentrations were measured in duplicate after 28 months via a colorimetric method following a calibration with artificial seawater Si(OH)<sub>4</sub> standards (two calibration lines; 0 to  $10 \,\mu\text{mol}\,1^{-1}$  and 0 to  $125 \,\mu\text{mol}\,1^{-1}$ ; Grasshoff et al., 1983; rsd < 2.5%) in the shore-based laboratory on the same samples that were analyzed for Si-isotopic composition. Since Si(OH)<sub>4</sub> concentrations of surface and some subsurface waters at stations north of the Polar Front (PF) were too low ( $<10 \,\mu mol \, l^{-1}$ ) to directly apply the Si purification procedure required for the Si-isotopic measurement (De La Rocha et al., 1996), a Si(OH)<sub>4</sub> preconcentration step was performed. The protocol used was adapted from the method described by Brzezinski et al. (2003b) and Reynolds et al. (2006) originally based on the MAGIC method (MAGnesium Induced Coprecipitation; Karl and Tien, 1992). This consisted of a single- (or two-) step scavenging of Si(OH)<sub>4</sub> by a brucite (Mg[OH]<sub>2</sub>) precipitate obtained by increasing the pH either with NaOH  $1 \mu mol l^{-1}$  (20 ml l<sup>-1</sup> seawater; Reynolds et al., 2006) or  $NH_4OH \ 13.5 \ \mu mol \ l^{-1}$  (6 ml  $l^{-1}$ seawater; Brzezinski et al., 2003b). The second scavenging was applied when the Si-recovery during the first step was incomplete. The precipitates were recovered by centrifugation and redissolved with HCl.

Silicon was co-precipitated with triethylamine molybdate (De La Rocha et al., 1996) with a minimum Si requirement of  $\sim 1.5 \,\mu$ mol. After combustion of the silicomolybdate precipitated in covered Pt crucibles at 1000 °C, the pure cristobalite phase was transferred to pre-cleaned PP vials. Dissolution of cristobalite was done in a dilute HF/HCl mixture as described in Cardinal et al. (2003). Isotopic measurements were carried out on a Nu Plasma MC-ICP-MS (MultiCollector-Inductively Coupled Plasma-Mass Spectrometer; ULB, Brussels) using Mg external doping in dry plasma mode following Abraham et al. (2008). The silicon isotopic compositions are expressed relative to a standard (NBS28) which is analyzed immediately after and before the sample, using the delta value (‰) as follows:

$$\delta^{30} \mathrm{Si} = \left( \frac{\left(\frac{^{30}\mathrm{Si}}{^{28}\mathrm{Si}}\right)_{\mathrm{sample}}}{\left(\frac{^{30}\mathrm{Si}}{^{28}\mathrm{Si}}\right)_{\mathrm{NBS28}}} - 1 \right) \cdot 1000 \tag{1}$$

The average precision and reproducibility of the measurements are  $\pm 0.1 \% (\pm 1 \text{ sd})$  for  $\delta^{30}$ Si (Reynolds et al., 2007). The accuracy of the measurements is checked on a daily basis on secondary reference materials (e.g. Diatomite) whose Si-isotopic composition is well known, based on an interlaboratory comparison exercise (Reynolds et al., 2007).

During the CLIVAR-SR3 cruise along the 145° E meridian (ACC; Australian sector; spring; Fig. 1b)  $\delta^{29}$ Si, by using <sup>29</sup>Si:<sup>28</sup>Si ratio instead of <sup>30</sup>Si:<sup>28</sup>Si ratio (Eq. 1), was measured instead of  $\delta^{30}$ Si (Cardinal et al., 2003, 2005), since the method used could not resolve interferences on <sup>30</sup>Si. To compare with the present BGH  $\delta^{30}$ Si data, the CLIVAR SR-3  $\delta^{29}$ Si values were converted to  $\delta^{30}$ Si values using the



**Fig. 2.** Interpolation of  $\Theta$  (°C, **a**), salinity (**b**), both with oxygen contours (µmol kg<sup>-1</sup>; white lines) distributions during BONUS-Goodhope (adapted from Speich et al., 2011).

theoretical conversion factor of 1.96 calculated from the kinetic fractionation law (Young et al., 2002).

### 3 Results

In the following discussion, the different water masses are defined by their potential temperature, salinity, and  $O_2$  properties (Fig. 2): Upper Circumpolar Deep Water (UCDW; oxygen minimum), Lower Circumpolar Deep Water (LCDW; deep salinity maximum), Antarctic Intermediate Water (AAIW; subsurface salinity minimum), Winter Water (WW, subsurface temperature minimum), SAZ-STZ thermoclines (between AAIW and the mixed layer, ML), and Antarctic Bottom Water (AABW; bottom increases in  $O_2$  content, decreases in temperature and salinity). As some water masses presented asymmetric distribution in both concentration and isotopic composition, the median and the interquartile range ( $\cdot 2/3$ , similar to 1 sd) has been chosen as metric for the "central values" and dispersion for the different water masses.

**Table 1.** Si(OH)<sub>4</sub> concentration and isotopic composition. Only the standard deviations of samples replicated are provided. The solid lines represent the mixed layer depth.

