Vertical export flux of metals in the Mediterranean Sea

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Abstract:

We examined metal (Al, V, Cr, Mn, Fe, Ni, Cu, Zn, Cd and Pb) and particulate organic carbon (OC) concentrations of the marine vertical export flux at the DYFAMED time-series station in the Northwestern Mediterranean Sea. We present here the first data set of natural and anthropogenic metals from sediment trap moorings deployed at 1000 m-depth between 2003 and 2007 at the DYFAMED site. A highly significant correlation was observed between most metal concentrations, whatever the nature and emission source of the metal. Cu, Zn and Cd exhibit different behaviors, presumably due to their very high solubility and complexation with organic ligands. The observed difference of atmospheric and marine fluxes in terms of temporal variability and elemental concentration suggests that dense water convection and primary production and not atmospheric deposition control the marine vertical export flux. This argument is strengthened by the fact that significant changes in metal concentrations of trapped particles.

Highlights

► Dense water convection and surface primary production control the vertical export flux of metals ► Atmospheric deposition events do not trigger significant vertical export fluxes ► all metals whether of natural or crustal origin are accumulated in the mixed layer until they are exported at the same time

1. Introduction

Identifying the factors controlling the vertical export flux of particulate matter from the surface ocean to the deep sea is of paramount importance to understand the mechanisms leading to the sequestration of carbon (Jickells et al., 1998). The question of whether the incorporation of suspended minerals drives the vertical export flux of particulate organic carbon (OC, used hereinafter for the particulate fraction only) in the ocean is still under debate (Armstrong et al., 2002 and Passow, 2004). The Mediterranean Sea is an ideal site to address this question. Strong physical forcing, intense coastalpelagic interactions, short water residence times and an equally strong influence of natural and anthropogenic continental sources on the marine biogeochemical cycles of metals make the Mediterranean Sea particularly sensitive to environmental and climatic changes (Martin and Milliman, 1997, Krahmann and Schott, 1998, Béthoux and Gentili, 1999 and Duarte et al., 1999). This particular sensitivity justifies that the Mediterranean Sea is a privileged ecosystem for the investigation of marine responses to anthropogenic metal inputs and warming climate (Durrieu de Madron et al., 2011). The atmosphere of the Northwestern Mediterranean Sea is characterized by a European signature disrupted by episodic Saharan dust events (Chester et al., 1997 and Heimbürger et al., 2010b). Atmospheric metal inputs to the Northwestern Mediterranean Sea originate from natural and anthropogenic emissions sources. The major source of natural metals in this region is the Saharan desert. Episodic but intense pulses characterize this particular source (Chester et al., 1997, Guerzoni et al., 1999, Marty et al., 2002, Heimbürger et al., 2010b and Ternon et al., 2010). Anthropogenic metals are mainly carried with air masses from Northern and Central Europe. As a result, metal concentrations in Mediterranean surface waters are higher than in the open ocean,

67 and those of the inflowing North Atlantic Ocean (Morley et al., 1997). Metal 68 distribution patterns in the water column suggest that their biogeochemical cycling is 69 mainly governed by atmospheric inputs (Béthoux et al., 1990; Migon et al., 2002; 70 Heimbürger et al., 2011). Three independent studies (Martín et al., 2009; Angelidis et 71 al., 2011; Heimbürger et al., 2012) show a recent increase of anthropogenic metals 72 in Mediterranean deep marine sediment records. This increasing metal trend in 73 sedimentary records presumably reflects the evolution of anthropogenic metal 74 emissions along the densely populated Mediterranean coast (~300 inhabitants per 75 km² (UNEP/MEDPOL, 2004; Laubier, 2005), in addition to metal inputs from long-76 ranged sources.

However, the problem remains of knowing which parameter controls the temporal variability of the vertical export flux of metal to marine sediments. Previous studies (Fowler et al., 1987; Migon et al., 2002) have shown that biological and biogeochemical processes occurring at the surface control the temporal variability of vertical OC export fluxes in the Northwestern Mediterranean Sea. Miquel et al. (1994; 2011) pointed out that the vertical mixing of the water column is a key factor determining the magnitude of the vertical OC export flux.

84 The ballasting theory pioneered by (Armstrong et al. (2002); Armstrong et al. (2009)) 85 suggests another mechanism, in which mineral material (atmospheric dust, biogenic 86 silica, and carbonate shells) determines the occurrence of vertical OC fluxes. 87 However, Passow (2004) proposed that, despite their conspicuous ballasting role, 88 mineral particles may not cause vertical OC fluxes. On the contrary, vertical OC 89 fluxes would determine the vertical export of mineral material. With the present paper 90 we aim to contribute to this debate by examining natural and anthropogenic metal 91 concentrations of the vertical export flux captured by a sediment trap moored at 92 1000m-depth at the DYFAMED site, Northwestern Mediterranean Sea.

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94 **2. Methods**

95 **2.1. Study site**

96 The DYFAMED (DYnamique des Flux Atmosphériques en Méditerranée) time-series 97 station (2350m-depth, 43°25'N, 7°52'E; Fig. 1) is a long-term monitoring station in 98 the open Ligurian Sea (Northwestern Mediterranean Sea) located 28 nautical miles 99 off continental France. The Ligurian Sea circulation is characterized by a permanent 100 cyclonic gyre (Lévy et al., 1998). The Ligurian Current creates a band ~ 30 km wide

101 and > 250 m deep, which is believed to separate the DYFAMED site from coastal 102 lateral inputs by a strong horizontal density gradient (Niewiadomska et al., 2008). 103 Therefore, atmospheric metal inputs are believed to be by far the most significant 104 source to the open Ligurian Sea(Migon et al., 2002). The DYFAMED site has been 105 used several times for the study of interactions between atmospheric deposition and 106 open surface waters (e.g. DYFAMED and MEDFLUX programs; see special issues 107 Deep-Sea Research II 49, 11 (2002) and 56, 18 (2009), respectively). DYFAMED is 108 now viewed as a reference site for monitoring of ongoing changes in the 109 Northwestern Mediterranean Sea.

