
Nanopolystyrene beads affect motility and reproductive success of oyster spermatozoa (*Crassostrea gigas*)

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Abstract :

Oysters are keystone species that use external fertilization as a sexual mode. The gametes are planktonic and face a wide range of stressors, including plastic litter. Nanoplastics are of increasing concern because their size allows pronounced interactions with biological membranes, making them a potential hazard to marine life. In the present study, oyster spermatozoa were exposed for 1 h to various doses (from 0.1 to 25 $\mu\text{g mL}^{-1}$) of 50-nm polystyrene beads with amine (50-NH₂ beads) or carboxyl (50-COOH beads) functions. Microscopy revealed adhesion of particles to the spermatozoa membranes, but no translocation of either particle type into cells. Nevertheless, the 50-NH₂ beads at 10 $\mu\text{g mL}^{-1}$ induced a high spermotoxicity, characterized by a decrease in the percentage of motile spermatozoa (–79%) and in the velocity (–62%) compared to control spermatozoa, with an overall drop in embryogenesis success (–59%). This major reproduction failure could be linked to a homeostasis disruption in exposed spermatozoa. The 50-COOH beads hampered spermatozoa motility only when administered at 25 $\mu\text{g mL}^{-1}$ and caused a decrease in the percentage of motile spermatozoa (–66%) and in the velocity (–38%), but did not affect embryogenesis success. Microscopy analyses indicated these effects were probably due to physical blockages by microscale aggregates formed by the 50-COOH beads in seawater. This toxicological study emphasizes that oyster spermatozoa are a useful and sensitive model for (i) deciphering the fine interactions underpinning nanoplastic toxicity and (ii) evaluating adverse effects of plastic nanoparticles on marine biota while waiting for their concentration to be known in the environment.

Keywords : Nanoplastics, oyster, spermatozoa, motility, reproductive success

Introduction

In 2017, 348 Mt of plastics were produced by human activities, a substantial increase from the 2 Mt reported in 1950 (PlasticsEurope 2018). This high consumption and the low efficiency of waste management have now led to a huge release of mismanaged items into the ultimate recipient: the oceans (Eriksen *et al.* 2014, Jambeck *et al.* 2015). Plastic release from coastal regions was estimated between 4.8 and 12.7 Mt in 2010, and projections forecast an increase of one order of magnitude within the next decade if mitigation measures are not developed (Jambeck *et al.* 2015). Plastic debris are pervasive in the marine environment due to their strong resistance to UV-light, temperature, mechanical degradation (*e.g.* wave action) and micro-organisms (GESAMP 2015). Their ubiquitous nature in aquatic environments raises significant and critical questions regarding whether plastic contamination is a new planetary boundary threat as well as an unsafe condition for human health (Jahnke *et al.* 2017, Wright and Kelly 2017).

Global inventories describing the state of this contamination have spotlighted a high amount of small debris termed microplastics (MP; < 5 mm). Microplastics (330 μm –5 mm) constitute 92% (485×10^{10} particles) of the total plastic items estimated at the surface layer of open oceans (Eriksen *et al.* 2014). However, recent assessments of the abundance of small MP (10 – 330 μm) have highlighted the underestimated state of the contamination. More than 90% of plastic debris exhibited a size smaller than 100 μm in surface water and sediments from the southern North Sea (Lorenz *et al.* 2019), the Arctic deep-sediments (Bergmann *et al.* 2017) and sea ice (Peeken *et al.* 2018). This is related to the release of small manufactured (primary) MP (*e.g.*, exfoliates, scrubbers) and the uninterrupted weathering of plastic pieces (secondary MP) in seawater down to nanoscale particles; the latter are defined as nanoplastics (NP) for which size limits remain unclear (< 1 μm or < 100 nm) (Koelmans *et al.* 2015, Gigault *et al.* 2018, Hartmann *et al.* 2019). Recently, Hartmann *et al.* (2019) proposed a compromise, suggesting the use of the term NP for particles smaller than 100 nm and sub-MP for particles between 100 nm and 1 μm . In addition, primary NP are suspected to enter directly in the environment *via* wastes of industrial and domestic activities (Zhang *et al.* 2012, Hernandez *et al.* 2017). Higher amounts of NP than MP are expected, but the current lack of analytical tools and methods prevents the estimation of their environmental concentrations in the ocean (Huvet *et al.* 2016, Mintenig *et al.* 2018, Schwaferts *et al.* 2019).