### Table 1. Continued.

Station	Depth	Si(OH) <sub>4</sub>	$\delta^{30} Si_{Si(OH)_4}$	sd
	m	$\mu$ mol l <sup>-1</sup>	%0	‰
Super 5	29	65.2	1.85	0.00
16 March 2008	88	64.2	2.00	0.00
		0.1.2	1.50	
$00.02^{\circ} \text{E}-57.32^{\circ} \text{S}$	151	84.1	1.50	0.03
	199	96.2	1.35	0.01
	251	101.5	1.37	0.05
	299	104.2	1.41	0.05
	400	110.5	1.44	0.36
	499	117.2	1.47	0.00
	700	116.5	1.44	0.20
	1003	120.0	1.27	
	1499	119.1	1.13	0.05
	1999	116.6	1.12	0.06
	2500	119.7	1.43	0.00
	3000	117.2	1.11	0.08
	3979	118.8	1.08	
Large 7	4	50.7	1.95	
14 March 2008	80	50.0	1.78	0.07
00.03° E–55.14° S	100	58.9	1.77	0.11
	149	74.4	1.54	
	199	79.7	1.53	0.17
	300	86.5	1.61	
	600	93.9	1.25	0.08
	1002	103.5	1.34	
	2096	122.7	1.38	
	2768	129.2	1.43	0.03
Super 4	11	22.2	2.42	
11 March 2008	79	22.3	2.37	0.06
00.00° E–51.87° S	100	21.9	2.16	0.08
	149	30.5	2.03	
	199	63.2	1.48	0.00
	300	86.0	1.41	0.09
	398	87.6	1.58	0.17
	553	88.9	1.37	0.16
	704	90.1	1.53	
	1201	101.9	1.35	
	1601	109.5	1.47	
	2001	121.6	1.30	
	2551	131.2	1.13	

Station	Depth	Si(OH) <sub>4</sub>	$\delta^{30} Si_{Si(OH)_4}$	sd
	m	$\mu$ mol 1 <sup>-1</sup>	%0	‰
Large 6	3	4.0	2.47	
8 March 2008	48	4.0	2.48	
01.18° E–50.22° S	98	5.1	2.47	
	129	20.1	2.03	
	200	41.3	2.03	
	251	48.2	1.57	0.15
	401	64.4	1.42	0.13
	602	75.7	1.39	
	801	79.3	1.46	
	999	79.1	1.37	0.15
	1501	78.4	1.44	
	2002	93.1	1.28	
	2500	115.4	1.36	
	2998	124.7	1.38	
	3596	125.9	1.62	
Large 5	9	1.8	2.77	
7 March 2008	70	1.9	2.84	0.08
02.50° E–49.02° S	100	2.0	1.99	0.03
	151	7.2	2.07	
	200	13.8	1.92	
	300	22.4	1.36	
	402	30.9	1.99	0.28
	600	44.6	1.63	
	799	65.6	1.53	0.01
	1000	68.2	1.58	
	2001	76.2	1.41	
	3001	105.7	1.37	
	4080	128.6	1.44	
Super 3	5	2.1	3.24	0.01
6 March 2008	41	2.1	2.85	
04.23° E-47.33° S	79	2.2	2.36	
	100	4.2	2.57	0.19
	149	9.2	2.30	
	200	10.4	2.42	
	401	25.3	1.93	0.35
	600	42.5	1.67	
	1002	67.8	1.43	
	1403	70.0	1.42	
	2000	76.6	1.34	
	2301	85.6	1.26	0.01
	2900	101.8	1.39	0.34
	3500	116.0	1.16	
	4101	122.6	1.37	
	4532	132.1	1.28	

In the mixed layer,  $\delta^{30} Si_{Si(OH)_4}$  and Si(OH)<sub>4</sub> concentrations displayed an inverse pattern with latitude (Table 1; Fig. 3a, b). <sup>30</sup>Si enrichment was associated with a decrease in Si(OH)<sub>4</sub> concentration from south to north, as also reported in earlier studies (Varela et al., 2004; Cardinal et al., 2005). This condition is mainly driven by the preferential uptake of <sup>28</sup>Si by diatoms (De La Rocha et al., 1997) along the northward advection of surface waters (Varela et al., 2004) followed by diatom export out of the mixed layer. This

study reveals heavier summer surface  $\delta^{30} \text{Si}_{\text{Si}(\text{OH})_4}$  signatures in agreement with an increased seasonal Si(OH)<sub>4</sub> depletion (Fig. 3) relative to spring results (Cardinal et al., 2005), for two different ACC transects (Australian and Atlantic sectors, respectively, Fig. 1).

Table 1. Continued	
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Table 1. Continued.

Station	Depth	Si(OH) <sub>4</sub>	$\delta^{30}$ Si <sub>Si(OH)</sub>	sd
	m	$\mu$ mol 1 <sup>-1</sup>	‰	‰
T	5	1.0	2.50	0.05
Large 4	5 30	1.0	2.50	0.05
5 March 2008	52	1.2	3.01	
05.52° E–46.01° S	61	1.1	2.44	
	90	2.9	2.72	
	151	6.9	2.41	
	300	16.3	1.90	
	501	29.5	1.70	
	749	51.5	1.68	0.07
	1000	68.3	1.45	
	1502	75.4	1.51	
	1999	73.3	1.36	0.09
	2497	87.1	1.36	
	3001	104.4	1.43	
	3500	113.7	1.40	
	4148	126.0	1.47	
Large 3	11	1.0	2.46	0.19
2 March 2008	28	0.9	2.78	0.23
06.53° E–44.54° S	101	4.4	2.32	0.12
	149	6.0	2.07	
	298	11.1	2.06	
	400	14.8	1.83	0.10
	1251	66.8	1.54	
	1997	71.4	1.51	
	3001	95.8	1.36	
	3601	110.5	1.22	
	4371	125.9	1.24	
Super 2	10	0.6	3.24	0.05
27 February 2008	40	1.0		
08.56° E–42.28° S	80	2.5	2.07	0.06
	100	3.2	1.89	0.08
	150	4.8	2.09	
	300	9.5	2.01	0.14
	601	19.0	1.83	
	1001	46.9	1.50	0.01
	1400	68.0	1.29	0.11
	2000	71.6	1.28	
	2501	66.1	1.42	0.09
	3500	87.1	1.30	
	4057	110.6	1.36	

