
Microbial Diversity and Biosignatures: An Icy Moons Perspective

Jebbar Mohamed ^{1,*}, Hickman-Lewis Keyron ², Cavalazzi Barbara ³, Taubner Ruth-Sophie ⁴, Rittmann Simon K.-M. R. ⁵, Antunes Andre ⁶

¹ Univ. Brest, CNRS, Ifremer, Laboratoire de Microbiologie des Environnements Extrêmes, 29280, Plouzané, France

² Dipartimento di Scienze Biologiche, Geologiche e Ambientali, Università di Bologna, Bologna, Italy

³ Centre de Biophysique Moléculaire, CNRS, Rue Charles Sadron, 45071, Orléans, France

⁴ Department of Geology, University of Johannesburg, APK Campus, Johannesburg, South Africa

⁵ Archaea Physiology & Biotechnology Group, Archaea Biology and Ecogenomics Division, Department of Ecogenomics and Systems Biology, Universität Wien, Wien, Austria

⁶ State Key Laboratory of Lunar and Planetary Sciences, Macau University of Science and Technology (MUST), Taipa, Macau SAR, China

* Corresponding author : Mohamed Jebbar, email address : mohamed.jebbar@univ-brest.fr

Abstract :

The icy moons of the outer Solar System harbor potentially habitable environments for life, however, compared to the terrestrial biosphere, these environments are characterized by extremes in temperature, pressure, pH, and other physico-chemical conditions. Therefore, the search for life on these icy worlds is anchored on the study of terrestrial extreme environments (termed “analogue sites”), which harbor microorganisms at the frontiers of polyextremophily. These so-called extremophiles have been found in areas previously considered sterile: hot springs, hydrothermal vents, acidic or alkaline lakes, hypersaline environments, deep sea sediments, glaciers, and arid areas, amongst others. Such model systems and communities in extreme terrestrial environments may provide important information relevant to the astrobiology of icy bodies, including the composition of potential biological communities and the identification of biosignatures that they may produce.

Extremophiles can use either sunlight (phototrophs) or chemical energy (chemotrophs) as energy sources, and different chemical compounds as electron donors or acceptors. Aerobic microorganisms use oxygen (O₂) as a terminal electron acceptor, whereas anaerobic microorganisms may use nitrate (NO₃⁻), sulfate (SO₄⁻²), carbon dioxide (CO₂), Fe(III), or other organic or inorganic molecules during respiration. The phylogenetic diversity of extremophiles is very high, leading to their broad dispersal across the phylogenetic tree of life together with a wide variety in metabolic diversity.

Some metabolisms are specific to archaea, for example, methanogenesis, an anaerobic respiration during which methane (CH₄) is produced. Also sulfur-reduction performed by some bacteria and archaea is considered as a primitive metabolism which is restricted to anoxic sulfur-rich habitats in nature.

Methanogenesis and sulfur reduction are of specific interest for icy moon research as it might be one of the few known terrestrial metabolisms possible on these celestial bodies.

Therefore, the adaptation of these intriguing microorganisms to extreme conditions will be highlighted within this review.

Keywords : Extremophiles, Prokaryotes, Metabolism, Diversity, Adaptation, Space explorat

1 Introduction

Most definitions of a “living” process or system (Cleland and Chyba 2002) refer to three distinct properties: the ability to self-organize complex macromolecular structures, the ability to harness the energy necessary to maintain separate organization from the environment (“metabolism”), and an ability to replicate the self and to proliferate more or less identically. Thus, self-organization, metabolism and self-replication are the characteristic “cornerstones” of any living entity. This conceptual definition is the result of a long historical process, which has forced biologists of subsequent eras to redefine their understanding of what is, and is not, alive.

Regardless, a general and consensual definition of “living” is a matter of concern to many fields—physicists, chemists and astrobiologists—all of whom seek to recreate life-like behaviors, or to identify its signatures. A challenge is the continual confrontation only with terrestrial biology composed of a remarkably common base of molecular components (nucleic acids, proteins, lipids), each of which is dedicated to a main specific function: the conservation and handling of information for nucleic acids, structural organization and biochemical catalysis for proteins, and spatial delineation of compartments for lipids. On Earth, the co-occurrence of these building blocks indicates an affiliation with the “living” world.

The issue of the origin of life is usually addressed from two different perspectives: the first examines the conditions under which the basic building blocks and macromolecules significant for life may have emerged on the early Earth, the second explores the origin of functional subsystems (metabolism, replication) and basic structural organization (i.e., the cell) of what is recognized as “alive”.

Experiments from the 1920s onward (Oparin 1924; Haldane 1929; Urey 1951; Miller 1953) led to the hypothesis of multiple “possible” scenarios of the origin of life (on Earth), often conflicting and irreconcilable, and typically almost mythological “stories” of physically and/or chemically possible processes, of which only small parts had an empirical basis. Despite their value in the inception of a debate on the origins of life, these hypotheses were often based on what we now know to be inaccurate or incomplete data (early Earth atmospheric composition, the genesis and dynamics of the Solar System, etc.). Early reports on the diversity of life (Woese 1979) also failed to consider the incredible metabolic sophistication that microbial life has shown in acquiring energy from its environment, even under the most inhospitable and extreme conditions. Clarification of the co-evolution of the

101 biosphere and geosphere has led to an increasing recognition of the fact that the two are
102 intimately associated and likely constrained the development of the other (Lovelock and
103 Margulis, 1974; Williams and Fraústo Da Silva 2003).

104 The Earth is generally thought of as a world inhabited by plants, animals and microor-
105 ganisms able to grow under conditions compatible with life as it is found in most terrestrial
106 and marine ecosystems (temperature: 10–40 °C, pH ~ 7, pressure: 1 atm, water availability,
107 minimal ionizing radiation level, etc.). Extremophiles have succeeded in inhabiting environ-
108 mental niches with physicochemical parameters outside this comfort zone.

109 Extreme environments are characterized by environmental parameters at the boundaries
110 of conditions that sustain and shape life in its various forms; whether terrestrial, oceanic,
111 cryospheric or deep endolithic, they are widespread on our planet. Far from being marginal
112 areas, they (especially the deep ocean and polar regions) represent, in terms of biomass
113 volume, the most important part of the Earth biosphere.

114 In these extreme environments, dominated by prokaryotic microorganisms (Bacteria and
115 Archaea), some organisms thrive under conditions that are at the limits of their physiological
116 and energy potential, whereas others have highly adapted genetic features that result in acru-
117 cial requirement of such conditions. When classifying microorganisms as extremophiles, the
118 concept of a “normal” environment is used as a reference. In assuming this anthropocentric
119 view, it should not be forgotten that “extreme” environments, which today seem so hostile,
120 appear to have predominated when the first life forms appeared on Earth. Nowadays, these
121 environments are still colonized by highly diverse microbial communities.

122 Depending on the prevailing physico-chemical parameters of the environment, ex-
123 tremophiles can be subdivided into different categories: hyperthermophiles ($T_{\text{opt}} \geq 80$ °C)
124 e.g. *Methanopyrus kandleri*, () the archaeon with the highest temperature life record); psy-
125 chrophiles ($T_{\text{opt}} \leq 15$ °C) such as the bacterium *Psychrobacter fulvigenes* (Romanenko et al.
126 2009) capable of growing at temperatures as low as -5 °C; acidophiles ($\text{pH}_{\text{opt}} \leq 3$) includ-
127 ing *Picrophilus oshimae* (Schleper et al. 1995), an archaeon that has shown optimal growth
128 at pH = 0.7; alkaliphiles ($\text{pH}_{\text{opt}} \geq 9$) such as *Bacillus pseudofirmus* (Nielson et al. 1995),
129 capable of growing at pH 11, halophiles such as the archaeon *Halobacterium salinarum*
130 (Ventosa and Oren 1996), which can survive in the presence of 5.5 M (32%) NaCl (its sat-
131 uration limit); and piezophiles, e.g. *Thermococcus piezophilus* the archaeon that holds the
132 record for withstanding the highest hydrostatic pressure (130 MPa, i.e. 1300 times atmo-
133 spheric pressure) (Dalmasso et al. 2016).

134 Extremophiles expand our understanding of biodiversity on Earth and our knowledge of
135 the limits of life. Deducing the mechanisms that enable extremophiles to persist under harsh
136 conditions not only provides a thorough knowledge of the functioning of living cells but can
137 also lead to interesting applications in biotechnology, particularly the economic utility of
138 extremophiles. Understanding the uncommon properties of extremophiles has led to ques-
139 tions about their origin (have these organisms recently adapted to the extreme conditions of
140 their environment or are they relics of organisms that existed on the Early Earth and that
141 had to face even harsher environmental conditions?). Understanding the limits of life on
142 Earth can provide hints of the diversity of potential extraterrestrial life (past or present). It
143 is therefore not surprising that astrobiology studies the properties of life in Earth’s extreme
144 environments.

145 This review is dedicated to describing the state of the art and raising questions about tax-
146 onomic and metabolic diversity and the evolution of microorganisms (archaea and bacteria),
147 notably extremophiles, and their biosignatures with an astrobiological perspective.

2 Extremophiles: Diversity, Adaptation and Biosignatures

Over the past forty years, research has dramatically altered our understanding of the limits of life in terms of its physical and chemical constraints. Organisms, mainly prokaryotes, have been found to live optimally at very high or very low temperatures, in hyperacid or alkaline environments, or in salt-saturated environments, for example. Other organisms are able to live or survive under conditions of extreme stress, for instance a lack of water, the presence of high concentrations of heavy metals, or exposure to significant radiation doses or extreme pressures. In the following, we review and offer perspectives on extremotolerances to hypersaline and high hydrostatic pressure environments that are of particular relevance to the icy oceanic bodies of the outer Solar System.

2.1 Hypersaline Biotopes

Habitats with salinities higher than average seawater (i.e. 3.5% total dissolved salts) are considered hypersaline. Many of these habitats result directly from the evaporation of sea water, and thus have similar relative proportions of ions; for example, they are dominated by sodium and chloride. Marine hypersaline environments are termed thalassohaline, in contrast to athalassohaline environments, which have non-marine ionic compositions and are associated with non-coastal water bodies (DasSarma and Arora 2001; Rodríguez-Valera 1988).

A profusion of hypersaline biotopes, distributed across the Earth, can be found in arid, coastal and even deep-sea settings (e.g. Antunes et al. 2011; DasSarma and Arora 2001; Oren 2002a, 2002b). In coastal regions, seawater often penetrates through seepage or narrow inlets creating small evaporation ponds. Well-known examples of such ponds are Solar Lake and Gavish Sabkha near the Red Sea coast, Guerrero Negro on the Baja California peninsula (Mexico), Lake Sivash near the Black Sea (Crimea), and Shark Bay in Western Australia. Such hypersaline evaporation ponds have also been found in Antarctica (e.g. Deep Lake, Organic Lake and Lake Suribati). Elevated salinities are usually found in natural inland hypersaline lakes such as the Dead Sea (Middle East) and the Great Salt Lake (USA), the two largest and best-studied such environments. A number of alkaline hypersaline soda brines also exist, including the Wadi Natrum lakes of Egypt, Lake Magadi in Kenya, the Great Basin lakes of the western United States (Mono Lake, Owens Lake, Searles Lake and Big Soda Lake), and several series throughout China and India.

The number of hypersaline sites is further increased by the numerous artificial solar salterns constructed for the production of sea salt, by subterranean brines and evaporite deposits and by the existence of several brine-filled deep-sea basins. Another type of hypersaline biotope is presented by the often-overlooked saline soils. These include desolate areas in such places as Death Valley (California, USA), Alicante (Spain), Iraq and even the Dry Valleys in Antarctica, amongst others (Ventosa et al. 1998).

Anoxic hypersaline basins, or deep-sea anoxic brines, are very special and rare environments in the oceans. They are formed as a result of tectonic activity and exposure of ancient salt deposits, existing under layers of sediments and originated from evaporated ancient seas (e.g. Antunes et al. 2011; Antunes 2017). The interaction of seawater with the underlying salt leads to the formation of brines which are 4 to 5 times more concentrated in salt than the surrounding seawater, creating highly saline “lakes” on the sea floor (Camerlenghi 1990). The presence of such basins often coincides with the presence of cold seep zones or, more rarely, hydrothermal vents, resulting in the release of methane, hydrogen sulphide and hydrocarbons. One of the characteristics of anoxic hypersaline basins is the

201 presence of multiple gradients, particularly at the interface between seawater and the hyper-
202 saline zone (brine), including salinity, temperature, free O₂, density and pH (Antunes et al.
203 2018). These physico-chemical gradients provide highly variable and specific environments
204 of interest for the growth of microorganisms. In addition, the density gradient formed at the
205 seawater/hypersaline zone interface acts as an organic and inorganic particle trap, providing
206 the significant amount of nutrients necessary for cell growth (Daffonchio et al. 2006). It is
207 possible to distinguish differences between the known brine lakes of different seas; in the
208 Mediterranean Sea, concentrations of Mg²⁺, SO₄²⁻ and K⁺ are high, whereas in the Red Sea,
209 concentrations of Ca²⁺ and Mn²⁺ are higher. On the contrary, lower ionic concentrations,
210 particularly Mg²⁺ and K⁺, exist in the Gulf of Mexico (Antunes et al. 2011).

211 The relevance of deep-sea brines in the context of the exploration of the oceans of the icy
212 moons the outer solar system is particularly worth highlighting as they have been recently
213 proposed as potential terrestrial analogues to conditions in such exooceans (Antunes et al.,
214 accepted).

215 216 **2.2 Biodiversity in Hypersaline Environments**

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218 Despite being considered extreme, hypersaline environments host a diverse variety of
219 organisms, including representatives from all three domains of life. In fact, microbial
220 densities can be so high in these locations that the thriving communities of pigmented
221 halophilic microorganisms (which includes a few bacteria but is composed mostly of
222 halophilic archaea and/or the β -carotene-rich green alga *Dunaliella*) often give the wa-
223 ter characteristic pinkish or even reddish hues. The inhabitants of saline environments
224 range from higher organisms to unicellular eukaryotic microorganisms, and a heterogeneous
225 group of prokaryotes, which constitute the predominant microflora (Rodríguez-Valera 1988;
226 Ventosa and Nieto 1995).