The over-representation of small plastic debris in oceans emphasizes the need for an assessment of their threat to biota and ecosystems (Galgani *et al.* 2013). Among particle characteristics, the size is suspected to play a major role in potential toxicity for marine life. Harmful effects were recently reported to be higher for NP (<100 nm) than for MP in terms of survival, lifespan, fecundity, fertilization, and embryo-larval development in aquatic organisms based on studies using model spherical particles of polystyrene (Lee *et al.* 2013, Jeong *et al.* 2016, 2017, Tallec *et al.* 2018). This difference in effects is likely related to nanoparticle properties, and particularly their large surface-to-volume ratio in comparison with bulk material, which promotes high reactivity with biological systems and leads to various effects (cytotoxic, genotoxic, reprotoxic) (Mattsson *et al.* 2015, Paul-Pont *et al.* 2018). Moreover, NP have the potential to translocate into tissues/organs as observed in scallop, which could increase their toxicity and bioavailability for higher trophic levels (Al-Sid-Cheikh *et al.* 2018).

Free-spawning invertebrates are numerous in aquatic environments (Lewis and Ford 2012). At the start of the life cycle, early life stages (ELS) are crucial because they participate actively in the population recruitment, the community structure, and consequently to the ecosystem balance (Lewis and Ford 2012, Reinhardt *et al.* 2015). Nevertheless, ELS have to face all anthropogenic contaminants, and the current consensus is that ELS are more sensitive than adults (Hutchinson *et al.* 1998, Gallo and Tosti 2019). In this regard, toxic effects of pesticides, pharmaceutical residuals, hydrocarbons, and nanoparticles on ELS of key marine invertebrates (mainly sea urchins, oysters and mussels) were identified (Ringwood *et al.* 2010, Kadar *et al.* 2011, Di Poi *et al.* 2014, Balbi *et al.* 2017, Vignier *et al.* 2017, Gallo *et al.* 2018, Mai *et al.* 2018, Tallec *et al.* 2018).

Among these free-spawning marine species, the Pacific oyster (*Crassostrea gigas*) is a keystone species in coastal ecosystems due to its ecological roles (*e.g.*, biogeochemical cycles, water quality, and reef building) and economic value for aquaculture (Grabowski *et al.* 2012, Bayne 2017). Our previous study demonstrated a hampered oyster fertilization following a simultaneous exposition of spermatozoa and oocytes to 50 nm nanopolystyrene beads displaying amine (50-NH₂) or carboxylic (50-COOH) functionalizations. The response showed a dose-dependent pattern (0–25 µg.mL⁻¹), but the implied toxic pathways were not identified (Tallec *et al.* 2018). Nevertheless, we suggested various explanations for the observed toxicity, including perturbation of antioxidant defenses, disruption of spermatozoa motility, and membrane destabilization. Moreover, in this previous work, we could not discriminate effect on spermatozoa and oocyte. The aim of the present study was therefore to explore detrimental

processes specifically in oyster spermatozoa following exposures to the 50-NH₂ and 50-COOH beads. The effects on sperm motility, reactive oxygen species (ROS) production, survival, membrane integrity/constitution and reproductive success (embryogenesis success) were investigated.

Materials and methods

Particle characterization

Spermatozoa were exposed to two types of commercial nanopolystyrene beads (nano-PS; 50 nm; no fluorescent labelling; Bangs Laboratories[®]/Polysciences): (i) amino-polystyrene (50-NH₂); and (ii) carboxylic-polystyrene (50-COOH). In parallel, the same nano-PS beads, but with a green (excitation 480 nm and emission 520 nm; 50-NH₂) or an orange (excitation 552 nm and emission 580 nm; 50-COOH) fluorescent label were used for confocal microscopy observations. The toxicity experiments were performed with non-fluorescent beads to avoid a confounding effect due to a possible leaching of the fluorophores (Catarino *et al.* 2019, Schür *et al.* 2019). As described in Tallec *et al.* (2018), commercial suspensions were stocked in ultrapure water (UW) with surfactant (Tween-20[®]; <0.1 %) at 4°C prior to the experiments. The Z-average (size; nm) and mean surface charge (ζ -potential; mV) of these particles were previously characterized by dynamic light scattering (DLS; nano-Zetasizer ZS; Malvern Instruments; UK) in UW and 1- μ m filtered seawater (FSW; pH 8.1 and 34 PSU) (Tallec *et al.* 2018). Briefly, the 50-NH₂ suspension exhibited a positive surface charge (UW: 44.0 ± 1.5 mV; FSW: 15.6 ± 2.7 mV) and a nanometer-scale size (UW: 53.3 ± 2.3 nm; FSW: 96.5 ± 2.0 nm) while the 50-COOH suspension had a negative surface charge (UW: -62.1 ± 0.4 ; FSW: -13.8 ± 0.8) and formed micrometer-scale aggregates in FSW (3735.0 ± 443.8 nm) in comparison to UW where no aggregation occurred (55.9 ± 0.4 nm). In addition, for the present study, the particles were placed in the experimental medium (EM) without spermatozoa (removed by a filtration on 0.2 μ m) to test potential effects of seminal fluid components on particle behavior (charge and aggregation state), as described in González-Fernández *et al.* (2018).