Most of the  $\delta^{30}Si_{Si(OH)_4}$  variation across the BONUS-GoodHope transect was limited to the upper 1000 m, with a clear north-south gradient showing heavier Si-isotopic values equatorward associated with a decrease in [Si(OH)\_4] (Figs. 3 and 4). Although the isotopic variation below 1000 m was small, there was still a slight decrease in values of  $\delta^{30}Si_{Si(OH)_4}$  coincident with increasing Si(OH)\_4 concentrations (Fig. 4), as reported also by Cardinal et al. (2005) and Fripiat et al. (2011a). The Upper Circumpolar Deep Water (UCDW) had a slightly heavier  $\delta^{30}Si_{Si(OH)_4}$  signature and a

Station	Depth m	$Si(OH)_4$ $\mu mol 1^{-1}$	$\delta^{30}$ Si <sub>Si(OH)4</sub> ‰	sd ‰
Large 2	101	1.2	1.96	
25 February 2008	200	3.1	1.89	
13.10° E–36.45° S	504	14.4	1.77	
	600	18.9	1.73	
	1000	48.4	1.55	0.07
	2001	64.0	1.48	
	2900	69.2	1.57	
	4200	114.9	1.44	
	4570	117.9	1.13	0.01
Super 1	4	2.2	3.04	0.24
21 February 2008	25	2.3	2.89	0.07
13.10° E–36.45° S	75	4.5	2.07	0.01
	220	4.8	1.99	
	420	9.7	1.79	
	701	26.0	1.36	
	1002	51.4	1.64	
	1499	68.8	1.48	
	2001	60.0	1.37	
	2559	62.7	1.40	0.01
	2900	65.9	1.18	
	3499	88.2	1.51	
	4199	112.2	1.30	0.14
	4600	116.0	1.06	
	5000	115.6	1.05	

lower Si(OH)<sub>4</sub> concentration than Lower Circumpolar Deep Water (LCDW) and Antarctic Bottom Water (AABW).

### 4 Discussion

### 4.1 General considerations

In the following discussion, we will follow the silicon pathway and the transformations encountered along the meridional as well as the circumpolar circulations of the ACC. The ACC represents the largest mass transport of all ocean currents, a slab of water more than 2000 m thick moving eastward (Tomczak and Godfrey, 2001), but mainly centered in several jets along frontal systems: the inner ACC fronts (Sokolov and Rintoul, 2009); SubAntarctic Front, Antarctic Polar Front, Southern ACC Front. The ACC fronts represent almost impermeable barriers, delimiting zones with relatively constant hydrological and biogeochemical properties (Sokolov and Rintoul, 2007) although cross frontal exchanges do occur locally, associated with sharp topographic features. This wind-driven meridional ACC circulation is characterised by: (1) the shoaling of deeper isopycnal surfaces (UCDW and LCDW) in the Southern ACC towards the Antarctic divergence (Fig. 2; Pollard et al., 2002, 2006),



**Fig. 3.** Interpolation of  $\delta^{30}$ Si (**b**, **d**) and [Si(OH)<sub>4</sub>] (**a**, **c**) distributions in the upper 1000 m during BONUS-GoodHope (this study; **a**, **b**) and CLIVAR-SR3 (145° E; **c**, **d**; from Cardinal et al., 2005). Interpolation is from Ocean Data View (R. Schlitzer, 2003, available at http://odv.awi.de). The positions of the fronts from Speich et al. (2011) at the time of sampling are also shown (dashed white lines).

(2) part of the upwelled CDW losing buoyancy near Antarctica to form AABW (Orsi et al., 1999) and the remaining gaining buoyancy and being advected northward (Antarctic Surface Water, AASW) via Ekman pumping (Fig. 2; Sloyan and Rintoul, 2001; Pollard et al., 2002), (3) the subsequent formation of intermediate water masses taking place (AAIW, and SubAntarctic Mode Water, SAMW), spreading and deepening northward of the PF (Fig. 2: Sarmiento et al., 2004). The latitudinal  $\delta^{30}$ Si variation in the water column follows this meridional circulation (Figs. 3 and 4).