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111 **2.2. Sampling**

112 Automated time-series sediment traps were moored at 1000-m depth at the 113 DYFAMED site between March 2003 and February 2007. Conical sediment traps (Technicap PPS-5, height 2.3m, collection area 1m²) were equipped with a 114 115 programmable 24-cup collector. The sampling cups contained a solution of 2% 116 buffered formaldehyde in filtered seawater (0.22µm) to prevent in situ microbial 117 degradation and grazing by swimmers. After recovery, the samples were stored in 118 the dark at 4°C. Swimmers were removed by successive sieving through 1500 and 119 600µm and followed by hand-picking under the binocular microscope. The remaining 120 samples were then desalted by rinsing on a Nuclepore filter (porosity 1µm) with buffered (pH 7) deionized Milli-Q[®] Millipore water (resistivity: 18MΩ.cm) three times 121 122 (JGOFS, 1996) and freeze-dried prior to analysis (Miguel et al., 1994; Miguel et al., 123 2011). Samples were weighed using a high precision balance (Sartorius).

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125 **2.3. Analysis**

126 Sediment trap samples were mineralized according to the following protocol: The 127 organic matrix was destroyed by oxidation (HNO₃) while the mineral aluminosilicate 128 matrices were destroyed with HF. Every freeze-dried sample was weighed in 7mL 129 Teflon flasks, and dissolved as follows: 1) each flask was filled with 1mL HNO₃ 65% 130 (suprapur, Merck), sealed and placed in a larger Teflon bottle (60mL). This 131 apparatus was left 6 hours in an oven at 150°C, after which bottles and flasks were 132 brought to room temperature and left open under laminar flow hood until a brown dry 133 residue remained. 2) 500 μ L HNO₃ 65% and 500 μ L HF 40% were added to the

remainder. The flasks and bottles were closed and put in an oven 6 hours at 150°C, prior to open evaporation at room temperature under laminar flow hood, until a white dry residue was obtained. This residue was ultrasonically dissolved in 5mL HNO₃ 0.1N and then diluted with HNO₃ 0.1N to 15mL.

138 Digested and dissolved metals were analyzed by ICP-OES, using a Perkin-Elmer 139 Optima 3000, axial torch instrument. An ultrasonic nebulizer (CETAC) was used for 140 sample introduction to improve the sensitivity (Desboeufs et al., 2003) to 0.01 to 0.1 141 ppb levels depending on the metal. A plasma power of 1250W and a sample flow of 142 1 mL.min⁻¹ were used. An external calibration was performed with multi-elemental standard solutions made by mixing 1g.L⁻¹ mono-elemental solutions provided by 143 144 Merck (Darmstadt, Germany). The accuracy was checked using dilutions of multi-145 elemental commercial solutions and SLRS-4 as certified reference material (CRM). 146 We also monitored the analysis with CRM geo-standards: GBW (National Research 147 Center for Certified Reference Materials, China), BCSS-1 and PACS (National 148 Research Council, Ottawa, Canada). A summary of analytical performances is given 149 in Table 1.

150 Metal concentrations (Al, V, Cr, Mn, Fe, Ni, Cu, Zn, Cd and Pb) were analyzed for 151 samples collected from March 2003 to February 2007. However, years 2005 and 152 2006 were affected by higher currents than usual. Currents >12cm.s⁻¹ are more likely 153 to bias quantitative collection of settling particles by sediment traps (Baker et al., 154 1988; Scholten et al., 2001; Buesseler et al., 2007). This situation occurred from 22 155 February to 23 June 2005, and during most of the year 2006. Vertical export fluxes 156 during those periods might be underestimated (Miguel et al., 2011) and the quality of 157 the trapped particles in terms of OC content and metal concentration has to be 158 observed critically.

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160 **2.4. Data treatment and statistical analysis**

To our knowledge no certified reference material for sediment trap material exists today and complete procedural blanks are hard if not impossible to realize. We carefully examined the raw data in order to investigate for possible artifacts and contamination. To do so we used enrichment factors (EFs) to trace the anthropogenic component of samples, either from anthropogenic inputs or contamination during sampling. We chose EFs standardized to Al, which is the most

167 commonly used soil dust reference. These ratios are usually defined for a given
168 element as (e.g. (Herut et al., 2001; Heimbürger et al., 2010b):

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 $EF = (M_{sample} / AI_{sample}) / (M_{background} / AI_{background})$ (1)