Suspensions of nano-PS for exposures

Stock suspensions of 50-NH₂ (100 mg.mL⁻¹) and 50-COOH (25 mg.mL⁻¹) beads were prepared at 1,000 μ g.mL⁻¹ in UW and then diluted to 25, 10, 1, and 0.1 μ g.mL⁻¹ (corresponding to 3.6×10^{11} , 1.5×10^{11} particles.mL⁻¹, 1.5×10^{10} and 1.5×10^9 particles.mL⁻¹, respectively) in EM (21°C) to obtain 9 treatments (2 particle types \times 4 concentrations + 1 control). The use of dose-

response experiments is recommended to assess harmful effects of small plastic debris in order to determine toxicity thresholds owing to the uncertainties about their current and future environmental concentrations (Koelmans *et al.* 2017, Paul-Pont *et al.* 2018). Confounding effects due to the presence of a surfactant (Tween-20, in our case) in the commercial nano-PS suspensions (Pikuda *et al.* 2018) were avoided by adding Tween-20 (Sigma-Aldrich®) to the control treatment at an equivalent final concentration to that in the 25 $\mu\text{g}\cdot\text{mL}^{-1}$ beads treatment (0.0001%).

Gamete collection and exposures

Mature oysters (18 months old) produced in 2015 according to Petton *et al.* (2015) were collected in the Marennes-Oléron basin (France) and transferred to the Ifremer experimental facilities (French Research Institute for Exploitation of the Sea) in June 2017. These oysters were acclimatized at 16°C in a 350 L raceway and fed *ad libitum* on a balanced mixture of two microalgae *Isochrysis* sp. (Tahitian clone: T-iso; CCAP927/14) and *Chaetoceros gracilis* (UTEX LB2658) until experiments. Oyster sex was determined under microscope (Olympus BX51; $\times 10$ -20 magnification) on a sub-sample of gonads (50 μL). The sperm were collected by stripping the gonads of male oysters and sieving the resulting mixture at 100- μm to eliminate debris (Steele and Mulcahy 1999). The mixture was diluted in 100 mL of FSW at 21°C for 1 h to start capacitation (*i.e.*, the final step of maturation occurring in seawater to trigger hyperactivated motility). Spermatozoa concentrations were calculated by flow cytometry (EasyCyte Plus cytometer; Guava Merck Millipore, USA) (Le Goïc *et al.* 2013) and then set at 10^7 cells. mL^{-1} in tubes filled with the different nano-PS treatments (0.1, 1, 10 and 25 $\mu\text{g}\cdot\text{mL}^{-1}$). Exposures lasted 1 h and were performed on six (for all measures except for lipid analyses; final volume = 3.5 mL) and four (for lipid analyses; final volume = 100 mL) replicates for each treatment; each replicate corresponding to a pool of spermatozoa from 3 males (total number of male oysters = 18).

Sperm motility and velocity

After the 1 h exposure, 25 μL of each treatment and replicate were diluted in 475 μL of FSW containing pluronic acid (1 $\text{g}\cdot\text{L}^{-1}$) (Boulais *et al.*, 2019). After 10 minutes, 12 μL of each suspension were placed in FastRead cells (Fischer Scientific®, USA), and the sperm were observed by light microscopy (Olympus BX51, Japan; $\times 20$ magnification with phase contrast) to acquire videos (Camera Qicam Fast 1394, 60 frames. sec^{-1} , 6 sec.treatment $^{-1}$). Five videos were acquired for each replicate. Videos were analyzed by the ImageJ Software using the

Computer-Assisted Sperm Analyzer (CASA) plug-in to characterize the percentage of motile spermatozoa (%) and the Velocity of the Average Path (VAP; $\mu\text{m}\cdot\text{sec}^{-1}$) (Wilson-Leedy and Ingermann 2007). The calibration settings developed for oyster spermatozoa are described in Boulais *et al.* (2015).

Flow Cytometry

All flow cytometry measurements were performed at the end of the exposures with an EasyCyte Plus cytometer (Guava Merck Millipore, USA), except for the acrosomal integrity, which was characterized using a FACSverse cytometer (BD Biosciences, USA). The two cytometers were equipped with a 488 nm argon laser; two fluorescence detectors were used with the EasyCyte Plus (green 525/30 nm and red 680/30 nm) and one for the FACSverse (Per-CP-Cy5.5-A 700/54 nm). The relative cell size and complexity were estimated based on the forward scatter (FSC) and side scatter (SSC) values (U.A.) (Le Goïc *et al.* 2013).