227 228 *2.2.1 Eukarya in Saline Environments*

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230 Within the domain Eukarya, halophiles are scarce, and mostly restricted to unicellular forms
231 (Oren 2002b; Trüper and Galinski 1986). A variety of plants (e.g. *Atriplex halimus*) can
232 survive in moderately high saline soils, although apparently no vertebrate has ever been
233 reported at salinities higher than 1 M NaCl (DasSarma and Arora 2001; Ollivier et al. 1994).
234 The most common multicellular eukaryotes in hypersaline environments are invertebrates,
235 with reported species including rotifers, tubellarian worms, copepods, ostracods, and insects.
236 Noteworthy among the insects are the well-known brine flies (*Ephydra hians* and *E. gracilis*)
237 and brine shrimp (*Artemia franciscana* and *A. salina*), with the latter playing an important
238 role in the nutrition of the pink flamingo and other birds (DasSarma and Arora 2001; Ventosa
239 and Nieto 1995).

240 Dense populations of unicellular green algae can be observed at moderately high salin-
241 ities, with most being moderate halophiles and only very few examples observed at higher
242 salinity level (e.g. *Dunaliella salina* and *Asteromonas gracilis*). The several species of the
243 genus *Dunaliella* are almost ubiquitous in hypersaline environments, being often the main
244 or only primary producer and serving as main food source for brine shrimps and larvae of
245 brine flies, while representatives of diatoms are also frequently found but rarely abundant
246 (DasSarma and Arora 2001; Oren 2002a).

247 Other eukaryotic representatives include a large variety of protozoa (e.g. *Porodon uta-*
248 *hensis*) as well as yeasts and other fungi (DasSarma and Arora 2001). These groups of
249 organisms are very often overlooked when looking at the microbiology of high salinity

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251 environment but our knowledge about their diversity has been getting increased attention
252 (excellently reviewed by e.g. Gunde-Cimerman et al. 2009; Hardy and Simpson 2017; Zajc
253 et al. 2017). <ref:hs17?>

254 2.2.2 Archaea in Saline Environments

255 Extreme halophiles are traditionally associated with the euryarchaeal class *Halobacteria*,
256 which was recently reorganized (Gupta et al. 2015) and split into 3 different orders—
257 *Halobacteriales*, *Haloferacales*, and *Natrialbales*—and is still undergoing taxonomic re-
258 structuring based on phylogenomic data (e.g. Gupta et al. 2016). As of September 2019, this
259 family of aerobic euryarchaeotes currently comprises 259 species with validly published
260 names, placed in 63 genera (Table 2). An interesting member of the *Halobacteriaceae* is
261 the more recently isolated first representative of the square haloarchaea of Walsby, *Halo-*
262 *quadratum walsbyi* (Bolhuis et al. 2004; Burns et al. 2004). This intriguing group of mi-
263 croorganisms was first reported by Walsby (1980) but remained elusive despite numerous
264 cultivations attempts and well-known widespread and abundant occurrence.

265 Extremely halophilic archaea are less common outside the *Halobacteria* but can also
266 be found within some euryarchaeal genera namely within the class *Methanomicrobia* (e.g.
267 *Methanosalsum*, *Methanohalobium*, *Methanohalophilus*, within the family *Methanosarci-*
268 *naceae* and *Methanocalculus*, within an unassigned family of the order *Methanomicrobiales*,
269 and the recently described genus *Methanonatronarchaeum* of the class *Methanonatronar-*
270 *chaeia*; Oren 2014; Ventosa et al. 2012; Sorokin et al. 2018). In addition to these, a few
271 other methanogenic genera are also known to include moderately halophilic species (Ven-
272 tosa et al. 1998). Aside from these methanogens, and the *Halobacteria*, no other archaeal
273 halophiles have been identified outside the Euryarchaea.

274 2.2.3 Bacteria in Saline Environments

275 Overall, halophilic bacteria are a very diverse and heterogeneous group. Phylogenetically
276 they are included in at least seven phyla: *Actinobacteria*, *Bacteroidetes*, *Cyanobacteria*,
277 *Firmicutes*, *Proteobacteria*, *Spirochaetes*, and *Thermotogae* (Ventosa et al. 2012).

278 Compared to the Archaea, fewer examples of extreme halophily are currently known
279 in Bacteria but their numbers have increased rapidly in the last few years. Some examples
280 of this wide diversity include the actinomycete *Actinopolyspora halophila*, several gamma-
281 proteobacteria of the genus *Halorhodospira*, and *Salinibacter ruber*, which is a member
282 of the *Cytophaga-Flavobacterium-Bacteroides* group (Antón et al. 2000, 2002; Kamekura
283 1998). *Salinibacter* is especially interesting due to its significant contribution to the biota
284 of NaCl-saturated saltern crystallizer ponds. The surprisingly numerous similarities with
285 the haloarchaea, specifically in osmotic adaptation strategy, point to a possible process of
286 convergent evolution (Antón et al. 2002; Oren 2004).

287 Moderately halophilic bacteria, however, are much more diverse, being present in many
288 of the major bacterial phylogenetic groups. The vast majority of the validly described mod-
289 erately halophilic bacteria are members of the Proteobacteria, with the gamma-subgroup,
290 namely the genera *Salinivibrio*, *Marinobacter*, and *Arhodomonas*, as well as members of
291 the family *Halomonadaceae*, being especially preponderant. The *Halomonadaceae* includes
292 some of the most versatile prokaryotes regarding their adaptability to a wide range of salin-
293 ities (Oren 2000; Ventosa et al. 1998). *Rhodospirillum salinarum*, an anaerobic phototroph,
294 and *Desulfovibrio halophilus* and *Desulfohalobium retbaense*, both anaerobic sulphate re-
295 ducers, are further examples of organisms with wide ranges of salinity tolerances within
296

301 the alpha- and delta-Proteobacteria, respectively (Galinski and Trüper 1994; Ollivier et al.
302 1994).

303 The *Halanaerobiales*, an order within the low G+C branch of the Gram-positive bacteria
304 includes the families *Halobacteroidaceae* and *Halanaerobiaceae*, other very important and
305 numerous groups of moderately halophilic bacteria (Rainey et al. 1995). Further representa- <ref:ra95?>
306 tives are found in the low G+C and high G+C Gram-positive bacteria, the cyanobacterial
307 branch, the *Cytophaga-Flavobacterium-Bacteroides* branch, and also within the spirochetes
308 and the actinomycetes (Ventosa et al. 1998).

310 2.2.4 Physiological Adaptations to High Salinity

311
312 Life at high salinity is not without its burdens. Increased salinity leads to a decrease in water
313 activity (i.e. the amount of water that is thermodynamically available) which, in accordance
314 with the natural tendency of systems to attain and maintain equilibrium and the permeabil-
315 ity of the cytoplasmic membrane to water, afflicts cells with osmotic stress (Brown 1990;
316 Csonka 1989; Vreeland 1987). Indeed, an unadapted organism placed in a saline environ-
317 ment (i.e. hyperosmotic conditions) will rapidly lose water, leading to decreased cell vol-
318 ume and/or turgor pressure and ultimately affecting its metabolism and macromolecules
319 (da Costa et al. 1998; Poolman and Glaesker 1998). Failure to adjust to these new condi-
320 tions results in cessation of growth, possibly due to molecular crowding and a consequent
321 reduction in diffusion rates of proteins and metabolites, which may eventually result in cel-
322 lular death (Kunte et al. 2002). Evolution has provided life with two different approaches to
323 deal with osmotic stress:

324 *The salt-in-cytoplasm strategy.* Using this strategy, the necessary thermodynamic ad-
325 justment of the cell is achieved through an increase in cytoplasmic salt concentration
326 (normally through an increased intake of K^+ and Cl^-). The resulting increase in in-
327 tracellular ionic strength requires several changes in cellular function, most markedly
328 at the level of the enzymatic machinery, resulting in a characteristic excess of acidic
329 amino acids and small amounts of hydrophobic amino acids (da Costa et al. 1998;
330 Oren 1999). The predominance of charged amino acids on the surface of enzymes and
331 ribosomes stabilizes their hydration shells under high ionic conditions. Moreover, most
332 of these enzymes are only functional at increased ionic levels (da Costa et al. 1998;
333 Galinski and Trüper 1994). The permanent character of these cellular modifications restricts
334 organisms that use this strategy to highly saline environments.

335 This salt-in-cytoplasm strategy was first discovered in aerobic, extremely halophilic ar-
336 chaea of the order *Halobacteriales* and is considered the typical archaeal strategy of os-
337 moadaptation (Kunte et al. 2002). This strategy is also used by anaerobic halophilic bacteria
338 of the order *Halanaerobiales* and the aerobic halophilic bacteria *Salinibacter ruber* (Oren
339 2000).

340 *The organic-osmolyte strategy.* This strategy relies on an increase in external salinity
341 being counteracted by the accumulation (either by *de novo* synthesis or uptake from the en-
342 vironment) of uncharged, highly water-soluble, organic solutes (Kempf and Bremer 1998)
343 (Fig. 1). These osmolytes do not disrupt metabolic processes and include sugars (e.g. tre-
344 halose), polyols (e.g. glycerol) and their derivatives, free amino acids (e.g. glutamate) and
345 their derivatives, betaines, and ectoines (Csonka 1989; da Costa et al. 1998; DasSarma and
346 Arora 2001; Galinski and Trüper 1994). This strategy allows organisms to keep their cyto-
347 plasm free of NaCl, to a large extent, while avoiding the need for major changes in cellular
348 machinery, thus providing a higher physiological flexibility. This explains the characteristi-
349 cally wide salt tolerance ranges associated with the use of this type of osmotic adaptation.

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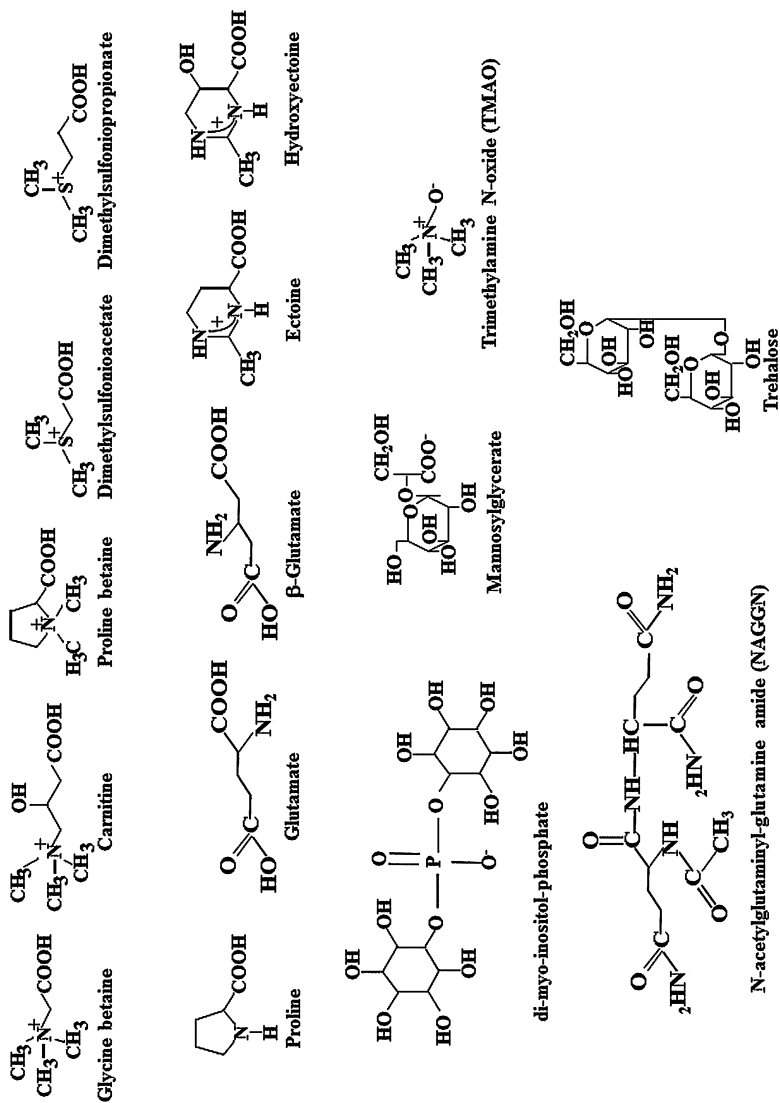


Fig. 1 Chemical structures of selected compatible solutes or osmolytes

401 The organic-osmolyte strategy is widespread among Bacteria, Eukarya and some Archaea.
402 Indeed, some methanogenic along with some haloalkaliphilic Archaea are known to use a
403 combination of both strategies (Desmarais et al. 1997).
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405 **2.3 High Hydrostatic Pressures Biotopes**