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172 where $(M_{sample} / AI_{sample})$ is the metal concentration of the sample relative to the AI 173 concentration. This ratio is normalized to the ratio (Mbackground / Albackground) of Earth's 174 continental crust (Wedepohl, 1995). Average metal concentrations in soil dust or 175 rocks are very difficult to define accurately, due to the inhomogeneous composition 176 of worldwide soils. Moreover, the (M_{background} / Al_{background}) ratio may be dependent on 177 grain size fractionation and chemical alteration during atmospheric transport, 178 biogeochemical cycling in the euphotic zone and settling of particles through the 179 water column. Such modifications might involve the solubilization of certain metals 180 from particles, photo-chemically induced redox reactions and aggregation processes 181 (Desboeufs et al., 2001). The use of EFs is permissible if they are used for the 182 comparison of different enrichments standardized to the same Albackground value 183 (Herut et al., 2001; Heimbürger et al., 2010b). We calculated EFs for each metal to 184 distinguish anthropogenic influences from natural ones. Suspiciously high EFs 185 occurred occasionally for Cu, Zn, Cd and Pb, always and only for the first samples of 186 a separate sediment trap deployment. Those metals are particularly prone to 187 contamination, thus handling, maintenance and setting of the sediment trap mooring 188 lines and their deployment might have corrupted them, although the collection of 189 settling particles always started at least 24 hours after completion of the deployment 190 of the mooring. We removed all of those occasional outliers from the data set.

Principal component analysis (PCA) was applied to extract the geochemical signatures of the chemical composition of the vertical export flux and to get insights into the underlying factors accounting for them. The analysis was performed by means of the Z-scores transformation of the raw data:

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196 Z-score=
$$(X_i - X_{avg})/X_{std}$$
 (2)

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where X_i is a given value of a variable in a sample, X_{avg} is the average of that variable and X_{std} is its standard deviation. Z-scores allow keeping the relative variation of the original data while reducing all variables to a similar range of

variation avoiding scaling effects. The best solution was obtained with a Varimax rotation, which is more restrictive with the variables associated to the principal components (i.e. maximizes the proportion of variance of the variables in the principal components). The square of the factor loadings was used as an estimation of the proportion of variance of each variable for each principal component. Statistical computations were performed with XLSTAT® software from Addinsoft.

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3. Results and Discussion

Temporal variability of vertical export flux, OC and metal concentrations are compared at 1000m-depth (Fig.2.). Mineralization processes and grazing are important features in the upper water column, whereas, at 1000m-depth, vertical export fluxes are considered to be net fluxes with minimal alteration (Martin et al., 1987; Guidi et al., 2009).

Mean metal concentrations and vertical metal export fluxes at the DYFAMED site 214 215 and various other moorings in the Mediterranean and Black Sea are shown in 216 Table 2. The variability of metal and OC concentrations is relatively low (RSD = 21-217 46 %) compared to the variability of vertical export flux (RSD = 147 %). This implies 218 that the temporal variability of TM and OC fluxes (calculated as the product of TM, or 219 OC, and vertical export flux) is almost entirely governed by the variability of the 220 vertical export flux. This is also the case for other Mediterranean sediment trap 221 moorings, suggesting that this is a general feature. Vertical export fluxes, OC and 222 metal concentrations are roughly in the same ranges as values presented by Migon 223 (2002) and Heimbürger et al. (2012), and also comparable to finding in other basins 224 of the Mediterranean Sea (Theodosi et al., 2010; Roussiez et al., 2012; Theodosi et 225 al., 2012) and the Black Sea (Theodosi et al., 2013). Miguel et al. (2011) suggested 226 that under a one-dimensional scenario, ~38% of OC leaving the 200m-depth horizon 227 is remineralized before reaching the 1000m-depth horizon. OC concentrations during 228 the selected sampling period varied between 2 and 19% (Fig. 2), a range typically 229 observed during the 20-year DYFAMED time-series (Miguel et al., 2011).

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3.1. Seasonal variability of the marine vertical export flux

232 The vertical export flux at DYFAMED follows a well-known seasonal pattern (Miquel

233 et al., 1994; Migon et al., 2002; Ternon et al., 2010; Miquel et al., 2011; Heimbürger

et al., 2013). This pattern was repeated over the duration of the sampling period andcharacterized by the following known sequence:

236 In winter (December to February), the cooling and evaporation of surface waters 237 lead to the formation of dense water. The convection of dense water leads to rapid 238 downward transport of dissolved (Copin-Montégut and Avril, 1993; Avril, 2002) and 239 particulate matter (flush-down effect), including metals (Béthoux and Gentili, 1999; 240 Heimbürger et al., 2013). The vertical export flux during dense water convection 241 contains a higher portion of atmospherically-deposited mineral material, which has 242 been accumulated in the surface layer during the preceding stratified period (Migon 243 et al., 2002; Heimbürger et al., 2010a). As a result, the OC concentration is the 244 lowest. The seasonal pattern of the vertical export flux shows reproducible peaks in 245 January-February, e.g. 2004 and 2005 (Fig. 2). During the mesotrophic period 246 (March to May), the vertical export flux is driven by primary production. Nutrients 247 brought to surface waters by the previous dense water convection trigger 248 phytoplanktonic blooms in spring, and generate moderate vertical export fluxes, e.g. 249 2003 and 2004 (Fig.2). The spring vertical export flux is characterized by moderate 250 OC concentrations, as a result of the combination of biogenic material and mineral 251 material. The intensity of the vertical export flux under mesotrophic conditions 252 directly depends on the intensity of the spring bloom conditioned by the dense water 253 convection in winter (Marty and Chiavérini, 2010).

254 Under oligotrophic stratified conditions (June to November, approximately), the vertical export flux is minimal (~40 mg.m⁻².d⁻¹) at the DYFAMED site (Sarthou and 255 256 Jeandel, 2001; Migon et al., 2002; Sternberg et al., 2007). During this period, there is 257 only little to no production of fecal pellets, which are known for their capacity to 258 transfer matter and elements to depth (e.g. (Fowler and Knauer, 1986; Marty et al., 259 1994; Carroll et al., 1998)). Vertical export flux during this period results almost 260 entirely from regenerated production (Marty et al., 2002). A higher percentage of 261 small-sized phytoplankton is exported out of the euphotic zone because grazing is 262 low (Guidi et al., 2009). The lower vertical export flux recorded in oligotrophic 263 conditions exhibits the highest OC concentrations (Fig. 2).