The spermatozoa membrane integrity was assessed with the “Live/Dead Sperm Viability kit” (Molecular Probes[®], USA) according to Le Goïc *et al.* (2013). Briefly, 10^6 spermatozoa were sampled and stained with SYBR-14 (final concentration 1 μM) and propidium iodide (PI; final concentration 10 $\mu\text{g}\cdot\text{mL}^{-1}$) for 10 minutes in dark condition at room temperature. This dual-staining allows a measurement of the percentage of live and dead spermatozoa; the SYBR-14 penetrates only cells without membrane damage (live spermatozoa; green detector), while the PI only enters into cells that have lost membrane integrity (dead spermatozoa; red detector) (Garner and Johnson 1995). Results were expressed as the percentage of live spermatozoa (%).

Intracellular reactive oxygen species (ROS) production was determined by incubating 10^6 spermatozoa with 2',7'-dichlorofluorescein diacetate (DCFH-DA; final concentration 10 μM ; Sigma-Aldrich[®]) for 50 minutes in dark condition at room temperature. DCFH-DA is hydrolyzed in the cells by esterases to form the DCFH, which emits a green fluorescence after oxidation by ROS to DCF (Lambert *et al.* 2003). The relative amount of ROS in cells was expressed as the mean level of green fluorescence (A.U.; green detector).

A posteriori, the acrosomal integrity was verified using the protocol of Volety *et al.* (2016) in control samples and in treatments that affected the spermatozoa's reproductive success. A sample of 2×10^5 spermatozoa was incubated with the LysoTracker Red DND-99 (Molecular Probes[®], USA; final concentration 1 mM) for 10 minutes in dark condition at room temperature. This probe specifically stains the acrosome, which is an acidified organelle (Volety *et al.* 2016).

A loss of acrosomal integrity is characterized by a decrease in the mean level of red fluorescence (A.U.; Per-CP-Cy5.5-A detector).

Lipid analyses

Lipid analyses were performed only on spermatozoa from the control and treatments that affected the spermatozoa motility. Samples (2.5×10^9 spermatozoa.treatment⁻¹; n = 4) were collected on GF/F glass-fiber filters (0.2 μ m; Whatman[®]) and then extracted in 6 mL chloroform: methanol (2:1 v/v), according to a procedure modified from Folch *et al.* (1957) (Da Costa *et al.* 2016). Samples were flushed with nitrogen gas and stored at -20°C until analyses. All materials were rinsed with acetone and/or burnt for 6h at 450°C before use to prevent contamination; all chemicals were HPLC grade.

Analyses of lipid classes

Lipid classes were separated by high-performance thin layer chromatography (HPTLC). Two HPTLC glass plates coated with silica (200 × 100 mm; Merck[®]60, Germany) were used: one to separate neutral lipids (NL) and the other to separate polar lipids (PL). The HPTLC plates were first cleaned with a mixture of methyl acetate:isopropanol:chloroform:methanol:KCl 0.25% (10:10:10:4:3.6; v/v) for PL and with hexane:diethyl ether (97:3; v/v) for NL according to Da Costa *et al.* (2016). Plates were activated at 120°C for 30 min (TLC Plate Heater, CAMAG[®]; Switzerland) and lipid extracts (2 μ L for PL and 5 μ L for NL) were spotted on them with an automatic TLC sampler ATS4 (CAMAG[®]). The PL were separated in a solution of methyl acetate:isopropanol:chloroform:methanol:KCl 0.25% (10:10:10:4:3.6; v/v). The NL were separated by 2 successive developments, first using a mixture of hexane: diethyl ether:acetic acid (20:5:0.5; v/v), followed by a solution of hexane:diethyl ether (97:3; v/v). Lipid classes were revealed by immersion of the plates in a solution containing 3% CuSO₄ and 8% H₃PO₄ (w/v in distilled water) and heated at 180°C (30 min). Data was acquired using a scanner densitometer at 370 nm (TLC Scanner 4 CAMAG[®]). The results were analyzed with the WinCATS software (v2.4; CAMAG[®]) by comparing the retention time and intensity of bands related to each class against known standards (Moutel *et al.* 2016). The lipid classes were expressed as the mass percentage of each class to the total lipid content which was expressed in mg per 10⁹ spermatozoa. Nine classes were identified: 3 NL classes (triglycerides, sterols and free fatty acids) and 6 PL classes (phosphatidylcholine [PC], phosphatidylethanolamine [PE], phosphatidylinositol + ceramide amino-ethylphosphonate [PI+CAEP], phosphatidylserine [PSer] and cardiolipin [CL]). Triglycerides correspond to storage lipids and free fatty acids is considered as a proxy of lipid degradation. Sterols, PE, PSer, PI+CAEP, and

CL constitute membrane lipids. The PC/PE ratio was used as a proxy for the membrane stability (Li *et al.* 2006).