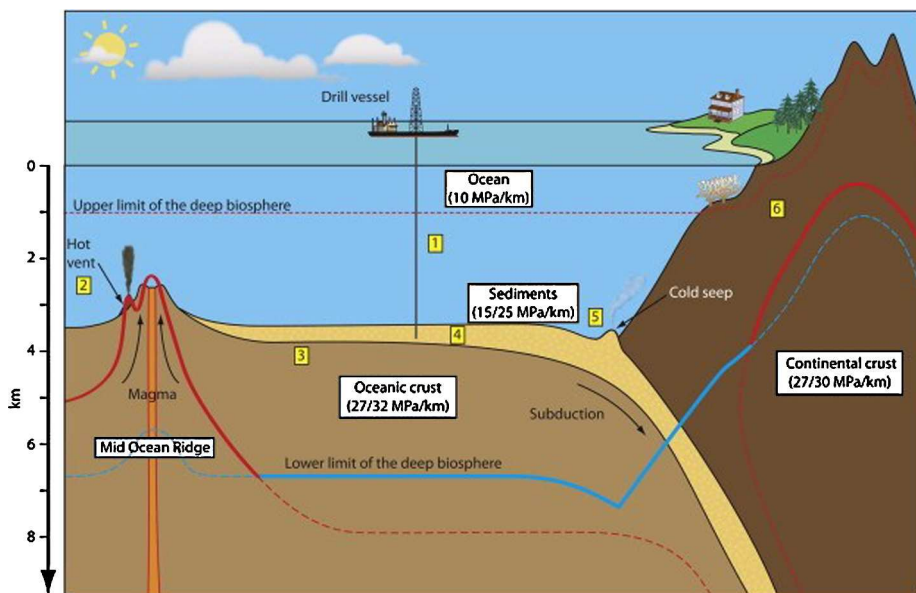
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407 The Twentieth Century was marked by technological and scientific breakthroughs that have
408 drastically modified the way we understand life on our planet. It was demonstrated that uni-
409 cellular prokaryotic life forms are able to inhabit virtually any environment on Earth, and
410 that they constitute life's largest diversity reservoir. The domain Archaea was created to
411 accommodate newly isolated prokaryotic organisms with specific features that make them
412 more similar to eukaryotes. Recent estimates also suggest that life dwells mostly under-
413 ground (Reith 2011; Colwell and D'Hondt 2013; Colman et al. 2017) and that this deep
414 biosphere, located in the continental subsurface and in the oceans below 1000 m in depth,
415 could represent up to 70% of all cells on Earth, and up to 50% of (Oger and Jebbar 2010)
416 the primary production of biomass. Most of these biotopes are oligotrophic in nature and
417 characterized by high hydrostatic pressures (HHP). Although the deep biosphere represents
418 the largest ecosystem on Earth, however, it is still poorly characterized in terms of diversity
419 and its mechanisms of adaptation to HHP.
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421 Amongst deep-biosphere biotopes, hydrothermal vents may be the most intriguing.
422 Discovered in 1979, they were shown, despite being hot oligotrophic and HHP environ-
423 ments, to harbor abundant primary productivity and diversity (Corliss et al. 1979). Pri-
424 mary production, in these environments, is based exclusively on the anaerobic chemical
425 harvest of the energy of the geologically sourced fluids seeping through the ocean floor.
426 Because of this, they are the only ecosystems on Earth not linked to photosynthesis, or
427 photosynthesis-derived products such as O₂. It has been postulated that deep-sea hydrother-
428 mal vent systems were the birth sites of life on Earth (e.g., Martin and Russell 2003;
429 Russell et al. 2010) and this item is described below in more details in Part 4 of this re-
430 view.

431 HHPs are ubiquitous in deep environments. Hydrostatic pressure increases with depth at
432 an approximate rate of 10 MPa (~100 atmospheres or 100 bar) per km in the water column
433 and 30 MPa per km in the crust. The definition of the deep biosphere is conveniently and
434 arbitrarily defined as water depths of 1000 m and more (Jannasch and Taylor 1984). Con-
435 sequently, all environments above 10 MPa qualify as high-pressure biotopes. HHP waters
436 encompass 88% of the volume of the oceans—which have an average depth of 3800 m—
437 and thus an average hydrostatic pressure of ca. 38 MPa, but reach 110 MPa in the trenches.
438 In contrast, the average geothermal gradient in the continental system is ca. 25 °C km⁻¹
439 (Oger and Jebbar 2010). The current temperature limit for life, 122 °C (Takai et al. 2008),
440 would thus place the “deep” limit for the putative continental biosphere at ca. 5 km below
441 ground on average, under maximal pressures of 150 MPa. Most of the Earth's prokaryotes
442 live in these subsurface oceanic and terrestrial environments. From current knowledge of
443 the deep-biosphere their cell number is estimated at 3.5×10^{30} and ca. $2-6 \times 10^{29}$ respec-
444 tively i.e. about 10 times that estimated for surface environments (Whitman et al. 1998;
445 Magnabosco et al. 2018). Thus, even though the maximal productivity of the high-pressure
446 continental or marine biosphere is orders of magnitude lower than that of the surface
447 biotopes, due to their extremely large volume, these high-pressure biotopes contribute sig-
448 nificantly to the production and recycling of organic carbon (Fig. 2) (Magnabosco et al.
449 2018).
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471 **Fig. 2** Schematic transversal section of the Earth highlighting the numerous settings of the deep biosphere.
472 1: deep-sea; 2: deep-sea hydrothermal vents; 3: deep oceanic crust; 4: sedimentary sub-sea-floor; 5: deep-sea
473 cold seep; 6: continental deep biosphere. The red and blue lines represent the currently known temperature
474 and pressure limits for life, respectively. Solid lines highlight the parameter which limits the depth of the
475 deep biosphere. The upper dashed red line symbolizes the arbitrary 10 MPa upper limit of the deep biosphere
(Oger and Jebbar 2010)

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477 2.3.1 Physical Characteristics of High-Pressure Biotopes

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479 The deep ocean is characterized by HHP, darkness, a stable average temperature of ca.
480 2 °C, low organic carbon and a relative constant oxygen concentration. It is estimated that,
481 at present, ca. 1% of the carbon fixed by photosynthesis on the ocean surface eventually
482 reaches the ocean floor, thus the major nutritional potential of the deep-sea is defined by a
483 relatively low input of organic carbon (Oger and Jebbar 2010). As a corollary, adaptations
484 to oligotrophy (life with limited access to nutrients) and psychrophily (optimal life at low
485 temperature) are common in these environments. In contrast to the deep-sea biosphere, the
486 deep-continental biosphere is considerably more diverse.

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488 2.3.2 Diversity of HHP-Adapted Microorganisms

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490 The field of piezomicrobiology has suffered largely from a requirement for expensive high-
491 pressure retention sample containment and culturing laboratory equipments. The first HHP-
492 adapted prokaryotes were bacteria isolated from deep-sea sediments by Zobell and Johnson
493 (1949). The first obligate piezophiles, e.g. organisms that cannot develop at ambient pressure
494 and temperature, were isolated in 1981 (Yayanos et al. 1981). The diversity of piezophiles
495 in the deep-sea is largely dominated by five genera of psychrophilic, heterotrophic bacteria
496 (*Colwellia*, *Moritella*, *Shewanella*, *Psychromonas*, and *Photobacterium*) from the gamma-
497 Proteobacteria (Fig. 3). In contrast, the diversity of prokaryotes isolated from hydrothermal
498 environments is dominated by archaeal and bacterial hyperthermophilic chemolithotrophs,
499 i.e., those capable of gaining energy from the chemical transformation of dissolved minerals

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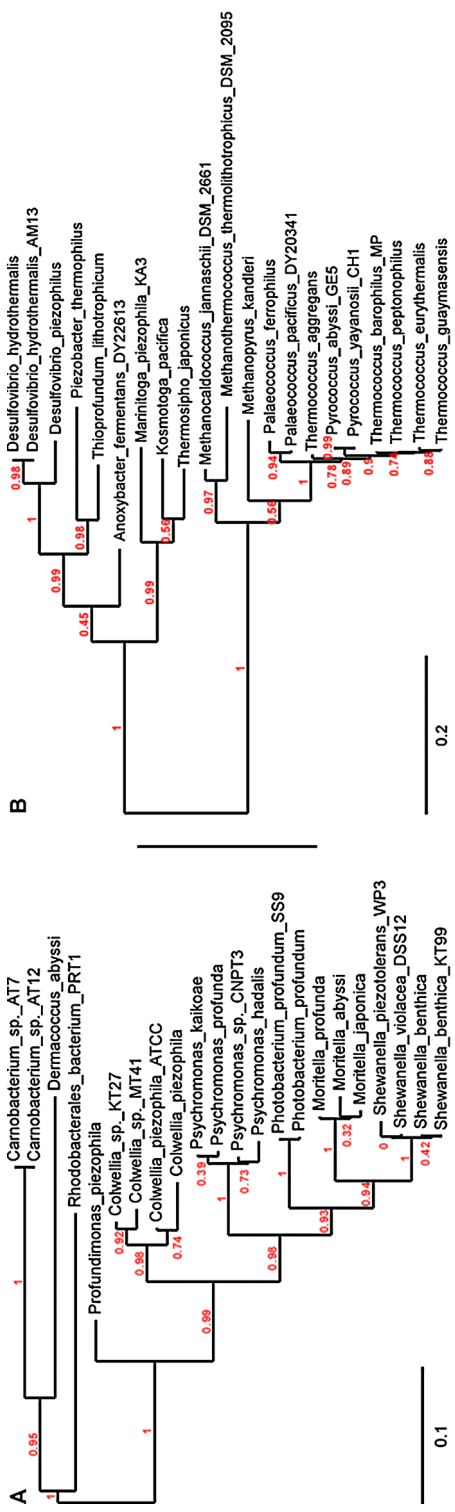


Fig. 3 Phylogenetic trees of (A) psychrophiles and (B) mesophiles and (hyper)thermophilic piezophiles. The trees were created using the SILVA database (Quast et al. 2013) and the Phylogeny.fr website with default parameters (Dereeper et al. 2008). The bar represents 0.1 or 0.2 nucleotide substitutions per position. The numbers above the branches are tree support values generated by PhyML using the aLRT statistical test

551 and able to fix dissolved carbonates into organic molecules (Jebbar et al. 2015). Discoveries
552 of abundant life in diverse high-pressure environments, including the deep oceans, hy-
553 drothermal vents, and crustal rocks, supports the existence of an adaptation of life to HHP,
554 and is consistent with the significance of HHP in the prebiotic synthesis of key biomolecules
555 and the origin of life on Earth (Hazen et al. 2002).

556

557 2.3.3 Effect of HHP on Biomolecules

558

559 Pressure affects both chemical equilibrium and reaction rates, depending upon the reaction
560 (ΔV) and activation (ΔV^\ddagger) volumes involved. The behaviour of systems under high pres-
561 sures is governed by Le Châtelier's principle, which states that the application of pressure
562 shifts equilibrium toward the state that occupies the smallest volume. It accelerates a process
563 in which the transition state has a smaller volume than that of the ground state, for example,
564 if the volume of a protein is smaller in its unfolded form, this protein will be denatured by
565 the application of HHP.

566 At HHP of greater than 400 MPa, most proteins tend to unfold (Aertsen et al. 2009).
567 Exposure to mild HHP (~ 200 MPa) often affects only the quaternary structure, leading to
568 the dissociation of oligomeric proteins. As a consequence, HHP modulates the activity of
569 enzymes. The enzymatic activities of proteins isolated from HHP-adapted organisms tend
570 to be less affected by HHP than those of surface organisms (Aertsen et al. 2009), however,
571 the true structure-function relationships underlying the pressure stability of proteins are still
572 unknown.

573

574 2.3.4 Effect of HHP on Biological Systems

575

576 Biological membranes play a fundamental role in the adaptation of microbes to their envi-
577 ronment. The membrane acts as a physical barrier to regulate influx and efflux activities,
578 it plays a central role in energy storage and processing *via* ion gradients, and it provides
579 a template for environmental sensing, multicomponent uptake and signaling pathways and
580 motility. Thus, maintaining optimal membrane biological function is crucial for any organ-
581 ism. Temperature-, pH-, salinity- or hydrostatic pressure-induced shortcomings in mem-
582 brane organization are a serious threat to the cell. Archaeal and bacterial membranes have
583 significant structural differences in spite of the fact that they perform identical functions.
584 The mechanisms used by these membranes to cope with harsh conditions and shifting en-
585 vironments are quite similar. Bacterial polar lipids, with only a few rare exceptions, are
586 based on straight chain hydrocarbons linked by ester bonds on the sn-1 and sn-2 positions
587 of glycerol. Archaeal polar lipids are composed of isoprenoid hydrocarbon chains bound
588 by ether bonds to the sn-2 and sn-3 positions of glycerol (Fig. 4). Polar headgroups consist
589 of phosphodiester-linked polar groups or sugar moieties on the sn-1 (archaea) or sn-3 (bac-
590 teria) positions of the glycerol backbone (sn-glycerol-1-phosphate, or G-1-P, structure and
591 sn-glycerol-3-phosphate, or G-3-P, structure).