During the BGH which took place in late summer 2008, the measured silicon properties of the mixed layer can be considered as representative of final summer conditions. This is supported by (i) a Si-uptake in the mixed layer in the lower range of published values,  $0.02 \pm 0.03 \,\mu$ mol Si  $1^{-1} \, d^{-1}$  (mean Si-uptake from <sup>30</sup>Si tracer incubations for the BGH transect, n = 27 triplicates), accounting only for  $0.6 \pm 0.9 \,\%$ 



**Fig. 4.** Interpolation of  $\delta^{30}$ Si (a) and [Si(OH)<sub>4</sub>] (b) distributions for the complete water column during BONUS-GoodHope. Interpolation is from Ocean Data View (R. Schlitzer, 2003, available at http://odv.awi.de). The positions of the fronts at the time of sampling (Speich et al., 2011) are shown as dashed white lines.

of the available silicic acid concentration (Fripiat, 2010), and (ii) satellite Chl-*a*/POC data indicating that the main productive period in the area occurred roughly in December (Moore and Abbot, 2000; Allison et al., 2010).

# 4.2 Isotopic constraints on the silicic acid leakage from the Southern Ocean to the low latitude areas

### 4.2.1 Southern ACC

We define the southern ACC as the area south of the Polar Front where isopycnal surfaces shoal and UCDW and LCDW upwell near the surface (Fig. 2). From Fig. 5a, it is apparent that the subsurface waters (WW and UCDW) in the southern ACC are mixing interfaces between LCDW and AZ mixed layers. Indeed, both AZ-PF Winter Water (WW) and UCDW fit well on a mixing curve with the AZ-PF mixed layer and the LCDW as end-members, in agreement with our observations for the Indian sector of the Southern Ocean (Fripiat et al., 2011a). The UCDW and AZ-PF mixed layers generate a homogeneous deep mixed layer at the onset of winter convective mixing: Antarctic Surface Water (AASW; Park et al., 1998). Due to surface warming in the summer, Antarctic Surface Water stratifies into an upper ML and a lower WW, with the latter maintaining the original AASW characteristics (Pondaven et al., 2000a; Altabet and François, 2001; Fripiat et al., 2011a; Cavagna et al., 2011).

Using the mixing curve with the UCDW and AZ-PF ML median values (Fig. 5a) and the following equation:

$$f_{\rm UCDW} = \frac{\delta^{30} {\rm Si}_{\rm AASW} - \delta^{30} {\rm Si}_{\rm ML}}{\delta^{30} {\rm Si}_{\rm UCDW} - \delta^{30} {\rm Si}_{\rm ML}}$$
(2)



**Fig. 5.** (a) and (b):  $\delta^{30}$ Si and Si concentration values for different water masses across the BONUS-GoodHope transect (this study). Some mixing curve are also shown: between LCDW and AZ-PF mixed layer (blue line), between PF-AZ AASW and PFZ mixed layer (dashed red line), between PF-AZ AASW and PFZ mixed layer (green line). When there is no indication of zone, it refers to the median of the whole BGH transect. (c): schematic view of the processes responsible for the observed  $\delta^{30}$ Si distribution in subsurface waters across BONUS-GoodHope north of the Southern Boundary, representing the mixing interfaces between surface and deeper waters. Identified mixing interfaces are the AASW = WW, the AAIW, and the low latitude thermoclines. The arrows represent the mixing end-members keeping the same colors than in (a) and (b). (d): schematic view of the mixing interfaces across the CLIVAR-SR3 transect north of the Southern Boundary (Cardinal et al., 2005). Median and interquartile range · 2/3 represent the "central value" and dispersion for the water masses. The number of computed data for BONUS Goodhope, respectively for Si(OH)<sub>4</sub> and  $\delta^{30}$ Si<sub>Si(OH)4</sub>, is for UCDW (24:24), WW (11:11), AAIW (11:11); STZ-SAZ thermocline (7:7); AZ-PF ML (8:8); PFZ ML (9:9); SAF-SAZ-STZ ML (6:5). For CLIVAR SR3, it is for UCDW (41:17), WW (12:8), AAIW (18:8), SAMW (11:3), AZ-PF ML (8:7); PFZ ML (4:2); SAF-SAZ-STZ ML (0:6).

we estimate the mean UCDW Si mass fraction contribution  $(f_{\rm UCDW})$  to AASW in the AZ (represented by the mean of AZ-PF WW) to be 0.89. This value is close to the values reported for: (1) the HNLC area east of the Kerguelen plateau with the same methodology (i.e. 0.78, Fripiat et al., 2011a); and (2) also at the 0° Meridian between the SB and SACCF with a model using concentrations of nitrate and phosphates in winter (i.e. 0.90, Hoppema et al., 2000). By integrating the WW Si(OH)<sub>4</sub> concentration over the water column between the surface and bottom waters

of the WW layer (i.e. the  $300 \pm 50$  m deep AASW) and multiplying that amount with the UCDW contribution, we calculate that UCDW contributes  $15.7 \pm 2.6$  mol Si m<sup>-2</sup> to AASW. Standard errors for the mass balance estimations were obtained from Montecarlo simulations (normal distribution, n = 1000). The  $100 \pm 25$  m deep summer ML will thus inherit  $(100/300) \cdot (15.7 \pm 2.6 \text{ mol Si m}^{-2})$  or  $5.2 \pm$  $1.3 \text{ mol Si m}^{-2}$  which becomes available at the start of the growth season. The main assumptions behind these estimations are: (1) the UCDW Si-supply to the AASW mainly occurs during winter when the AASW is generated by deep convective mixing (Fig. 5a), and (2) the WW composition at the end of summer is still representative of the original AASW which represents the Si-source for the summer mixed layer (Pondaven et al., 2000a; Altabet and François, 2001; Fripiat et al., 2011a).