The described seasonal pattern of the vertical export flux may be interrupted by occasional bursts, caused by meanders passing the Ligurian current, fall blooms, and intense meteorological conditions. Two examples are given here:

1.) Slightly higher vertical export fluxes were observed in July 2005. Episodes of cold
northwesterly winds (Mistral) observed during this period (data from Météo-France)
have likely mixed the surface layer and advected nutrient-rich waters from below
(Andersen and Prieur, 2000; Marty et al., 2008).

271 2.) Small increases of the vertical export flux were recorded between October and 272 December. For example, a peak was observed in fall 2005 (Fig. 2), while vertical 273 export fluxes were negligible in 2003 and 2004 during the same period. Strong 274 easterly winds were observed early October 2005 and yielded a decrease of sea 275 surface temperature of 3.6°C (from 20.9 to 17.3°C) within 5 days at the Météo-276 France buoy ODAS located nearby the mooring line. Such a rapid temperature 277 decrease suggests wave-induced mixing below the surface mixed layer. The mixing 278 with nutrient-rich waters from below the thermocline may have supplied nutrients to 279 the depleted surface layer and triggered a fall bloom. This fall bloom was probably 280 dominated by nano- and picophytoplankton (Heimbürger et al., 2010a) and resulted 281 consequently only in a small increase of the vertical export flux (Marty et al., 2009). 282 This seasonal pattern suggests that, apart from dense water convection episodes 283 associated in spatially restricted Mediterranean areas (among which is the Ligurian 284 Sea), the vertical export of atmospherically-deposited material is almost entirely 285 driven by the magnitude and variability of primary production. This statement is in 286 agreement with many studies that pointed out the prominent role of primary 287 production in the removal of mineral material (e.g. (Fowler et al., 1987; Buat-Menard 288 et al., 1989; Jickells et al., 1998; Grotti et al., 2001; Hamm, 2002; Migon et al., 2002). 289 Marty and Chiavérini (2010) stated that the efficiency of the vertical transfer under 290 mesotrophic conditions directly depends on the intensity of the water column mixing 291 and the subsequent diatom bloom. However, several recent papers (e.g. (Armstrong 292 et al., 2002; Francois et al., 2002; Klaas and Archer, 2002; Armstrong et al., 2009; 293 Lee et al., 2009; Ternon et al., 2010) have postulated that mineral particles, 294 originating mainly from atmospheric dust deposition and biogenic minerals 295 (carbonates and silica) control the sedimentation rate of biogenic material by 296 ballasting, particularly when zooplankton fecal pellet production is low (Lee et al., 297 2009). This process presumably increases the density of OC aggregates and, 298 therefore, their sinking rates. It was also suggested that mineral particles protect OC 299 from oxidation and remineralization (De La Rocha and Passow, 2007).

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301 3.2. Metal concentrations of the marine vertical export flux

302 The present paper contributes to the ongoing debate of whether atmospheric 303 deposition or primary production/dense water convection controls the marine vertical 304 export flux. We examined sediment trap samples from the DYFAMED site, which 305 have been analyzed previously for OC and AI (Ternon et al., 2010), but not for the 306 selection of anthropogenic and natural metals here presented here. The authors 307 concluded that atmospheric deposition events drive the marine vertical export flux. 308 The outstanding feature of the present dataset is that nearly all metals, whether of 309 crustal or anthropogenic origin, are highly correlated (Table 3). This suggests that all 310 metals are exported to depth at the same time, independently of their different 311 depositional seasonal pattern. However, this does not hold true for Cd, Zn and Cu, 312 which suggests that these metals are less efficiently associated with sinking 313 particles. Metal solubility can be proposed as an explanation: Despite possible 314 adsorption onto lithogenic material such as metal oxide surfaces at seawater pH (Fu 315 and Allen, 1992), Cd is among the most soluble metals studied here (Migon, 2005). 316 This also applies to Zn (Kersten et al., 1991). The ability of Cu to complex with 317 dissolved organic ligands (Muller, 1996) might also lead to lower contribution to the 318 pool of sinking particles. The solubility of those three metals might explain losses in 319 sediment traps.

An alternative explanation could be related to the processing of sediment trap samples. Rinsing the particles by deionized water may have led to osmotic bursting of planktonic cells, thereby releasing metals within the cytoplasm of algal cells. Zinc, Cu, and Cd reside in phytoplankton cytoplasm to a significantly greater extent than most of the other metals that were assessed (these other metals are typically bound to cell walls and membranes; (Sunda, 2012)). The discrepancy between these two groups of metals can likely be explained by this artifact.