592 Following the observation that the lipids in the membrane of *E. coli* cells grown un-
593 der temperatures of 43 °C and 15 °C were different (Marr and Ingraham 1962; Sinensky
594 1971), yet the corresponding membranes had similar physical characteristics at their respec-
595 tive growth temperatures, Sinensky simulated the homeoviscous adaptation basis (Sinensky
596 1974; Oger and Cario 2013). According to this approach, organisms adjust the lipid compo-
597 sition of their membrane to facilitate the preservation of the appropriate membrane fluidity
598 in order to work optimally. This concept—in a broader sense—is understood to encompass
599 adaptation to proton/water permeability and the dynamic character of plasmic membranes

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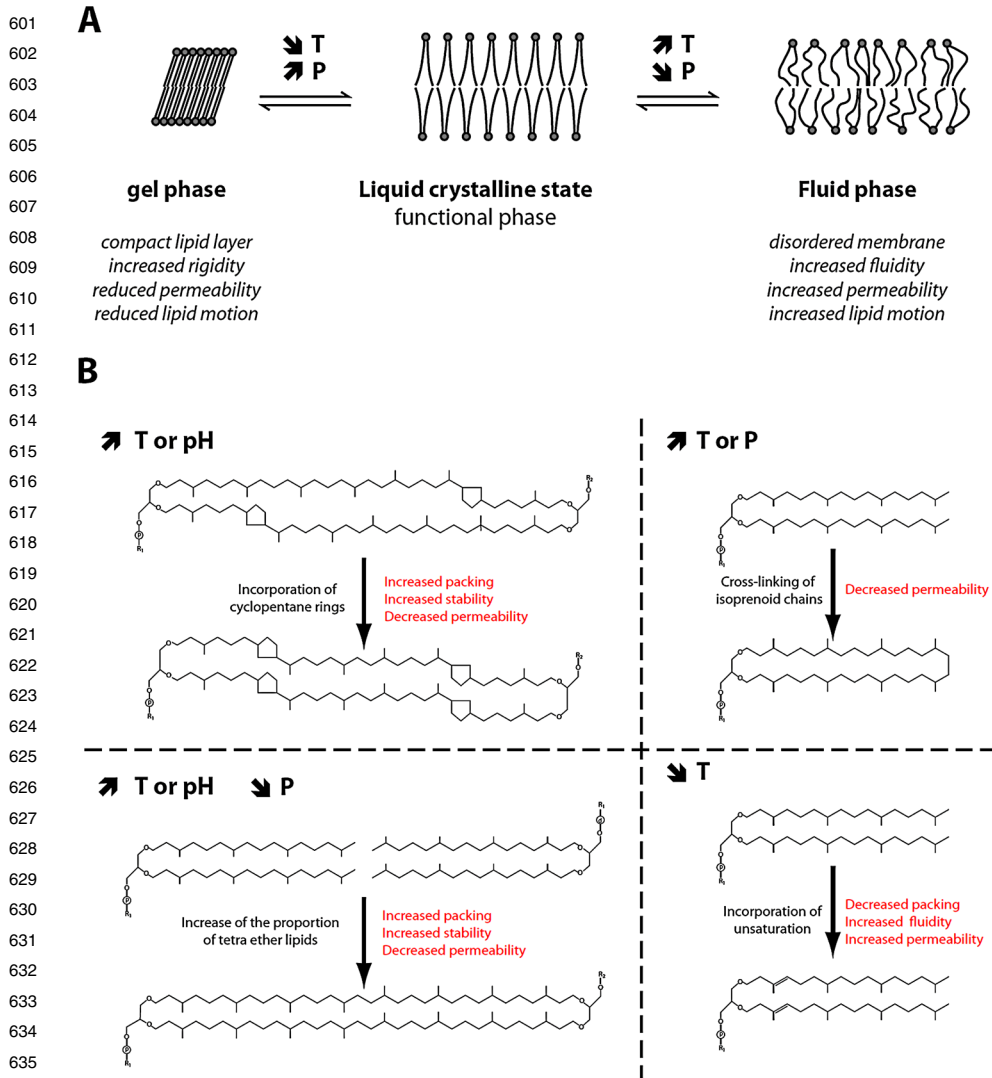


Fig. 4 Homeoviscous adaptation in Archaea. **(A)** In its functional state, the membrane is in a liquid crystalline state. Upon increasing temperature or decreasing hydrostatic pressure, lipid motion increases and the membrane enters the fluid phase. Conversely, when temperature drops or hydrostatic pressure increases, the lipid molecules pack more tightly and enter a gel phase. Membranes in both gel and fluid phases have impaired membrane function. **(B)** Known mechanisms of membrane lipid composition adaptation in Archaea (Oger and Cario 2013; Cario et al. 2015; Jebbar et al. 2015)

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(Oger and Cario 2013). Homeoviscous adjustment should also be regarded as a manner of adapting the composition, and therefore the functionality, of the membrane to abrupt shifts in the environment, or to stresses, including those of temperature, salinity, osmotic stress, pressure and pH. Under optimal physiological conditions, membranes are rather fluid and formed of disordered liquid crystalline phases. As temperature decreases or hydrostatic pressure increases, lipids in the membrane may undergo a transition from fluid to gel phase. If the

651 temperature is higher or if the pressure is lower than the optimal physiological conditions,
652 the movement rate of lipids in the membrane is greater, and this can affect the membrane's
653 stability and its inherent permeability (Fig. 4). As one might expect, the disruption of the
654 lipid phase state has a significant impact on the structure and function of the membrane (Lee
655 2003, 2004). The shift to the gel phase can lead to the aggregation of membrane proteins,
656 which are excluded *de facto* from the gel phase areas, thereby limiting the diffusion and
657 activity of proteins in the membrane and slowing down the flow of transported solutes, but
658 enhancing the permeability of cations and water.

659 The adjustment of the characteristics of the bacterial membrane is made according to
660 four main processes: (1) the change in acyl chain length, i.e., an increase in the length
661 of the two-carbon chain causes an increase in the lipid phase transition temperature from
662 10 °C to 20 °C, and decreases membrane permeability to protons and water (Winter 2002);
663 (2) the build-up of unsaturated fatty acids, since the introduction of a single unsatura-
664 tion can shift the fluid/gel transition from 10 °C to 20 °C (Russell and Nichols 1999;
665 Winter 2002); (3) the accumulation of specific polar groups such as phosphatidylcholine
666 (PC) or phosphatidylglycerol (PG) instead of phosphatidylethanolamine (PE), indeed, the
667 presence of PC as a polar head group results in a significant change in the fluid/gel transition
668 temperature (Yano et al. 1998; Winter 2002; Mangelsdorf et al. 2005; Winter and Jeworrek
669 2009), partially due to the diminished hydration and steric volume of ethanolamine com-
670 pared to choline, and partially to the capacity of PE and the failure of the PC groups to form
671 hydrogen bonds; and (4) the buildup of branched-chain fatty acids.

672 Archaeal lipid membranes usually have a considerably lower phase transition tem-
673 perature than bacterial acyl fatty ester lipids (Yamauchi et al. 1993). The adaptation of
674 the archaeal membrane to extreme environments may be attributed in part to the spec-
675 ific structure of its lipids. Although membranes consisting of fatty acyl ester lipids are
676 in the gel phase or liquid crystal phase according to their fatty acid composition, it is
677 presumed that Archaeal polar lipid membranes of archaeol and caldarchaeol are in the
678 liquid crystal phase over a wide temperature range of 0–100 °C (Stewart et al. 1990;
679 Dannenmuller et al. 2000).

680 The adaptability of the archaeal membrane is very similar in its physics to that of
681 the bacterial membrane, albeit using slightly different mechanisms to attain the same ef-
682 fects. There exist several different routes, as follows. (1) The incorporation of cyclopent-
683 ane rings along the isoprenoid chain as a function of fluctuating temperature (De Rosa et
684 al. 1980a, 1980b; Ernst et al. 1998; Uda et al. 2001, 2004) or pH (Shimada et al. 2008)
685 increases the packing efficiency of the membrane lipids (Gliozzi et al. 1983), which in-
686 creases membrane stability as a function of increasing temperature or salinity and decreas-
687 ing pressure or pH, and consequently lowers membrane permeability (Chong et al. 2012).
688 (2) The regulation of the tetraether-to-diether lipid ratio (Sprott et al. 1991; Lai et al. 2008;
689 Matsuno et al. 2009; Baumann et al. 2018; Taubner et al. under review; Taubner et al.
690 (under review)), since increasing tetraether lipids will stabilize membranes by forming
691 monolayer-type membranes or domains in the membrane, and consequently helping to
692 regulate the flux of solutes and protons across the membrane. (3) The crosslinking of
693 the two acyl-chains of the lipids yields macrocyclic archaeol or caldarchaeol derivatives
694 by a covalent bond between the isoprenoid chains, which also reduces molecular mo-
695 tion to create a more closely packed structure and increases membrane stability, creating
696 an efficient barrier against water, proton and solute leakage (Dannenmuller et al. 2000;
697 Mathai et al. 2001). (4) The increase in unsaturation along the isoprenoid chains of
698 the lipids as a function of temperature (Nichols et al. 2004) or salinity (Dawson et al.
699 ?da12); although this has, to date only been described in the psychrophilic methanogen

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<ref:da12?>

701 *Methanococcoides burtonii* (Franzmann et al. 1992; Nichols et al. 2004), unsaturated lipids
702 have been characterized in several species of hyperthermophiles (Hafenbradl et al. 1993;
703 Gonthier et al. 2001), which might indicate the occurrence of a similar adaptive strategy in
704 deep-sea hydrothermal vent organisms.

705 The adaptation of bacterial and archaeal membranes to harsh environments is clearly visi-
706 ble in the most common lipids, however, responding to variations in environmental stressors
707 might involve only a fraction of the adaptive traits mentioned above. Indeed, in order to
708 be effective, the membrane composition adaptation response needs to be very rapid. The
709 routes described require different timeframes, thus certain adaptive mechanisms will prevail
710 over others. For example, the increasing unsaturation of membrane lipids will decrease the
711 gel/fluid transition temperature to the same extent as the shortening of an acyl chain or the
712 substitution of a phosphatidylcholine by a phosphatidylethanolamine polar head, but will
713 be quicker because it is performed inside the cytoplasmic membrane on existing lipids by a
714 membrane protein (Kasai et al. 1976; Cybulski et al. 2002; Aguilar and de Mendoza 2006;
715 Beranova et al. 2008), whereas the other actions would require *de novo* lipid synthesis.
716

717 2.3.5 Adaptations to HHP in Piezophiles

718
719 DeLong and Yayanos (1985) showed that deep-sea organisms harbor an unusually high pro-
720 portion of mono- and poly-unsaturated fatty acids. This leads to highly disordered phospho-
721 lipid bilayers that are less permeable to water molecules and are proposed to maintain the
722 plasma membrane in a functional fluid state despite the rigidification effect of pressure. The
723 genes responsible for the synthesis of these unsaturated lipids have been shown to be up-
724 regulated by HHP in the moderate piezophile *Photobacterium profundum* strain SS9, and
725 are induced by HHP as part of the HHP-induced stress response in yeast (Allen et al. 1999;
726 Abe 2015). These results have led workers to propose that adaptation to HHP involves the
727 expression of HP-specific genes. This view is supported by genome-wide comparisons of
728 gene expression in piezophile and piezosensitive strains of the *Photobacterium* complex
729 (Campanaro et al. 2005).

730 In *P. profundum* SS9, transporters are mainly up-regulated at sub-optimal growth pres-
731 sure, e.g. 0.1 MPa in comparison to the pressure optimum of 28 MPa. Bartlett and colleagues
732 (Lauro et al. 2008) speculated that SS9 transporters evolved a novel protein structure to adapt
733 to elevated pressures, and that their up-regulation at 0.1 MPa could compensate for a reduc-
734 tion of functionality at lower pressures. Kasahara et al. (2009) first demonstrated a weak
735 HHP adaptation in the 3-isopropylmalate dehydrogenase of piezophilic *Shewanella* strains.
736 Thus, adaptation to HHP may result from an evolution of proteins towards an optimal ac-
737 tivity under HHP. The observation of the growth of *T. piezophilus* at 130 MPa, and that of
738 the dissociation of ribosomes in *E. coli* at ca. 30 MPa, clearly supports the necessity for
739 HHP-adapted ribosomes in the piezophilic strain.

740 Piezophilic *Shewanella* express a specific cytochrome protein complex under HHP
741 (Tamegai et al. 1997). The importance of specific piezo-adaptation in the respiratory chain is
742 further suggested by the presence of three complete sets of *cbb3* cytochrome oxidase genes
743 in the *P. profundum* SS9 genome (Vezi et al. 2005). A large-scale transposon mutagenesis
744 of *P. profundum* revealed several HHP-specific loci, most of which are involved in chromo-
745 somal partitioning and ribosomal function (Lauro et al. 2008). Therefore, adaptation to HHP
746 may require specific genes.

747 In the deepest parts of the oceans and, if present, on ocean worlds, hydrothermal vent
748 ecosystems are characterized by large fluctuations in salinity and temperature, from 0.1 to
749 twice the salinity of seawater and from fluid temperatures as high as 350 °C at the heart of the
750

vent, to 2 °C, the average temperature of the surrounding deep ocean waters. Hydrothermal vent environments in the deep sea are also subject to extremely high hydrostatic pressures up to 50 MPa, i.e., 500 times the atmospheric pressure, based on values measured at the deepest known hydrothermal vent field in the Cayman Trough of the Caribbean Sea (Dalmaso et al. 2016).

Deep sea hydrothermal vents are among the ecosystems on Earth where polyextremophilic conditions or multi-stress situations are encountered by living organisms, such as high or low temperature, high salinity, high hydrostatic pressure and nutrient starvation within the same environment. Organisms, whether eukaryotes or prokaryotes, thriving in these areas have evolved mechanisms to adapt to these harsh conditions. It is known that an increase in hydrostatic pressure affects many cell functions involving macromolecules, including growth, cell division and protein synthesis (Bartlett 2002). High salinity and high and low temperatures have in common that they may trigger a cell dehydration effect and the loss of internal water, thus compromising the ability of the cell to survive. An increase in hydrostatic pressure does not result in changes in the pressure differential across the cell membrane, whereas increased salinity may trigger an increase in osmotic pressure outside the cell that provokes a change in turgor pressure. To maintain the appropriate cell turgor and restore the cell volume, organisms accumulate low-molecular-weight osmolytes that are mainly organic solutes. These organic solutes are also accumulated by many organisms in cold and heat stresses, and possibly under high hydrostatic pressure (Martin et al. 2002; Yancey 2005). The solutes are amino acids and derivatives, polyols, sugars and derivatives, methylamines, and methylsulfonium compounds (Fig. 1). Organic osmolytes fall into several chemical categories: amino acids (glycine, alanine, proline, α -glutamate, β -glutamate, and N-acetyl- β -lysine), and derivative N-methyl-substituted amino acids (e.g., glycine betaine, homobetaine, carnitine, proline betaine, trimethylamine oxide), ectoine and hydroxyectoine, methylsulfonium solutes (dimethylsulfoniopropionate and dimethylsulfonioacetate), and small carbohydrates including monosaccharides (glucose), disaccharides (trehalose, sucrose, mannosucrose), sugar derivatives (glucosylglycerol, mannosylglycerate, glucosylglycerate), polyols (glycerol, inositol, sorbitol), and cyclitols (di-myoinositol-phosphate) (Empadinhas and da Costa 2006; Neves et al. 2005; Wood et al. 2001; Jebbar et al. 1992; Essendoubi et al. 2007; Yancey 2005; Kempf and Bremer 1998). Some solutes are widespread, for example glycine betaine, which is found in all domains of the tree of life, and carbohydrate osmolytes that occur in bacteria, archaea, fungi, algae, plants, mammalian kidneys and possibly deep-sea invertebrates. Other solutes are restricted to a small number of organisms, for example those thriving in hot environments (Empadinhas and da Costa 2006). Most organic osmolytes are neutral (either zwitterionic or lacking charges) at optimal physiological pH, although some (i.e. mannosylglycerate and di-myoinositol-phosphate in hyperthermophilic prokaryotes) are negatively charged and must be paired with potassium to achieve neutrality.