Assuming that steady state conditions apply (i.e. supply equals export at the annual scale), the effective annual net bSiO<sub>2</sub> production should be equal to the annual vertical UCDW Si-supply of  $5.2 \pm 1.1 \text{ mol Si m}^{-2}$ , which is again similar to the value of  $4.0\pm0.7$  mol Si m<sup>-2</sup> we reported earlier for the Indian sector (Fripiat et al., 2011a). This value exceeds the seasonal depletion estimated from a simple mixed layer mass balance which is  $3.6\pm0.9$  mol Si m<sup>-2</sup> yr<sup>-1</sup> (i.e. difference between WW and ML Si(OH)<sub>4</sub> concentration). Such imbalance indicates that some  $1.6 \pm 0.4 \text{ mol Si m}^{-2} \text{ yr}^{-1}$  are being supplied to the mixed layer during the summer stratification period. This flux is indistinguishable from the one observed in the Indian sector (Fripiat et al., 2011a). Sokolov and Rintoul (2007) stress the homogenous character of biogeochemical properties within the different ACC zones. Therefore, the discrepancy between net bSiO<sub>2</sub> production and seasonal Si(OH)<sub>4</sub> depletion is likely matched by a vertical supply of Si, rather than a horizontal supply, which would advect waters with similar Si-conditions. Note that Hoppema et al. (2000) estimate also at 0° meridian a seasonal silicic acid depletion, between the SACCF and SB, at  $4.5 \pm 0.7$  mol Si m<sup>-2</sup> yr<sup>-1</sup>. The slightly higher value from the latter study is probably due to the fact that this estimation has been done in the Southern AZ instead of the whole AZ.

Jin et al. (2006) estimated opal export out of the euphotic layer in the Southern Ocean in the range of 1 to  $9 \text{ mol Si m}^{-2} \text{ yr}^{-1}$ , which encompasses our estimate of net  $\text{bSiO}_2$  production. Integrating our estimation over the whole AZ area ( $14.10^6 \text{ km}^2$ , JGOFS synthesis report, 2001), net  $\text{bSiO}_2$  production is  $73 \pm 18 \text{ Tmol Si yr}^{-1}$  including  $22 \pm 5 \text{ Tmol Si yr}^{-1}$  originating from silicon supply during summer months. Considering a global net biogenic silica production of  $171 \pm 35 \text{ Tmol Si yr}^{-1}$  (mean from Nelson et al., 1995; Usbeck, 1999; Heinze et al., 2003; Jin et al., 2006), the AZ therefore contributes  $42 \pm 13$ % to this net production.

### 4.2.2 Northern ACC and STZ

The northern ACC is defined here as the area north of the PF and includes the STZ, although the latter is not normally considered part of ACC. The subsurface waters in the PFZ (halocline) fall on a mixing curve between the PF-AZ AASW and PFZ-PF ML (Fig. 6a). The subsurface layers in the PFZ could be a precursor of AAIW, spreading and deepening northwards through mesoscale frontal circulation (i.e. via meanders and eddies; Arhan et al., 2011). In this sector of the Southern Ocean, AAIW-type waters were observed north of the SAF (salinity minimum layer, ~500 to 1000 m), these are



**Fig. 6.** (a): mixing curve between AZ-PF WW and PFZ-PF ML (black line). (b): mixing curve between STZ-SAZ mixed layer and AAIW. The end-members are estimated taking the median values for the considered water masses.

mainly of an Indian Ocean origin (Reid, 2003). Nevertheless, the AAIW north of the SAF is not significantly different from the PFZ subsurface waters in terms of Si-properties (t-test, *p*value > 0.3, both for concentration and isotopic composition; Fig. 6a). This could reflect the occurrence of similar formation dynamics such as mixing of AASW with highly Sidepleted mixed layers to the north. Whereas the involvement of AASW in the formation of AAIW is well established, the exact mechanisms are still unclear (Molinelli, 1981; England et al., 1993; Sloyan and Rintoul, 2001; Morrow et al., 2004). However, Tomczak and Godfrey (2001) conclude that both circumpolar cross-frontal mixing and PFZ local deep convective mixing play a role, which is in agreement with results published here.

The fact that the central value of AAIW falls off the mixing curve between the AZ-PF AASW (=AZ-PF WW) and PFZ ML end-members (Fig. 5a), may reflect that the AZ-PF AASW is not representative of AASW flowing into the PFZ via cross frontal exchanges. However, when selecting strictly the AASW (=PF WW), with Si-properties at the PF as end-member conditions, AAIW does fall on the mixing curve with PFZ-ML. Therefore PF AASW is a better end-member representative of the cross frontal exchange process.