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328 3.3. Atmospheric metal inputs from natural and anthropogenic sources

Pb concentrations are significantly correlated with those of Al, V, Cr, Mn, Fe, and Ni. This is presumably due to its strong affinity for suspended particulate matter, principally Al, Fe and Mn oxides, organic matter, carbonates and clay (Kersten et al., 1991). Among the highly correlated metals (Al, V, Cr, Mn, Fe, Ni, and Pb), one can find crustal (Al, Fe), intermediate (Cr, Mn), and anthropogenic elements (V, Ni, and Pb). Natural and anthropogenic metals depict distinct seasonal deposition patterns

335 related to their different emission sources (Heimbürger et al., 2010b). For example, 336 the deposition of crustal metals is strongly associated with Saharan dust episodes 337 (Bonnet and Guieu, 2006; Heimbürger et al., 2010b; Ternon et al., 2010). Saharan 338 dust events are generally observed in spring and summer in the western Mediterranean (Moulin et al., 1997; Heimbürger et al., 2010b). Atmospheric 339 340 deposition of anthropogenic metals does not exhibit this pulsed character, and most 341 of the anthropogenic inputs occur in winter, when polluted air masses from Northern 342 Europe influence the Ligurian Sea (Barnaba and Gobbi, 2004; Duncan and Bey, 343 2004; Heimbürger et al., 2010b). Anthropogenic metals have various emission 344 sources that are temporally variable. This may also lead to different seasonal 345 patterns of their atmospheric deposition. For example, V and Ni emissions result 346 mainly from oil combustion, whereas major Pb emissions results from steel 347 metallurgy plants, mining complexes in Eastern Europe, leaded petrol carried by 348 long-range atmospheric transport from North Africa, Middle East or Eastern Europe 349 (Bollhöfer and Rosman, 2001; Migon et al., 2008). Individual metal deposition to the 350 Northwestern Mediterranean Sea surface is temporally variable, according to the 351 nature and the individual seasonal patterns of their emission sources (Heimbürger et 352 al., 2010b).

353 Different seasonal variations in the supply of atmospheric deposition to surface 354 waters and the rate of particle removal from surface waters result in rather constant 355 metal concentrations of the vertical export flux through the year. Similar observations 356 were made for dust deposition at the BATS time-series (Jickells et al., 1998). 357 Compared to the BATS time-series site (Sargasso Sea), the DYFAMED time-series 358 site (Northwestern Mediterranean Sea) receives much higher loads of atmospheric 359 metal deposition and the water column is mixed every winter by dense water 360 convection. Vertical export flux in the Mediterranean Sea is driven by dense water 361 convection in winter (flush-down effect) and by primary production in spring (Miguel 362 et al., 1994; Migon et al., 2002; Miguel et al., 2011).

We hypothesized that atmospheric metal deposition accumulates in the marine surface layer in the absence of dense water convection and low biological activity (under stratified oligotrophic conditions). Indeed, individual dust particles have, based upon Stokesian calculations (Stokes, 1901), a negligible settling velocity (<5 m.d⁻¹; (Buat-Menard et al., 1989)). Therefore, the transfer of particulate metals from the sea surface to 1000m-depth without the driving force of hydrology or biology

369 would require at least 200 days. This is consistent with particulate metal residence 370 times calculated at the BATS time-series site (Jickells et al., 1984). This time being 371 longer than the oligotrophic period (on average, ~ 5 months in the Northwestern 372 Mediterranean Sea), atmospheric deposition (dissolved matter and particles that do 373 not dissolve) cannot be removed from surface waters without packaging onto large 374 organic particles or aggregates, incorporation into fecal pellets (Fowler and Knauer, 375 1986; Wang and Fisher, 1998), or adsorption onto planktonic debris and fecal pellets 376 during their sinking (Fisher et al., 1991).

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378 3.4. Saharan dust inputs

In spite of Stokesian considerations, we investigated if field observations could 379 380 suggest that atmospheric deposition of mineral loads is capable of causing vertical 381 export fluxes, either by direct sinking of mineral particles, or indirectly by atmospheric 382 fertilization of surface waters. Ternon et al. (2010) concluded that atmospheric 383 deposition events drive the marine vertical export flux at the DYFAMED time-series site. Their interpretation is based on the fact that "...for this 4-years time-series, high 384 385 OC fluxes were related to high marine lithogenic fluxes (their Fig. 8), forming high 386 export events...". Both variables, OC flux and the marine lithogenic flux, are 387 calculated as the product of the marine vertical export flux and the concentration of 388 OC and AI, respectively. That means that both flux variables contain a similar 389 variable, and are therefore not independent. If we examine their atmospheric data 390 closely we can identify 4 Saharan dust events (atmospheric flux of >1000mg.m⁻².d⁻¹). 391 Only one single Saharan dust event actually resulted in significant marine vertical 392 export flux, taking the same threshold value. This exceptionally strong input of 393 mineral dust observed over the Ligurian Sea occurred in February 2004, when 22,210 mg.m⁻² were deposited during a single event (Bonnet and Guieu, 2006). 394 395 Marine vertical export fluxes increased during this period (Ternon et al., 2010). This 396 Saharan dust event occurred during the dense water convection period and the 397 marine vertical export fluxes remained elevated for the entire dense water 398 convection period (Heimbürger et al., 2013). It is thus difficult to clearly determine 399 what was the driving force (hydrology or atmospheric deposition) for this vertical 400 export flux event. None of the 3 other Saharan dust events that occurred during our 401 sampling period (Ternon et al., 2010) actually resulted in significant marine vertical 402 export flux. For instance, a significant atmospheric dust episode that occurred during

403 oligotrophic stratified conditions in summer 2006 over the Ligurian Sea (Heimbürger
404 et al., 2010b; Ternon et al., 2010) did not yield increased marine vertical export
405 fluxes.