These solutes are often called “compatible solutes”, a term that refers to compounds that can accumulate at very high levels without perturbing cell metabolism or enzyme activity (Brown 1976). Many such solutes have protective properties, such as cell metabolic protection, and serve as antioxidants that scavenge free radicals and reactive oxygen species generated under stress treatments (Cushman 2001; Sunda et al. 2002; Yancey 2005). They can also stabilize macromolecular structures (proteins, membranes) only when stresses such as high salinity, high temperature, freezing and high hydrostatic pressure are present and directly destabilize cell components (Singer and Lindquist 1988; Story and Story 1996; Rudolph and Crowe 1985; Santos and da Costa 2002; Kelly and Yancey 1999).

In many bacteria and archaea, it has been demonstrated that a number of compatible solutes are accumulated by the cell in response not only to salt stress but also as a means

801 to counteract the destabilizing effects of heat and chill stresses on cell macromolecules
802 (Kuhlmann et al. 2008; Empadinhas and da Costa 2006; Holtmann and Bremer 2004). De-
803 spite this, the compatible solute counteraction of the destabilizing effect of high hydro-
804 static pressure on macromolecules is not obvious and poorly demonstrated, particularly in
805 prokaryotic cells. Yancey and coworkers have shown that the organic osmolyte trimethy-
806 lamine oxide (TMAO) occurs at high levels in many deep-sea animals in comparison to re-
807 lated shallow-water species (Gillett et al. 1997, Kelly and Yancey 1999). Since hydrostatic
808 pressure is the only physico-chemical parameter that is linear with depth, these authors sug-
809 gested that TMAO might counteract the effects of high hydrostatic pressure. In the deep-sea
810 bacterium *P. profundum* strain SS9, cells accumulate mainly glutamate and glycine betaine
811 at atmospheric pressure (0.1 MPa), whereas at optimal growth pressure (28 MPa), cells pref-
812 erentially increase intracellular concentrations of β -hydroxybutyrate and β -hydroxybutyrate
813 oligomers termed “piezolytes” for solutes that are accumulated at high hydrostatic pressures
814 (Martin et al. 2002). In addition, another study on marine bacteria has shown that adaptation
815 to high salinity synergistically enhances cell survival at high hydrostatic pressures, which
816 suggests the involvement of osmolytes in counteracting both stresses in these prokaryotes
817 (Kaye and Baross 2004).

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818 In hyperthermophilic piezophiles, it is evidenced that adaptation to HHP involves a global
819 change in the expression of genes in some metabolic pathways (amino acid biosynthesis, hy-
820 drogen metabolism), rather than the expression of a stress response *per se* (Vannier et al.
821 2015). In *Thermococcus barophilus*, for example, adaptation to HHP involves osmolyte
822 accumulation to maintain proper protein folding and activity (Cario et al. 2016). Manno-
823 sylglycerate (MG) is primarily accumulated as a compatible solute in response to salinity
824 stress, but in contrast to other Thermococcales, MG also accumulates in response to thermal
825 stresses, and its accumulation peaked in the case of combined stresses. The accumulation
826 of MG has been found to drastically increase under sub-optimal hydrostatic pressure condi-
827 tions, demonstrating that low pressures are perceived as a form of stress in this piezophile,
828 and that the proteome of *T. barophilus* is sensitive to low-pressures. MG accumulation is
829 strongly reduced under supra-optimal pressure conditions, clearly demonstrating the struc-
830 tural adaptation of this proteome to high hydrostatic pressure. There is direct and indirect
831 evidence for the structural adaptation of the proteome to HHP, although the specific signa-
832 ture of this adaptation at the genome level remains elusive.

833 This section provided an in-depth overview of the biodiversity of micro-organisms in ex-
834 treme hypersaline environments and also where high hydrostatic pressure prevails. Molec-
835 ular signatures and cellular and physiological responses to extreme salinity and high hy-
836 drostatic pressures were also examined. Among the microorganisms associated with these
837 extreme environments described above are the methanogenic archaea that have successfully
838 colonized all of the earth’s ecosystems. Methanogenesis and methanogens are described in
839 more detail in the following paragraph.

840
841

842 **3 Methanogens as Model Organisms for Icy Moon Related Cultivation:** 843 **Adaptation to Extreme Conditions** 844

845 McKay et al. (2008, 2012) determined that only three microbial ecosystems on Earth could
846 serve as analogues for a potential ecosystem on an icy moon. These ecosystems do not rely
847 on photosynthesis, on any by-product of photosynthetic metabolism, nor are they dependent
848 on O₂. One of these ecosystems is based on sulfur-reducing bacteria, and the other two are
849 based on methanogenic archaea (methanogens). In the following section, we will focus on
850

851 methanogens and their adaption to extreme conditions. A more detailed review about that
852 topic can be found in Taubner et al. (2015).

853 Besides Earth, methane (CH₄) has been detected on every planet of the Solar System, on
854 the dwarf planets Pluto, Makemake, and Eris (Formisano et al. 2004, Mumma et al. 2009,
855 Webster et al. 2015), and on the icy moons Titan (Niemann et al. 2005) and Enceladus (Waite
856 et al. 2009, 2014, 2017). Most of the CH₄ found on Earth is of biogenic origin (Liu et al.
857 2008). Methanogens are the overwhelmingly dominant producers of CH₄ as metabolic end
858 products of their carbon- and energy-yielding reactions (Thauer et al. 2008; Liu et al. 2008;
859 Taubner et al. 2015; Rittmann et al. 2015), however, some (aerobic) marine microorgan-
860 isms were also shown to produce CH₄ from methylphosphonic acid (Karl et al. 2008;
861 Metcalf et al. 2012; Carini et al. 2014). Methanogens are a phylogenetically and metaboli-
862 cally diverse group of prokaryotic organisms from the domain Archaea. Within the do-
863 main Archaea, methanogens belong exclusively to the phylum Euryarchaeota. All char-
864 acterized methanogens are known to be obligate anaerobic chemolithoheterotrophs or
865 chemolithoautotrophs. Moreover, methanogens might resemble amongst the oldest life
866 forms that emerged on Earth (Grassineau et al. 2006; Ueno et al. 2006; Martin et al. 2008),
867 but this is still under discussion (Brochier-Armanet et al. 2011; Blank 2009). Methanogens
868 are used as astrobiological study objects because of both their metabolic versatility and abil-
869 ity to withstand extreme environmental conditions (Cavicchioli 2006; Huber et al. 1989;
870 Taubner et al. 2018); they are further characterized by a variety of unusual morphological
871 and ecophysiological features. In this subsection, we review and discuss methanogens with
872 respect to temperature, pressure, pH and osmolarity, and highlight recent studies performed
873 with methanogens in an astrobiological context.

874 3.1 Adaptions to Temperature

875 Individual methanogenic strains are viable within a temperature window for growth of ap-
876 proximately 45 °C, however, the biochemical pathways of methanogenesis *per se* are not
877 restricted to a certain temperature, but are generally functional at temperatures from be-
878 low 0 °C (Cavicchioli 2006) up to 122 °C. This allows individual strains of methanogens to
879 grow from psychrophilic to hyperthermophilic growth conditions (Nakamura et al. 2006;
880 Ma et al. 2006; Lü and Lu 2012; L'Haridon et al. 2003; Jones et al. 1983a; Jiang et al. 2005;
881 Jeanthon et al. 1998; Jeanthon et al. 1999; Cheng et al. 2007; Parshina et al. 2014;
882 von Klein et al. 2002; Franzmann et al. 1997; Wagner et al. 2013; Schirmack et al. 2014;
883 Takai et al. 2008). Metabolic reactions occurring at the highest temperatures were observed
884 for *M. kandleri* strain 116 when grown at 122 °C (Takai et al. 2008).

885 Bodies of the outer Solar System, which could possibly support methanogenic life, fall
886 within the temperature range of psychrophilic methanogens. Recent results advocate the
887 possibility that hydrothermal vents might exist on icy moons, such as Enceladus (Hsu et al.
888 2015) or Europa (e.g., Zolotov and Kargel 2009), which would widen the growth temper-
889 ature range for methanogens in the subsurface water reservoirs of these bodies. However,
890 as these potential warm to hot spots at the bottom of the subsurface oceans are most likely
891 locally restricted, we will focus on psychrophilic methanogens in the following.

892 To distinguish different levels of psychrophily, psychrophilic methanogenic strains
893 were classified according to their temperature niche adaptation, which can be narrow
894 or wide (Cavicchioli 2006; Dong and Chen 2012), respectively denoted as “stenopsy-
895 chrophile” and “eurypsychrophile” organisms (Cavicchioli 2006; Siddiqui et al. 2006;
896 Feller and Gerday 2003). Stenopsychrophiles are considered true psychrophiles and are
897 only able to grow within a narrow temperature range. Compared to stenopsychrophiles, eu-
898 rypsychrophiles can tolerate a larger temperature interval, tolerate a higher mean optimum
899

Table 1 Summary of presently known psychrophilic strains and their main temperature and pH features

Strain	T [°C]			pH			Ref.
	min	opt	max	min	opt	max	
<i>Methanospirillum psychrodurum</i>	4	25	32	6.5	7	8	Zhou et al. (2014)
<i>Methanosarcina baltica</i>	3	21	28	6.3	7.2	7.5	von Klein et al. (2002)
<i>Methanosarcina lacustris</i>	1	25	35	4.5	7	8.5	Simankova et al. (2001)
<i>Methanobolus psychrophilus</i>	0	18	25	6	7–7.2	8	Zhang et al. (2008)
<i>Methanogenium marinum</i>	5	25	25	5.5	6–6.6	7.7	Chong et al. (2002)
<i>Methanogenium frigidum</i>	0	15	17	6.3	7.5–7.9	8	Franzmann et al. (1997)
<i>Methanohalobium evestigatum</i>	50	n.a.	n.a.	n.a.	7.4	n.a.	Zhilina and Zavarzin (1987)
<i>Methanogenium cariaci</i>	15	20–25	35	6	6.8–7.2	7.5	Romesser et al. (1979)
<i>Methanogenium boonei</i>	5	19.4	25.6	6.4	n.a.	7.8	Brauer et al. (?br11)
<i>Methanoculleus marisnigri</i>	15	20–25	48	6	6.2–6.6	7.6	Maestrojuán et al. (1990)
<i>Methanoculleus chikugoensis</i>	15	25	40	6.7	6.7–7.2	8	Dianou et al. (2001)
<i>Methanococcoides alaskense</i>	2.3	23.6	28.4	6.3	n.a.	7.5	Singh et al. (2005)
<i>Methanococcoides burtonii</i>	1.7	23.4	29.5	6.8	n.a.	8.2	Franzmann et al. (1992)
<i>Methanospirillum stamsii</i>	5	20–30	37	6	7.0–7.5	10	Parshina et al. (2014)
<i>Methanosarcina soligelidi</i>	0	28	54	4.8	7.8	9.9	Wagner et al. (2013)

growth temperature, and can (sometimes) be cultivated when exposed to elevated temperatures. Psychrophilic methanogens have been used in many research ventures, as they are important organisms in cold habitats on Earth (Cavicchioli 2006; Dong and Chen 2012). A list of psychrophilic methanogenic strains and their respective temperature niche can be found in Table 1.

The temperature adaptation mechanisms of methanogens were identified at different levels. At the protein level, cold adaptation mechanisms were examined in *Methanococcoides burtonii*. Here, the archaeal elongation factor 2 (EF2) proteins were found to be active at low growth temperatures but unstable at high growth temperatures (Siddiqui et al. 2002; Thomas and Cavicchioli 2000; Thomas et al. 2001). Moreover, proteins interacting with EF2 of *M. burtonii*, but also compatible solutes, are involved in activating as well as stabilizing protein machinery under low growth temperatures (Thomas et al. 2001). Another study showed that in *M. burtonii*, a putative DEAD box RNA helicase gene (*deaD*) was abundantly expressed at 4 °C (Lim et al. 2000). Additional characteristics for cold adaptation in methanogens include the increased presence of dihydrouridine in tRNAs of *M. burtonii* compared to the presence of dihydrouridine in other archaeal strains (Noon et al. ?no03?). Unlike adaptations to cold in thermophiles, *M. burtonii* did not show decreased modification of its tRNAs, but exhibited few modifications (comparable to bacteria), in particular dihydrouridine incorporation into tRNA.

A genome comparison of the psychrophilic methanogens, *M. burtonii* and *Methanogenium frigidum* was performed to identify characteristics which distinguish cold adaption mechanisms in these organisms from other archaea. Predicted and modelled proteins from *M. burtonii* and *M. frigidum* comprise a higher quantity of non-charged polar amino acids present in the solvent-accessible area of proteins. Specifically, glutamine and threonine were detected in higher abundance. Moreover, a lower content of hydrophobic amino acids, in particular leucine, were noted. Finally, two hypothetical proteins with CSD-folds and a unique winged helix DNA-binding domain protein were identified in *M. burtonii*, together with a

951 cold shock domain (CSD) protein (homologue of CspA) in *M. frigidum* (Saunders et al.
952 2003). In another study, a proteomics approach was taken to analyze the functional charac-
953 teristics of *Methanosarcina barkeri* during a low-temperature down shock response (from
954 37 °C to 15 °C) and for its low-temperature adaptation strategies at 15 °C. In a combined
955 approach using growth studies and proteomics insights into the low-temperature adapta-
956 tion capacity of *M. barkeri* could be obtained (Gunnigle et al. 2013). Astrobiologically
957 oriented experiments have been performed to examine the temperature-dependent starva-
958 tion features of selected *Methanosarcina* species including *M. solegelidi* SMA-21, finding
959 that this methanogen tolerated freezing with a survival of 98.5% in comparison to, e.g.,
960 *Methanobacterium* sp. MC-20, which exhibited only 1% survival under the same conditions
961 (Morozova and Wagner 2007). *M. soligelidi* SMA-21 showed a high survival potential at
962 4 °C and at 28 °C compared to other methanogens tested (Morozova and Wagner 2007).