We estimate the AASW Si mass fraction contribution to AAIW to reach 0.97 (adapted Eq. 2). Such a high value highlights that almost all the Si in the AAIW originates from the AASW layer with only a very low residual Sicontribution from the summer PFZ mixed layers. Assuming steady state conditions once more, the annual AASW Sisupply into the AAIW must be balanced by the net bSiO2 production. Considering that the thickness of the AAIW and the ML layers are  $750\pm250$  m and  $80\pm20$  m (based on T-S water properties), respectively, we calculate the net  $bSiO_2$ production at  $1.1 \pm 0.4 \text{ mol Si m}^{-2} \text{ yr}^{-1}$ . The seasonal Sidepletion (from the difference between AAIW and PFZ ML Si(OH)<sub>4</sub> concentrations) is  $1.0 \pm 0.3$  mol Si m<sup>-2</sup> yr<sup>-1</sup>. This value is close to the net bSiO<sub>2</sub> production estimate implying the summer Si-supply to be small and poorly constrained  $(0.1 \pm 0.5 \text{ mol Si m}^{-2} \text{ yr}^{-1})$ . Based on the compilation of Siuptake and bSiO<sub>2</sub> dissolution rates from tracer incubation experiments, Fripiat et al. (2011b) noted that the mean net  $bSiO_2$  production in the PFZ was 1.2 mol Si m<sup>-2</sup> yr<sup>-1</sup>, in accordance with results published here. Integrating our estimation,  $1.1 \pm 0.4$  mol Si m<sup>-2</sup> yr<sup>-1</sup>, over the whole PFZ area  $(3.10^6 \text{ km}^2, \text{ JGOFS synthesis report, 2001})$ , the net bSiO<sub>2</sub> production is  $3 \pm 1$  Tmol Si m<sup>-2</sup> yr<sup>-1</sup> representing  $2 \pm 1$  % of the global net biogenic silica production (Nelson et al., 1995; Usbeck, 1999; Heinze et al., 2003; Jin et al., 2006). Thus diatom activity in the PFZ removes only a small fraction of the Si supplied to this area. The lower Si(OH)<sub>4</sub> concentration of AAIW mainly reflect the highly Si(OH)<sub>4</sub> depleted PF and PFZ surface water, imposing low Si(OH)<sub>4</sub> contents on the resulting mixing product (AAIW) without inducing changes in  $\delta^{30}$ Si (Figs. 5a, c, 6).

Analysing the nitrate isotopic distribution, Sigman et al. (1999) suggested that the main source for nitrate in the PFZ was AASW advected northwards by Ekman drift. They report the AAIW to be only slightly different from AASW in terms of NO<sub>3</sub> concentration and isotopic composition, which contrasts with our observations for Si, especially regarding Si(OH)<sub>4</sub> concentration, which is significantly lower in AAIW than in AASW. The large decoupling between the seasonal depletion of Si(OH)<sub>4</sub> and nitrate in the mixed layer could explain the difference between this study and Sigman et al. (1999) regarding the degree of AASW alteration during formation of AAIW. Nitrate concentration is still high at the PF and PFZ (30 to  $20 \,\mu\text{mol}\,l^{-1}$ ) compared to [Si(OH)<sub>4</sub>]  $(<5 \mu mol l^{-1})$ , while Si and N contents of PF WW are both  $\sim$ 35 µmol l<sup>-1</sup>. As a result, nitrate biogeochemical properties of AAIW appear closer to those for AASW, leading Sigman et al. (1999) to conclude that unaltered AASW is the main source for the AAIW nitrate pool.

The subsurface waters in SAZ and STZ are characterized by a thermocline between the mixed layer and the AAIW be-



**Fig. 7.**  $\delta^{30}$ Si vs. Si(OH)<sub>4</sub> for samples with Si < 60 µmol Si l<sup>-1</sup> in the ACC and the Pacific Ocean: ACC south of the PF are represented by black dots (Varela et al., 2004; Cardinal et al., 2005; Fripiat et al., 2011c; Cavagna et al., 2011; this study), ACC north of the PF by white dots (Cardinal et al., 2005; this study), the Equatorial Pacific by crossed white squares (Beucher et al., 2008), and the north Pacific by white squares (De La Rocha et al., 2000; Reynolds et al., 2006).

low (salinity minimum layer). Despite the fact that values fall slightly below the mixing curve (Figs. 5b and 6b), the Si characteristics clearly identify the thermocline in the SAZ and STZ as being a mixing interface between AAIW and the SAZ-STZ ML (Fig. 6b). The slight offset could be explained by the fact that the entire subtropical region, just north of our representative SAZ station (S2; Fig. 1a), is impacted by the Agulhas Current. This is the western boundary current flowing southward along the Southern Eastern Africa continental slope before retroflecting eastward and forming intense anticyclonic eddies (Agulhas rings), constituting the STF in the Indo-Atlantic sector and potentially crossing the STF into the SAZ region (Dencausse et al., 2011). These rings supply Indian central waters and AAIW of Indian origin into this region. Since the variety of the mixing end-members was not sampled here, it is not possible to estimate the net bSiO<sub>2</sub> production and the summertime Si-supply as has been possible for the PFZ and the AZ.

The spreading and deepening northward of the Southern Ocean water masses (AAIW and SubAntarctic Mode Water (SAMW), the latter water mass was measured during CLIVAR-SR3 although not BGH) is propagated towards the Equatorial Pacific, as clearly seen from their similarity with the Si isotopic composition of subsurface waters in the Equatorial Pacific (Beucher et al., 2008, Fig. 7). This indicates that AAIW-SAMW waters are the main Si-suppliers to the surface waters at lower latitudes, as has also been shown by Sarmiento et al. (2004).