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407 3.5. Metal enrichment factor assessment

408 To further investigate the relationship of atmospherically-deposited particles and the 409 marine vertical export flux we chose to study their chemical composition using EFs. 410 Anthropogenic sources can be distinguished from natural ones using EFs. We 411 calculated EFs for each metal relative to the AI concentration (M_{sample} / AI_{sample}) and 412 normalized to the ratio (M_{background} / Al_{background}) of Earth's continental crust 413 (Wedepohl, 1995). Summary statistics of EFs and EF of Mediterranean aerosols are 414 given in Table 4. EFs of all metal of the marine vertical export flux (1.23 to 8.93) are 415 much lower compared to those of Mediterranean aerosols (2.59-597) (Heimbürger et 416 al., 2010b). This, and the fact that vertical export flux metal EFs show only very low 417 temporal variability, suggest again a common transport mechanism for all studied 418 metals and that intense deposition events are smoothed out. The consistently lower 419 metal EFs of the marine vertical export flux compared to those of Mediterranean 420 aerosols may also indicate that a part of the atmospheric deposition is solubilized 421 and may enrich dissolved metal concentration of the Mediterranean surface waters 422 (Heimbürger et al., 2011).

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424 3.6. Principal components analysis

425 We applied principal component analysis after z-score transformation and Varimax 426 rotation to extract the geochemical signatures of the chemical composition of the 427 vertical export flux and to get insights into the underlying factors accounting for them. 428 The square of the factor loadings was used as an estimation of the proportion of 429 variance of each variable for each principal component (Table 5). Factor 1 explains 430 58 % of the variation with the main contribution of OC and all metals except Cu, Zn 431 and Cd. This means that over half of the variability of the entire data set can be 432 described by this factor, which composed by the variables Al, V, Cr, Mn, Fe, Ni, and 433 Pb. We suggest that the concomitant vertical export of those metals is the process 434 that gives the vertical export flux this distinct geochemical signature. The negative 435 OC value explains the relative dilution effect of the OC contribution to metal 436 concentrations of the vertical export flux. Factor 2 explains 21 % of the variation and

is principally determined by OC, Cu and Zn. This confirms once again that all metals
(except Cu, Zn and Cd, see discussion above) are accumulated in the surface
waters, homogenized and exported to depth at the same time.

440 Aggregation and coagulation processes, combined with sedimentation, strongly 441 impact the amount of mineral matter that is packaged with biogenic material. They 442 also impact the sinking velocity of exported particles and, therefore, the fate and 443 biogeochemical cycling of inorganic material (including metals) in the water column 444 (Armstrong et al., 2002; Burd and Jackson, 2009). Those processes presumably 445 occurred at the DYFAMED station only when biogenic material was present in 446 sufficient concentration. However, our results suggest that dense water convection is 447 the major driver of metal vertical export flux and that aggregation cannot occur 448 without a minimal concentration of organic matter. This is consistent with the role of 449 biological production as a factor that determines the occurrence of vertical export 450 fluxes after spring bloom, while atmospheric deposition is likely to supply pelagic 451 waters with mineral material, but seems unable to trigger significant vertical export 452 fluxes.

453

454 **4. Conclusions**

455 Our results suggest that the marine vertical export flux of metals is controlled by 456 hydrology and biology, and not by atmospheric deposition. This statement is in 457 agreement with studies by (Deuser et al. (1983); Jickells et al. (1984); Buat-Menard 458 et al., 1989; 1998; Passow (2004)) and De La Rocha and Passow (2007). For 459 instance, Deuser et al. (1983) noticed early on the decoupling of atmospheric 460 deposition and marine vertical export flux at the BATS time-series site. At the BATS 461 time-series site, the removal of mineral particles from the surface ocean is controlled 462 by biology. We show that the vertical export flux at the DYFAMED time-series site is 463 controlled by both hydrology and biology. The hydrology of the Northwestern 464 Mediterranean Sea is strongly constrained by meteorological conditions (winter 465 temperature, wind events, rain events) and it conditions biology (Marty and 466 Chiavérini, 2010). As a consequence, the marine vertical export flux in the 467 Northwestern Mediterranean Sea is strongly dependent on climatic and 468 meteorological conditions as well. Therefore, the understanding of the 469 interannual/decadal variability of vertical export fluxes of elements in relation with 470 climatic and meteorological changes requires i) reliable measurements of vertical

export fluxes, including the use of proxies such as ²³⁴Th and ²³⁰Th to minimize 471 472 possible bias in the measurements (Rutten et al., 2000; Roy-Barman et al., 2009), 473 and ii) good knowledge of physical (climatic and meteorological) parameters that 474 determine the magnitude of dense water formation, as well as their 475 interannual/decadal variability (Stabholz et al., 2013). This is important in terms of 476 ongoing global change, because any alteration of the climatic/meteorological 477 conditions would significantly impact the marine vertical export flux, and this would 478 ultimately determine the evolution of metal cycling in the Northwestern 479 Mediterranean, independently from changes in the atmospheric (metal) deposition.

We showed that the marine vertical export flux of metals to deep marine sediments is driven by dense water convection and primary production. The study metal deposition by means of deep marine sediment records requires therefore a profound understanding of the seasonal and long-term variability of primary production and hydrology.

Furthermore, we recommend comparing elemental concentrations rather than
 elemental fluxes in marine vertical export studies.

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- 488

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498

499 Tables

Table 1 Certified reference material (CRM) validation results, expressed in μ g.g⁻¹. The median relative standard deviation (RSD, in %) is calculated from full replicates including the mineralization step, 10 for GBW, 5 for BCSS-1 and PACS. The median recovery observed between certified and measured values is expressed in %.

504 Values noted * are close to the detection limit. Mean RSD values can be used to 505 determine the uncertainty of the analytical method.