963 Methanogens possess other physiological adaptation mechanisms to changes of growth
964 temperature, for instance the ability to modify cytoplasmic membrane lipids to maintain
965 membrane fluidity. A prerequisite is that the lipid membrane of organisms must be kept in
966 the liquid crystalline phase in order to stay functional, which was found in methanogens
967 over the temperature range between 0 and 100 °C (Koga 2012).

968 Membrane fluidity maintenance in psychrophilic methanogens is achieved through
969 growth temperature-mediated lipid saturation instead of the unsaturation mechanisms that
970 occur in bacteria (Nichols et al. 2004). In methanogens, lipid unsaturation is performed by
971 geranylgeranyl reductase. Notwithstanding, the cytoplasmic lipid composition in general,
972 and its unsaturation properties of methanogens in particular, are unclear indicators as to
973 whether a methanogen is adapted to a psychrophilic or a thermophilic lifestyle (Koga 2012).

974 The core lipids of *M. thermoautotrophicus* growing at its optimal growth temperature
975 of 65 °C are composed of archaeol and caldarchaeol, whereas the core membrane lipids of
976 *M. kandleri*, growing at 90 °C, are archaeol (Koga 2012). The core lipids of *Methanocaldo-*
977 *coccus villosus* and *Methanothermococcus okinawensis* are archaeol and macrocycle (with
978 minute abundances of tetraether lipids) (Baumann et al. 2018) but, upon increasing the
979 growth temperature of *Methanocaldococcus jannaschii* from 45 °C to 65 °C, the lipid mem-
980 brane composition changes from mainly archaeol to macrocycle as well as caldarchaeol
981 (Koga 2012; Sprott et al. 1991). Moreover, the presence of double bonds in isoprenoid chains
982 is not indicative of adaptation to lower growth temperatures (Koga 2012).

983 The above-mentioned results indicate that mechanisms at the genome level (e.g. the
984 expression of *deadD* at suboptimal growth temperature), at the proteome level (e.g. ac-
985 tivity of EF2), and in the lipid membrane composition distinguishes the adaptations
986 of stenopsychrophilic, eurypsychrophilic and thermophilic methanogens. Other described
987 physiological characteristics of methanogens to cope with adaptations to psychrophilic
988 cultivation conditions include the uptake of compatible solutes (Dong and Chen 2012;
989 Cavicchioli 2006, Grochowski et al. 2008). A discussion on the role of compatible solutes
990 as osmoprotective compounds is given below. <ref:gr08?>

991

992 3.2 Adaptions to Pressure

993

994 Methanogens are known to grow under low- and high-pressure conditions. The cultivation
995 of methanogens under high-pressure conditions offers an opportunity for astrobiological
996 studies, for example, the investigation of physiological responses and metabolic adaptations,
997 and for investigating the ecology of hydrothermal vent systems proposed for ocean worlds.
998 Methanogens are known to grow at more than 20 MPa of pressure (Jeanthon et al. 1998,
999 1999, 2012). The cultivation of methanogens under high-pressure conditions of up to 300

1000

kPa can be easily performed in closed batch cultivation in either serum bottles (Taubner et al. 2016) or in sophisticated cultivation devices such as bioreactors (Nishimura et al. 1992; Seifert et al. 2014). The cultivation of methanogens under low-pressure conditions and at pressures beyond 300 kPa requires special equipment (Kral et al. 2011; Kral and Altheide 2013; Park and Clark 2002; Miller et al. 1988; Boonyaratanakornkit et al. 2006; Taubner et al. 2018). Low-pressure experiments are relevant for astrobiology, to represent the lower above ground pressure present on Mars and other Solar System bodies.

With respect to growth, substrate uptake, and CH₄ production kinetics, two methanogens (KN-15 and *M. marburgensis*) examined under moderate high-pressure conditions in fed-batch or continuous culture mode in bioreactors (Nishimura et al. 1992; Seifert et al. 2013; Seifert et al. 2014) have shown that the point at which growth kinetics changed from exponential growth to linear growth (and the specific growth rate (μ)) of strain KN-15 increased with increasing pressure (Nishimura et al. 1992). The results obtained for *M. marburgensis* showed that CH₄ production is gas-limited and, although applying high-pressure conditions, the maximum physiological capacity of the organism to produce CH₄ was not reached (Seifert et al. 2014).

M. jannaschii was cultivated under gas-limited conditions and it was found that the tested strain exhibited a stress response under both high-pressure and low-pressure cultivation conditions at the transcriptional level (Boonyaratanakornkit et al. 2006). High-pressure and decompression experiments were also performed using *M. jannaschii*, employing a high-pressure bioreactor. When rapid decompression from approximately 26 MPa to atmospheric pressure was performed, the cell envelopes of *M. jannaschii* ruptured, however, when the decompression time was increased from 1 s to 5 min, the rupture of *M. jannaschii* cell envelopes decreased significantly (Park and Clark 2002). In another study, *M. jannaschii* was used to investigate growth and CH₄ production kinetics at high-pressure conditions and at different temperatures, in the presence of He or Ar in addition to H₂/CO₂. It was found that the high-temperature limit for CH₄ production kinetics of *M. jannaschii* increased with increasing pressure (Miller et al. 1988).

Additional high-pressure and high-temperature investigations using *Methanococcus thermolithotrophicus* were accomplished in 10 mL nickel tubes in series of connected autoclaves. This experimental setup was used to expose the organism of choice to temperature and high-pressure changes of 400 °C and 400 MPa over 10 min to investigate optimum pressure levels (Bernhardt et al. 1987). A pressure of 50 MPa was found to be optimal for the growth of *M. thermolithotrophicus*, whereas applying overpressure of >75 MPa resulted in increased cell lysis and in changes of morphology and in changes of growth kinetics (Bernhardt et al. 1987).

In a recent study mimicking the concentrations of gaseous and liquid inhibitors as well as high-pressure conditions on Enceladus (Taubner et al. 2018), different pressure conditions with and without gaseous inhibitors were applied to evaluate the viability of methanogens in these environments. It was shown that the methanogenic strain *M. okinawensis* produced CH₄ at pressures up to 9 MPa, but only in the presence of molecular nitrogen in the gas phase. Using a H₂/CO₂ gas phase at this high pressure results in a high CO₂ partial pressure which significantly lowers the pH of the medium. Under putative Enceladus-like conditions including potential gaseous and liquid inhibitors like carbon monoxide (CO), ethene (C₂H₄), formaldehyde (CH₂O), or methanol (CH₃OH), CH₄ production was observed up to 5 MPa. The CH₄ production kinetics did not change due to the presence of gaseous and liquid inhibitors during experiments between 300 kPa to 5 MPa (Taubner et al. 2018). A simultaneous bioreactor system (SBRS) was developed, consisting of four identical tabletop bioreactors that are suitable for performing gas conversion and gas production kinetics at

1051 pressures up to 50 bar and temperatures up to 145 °C. *M. marburgensis*, *M. palustre*, and
1052 *M. thermagregans* were successfully cultivated at 1 MPa and/or 5 MPa and differences in
1053 the CH₄ production kinetics of these organisms were detected (Pappenreiter et al. 2019).
1054 The SBRS system facilitates throughput high-pressure astrobiological research, which is of
1055 timely relevance in assessing the possibility of high-pressure habitats on outer Solar systems
1056 bodies.

1058 3.3 Adaptations to pH

1059
1060 Most of the >150 characterized methanogens grow at neutral pH values (Taubner et al.
1061 2015). Methanogens such as *M. okinawensis* (Takai et al. 2002; Taubner et al. 2018) and
1062 *M. marburgensis* (Bernacchi et al. 2014) tolerate a broader pH range down to values of
1063 3.5 and 4.5, respectively. Furthermore, there are also other methanogens known to be
1064 able to grow under acidic pH conditions (Bräuer et al. 2011; Cadillo-Quiroz et al. 2009;
1065 Ver Eecke et al. 2013). However, from an astrobiological viewpoint, at least Enceladus' sub-
1066 surface ocean is rather alkaline (Glein et al. 2015). Currently six alkaliphilic methanogens
1067 have been characterized and are available in pure culture (Table 2). The most alkaliphilic
1068 methanogens are *Methanocalculus natronophilus* (Zhilina et al. 2013), *Methanocalculus*
1069 *alkaliphilus*, and *Methanosalsum natronophilum* (Sorokin et al. 2015). *M. natronophilus*
1070 was isolated from the sediments of a collector in the vicinity of a soda lake. The strain
1071 utilizes CO₂ and H₂ or formate as an energy source and acetate as a carbon source. *M. al-*
1072 *kaliphilus* and *M. natronophilum* were enriched from hypersaline soda lake sediments at
1073 pH 10. All three alkaliphilic methanogens grow at pH between 8.2 and 10.0 and optimally
1074 around pH 9.0–9.5. *M. alkaliphilus* utilizes formate or H₂ as an electron donor and acetate
1075 as a carbon source, whereas *M. natronophilum* metabolizes methanol, methylamines, and
1076 dimethyl sulfide. Another alkaliphilic and slightly thermophilic methanogen, *Methanona-*
1077 *tronarchaeum thermophilum* was recently characterized. This methanogen comprises a new
1078 euryarchaeal class, the *Methanonatronarchaea*. This organism grows between pH values of
1079 8.2–10.2 and optimally between pH 9.5–9.7. *M. thermophilum* utilizes methanol, methy-
1080 lamines and dimethylsulfide as electron acceptors and formate or H₂ as electron donors
1081 (Sorokin et al. 2018). A list of methanogens cultivable in either acidic or alkaline conditions
1082 is shown in Table 2.

1084 3.4 Adaption to Osmolarity

1085
1086 All known methanogens depend on low intracellular salt concentrations to maintain cellu-
1087 lar integrity and the functioning of homeostatic processes. For the maintenance of cellular
1088 functions at higher extracellular concentrations of salt, some methanogens are known to
1089 accumulate compatible solutes to reduce the difference of osmotic potentials between the
1090 cytoplasm and the environment. Compatible solutes are osmoprotective molecules (Fig. 1)
1091 and do not alter the metabolic and cellular processes, even when accumulated in high con-
1092 centrations (Jones et al. 1983a, 1983b).

1093 Trimethylglycine (glycine betaine) and β -glutamate were shown to act as compatible so-
1094 lutes in methanogens, whereby the former can be assimilated by some methanogens from
1095 the growth medium (Grochowski et al. 2008; Robertson et al. 1990; Lai et al. 1991). Addi-
1096 tionally, an adenosine derivate was proposed to act as a compatible solute in *Methanobolus*
1097 *psychrophilus* R15. Furthermore, it was suggested that some of the compatible solutes de-
1098 scribed for methanogens could possibly possess cryoprotective functions (Dong and Chen
1099 2012). The main compatible solute utilized by methanogens is trimethylglycine (Robertson
1100

Table 2 Summary of currently known methanogens cultivable in either very acidic or alkaline conditions

Strain	T [°C]			pH			Ref.
	min	opt	max	min	opt	max	
<i>Methanospirillum stamsii</i>	5	20–30	37	6	7–7.5	10	Parshina et al. (2014)
<i>Methanocalculus natronophilus</i>	14	30–37	45	8	9–9.5	10.2	Zhilina et al. (2013)
<i>Methanospirillum hungatei</i>	20	37–45	50	6.5	7–9	10	Iino et al. (2013)
<i>Methanobrevibacter millerae</i>	33	36–42	43	5.5	7–8	10	Rea et al. (2007)
<i>Methanobrevibacter olleyae</i>	28	28–42	42	6	7.5	10	Rea et al. (2007)
<i>Methanotorris igneus</i>	45	88	91	5	5.7	7.5	Burggraf et al. (1990)
<i>Methanosphaerula palustris</i>	14	30	35	4.8	5.5	6.4	Cadillo-Quiroz et al. (2009)
<i>Methanoregula boonei</i>	10	35–37	40	4.5	5.1	5.5	Bräuer et al. (2011)
<i>Methanothermococcus okinawensis</i>	40	60–65	75	3.5	6–7	8.5	Takai et al. (2002), Taubner et al. (2018)
<i>Methanonatronarchaeum thermophilum</i>	30	50	60	8.2	9.5–9.7	10.2	Sorokin et al. (2018)

et al. 1990), which is used by e.g. *Methanosarcina thermophila* TM-1 (Proctor et al. 1997) and can be accumulated through an uptake system composed of a single, high-affinity H⁺- and/or Na⁺-driven transporters (Proctor et al. 1997). *M. thermophila* TM-1 can adapt to different osmolarities by synthesizing α -glutamate and N- ϵ -acetyl- β -lysine, or by accumulating trimethylglycine or K⁺. In *Methanohalophilus portucalensis* FDF1, the compatible solutes α -glutamate, β -glutamine, and N- ϵ -acetyl- β -lysine were described as osmoprotectives (Lai et al. 1991, 2000), whereas trimethylglycine was preferentially taken up from the medium as an osmoprotective compound instead of being produced *de novo* (Lai et al. 2000).