### 4.2.3 Paleoceanographic implications

The fact that the mesopelagic layers ( $\sim 100-1000 \text{ m}$ ) with their Si content and isotopic characteristics are the source region for Si to most of the surface waters, has significant implications for the paleoceanographic interpretation of the  $\delta^{30}$ Si records preserved in the sediments. Different ACC mixing interfaces have been identified: AASW, AAIW, both characterizing the halocline and the thermocline water northwards. The residual silicic acid concentrations of the endmembers control the  $\delta^{30}$ Si alteration of the mixing products. Except for AASW, all the mixing interfaces have a highly Sidepleted mixed layer as end-member. As a result, the silicic acid AASW concentration is diluted across the different interfaces without significantly changing the AASW  $\delta^{30}$ Si signature (Figs. 5c, 7). Si-utilization in the northern ACC and at lower latitudes will thus yield a similar surface water  $\delta^{30}$ Si range as in the southern ACC, although with a lower silicic acid concentration (Figs. 5c, 7). The North Pacific Ocean has significantly lighter  $\delta^{30}$ Si values (Reynolds et al., 2006) and consequently, for similar levels of Si-utilization, the  $\delta^{30}$ Si of the resulting surface water will be lighter. From a palaeoceanographic perspective, the mesopelagic variation in  $\delta^{30}$ Si recorded in the Southern ACC (within and south of the PF) induces a complexity since glacial circulation and hydrology is expected to be different from today (Toggweiler et al., 2006; Watson and Naveira Garabato, 2006). It is probable that different oceanic circulations altered the Si- properties of water masses and subsequently the  $\delta^{30}$ Si of the source regions feeding surface waters.  $\delta^{30}$ Si in the subsurface varies between +1.3 to +2.0 ‰ (Fig. 3b, d), a range which is significant with regard to the isotopic difference observed for Southern Ocean sediments between glacial/interglacial periods ( $\sim$ +0.2 to +1.5 ‰ in the AZ, De La Rocha et al., 1998; Brzezinski et al., 2002, and  $\sim+1$  to +2 ‰ in the SAZ, Beucher et al., 2007). These dynamic effects must be assessed through, for example, a process-oriented modelling effort.

### 4.3 Inter-oceanic leakage of Si-isotopes

Along the pathway of the global ocean overturning circulation (OC), silicic acid concentrations increase in North Atlantic Deep Waters (NADW) during the processes of the OC towards the North Pacific and North Indian as the North Pacific Deep Water (NPDW) and the Indian Ocean Deep Water (IODW), respectively (Fig. 8a). This leads to a progressive accumulation of dissolved Si from bSiO<sub>2</sub> dissolution (Sarmiento et al., 2004, 2007), which is supported by the differing values of the  $\delta^{30}$ Si of silicic acid which decreases from ~+1.4 ‰ in NADW to ~+0.6 ‰ in NPDW (Fig. 8a; De La Rocha et al., 2000; Reynolds et al., 2006). Similar trends are also observed for  $\delta^{13}C_{DIC}$  due to the remineralization of organic matter (Kroopnick, 1985). However, the lack of a  $\delta^{30}$ Si gradient between the NADW and Circumpolar Deep

Waters (CDW, Fig. 8a) probably results from an almost complete utilisation of silicic acid in most of the Atlantic Ocean surface waters (Conkright et al., 2002). As a result, the integrated  $\delta^{30}$ Si of sinking biogenic silica converges to the one of the source, i.e. silicic acid supply, so that the  $\delta^{30}$ Si of the bSiO<sub>2</sub> rain-out from the surface is similar to that of silicic acid in subsurface waters (initially mostly AAIW), ultimately sinking in the North Atlantic to form NADW (Sarmiento et al., 2004; Reynolds, 2009). The subsequent Si-isotopic lightening from the CDW to the NPDW (Fig. 8a) can be attributed to the dissolution of settling diatoms in areas where silicic acid is only partly used (Pondaven et al., 2000b; Brzezinski et al., 2001). Consequently values integrate  $\delta^{30}$ Si signatures significantly lighter than the source, as is the case for the Southern Ocean and North Pacific. In contrast with Reynolds (2009) who, based on modelling approaches, reports that the Si isotopic gradient is centered in the Southern Ocean deep waters, the measured isotopic gradient in deep waters ( $\sim 1.0$  ‰) is in fact sharper along the deep Pacific basins, while the isotopic shift observed along the ACC pathway shows only a slight, though significant (t-test, pvalue < 0.001) <sup>30</sup>Si depletion of  $\sim 0.2$  ‰ (Fig. 8b).

Toggweiler et al. (2006) partitioned the global ocean's overturning circulation in two cells, the northern and southern cells which occupy distinct domains in the ocean's interior as detailed in the following. In the modern Ocean these cells mix together to form CDW, which comes up to the surface along the southern flank of the ACC. The middepth northern cell is the domain of North Atlantic Deep Water (NADW), presenting lower Si(OH)<sub>4</sub> concentrations and heavier  $\delta^{30}$ Si values (Fig. 8a). The lower ventilated southern cell is the domain of the deep, bottom waters (AABW, IODW, and NPDW), characterised by higher Si(OH)<sub>4</sub> concentrations and lighter  $\delta^{30}$ Si values (Fig. 8a). This partioning could partly explain the isotopic gradient in the CDW via isopycnal mixing from the Atlantic sector, where NADW enters the Southern Ocean, compared to the Australian sector, where waters are already potentially affected by the mixing with the IODW (Reid, 2003), probably bearing a lighter Siisotopic composition as for NPDW. Moreover, the partitioning of the global ocean's overturning circulation in two cells, with different ventilation rates (Toggweiler et al., 2006), could act as a positive feedback for the isotopic expression of biogenic silica dissolution effects in the Pacific Ocean as light silicic acid from bSiO<sub>2</sub> dissolution is trapped in the Pacific Ocean, presenting lower ventilation (Sarmiento et al., 2007).