506

507 **Table 2.** Mean metal concentrations and fluxes at the DYFAMED site and various 508 other moorings in the Mediterranean and Black Sea. Variability of metal 509 concentrations is relatively low compared to the variability of vertical export flux. 510 *values have been calculated from the available data given in the cited papers

511

Table 3. Pearson intercorrelation matrix of the z-scores of OC and metal concentrations of sediment trap material from 2003-2007 (number of variables 11, number of observations 91, missing values have been pairwise deleted). Values in bold are different from 0 with a significance level alpha=0.05.

516

517 **Table 4.** Enrichments factors of all metals of the vertical export flux and EFs of518 Mediterranean aerosols (Heimbürger et al., 2010b).

519

Table 5. Factor loadings after Varimax rotation. Factor 1 explains 58 % of the variation with the main contribution of OC and all metals except Cu, Zn and Cd. The negative OC value explains the relative dilution effect of the OC contribution to metal concentrations of the vertical export flux. Factor 2 explains 21 % of the variation and is principally determined by OC, Cu and Zn.

525

526 Figure captions

527 **Figure 1** Map of the Northwestern Mediterranean Sea and the location of the time-528 series sampling station DYFAMED.

529

Figure 2 Temporal variability of the vertical export flux (mg.m⁻².d⁻¹), particulate metal (Al, V, Cr, Mn, Fe, Ni, Cu, Zn, Cd and Pb) and organic carbon (OC) concentrations (μ g.g⁻¹ and %, respectively) of the sediment trap moored at 1000m-depth at the DYFAMED time-series site between 2003 and 2007. The vertical export flux is shown in each inset as bar plots (right axis). Particulate metal and OC concentrations are shown as black line with dots (left axis).

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CRM	Metal	AI	V	Cr	Mn	Fe	Ni	Cu	Zn	Cd	Pb
GBW	Measured	78900	107	57	3106	39100	163	498	152	0.2	27
	RSD	3	5	7	6	6	9	5	6	30*	8
	Certified	72800	112	58.4	3330	46000	150	424	160		29.3
	Recovery	108	96	97	93	85	109	117	95		92
BCSS-1	Measured	67600	93	105	231	30900	63	27	108	0.4	22
	RSD	3	5	5	5	6	5	8	12	7	18
	Certified	62600	94.4	123	229	33400	55.3	18.5	119	0.25	22.7
	Recovery	108	98	86	101	93	113	147	91	163*	98
PACS	Measured	70200	132	106	469	43400	57	508	824	3	400
	RSD	2	5	2	1	1	9	2	6	4	5
	Certified	64700	127	113	470	48700	44.1	452	824	2.38	404
1	Recovery	108	104	94	100	89	130	112	100	109	99
Mean	Mean RSD	3	5	5	4	4	8	5	8	6	10
	Mean										
	recovery	108	99	92	98	89	117	125	95	109	96

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Stud	y ition	th	is stu	dy	Migon et al., 2002	Heimb ürger et al. 2013 WMED	Roussiez et al., 2012	Theodosi et al., 2010				The	eodo: I., 20	si et 13					The	eodosi	et al.,	2013	
Stud	У	ים	WMED FAME) ED	WMED DYFAMED	DYFA MED	WMED GoL	EMED				I	EME	D						Blac	k Sea		
perio	od 	2	2003-0	7	1997-98	2005- 06	2003	1999- 2005				2	007–	08						200	7-09		
trap	depth		1000		200	(20m.	20m a b	1715		700			1200			2000			020			1020	
(11)			1000	RS	200	a.p.j	50111.a.D.	1715		700			1200	,		2000	RS		930	RS		1930	RS
		4	S D 1	(%)	RSD SD (%)					SD	RSD (%)		SD	RSD (%)	4	S D 1	(%)	2	S D 1	(%)	4	S D 6	(%)
	AI	5 0 7	6 0 8	22		58400			433 00	204 00	47	4150 0	1290 0	31	2 3 0 0	9 0 0	40	2 0 0	2 0 0	67	9 0 0	9 0 0 0	14 4
	v	7 5	2 0	27		108		5.75	56	14	25	56	14	25	5 2	1 1	21	2 5	1 4	56	2 7	1 6	59
(hg.g ^{.1})	Cr	7 9 6	2 6 2	33		94	76	20.5	100	110	110	120	180	150	6 4 5	1 9 1 7	30	5 5 2	2 9 1	53	7 7 3	7 3 2	95
-tration	Mn	9 2 3	4 2 5	37		785		30.1	550	190	35	520	180	35	0 2 0	0 4	34	4 0 1 3	0 6	42	0 1 5	0 9	70
concen	Fe	3 0 4	2 4 1	22		26300		1400	226 00	520 0	23	2190 0	5400	25	1 0 0	5 2 0	22	8 0 0	9 0 0	50	5 0 0	1 0 0	59
Meta	Ni	5 5	3	25		57	38		21	18	86	17	13	76	5	0	67	9	5	38	2	1	2
	Cu	7 6 1	1 8	24		47	27	6.85	26	16	62	33	17	52	4 0	1 5	38	7 6	3 2	42	1 0 0	1 1 0	11 0
	Zn	5 1 0.	5 6 0.	37		107	113	27.4				~		2	1								
	Cd	2 8 1 3	1 2 0	43		0.11	0.17	0.11	1.0	1.2	120	1.5	1.3	87	1. 3 4	1. 7 1	13 1	2. 2 3	2. 0 5	91 15	4. 0 3	8. 6	21 5 16
Vert	Pb	8	8	21		44	37	4.66	38	12	32	40	13	33	3	6	37	5	5	7	9	3	2
expo (mg. 1)	ort flux m ⁻² .d	1 0 5	1 5 4	14 7	77 77 100	105*	3		63*	2		57*			5 3*			1 2 0*			5 2*		
	AI	5 7 4 9	9 6 2 1	16 7	389 422 2 3 109	6132*	0	0	260 0	220 0	85	2200	2200	100	2 0 0 0	1 9 0 0	95	2 7 0 0	3 2 0 0	11 9	1 7 0 0	1 2 0 0	71
	v	9. 1 1 9	4. 9 1	16 3	1	11.3*	5		3.7	3.2	86	3	4	106	2. 9	2. 9	10 0	2. 8	3. 4	12 1	1. 9	1. 9	10 0
d ⁻¹)	Cr	4 7	4. 5 1	15 4	0	9.87*			5.4	5.1	94	6	10	182	3. 1	2. 9	94	6. 4	7. 2	11 3	4. 2	3. 2	76
(µg.m ⁻² .	Mn	7 4 2	0 3 4	13 9	5	82.4*			36	30	83	32	36	113	2 7 1	2 6 1	96	2 4 1	2 2 1	92	1 8 1	1 6	89
port flux	Fe	8 1 5 6	7 7 5 8	17 0		2762*			150 0	140 0	93	1400	1500	107	1 5 0	1 7 0	10 2	6 0 0	9 0 0	11 9	0 0 0	9 0 0	90
netal ex	Ni	0 9 6.	4 4 7	13 9		5.99*			1.7	1.7	100	1.4	1.9	136	1. 2	1. 5	12 5	4. 6	4. 9	10 7	3. 4	2. 8	82
/ertical r	Cu	8 7 1	8 8 1	11 5		4.94*			1.9	1.9	100	2.1	2.5	119	2. 1	2. 0	95	8. 2	7. 0	85	6. 0	5. 9	98
_	Zn	3. 3 0.	4. 9 0.	11 2	18.712.1 65	11.2*									0	0		0	0		0	0	
	Cd	U 2 4 9 3.	U 3 9 0 5.	15 6	0.080.06 75	0.0116 * 4.			0.04 4	0.04 7	107	0.04 4 2	0.04 1 2	93	0. 0 3 0	0. 0 2 4	80	0. 2 8 0	0. 5 7 0 5	20 4	0. 1 3 0	U. 1 3 0	10 0
	Pb	7 6	0 0	13 3	4.9 4.9 100	62 *		. <u>.</u>	2.2	1.7	77	1	0	9 5	1. 9	1. 6	84	6. 0	5. 0	91 7	1. 8	1. 6	89