Analyses of fatty acids (FA)

To determine FA composition of neutral and polar lipid fractions, 1 mL of lipid extracts was evaporated to dryness under N_{2(g)} and then resuspended in 3 × 500 µL of chloroform:methanol (98:2; v/v). Samples were deposited at the top of a micro-column (40 × 5 mm) using silica gel 60 previously heated for 6 h at 450°C and moistened with 6% deionized water (Le Grand *et al.* 2014). Thereafter, PL and NL fractions were eluted and separated by passing 10 mL of methanol:chloroform (98:2; v/v) and 20 mL of methanol, respectively, through the column. An internal standard (free fatty acid 23:0, 2.3 µg) was added at this point to quantify fatty acids in samples (Da Costa *et al.* 2016). The eluted fractions were transesterified to obtain fatty acid methyl esters (FAME): each fraction was evaporated to dryness under N_{2(g)}, resuspended in 800 µL of MeOH-H₂SO₄ (3.4% v/v), and heated at 100°C for 10 min. Hexane (0.8 mL) and distilled-water saturated with hexane (1.5 mL) were added, and the aqueous phase was removed after centrifugation (738 × g; 60 sec; 20°C). The FAME were rinsed twice with distilled-water saturated with hexane (1.5 mL). Finally, FAME were analyzed to determine the FA composition of each fraction with a Varian CP8400 gas chromatograph (HP, USA) (parameters are described in Mathieu-Resuge *et al.* (2019)). The FA composition was expressed as the mass percentage of each FA to the FA content per fraction (neutral and polar).

Microscopy analyses

Transmission electron microscopy

Transmission electron microscopy (TEM; JEOL 100 CXII, Japan) observations were performed to assess the potential fine interactions between nano-PS beads and spermatozoa, including translocation, as well as the integrity of the internal structures (nucleus, mitochondria, acrosome, and flagellum) of oyster spermatozoa according to Suquet *et al.* (2010). An aliquot of spermatozoa from the control treatment and from the highest exposed concentration (25 µg.mL⁻¹) for both 50-NH₂ and 50-COOH beads were immersed in successive buffers: (1) 6% glutaraldehyde:7% NaCl:0.4 M cacodylate (2:1:1; v/v) for 2 h (2) 0.4 M cacodylate:8% NaCl:UW (1:1:2; v/v) with three successive 15 min baths (3) 4% osmium tetroxide:0.4M cacodylate:10% NaCl: UW (1:1:1:1; v/v) for 1h. The samples were centrifugated (100 × g, 10 min, 4°C) between each step to remove the buffers. The samples were then embedded in 1% agar and dehydrated in successive mixtures: (1) 50% EtOH (15 min); (2) 70% EtOH (30 min);

(3) 90% EtOH (3 baths of 10 min); (4) 100% EtOH (3 baths of 10 min); (5) propylene oxide 99% (3 baths of 10 min). Ultimately, the samples were embedded in hard resin (Embed 812 kit; Electron Microscopy Sciences[®]), cut into ultrathin sections (70 nm), and double stained with lead citrate (250 mg.mL⁻¹) and uranyl acetate (7%) before observations.

Scanning electron microscopy

Scanning electron microscopy (SEM; FEI QUANTA 200, USA) observations were performed to identify potential external modifications of spermatozoa during exposures to nano-PS beads. An aliquot of spermatozoa from the control treatment and the highest exposed concentration (25 µg.mL⁻¹) for both 50-NH₂ and 50-COOH beads were immersed in a mixture of 6% glutaraldehyde:7% NaCl:0.4M cacodylate (2:1:1; v/v) for 2 h. The samples were then dehydrated in a graded EtOH series (50%, 70%, 90% and 100%; 10 min each) and desiccated in a critical-point dryer (Leica CPD 300; Germany). Lastly, samples were coated with gold using an SCD 040 sputter coater (Blazers, Liechtenstein).

Confocal microscopy

A confocal laser scanning microscope (CLSM; Zeiss LSM 780, Germany) was used to observe the interactions between fluorescent nano-PS beads and spermatozoa. An aliquot of spermatozoa exposed to fluorescent 50-NH₂ or 50-COOH (25 µg.mL⁻¹) beads, following the same experimental design as the toxicity experiments was sampled after 1 h of exposure and fixed with 6% glutaraldehyde at 4°C for 48 h. An excitation/emission of 488 nm/543 nm was used to visualize the 50-NH₂ beads and an excitation/emission of 514 nm/612 nm for 50-COOH beads. Spermatozoa were stained with 1% DAPI (40,6-diamidino-2-phenylindole; 10 min in dark condition; excitation/emission 405 nm/455 nm) to perform three-dimensional observations.