Many experiments examining the effect of osmolarity have used *Methanosarcinales*. NaCl concentrations from 0.05 to 1.0 mol L⁻¹ were used to examine the effect of osmolarity on growth kinetics and changes of morphology in *Methanosarcina* spp. (Sowers et al. 1993) and NaCl concentrations between 0.4 to 1.0 mol L⁻¹ disintegrated the methanochondroitin and sheath, which resulted in growth of *Methanosarcina* spp. as single cells. Furthermore, all tested *Methanosarcina* spp., which were encapsulated by a methanochondroitin layer, exhibited enhanced stability to <0.2 mol L⁻¹ NaCl osmolarity and grew at higher temperatures compared to the control group (Sowers et al. 1993).

An adaptation to high salt concentrations was shown with *Methanosarcina mazei* Gö1. The strain was able to tolerate up to 1 mol L⁻¹ salt through the uptake and accumulation of trimethylglycine from the growth medium. The osmoprotectant transporter A (OpuA) was involved in trimethylglycine uptake from the medium and its expression was demonstrated to be salt-induced (Roeßler et al. 2002).

Methanohalophilus spp. strains grown at different NaCl concentrations between 0.7 to 3.4 mol L⁻¹ demonstrated that the strains accumulated K⁺, however, the osmoprotective β -glutamate was detected when the strains were grown at NaCl concentrations of <1.5 mol L⁻¹ (Lai et al. 1991).

The alkaliphilic methanogens *M. natronophilus* (Zhilina et al. 2013), *M. alkaliphilus* and *M. natronophilum* (Sorokin et al. 2015), and *M. thermophilum* (Sorokin et al. 2018)

are slightly halophilic, extreme halotolerant, and extreme halophilic, respectively, i.e., they exhibit polyextremophily. Optimal growth of *M. natronophilus* requires carbonate concentrations of 0.7–0.9 mol L⁻¹ and Na⁺ at concentrations of 1.4–1.9 mol L⁻¹. *M. alkaliphilus* is characterized as a moderately salt-tolerant strain within the range from 0.2 to 1.5 mol L⁻¹ total Na⁺ in carbonate buffer at a pH of 9.5. *M. natronophilum* is highly salt-tolerant in a range from 0.5 to 3.5 mol L⁻¹ total Na⁺ growing also in carbonate buffer at a pH of 9.5. The recently described *M. thermophilum* is an extremely halophilic organism growing at total Na⁺ concentrations between 3 and 4.8 mol L⁻¹ with an optimum at 4 mol L⁻¹, and its cells lyse at a Na⁺ concentration below 2 mol L⁻¹. *M. thermophilum* accumulates K⁺ as its main compatible solute.

The above described characteristics and adaptations towards low- and high-pressure conditions, psychrophily and (hyper)thermophily, acidiphily and alkaliphily and osmolarity reveal that methanogens thrive under a variety of extreme growth conditions, but also during multi-factorial stress conditions (e.g. simultaneous multivariate concentrations of gaseous and liquid inhibitors, low pH, and pressure influences) and they respond to environmental disturbances in numerous ways (Kral et al. 2011; Taubner et al. 2015, 2018). Further, as already mentioned in Sect. 2.1.1 of this review, no other archaeal halophiles than methanogens have been identified outside the Euryarchaea. Only one recently characterized halophilic methanogen, *M. thermophilum*, uses the salt-in-cytoplasm strategy for osmoprotection and other methanogens employ the organic-osmolyte strategy to deal with osmotic stress. Hence, with respect to their potential to adapt to changing and extreme environmental conditions, methanogens are considered to be among the ideal candidates for further astrobiological studies.

4 Origins of Life and Biosignatures on Icy Worlds

4.1 Ocean World Settings for the Origins of Life

In any origins of life scenario, prebiological chemical complexification would have necessitated the confined or compartmentalized reaction of ions and molecules within an aqueous geological setting characterized by gradients and disequilibria (Russell et al. 2010). Over time, prebiotic complexification would have driven the system closer to obtaining life-like characteristics (e.g., metabolism, replication). The step(s) linking the immediate life-like precursor to the first living entity (i.e., a pioneer organism) are the most contentious (e.g. Wächtershäuser 1988; Martin and Russell 2003; Russell et al. 2010), and will not be covered in this review. A number of geological settings for the origins of life have been proposed. These settings are generally hydrothermal (submarine or subaerial hot springs) or hydrothermally influenced (the hydrothermal-sedimentary reactor hypothesis), though others are passive and rely on providing naturally reactive mineral-rich environments as “reactor flasks” for organic molecules sourced from elsewhere, for example, the pumice raft hypothesis (Brasier et al. 2011). Further possibilities, as yet incompletely explored, are the geodynamic nuclear reactor hypothesis (Ebisuzaki and Maruyama 2017) or the hydrodynamically driven volcanic-hosted splash pool hypothesis (Fox and Strasdeit 2013). A review of all of these hypotheses, culminating in a suggestion that the cycling of organic-rich fluids through the volcanogenic sediments in the vicinity of hydrothermal fields (the hydrothermal-sedimentary scenario for the origin of life), is given in Westall et al. (2018).

Despite the wide range of propositions behind the mechanics of the origins of life, most hypotheses for the geological setting of this process focus on either submarine or subaerial

1201 hydrothermal environments. Such settings are supported by both the top-down and bottom-
1202 up approaches to the origins of life. From the viewpoints of organic chemistry and geochem-
1203 istry, hydrothermal settings produce, and have the potential to concentrate within geologi-
1204 cal (mineral) edifices, the range of simple organic monomers and polymers whose gradual
1205 complexification theoretically leads to ‘protocells’ and cellular life (Russell et al. 2010;
1206 Lane and Martin 2012; Westall et al. 2018). From the biological viewpoint, estimations of
1207 the nature and metabolism of the earliest life invariably find a root in thermophilic, metal-
1208 rich settings (Nisbet and Fowler 1996; Williams and Fraústo Da Silva 2003; Gaucher et al. 2003;
1209 Gaillardet and Lane and Martin 2012). Whether life originated in the submarine or subaerial arena
1210 is a topic of contention, focused on the specific characteristics and parameters of the organic
1211 chemistry and geochemistry possible in these settings that may lead to prebiotic chemical
1212 complexification.

1213 Certainly, it is necessary that the geological environment of the origin of life was able to
1214 naturally produce or directly receive a wide range of organic molecules. Chemical complex-
1215 ification requires that this production is harnessed by a combination of gradient-driven com-
1216 partmentalized or confined milieu composed of mineral surfaces, preferably chiral (Martin
1217 and Russell 2003; Hazen and Sverjensky 2010; Dass et al. 2018). Gradients in temperature,
1218 salinity, redox state, and pH are natural disequilibrium drivers that are implicated in chemi-
1219 cal evolution, and mineral surfaces are considered equally necessary for abiogenesis, given
1220 their ability to chelate and determine the conformation of molecules concentrated at their
1221 surfaces (Hazen and Sverjensky 2010). Whether mineral phases act purely as the catalytic
1222 forces of conformation, or have morphologies at the microscale that drive the concentration
1223 of organic molecules and favor forward reaction dynamics (e.g. Parsons et al. 1998), the
1224 role of minerals in systems chemistry models for the origins of life is irrefutable. These three
1225 factors—organic molecule production, reaction concentration and substrate availability—
1226 are necessary prerequisites for an environment to be considered as a potential theatre for the
1227 origins of life (Westall et al. 2018).

1228 Further constraints, such as whether the fluid dynamics of the environment are appro-
1229 priate for long-term turbid mixing of molecule-mineral mixtures, and whether temperatures
1230 favor forward complexification or backward molecular simplification reactions are yet to
1231 be fully assessed (Westall et al. 2018). The timescales involved in the environmental and
1232 organic chemistry processes and the lifetime of the environment itself are further param-
1233 eters to be considered. In this regard, subaerial hot spring systems are less compelling than
1234 their submarine equivalents; however, hydrothermal fields of all types may endure for over
1235 several million years (Martin and Russell 2007; Westall et al. 2018; Cavalazzi et al. 2019).
1236 On the early Earth, geological longevity of subaerial environments would have been limited
1237 by periodic destruction by impactors, which were incident upon the Earth at a frequency
1238 up to hundreds or thousands of times higher than at present (Koeberl 2006; Sleep 2018;
1239 Pearce et al. 2018). Notwithstanding, the recent re-evaluation of the severity of the Late
1240 Heavy Bombardment means that planet-sterilizing impact events may have been very un-
1241 common (Zellner 2017). Recent schemes for long-term chemical evolution in subaerial
1242 hot springs (Van Kranendonk et al. 2017) would, however, be significantly limited by such
1243 temporal constraints, since environments that were not protected by an oceanic covering
1244 would have been susceptible to irrevocable alteration and destruction on potentially short-
1245 term periodic cycles. For this reason, ocean worlds deserve recognition as hosting habitable
1246 environments that may have allowed the origin and proliferation of life (Lammer et al. 2009;
1247 Barge and White 2017). Enceladus is the ideal test case, given that the proposed conditions
1248 at its ocean floor or in its plumes may be simulated in the laboratory as part of experiments
1249 with astrobiological application (Barge and White 2017; Taubner et al. 2018).

Both Enceladus and Europa are thought to have an internal structure that includes a contact zone between the ocean layer and the underlying silicate crust (Kargel et al. 2000; Chyba and Phillips 2001; McKay et al. 2018). Devolatilization of the rocky crust or mantle of these moons would conceivably lead to hydrothermal effluent generation, which could have produced local oceanic conditions conducive to habitability and prebiotic chemistry through providing an aqueous environment with an adequate energy source, the production of bio-essential elements and organic monomers and oligomers, and disequilibrium conditions in the form of temperature and pressure gradients (Kargel et al. 2000; Lammer et al. 2009). Ocean worlds are therefore potential localities for a second, possibly independent, origin of life in the Solar System. Assuming that life could have arisen on the icy moons of the outer Solar System, the question for palaeobiologists becomes one of the nature of traces of life that might be preserved and detectable. This poses a range of challenges very different to the extant life discussed thus far.

4.2 Ancient Traces of Life and Their Lessons for Biosignature Detection on Ocean Worlds

Robust evidence for life on Earth dates back to at least 3.481 Ga, based on critical studies of the stromatolites of the Dresser Formation (Walter et al. 1980; Van Kranendonk et al. 2006; Hickman-Lewis et al. 2019). The Dresser Formation stromatolites do not preserve unambiguous evidence of the microfossil architects themselves, for which further detailed studies beyond current carbon isotope work (Ueno et al. 2001, 2006) are required, but rather lamination characteristics in the organo-sedimentary structure that are demonstrated to have biological morphogenesis (Hickman-Lewis et al. 2019). Nonetheless, it is highly probable that these stromatolites are photosynthetic in origin, and are thus not of direct relevance to the habitable realms of ocean worlds, which demand chemosynthetic metabolic networks. Aside from the Dresser Formation, the oldest generally accepted fossiliferous horizons—those that have undergone and resisted some level of scientific criticism—are the 3.44 Ga Kitty’s Gap Chert (Westall et al. 2006, 2011), the ~3.43 Ga Strelley Pool Formation (Hofmann et al. 1999; Allwood et al. 2007) and the 3.42 Ga Buck Reef Chert (Tice and Lowe 2004; Tice 2009; Greco et al. 2018), although the first two examples are not without their critics (Lowe 1994; Lindsay et al. 2005; Wacey 2009). Comparably ancient Palaeoarchaeon fossiliferous material has more recently been described from the 3.47 Ga Middle Marker horizon (Hickman-Lewis et al. 2018), the 3.46 Ga stratiform Apex chert (Hickman-Lewis et al. 2016) and the 3.27 Ga Mendon Formation (Trower and Lowe 2016). With continued study at ever higher resolutions and ever more careful scrutiny, these and other examples may or may not emerge as widely accepted biosignatures. At the limit of the geological record, proposed Eoarchaeon biosignatures have been described from the >3.7 Ga Isua supracrustal belt of Greenland (Rosing 1999; Nutman et al. 2016; Hassenkam et al. 2017), and the >3.7 Ga Nuvuagittuq greenstone belt (Dodd et al. 2017) and >3.8 Ga Saglek Block (Tashiro et al. 2017) of Canada, but these are considerably more controversial. The iron-rich filament-like structures in hydrothermal deposits described by Dodd et al. (2017) have been reconsidered as volcanic glass (Wacey et al. 2018), whereas the putative stromatolites described by Nutman et al. (2016) have been found to more closely resemble metamorphosed carbonate sedimentary textures resulting from compression (van Zuilen 2018; Allwood et al. 2018). The carbon isotopes described by Rosing (1999) in Greenlandic sediments are, however, distinctly more robust, since the average value, $\delta^{13}\text{C} = -19\text{‰}$, is consistent with the simple Chloroflexus-like microbial consortia coupled with archaeal methanogens, i.e., an ecosystem dominated by the

3101 acetyl-CoA or propionyl-CoA pathway, that is suspected to dominated the primary product-
3102 tivity on Earth prior to the advent of oxygenic photosynthesis (Nisbet and Fowler 1999).