R^2	OC	AI	V	Cr	Mn	Fe	Ni	Cu	Zn	Cd	Pb
00	1										
AI	-0.71	1									
V	-0.72	0.98	1								
Cr	-0.62	0.91	0.94	1							
Mn	-0.66	0.75	0.80	0.83	1						
Fe	-0.65	0.98	0.96	0.92	0.74	1					
Ni	-0.45	0.81	0.85	0.94	0.79	0.84	1				
Cu	0.35	-0.22	-0.17	-0.14	-0.04	-0.18	0.01	1			
Zn	0.44	-0.34	-0.29	-0.25	-0.27	-0.29	-0.13	0.72	1		
Cd	0.60	-0.37	-0.36	-0.36	-0.42	-0.30	-0.26	0.20	0.31	1	
Pb	-0.24	0.53	0.61	0.67	0.72	0.58	0.72	0.39	0.14	-0.12	1
											R
										- 2	1
										1	11
										2	1997 - N
										1	P.)

	V	Cr	Mn	Fe	Ni	Cu	Zn	Cd	Pb
Mean	2.28	3.61	1.98	1.23	4.76	8.93	4.87	4.81	3.14
SD	0.16	0.54	0.52	0.06	0.60	2.97	2.30	2.79	0.62
RSD	7	15	26	5	13	33	47	58	20
MIN	1.91	2.62	0.62	1.08	2.78	3.80	1.91	1.20	1.75
MAX	2.65	5.02	2.92	1.47	6.68	16.42	12.39	17.76	4.62
N	91	91	91	91	89	86	83	88	89
EF of MED aerosol*			2.59	2.63	66.9	172	164	597	109

	F1	F2
00	-0.65	0.53
AI	0.92	-0.25
V	0.96	-0.18
Cr	0.96	-0.10
Mn	0.89	-0.08
Fe	0.93	-0.17
Ni	0.92	0.07
Cu 🦱	0.00	0.89
Zn 🔪	-0.18	0.84
Cd	-0.38	0.45
Pb	0.77	0.46

- 842 843 Highlights for "Vertical export flux of metals in the Mediterranean Sea"
- Lars-Eric HEIMBÜRGER^(1, 2, 3,*), Christophe MIGON^(1, 2), Rémi LOSNO⁽⁴⁾, Juan-Carlos 844
- MIQUEL⁽⁵⁾, Benoît THIBODEAU^(1, 2, 6), Marion STABHOLZ⁽⁷⁾, Aurélie DUFOUR^(1, 2) & Nathalie 845
- LEBLOND^(1, 2) 846
- 847
- 848 (1) Dense water convection and surface primary production control the vertical export flux of
- 849 metals
- 850 (2) Atmospheric deposition events do not trigger significant vertical export fluxes
- .ux .ixel. (3) all metals whether of natural or crustal origin are accumulated in the mixed layer until 851
- 852 they are exported at the same time
- 853 854