Assessment of the reproductive success

At the end of the exposure, spermatozoa from each treatment and replicate were placed in 2 L glass beakers filled with 1.5 L of clean FSW containing 20,000 control oocytes pooled from 3 females per replicate (n = 6 replicates; total number of female oysters = 18). The oocytes were collected by stripping the gonads and sieving at 100 µm and then 20 µm with sterile equipment to remove debris. The spermatozoa-to-oocyte ratio was 100:1, following the same procedure described in Tallec *et al.* (2018). After 48 h, the beakers were sieved at 40 µm to collect larvae, and samples were fixed with a formaldehyde-seawater solution (0.1% final) to estimate the D-

larval yield by light microscopy (Zeiss Axio Observer Z1; $\times 10$ – 63 magnification). The D-larval yield was defined as: (number of normal D-larvae / number of oocytes) $\times 100$ (Song *et al.* 2009). Abnormal D-larvae referred to shell/mantle malformations and developmental arrests (Mottier *et al.* 2013).

Statistical analyses and graphical representations

Statistical analyses and graphical representations were conducted using the R Software (R Core Team, 2016). All data were screened for normality (Shapiro-Wilk test) and homoscedasticity (Levene test). Percentages were analyzed after angular transformations, except for the percentage of motile spermatozoa, which was subjected to a fourth-root transformation to respect statistical rules. Particle behavior was compared between the experimental media (EM) and FSW with the t-test method. Motility, flow cytometry analyses, and lipid classes among treatments were analyzed by one-way ANOVA or the Kruskal-Wallis test, followed by pairwise comparison (Tukey's *post hoc* or Conover's *post hoc* test for parametric or non-parametric analyses, respectively). Modifications of FA composition in the polar and neutral fractions were identified by one-way analysis of similarities (ANOSIM) using a Bray-Curtis similarity matrix to separate clusters (R = 1: perfect separation; R = 0.5: satisfactory separation, R = 0: poor separation cluster) (Long *et al.* 2018). Correlations were made with the Pearson's method between the percentage of motile spermatozoa or VAP and the D-larval yield. The significance threshold of the statistical analyses was fixed at 0.05. Data were expressed as the means \pm standard deviation (SD).

Results

Characterization of nanopolystyrene beads in seminal media

The commercial size (50 nm) of both 50-NH₂ and 50-COOH beads was verified by TEM (Figure S1). In the experimental medium (EM), the 50-NH₂ suspension showed a size of 100.2 ± 2.0 nm and a ζ -potential of 14.7 ± 3.2 mV (Figures 1A and 1B). The 50-COOH suspension formed microscale aggregates characterized by a size of $7,435 \pm 538$ nm and a ζ -potential of -12.5 ± 1.5 mV (Figures 1C and 1D). The size of the 50-COOH suspension increased significantly by 2.3-fold in EM (df = 3.5, p-value = 0.002) compared with FSW (data from Tallec *et al.*, 2018).

[Figure 1 near here]

Effects on spermatozoa motility

Compared with the control treatment, both 50-NH₂ and 50-COOH beads significantly altered spermatozoa motility (Figure 2). At 10 and 25 µg.mL⁻¹, 50-NH₂ beads significantly reduced the percentage of motile spermatozoa (-79 and -99%, respectively; F-value = 28.18, p-values < 0.001) and the VAP (-62 and -94%, respectively; Kruskal-Wallis chi-squared = 21.75, p-values < 0.001) (Figures 2A and 2B). Exposures to 50-COOH beads only caused a significant effect at the highest concentration (25 µg.mL⁻¹) with reductions of the percentage of motile spermatozoa (-66%; F-value = 13.18, p-value < 0.001) and the VAP (-38%; F-value = 5.22 p-value = 0.01) (Figures 2C and 2D).

[Figure 2 near here]

Morphology and functional characteristics of spermatozoa exposed to nano-PS

The relative size and complexity (FSC and SSC, respectively) of the spermatozoa were not affected (p-values > 0.05) by exposures to 50-NH₂ beads (Table 1). The FSC was not statistically different among all concentrations of 50-COOH beads (p-value > 0.05). At 10 and 25 µg.mL⁻¹ of 50-COOH beads, the SSC increased significantly (+19% and 33%, respectively; F-value = 20.77, p-values = 0.005 and < 0.001, respectively) compared with the control treatment. The functional characteristics of spermatozoa (percentage of live cells, acrosomal integrity and ROS production) were not significantly altered (p-values > 0.05) by exposures to 50-NH₂ and 50-COOH beads at any concentration (Table 1).