3103 In all cases, putative fossil biosignatures must meet three benchmark criteria: (i) they
3104 should possess features or signatures (morphological, structural and geochemical) consist-
3105 ent with biology; (ii) they should be syngenetic with their host rock; and (iii) the host
3106 rock should evidence a geological setting consistent with habitability. The order in which
3107 this assessment is made is of little consequence to proposing a biosignature, since failure
3108 to meet even one of the three criteria is sufficient to preclude a feature being adjudged
3109 of palaeobiological significance. Having been submitted to several billion years of processes
3110 capable of changing them beyond recognition, the identification and analysis of
3111 the most ancient biosignatures can be fraught with difficulties. On Earth, the fundamen-
3112 tal dichotomy between Eoarchaean and Palaeoarchaean-Mesoarchaean fossil-like remains
3113 is the difference in metamorphic grade between the host rocks, which can be approxi-
3114 mately summarized as “up to amphibolite facies” and “up to greenschist facies”, respec-
3115 tively. Consequently, the stratigraphic, geochronological and eventual biogeochemical con-
3116 straints able to be placed upon Eoarchaean traces of life are far less robust than those for
3117 the Palaeoarchaean (Whitehouse et al. 2019). This has resulted in a range of putative Eoar-
3118 chaean biosignatures postulated based on geochemistry and geochronology, and for which
3119 biogeochemistry is criticized subjectively, contrasted with a range of Palaeoarchaean biosig-
3120 natures evaluated by microbial palaeontology and biogeochemistry, and whose fossilifer-
3121 ous nature can be assessed objectively. In all biosignatures assessment, the first task is to
3122 determine whether the purported signature is truly of biogenic origin and not an abiotic
3123 look-alike (biomorph) or artefact. Microbial structures and constructs, both macroscopic
3124 and microscopic, often have very simple shapes that can be imitated by abiotic processes
3125 (Garcia-Ruiz et al. 2003; Westall et al. 2006; Rouillard et al. 2018): spheroidal micro-
3126 fossils may be easily confused with spheroidal mineral precipitates, such as silica, while
3127 a sheet-like concentration of abiotic organic material could, without microscopic assess-
3128 ment, superficially resemble a biofilm. Disseminated organic matter in ancient sediments,
3129 especially when significantly degraded, needs to be distinguished from abiotic organic mat-
3130 ter of hydrothermal or other origin. A noteworthy case study of controversial biogenicity
3131 is presented by the microfossil-like objects of the 3.46 Ga “Apex Chert,” Western Aus-
3132 tralia. Although initially interpreted as organisms with a cyanobacterial affinity (Schopf
3133 and Packer 1987; Schopf 1992), later studies of the same material gradually unravelled
3134 the case for their biogenicity (Brasier et al. 2002, 2005, 2006; Wacey et al. 2016).
3135 Although of superficially microfossil-like morphology (filamentous, apparently septate),
3136 high-resolution FIB-SEM work demonstrated that this morphology results from aluminous
3137 clay minerals onto which carbon had become fortuitously adhered (Brasier et al. 2015;
3138 Wacey et al. 2016). Recent isotopic studies suggesting morphotype-specific carbon isotope
3139 fractionation indicative of a mixed methanogen-methanotroph community (Schopf
3140 et al. 2017) mean that this particular controversy is ongoing. Such cases of controver-
3141 sial or mistaken biogenicity plague biosignatures of all sizes, up to and including stromatolites.
3142 A famous (albeit extreme) example thereof is the “Taylor stromatolite,” a complex laminar-domical
3143 structure closely resembling modern stromatolites but having been created by coincidence
3144 during paint spraying in the mid-Twentieth Century. Similar supposed abiological examples
3145 are known from the geological record, and especially ancient stromatolitic occurrences, such as
3146 the 3.481 Ga Dresser Formation and 3.43 Ga Strelley Pool Formation stromatolites, have
3147 been routinely subject to strong criticism (Lowe 1994; Lindsay et al. 2005) in spite of bearing
3148 many biological characteristics (Walter et al. 1980; Van Kranendonk 2007; Hickman-Lewis
3149 et al. 2016, 2019). At the time of writing, scientific consensus on these stromatolites
3150 suggests that their origin is biological.

Having established the biogenicity of the feature, the second task is to establish its syngeneticity with the host rock. Microbes may infiltrate cracks and fissures in rocks of various ages (as chasmoliths or endoliths) and can become fossilised in their endo-/chasmolithic habitats. Westall and Folk (?wf03), for example, demonstrated that organisms previously considered syngenetic within ~3.8 Ga rocks from the Isua supracrustal belt are in fact Holocene endolithic cyanobacteria. The case for syngeneticity in carbonaceous microfossils on Earth is often strengthened by Raman spectroscopy demonstrating that the carbonaceous material and its host rock have equivalent thermal histories (e.g. van Zuilen et al. ?zu07; Marshall et al. ?ma07)

The third, governing consideration in biogenicity is the environment of formation, i.e., does the purported biosignature occur in a geological context consistent microbial habitability? Most such proof in ancient successions relies on a combination of sedimentology and trace and rare earth element geochemistry (e.g., Lowe and Byerly ?lb99; Hofmann and Bolhar ?hb07) and shows that early Earth environments were strongly influenced by volcanogenic inputs and hydrothermal fluids that are manifested as silicification zones in basalts beneath chert horizons.

4.3 Fossil Microbial Biosignatures Relevant to Ocean Worlds

As highlighted in the earlier sections of this review, the diversity of microbial biosignatures of relevance to ocean worlds is vastly reduced when compared to that of Earth due to the fact that all habitable environments on ocean worlds, particularly those at the seafloor, many tens of kilometres beneath the outer icy covering, would have been polyextremophilic. Accordingly, ‘highly evolved’ Palaeoarchaeal microbial mat communities may reflect a degree of biological complexity beyond that possible on Enceladus or Europa. Proposed primitive, uncomplicated biofilm communities that may be evidenced in the Eoarchaeal fossil record may be of more relevance, reflecting hyperthermophile, non-photosynthetic autotrophic communities (Nisbet and Fowler ?nf96, 1999; Rosing 1999). Methanogens are among the proposed earliest independent lineages in the tree of life, diverging from Euryarchaeota before 3.51 Ga and perhaps as early as 3.8–4.1 Ga (Battistuzzi et al. ?ba04; Wolfe and Fournier ?wf18). Having numerous extremotolerances that make them suitable candidate organisms for ocean world biomes (Taubner et al. 2015, 2018), understanding biosignatures of methanogenic life in the fossil record may be informative for their detection on ocean worlds. The obvious caveat to this section of the review is that it would be extremely challenging to access and analyse an extinct biosphere within the crust of either Europa or Enceladus.

Most evidence for methanogenesis in the fossil record relies upon carbon isotope ratios measured by in situ secondary ion mass spectrometry, since methanogenesis is characterized by a range of $\delta^{13}\text{C}$ values mostly falling between -5 and -41% , i.e., with values slightly to significantly more negative than other major metabolic pathways—rubisco-mediated photosynthesis, sulphate reduction, photoferrotrophy—evidenced in the fossil record at the same time (Schidlowski ?sc88; Vieth and Wilkes ?vw09). Their more negative carbon isotope fractionations can thus be used to indicate the presence of both Bacteria and Archaea in fossilized biomass (Hayes ?ha94; Nisbet and Fowler 1999). The carbon isotope record therefore provides independent support for the molecular clock estimations of methanogenesis as early as 3.8 Ga by virtue of highly ^{13}C depleted carbonaceous material in Greenlandic rocks (Grassineau et al. ?gr05). Extreme depletions of up to -60% in carbonaceous material from Palaeoarchaeal horizons have been used as implicit evidence for coupled methanogenesis and methanotrophy in widespread microbial ecosystems (Hayes ?ha94; Schopf et al. ?sc17; Lepot et al. ?le19).

1401 For ocean worlds, traces of planktonic life might also be among the key biosignatures
1402 for a fossil biosphere. Prior to the advent of oxygenic photosynthesis, planktonic life seems
1403 essentially limited to the hypothesised modes of life of large spheroidal and lenticular mi-
1404 crofossils described from numerous horizons in the East Pilbara terrane and the Barber-
1405 ton greenstone belt, although this is likely a function of preservational potential. Indeed,
1406 planktonic life away from hydrothermal vents is considered to have had the opportunity
1407 to proliferate once organisms had adapted to oligotrophy (e.g., Nisbet and Fowler 1999;
1408 Brasier et al. 2006). In Archaean metasediments, spheroidal and lenticular microfossils
1409 up to several hundred microns in size and with interpreted robust cellular morphologies
1410 (thickened, spore-like cell walls) span more than 400 Ma of geological history, from the
1411 3.4 Ga Strelley Pool Formation (Sugitani et al. 2015) and Kromberg Formation (Walsh 1992;
1412 Oehler et al. 2017) to the 3.0 Ga Farrell Quartzite (Sugitani et al. 2007). These microfossils
1413 are characterised by strongly negative carbon isotope fractionation ($\delta^{13}\text{C} = -30$ to -45%),
1414 consistent with biological origin and sufficiently restricted in range as to preclude origin in
1415 abiogenic chemical reactions such as the Fischer–Tropsch type processes (House et al. 2013;
1416 Oehler et al. 2017). The highly negative depletions may also be consistent with methane cy-
1417 cling, but this has yet to be unambiguously demonstrated (Oehler et al. 2017). Particularly
1418 enigmatic amongst these microfossil-like objects are the lenticular microfossils. The near-
1419 equant morphologies of lenticular microfossils, together with the flange-like appendages
1420 that characterise their equatorial regions, have been used as specific evidence for their hav-
1421 ing a planktonic stage in their life cycle (Sugitani et al. 2007, 2015; Oehler et al. 2017).
1422 Fluid dynamic modelling of virtual flanged cells has demonstrated both that the presence
1423 of the flange reduces sedimentation velocity and enlarges cell volume, two factors increas-
1424 ing their propensity for suspension and dispersion as part of a planktonic lifestyle (Kozawa
1425 et al. 2018). Dispersion may further be inferred from the widespread distribution of these
1426 fossils in space, i.e. across two Archaean landmasses (the Pilbara and Kaapvaal regions). Al-
1427 though many of these microfossils are solitary occurrences, some pairs, clusters and chains
1428 of lenticular objects have been described, particularly in the examples from Western Aus-
1429 tralia, strongly increasing the case for their biogenicity (see Sugitani 2018).

1430 In contrast to coccoidal and filamentous microfossils from the same formations (Walsh
1431 1992; Walsh and Lowe 1999; Westall et al. 2001), lenticular and large spheroidal microfos-
1432 sils typically show no strict association with stromatolitic or mat-like laminations, which
1433 imply that they are not involved in the mat-building process. Although this can be seen as
1434 implicit support for a planktonic lifestyle, instances of lenticular carbonaceous objects from
1435 the Middle Marker horizon do indeed occur within microbial mats (Hickman-Lewis et al.
1436 2018). While this does not argue against their biogenic origin, their evident simultaneous
1437 formation with microbial mats in this unique case warrants further investigation.

1438 The thick cell walls that characterize these organisms have been argued to be beneficial
1439 to open ocean modes of life. Oehler et al. (2017) interpreted that such thick walls may
1440 have enabled the cells to withstand high levels of UV radiation, metal toxicity, or sudden
1441 evapotranspiration events and associated salt stress that may have characterised early Earth
1442 habitats (see Lowe et al. 2014; Lowe and Byerly 2015). The potential for dispersion and
1443 longevity may also have permitted robust, lenticular cell-like objects to withstand local-
1444 scale environmental stresses inherent to ocean worlds in ways that more fragile organisms
1445 with thinner cell walls could not. The application of the microfossil record to ocean worlds
1446 remains very much an open topic; indeed, limited, if any, discourse on the subject had been
1447 attempted before this review.

1448 The true challenge of a correlative microscopy approach in palaeobiology applied to the
1449 putative biomes of an oceanic celestial body such as Europa and Enceladus lies in the dif-
1450

<ref:br06?>

1451 faculty inherent in accessing samples. At present, no mission objectives involve the assess-
1452 ment of the crust of an ocean world due to the near-insurmountable challenge of reaching
1453 the required localities. This section of the review may therefore be little more than intel-
1454 lectual discourse. Nonetheless, one can state that deducing the geology and geochemistry
1455 of putative hydrothermal vent deposits on Europa or Enceladus would open up the possi-
1456 bility to appraise the habitable niches of ocean worlds and consider the likelihood of a
1457 fossilised biosphere of purely chemotrophic life. Such a biosphere may be an excellent—
1458 and indeed a truly pristine—analogue for the most primitive (hyper)thermophile biospheres
1459 on the Hadean-Eoarchean Earth.

1461 5 Conclusions

1463 Life, especially in the form of microorganisms, has achieved colonization of almost all ar-
1464 eas on Earth, even the most hostile and extreme parts of the planet. Organisms have adapted
1465 by tailoring their cellular constituents to operate also at the boundaries and limits of life.
1466 Microorganisms have been successful in diversifying their metabolisms and taking benefit
1467 of the resources available in environments which might be low in the levels of nutrients
1468 or extreme in its physical conditions. In these harsh environments they manage to generate
1469 enough energy to ensure a minimum of maintenance of cellular constituents and even to
1470 proceed reproduction. This metabolic adaptation and diversity of microorganisms has been
1471 illustrated by the ability of microorganisms to produce energy from different types of sub-
1472 strates, to produce different types of molecules, such as those that make membranes more
1473 robust or those that are characteristic in the response to different stresses whose harmful
1474 effects normally cause the denaturation of most cellular components and finally leads to
1475 cellular death.

1476 The extent of microbial diversity on Earth is far from being fully elucidated, particularly
1477 in remote and extreme environments such as deep sea and subsurface sediments. Organ-
1478 isms living in extreme environments and in particular microorganisms have, over the evo-
1479 lutionary process, developed a large variety of adaptive strategies. As a result, they present
1480 a repertoire of original metabolic pathways and biomolecules that allow them not only to
1481 survive in extreme conditions, but often to grow in an optimal way in extreme ecological
1482 niches. Metabolic markers such as membrane lipids (saturated and polyunsaturated fatty
1483 acids, archaeol and caldarchaeol, etc.), compatible solutes (amino acids and derivatives,
1484 sugars and derivatives, polyols) or gas production (e. g. methane), witnesses of biological
1485 activity, have been detected in increasingly improbable environments previously considered
1486 sterile: thermal springs, hydrothermal vents, acidic lakes, alkaline lakes, hypersalines, deep
1487 marine sediments, oil reservoirs, glaciers, etc. The physico-chemical and energetic charac-
1488 teristics of some extreme terrestrial environments are analogous to those of other planets and
1489 icy moons in the Solar System, which raises the question of the past or present existence of
1490 life on these planets and icy moons, or the fulfilment of all the conditions for another origin
1491 of life. Biomolecules or biosignatures such as those listed in this chapter can be traced to
1492 detect early clues to potential extraterrestrial biological activity.

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