An unexpected feature is the much heavier isotopic composition in the eastern Equatorial Pacific and in the Cascadia basin (Beucher et al., 2008), compared to the western northern Pacific (Fig. 8a; De La Rocha et al., 2000; Reynolds, 2009). It is evident that more data from the deep Pacific and Indian Oceans are required in order to resolve the deep Siisotope distribution pattern and to identify the mechanisms driving it.



**Fig. 8.** (a): deep water distribution in the different oceanic basins: NADW (De La Rocha et al., 2000), CDW (Cardinal et al., 2005; this study), AABW (Cardinal et al., 2005; this study), NPDW (De La Rocha et al., 2000; Reynolds et al., 2006; Beucher et al., 2008). (b): CDW distribution for the Atlantic and Australian sectors, respectively this study and Cardinal et al. (2005). (c): AZ-PF WW = AASW distribution for the Atlantic and Australian sectors, respectively this study and Cardinal et al. (2005). (d): SAZ-STZ thermoclines, AAIW, and SAMW for the Atlantic and Australian sectors, respectively this study and Cardinal et al. (2005).

The isotopic gradient in deep waters, along the global overturning circulation, also seems to be expressed in the  $\delta^{30}$ Si values in the ACC subsurface waters between the Atlantic and the Australian sectors. Indeed, WW which is representative of winter AASW (Sect. 4.3) also displays a Siisotopic composition getting lighter from the Atlantic to the Australian sector (Fig. 8c; regression lines with the same slope (Mann-Whitney Rank Sum Test, p-value = 0.331) but significantly different y-intercept (Mann-Whitney Rank Sum Test, p-value < 0.001) as observed for the AAIW, the SAMW, and the SAZ-STZ thermoclines (Fig. 8d; t-test, pvalue = 0.011). If these trends are confirmed by a larger data set, they are to be taken into account when interpreting sedimentary archives and require the use of an appropriate correction factor. There is also need to further verify whether the observed trend in isotopic compositions can be confirmed along the ACC pathway in the Pacific.

### 5 Conclusions

This manuscript provides the first summer isotopic compositions of silicic acid ( $\delta^{30}$ Si) across the whole ACC and adjacent subsystems, the SubTropical Zone and the Weddell Gyre, for the full-depth water column in the Atlantic sector.

In a similar way as reported in Fripiat et al. (2011a), we used the Si-isotopic constraint to assess (1) net  $bSiO_2$  production and (2) summertime Si-supply into the mixed layer: estimated at (1)  $5.2\pm1.3$  and  $1.6\pm0.4$  mol Si m<sup>-2</sup> and (2)  $1.1\pm0.3$  and  $0.1\pm0.5$  mol Si m<sup>-2</sup>, respectively for the mean Antarctic Zone and Polar Front Zone. These results are within the range of other published data. Integrating our estimation over the whole AZ and PFZ area (JGOFS synthesis report, 2001), the net  $bSiO_2$  production contributes  $42\pm13$  and  $2\pm1\%$  respectively, to the global net biogenic silica production of  $171\pm35$  Tmol Si yr<sup>-1</sup> (Nelson et al., 1995; Usbeck, 1999; Heinze et al., 2003; Jin et al., 2006).

 $\delta^{30}$ Si in the subsurface waters is mainly controlled by mixing processes, transmitting the isotopic imprint resulting from surface water Si-uptake down to depth. Different mixing interfaces have been determined: the Antarctic Surface Water, the Antarctic Intermediate Water, and the low latitude thermoclines. The level of Si-depletion is an important factor controlling the alteration of  $\delta^{30}$ Si in the mixing interface. Indeed, a highly Si-depleted water mass dilutes the silicic acid concentration of the mixing product, without significantly affecting  $\delta^{30}$ Si. Except for the AASW, which is the ultimate source for the subsurface waters in the ACC, each mixing interface has a highly depleted mixed layer as an end-member. Overall the AASW silicic acid concentration strongly decreases northward across the different mixing interfaces, keeping the initial AASW  $\delta^{30}$ Si signature (Cardinal et al., 2005; Beucher et al., 2008; this study). Consequently, almost all surface waters in the modern ocean within and north of the PF are characterised by similar  $\delta^{30}$ Si signatures, though silicic acid concentrations differ. Such observations contrast significantly with the model outputs from Wischmeyer et al. (2003) or Reynolds (2009), and point out the absence of knowledge of the processes involved in modern Si-isotopic distributions.

Ocean hydrology and circulation is thought to be different along glacial/interglacial timescales. Modification of the ocean's circulation could therefore significantly affect the  $\delta^{30}$ Si of subsurface waters. Through a process-oriented modelling effort, this mixing induced variability in  $\delta^{30}$ Si has to be better constrained for the proper use of this proxy in paleoceanography. Furthermore, as has been highlighted here, a progressive decrease in  $\delta^{30}$ Si values is observed along the eastward ACC transport of water masses between the Atlantic (this study) and the Australian sector (Cardinal et al., 2005). This probably results from: (1) the progressive accumulation of Si from dissolving biogenic silica along the eastward ACC transport and (2) isopycnal mixing between deep water masses in each of the oceanic basins. Should this ACC trend be further confirmed by new Pacific ACC transects, this isotopic effect would then need to be corrected when the isotopic composition of sedimentary biogenic silica is interpreted.

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