[Table 1 near here]

Lipid composition of spermatozoa exposed to nano-PS

Neither type of nano-PS beads affected the total lipid content (mean value = 3.99 ± 0.17 mg.10⁹ spermatozoa⁻¹; p-value > 0.05; Table 2). The overall percentages of membrane lipids (mean value = $98.23\% \pm 0.03\%$) and storage lipids (mean value = $1.77\% \pm 0.03\%$) were statistically similar (p-values > 0.05) among all treatments. Statistical analyses revealed a significantly higher relative proportion (+11.5%; F-value = 19.09, p-value = 0.003) of phosphatidylserine at the highest concentration (25 µg.mL⁻¹) of 50-NH₂ beads in comparison to other treatments including control. Other class proportions remained unchanged among all treatments (p-values > 0.05) (Table 2). No changes were observed in the PC/PE ratio (mean value = 1.58 ± 0.02 ; p-

value > 0.05). One-way analysis of similarities (ANOSIM) revealed no statistical differences in the FA composition among treatments, for either the NL ($R = -0.2$ and $p\text{-value} > 0.05$; Table S1) or PL ($R = 0.18$ and $p\text{-value} > 0.05$; Table S2).

[Table 2 near here]

Interactions between nano-PS beads and spermatozoa

Confocal laser scanning microscopy observations revealed that the 50-NH₂ and 50-COOH beads appeared to adhere to the external membrane of spermatozoa head (Figures 3B, 3C and 3D), notably on the acrosomal area. For the 50-COOH beads treatment, spermatozoa were embedded in aggregates of 50-COOH beads (Figure 3E; Figure S2). This pattern of large aggregates containing trapped spermatozoa was not observed for the 50-NH₂ beads treatment. Adhesion of nano-PS beads to the spermatozoa was observed by both TEM and SEM, but only with the 50-NH₂ beads (Figures 4 and 5). No alterations of internal or external structures were remarked in spermatozoa exposed to either nano-PS beads when compared to the control treatment. Neither 50-NH₂ nor 50-COOH beads were detected inside spermatozoa with the TEM observations and the orthogonal views obtained by confocal microscopy (Figures S3 and S4).

[Figure 3 near here]

[Figure 4 near here]

[Figure 5 near here]

Effects on the reproductive success

The 50-NH₂ beads induced significant decreases in the D-larval yield at 10 and 25 $\mu\text{g}\cdot\text{mL}^{-1}$ (-59% and -95%, respectively; $F\text{-value} = 24.41$, $p\text{-value} = 0.01$ and < 0.001 , respectively) (Figure 6A). By contrast, exposures to 50-COOH beads had no effects in the D-larval yield ($p\text{-value} > 0.05$; Figure 6B) compared with the control treatment. Significant positive correlations were found in response to 50-NH₂ beads exposures between: (i) the D-larval yield and spermatozoa motility ($p\text{-value} < 0.001$; $R^2 = 0.68$; Figure 7A); (ii) the D-larval yield and VAP ($p\text{-value} < 0.001$; $R^2 = 0.76$; Figure 7B).

[Figure 6 near here]

[Figure 7 near here]

Discussion

Adverse effects on oyster spermatozoa depend on nanoparticle type. Healthy spermatozoa are crucial for ensuring the sustainability of *C. gigas* through the transfer of genetic heritage to oocytes and offspring. Among the parameters that define the quality of spermatozoa, the motility (*e.g.*, the percentage of motile spermatozoa and the swimming speed) and reproductive success are predominant (Taylor *et al.* 2014). Here, despite a very short exposure of 1 h, the 50-NH₂ beads reduced the percentage of motile spermatozoa and their VAP, leading to a decrease in the D-larval yield, which was used as a proxy for reproductive success. While the present study showed that toxic effects on spermatozoa appeared from 10 µg.mL⁻¹ of 50-NH₂ beads, embryotoxicity was observed from 0.1 µg.mL⁻¹ in oyster embryos exposed to the same particles (Tallec *et al.* 2018). Higher embryotoxicity than spermiotoxicity has been previously observed in oysters and sea urchins embryos/spermatozoa exposed to PAH-contaminated sediments and xenobiotics (Geffard *et al.* 2001, Manzo *et al.* 2006). This difference can be related to the set of intense metabolic and morphological modifications (*e.g.*, numerous cleavages, gastrulation, organogenesis, shell calcification) triggered during embryogenesis (Fitzpatrick *et al.* 2008).

By contrast, the 50-COOH beads seem to have only a transitory effect on oyster spermatozoa, *i.e.*, a decrease in the motility at the highest concentration, without negative outcomes for their reproductive success as suggested by the similar D-larval yield among conditions derived from control or exposed spermatozoa. These observations suggest an involvement of different toxic pathways between 50-NH₂ and 50-COOH beads on oyster spermatozoa, in agreement with previous studies (Della Torre *et al.* 2014, Manfra *et al.* 2017, Tallec *et al.* 2018). For instance, the 50-NH₂ beads caused a strong decrease in the embryogenesis success of the oyster *C. gigas* (EC₅₀ = 0.15 µg.mL⁻¹), the sea urchin *Paracentrotus lividus* (EC₅₀ = 2.61 µg.mL⁻¹) and the survival of the rotifer *Brachionus plicatilis* (EC₅₀ = 6.62 µg.mL⁻¹) whereas 50-COOH beads had a lower toxicity (EC₅₀ = 11.60 µg.mL⁻¹ for the embryogenesis success of *C. gigas*) or no effect (survival of *B. plicatilis*, embryogenesis of *P. lividus*). In the one hand, the aggregation of 50-COOH beads in the experimental medium decreases the surface-to-volume ratio of the particle suspension, decreasing particle reactivity and therefore toxic effects on cells in comparison to 50-NH₂ that remained unaggregated in seawater. On the other hand, the fluctuating toxicity between 50-NH₂ and 50-COOH beads is also linked to the surface properties of the nanoparticles (*e.g.*, surface charge), which are key-factors affecting the behavior and interactions with biological membrane (Cho *et al.* 2009, Nangia and Sureshkumar 2012, Della

Torre *et al.* 2014, Tallec *et al.* 2019). The positive charge of the 50-NH₂ beads in seawater favor interactions with cells, thereby increasing their toxic effects compared with the negatively-charged 50-COOH beads (Della Torre *et al.* 2014). The stronger interactions between 50-NH₂ beads and membrane than 50-COOH beads could explain results of TEM and SEM for which only 50-NH₂ beads are found to attach on spermatozoa despite numerous cleaning steps performed during the sample preparation.

The two types of nanobeads used in the present study allowed to test effects of the surface charge and the behavior of particles on oyster spermatozoa. Although the 50-NH₂ beads are excellent model particles to study effect of nanoparticles on marine model owing to its high stability in seawater, it is expected that nanoplastics in environment have carboxylated surfaces due to the oxidation processes occurring in seawater (Fotopoulou and Karapanagioti, 2012). This combined to the effects of the 50-COOH beads observed on oyster spermatozoa only at the highest concentration, *i.e.* 25 µg.mL⁻¹ (3.6×10^{11} particles.mL⁻¹), suggests a limited/negligible environmental risk of nanopolystyrene beads for this type of cell comparing to the highest concentrations of microplastics reported at sea (to our knowledge 8 particles.mL⁻¹; Brandon *et al.*, 2020) even if the nanoplastic concentrations in ocean are expected to be higher than microplastics (Wagner and Reemtsma, 2019).

The 50-NH₂ beads toxicity on oyster spermatozoa is not related to membrane breakage or modification of cell defenses. Numerous studies have reported cytotoxicity (*e.g.*, a decrease in cell viability, an increase of ROS production) in cell lines (*e.g.*, macrophage, brain astrocytoma) and aquatic organisms (*e.g.*, cyanobacteria) after exposures to 50-NH₂ beads at concentrations close to the ones used in the present study, signaling a perturbation of cell defenses (Xia *et al.* 2008, Bexiga *et al.* 2011, Feng *et al.* 2019). In oyster spermatozoa, no variations in ROS production or other markers of cytotoxicity (*e.g.*, acrosomal integrity) were observed, despite a significant decrease in key sperm functions (motility and reproductive success) induced by 50-NH₂ beads. A similar pattern (*i.e.*, a decrease in spermatozoa quality without overproduction of ROS) was previously described in bovine spermatozoa exposed to gold nanoparticles at 10 µg.mL⁻¹ (Taylor *et al.* 2014). These authors suggested that membrane perturbations due to the attachment of gold nanoparticles could explain the decreases in motility and fertilization success. In this regard, one pathway for toxicity of nano-PS proposed in the literature is related to membrane disturbance (Rossi *et al.* 2014, Jeong *et al.* 2018, Feng *et al.* 2019). For instance, 50-NH₂ beads at 2.5 and 4 µg.mL⁻¹ induced severe alterations in *Synechococcus elongatus* membrane (increase in permeability and loss of integrity) (Feng *et al.* 2019). The present

experiment was therefore designed to test this hypothesis on oyster spermatozoa. Although flow cytometry analyses suggested no effects of the 50-NH₂ beads on spermatozoa membrane integrity, the increase in the phosphatidylserine relative level observed at 25 µg.mL⁻¹ (+11.5%) suggests a membrane reorganization in response to the adhesion of 50-NH₂ beads on spermatozoa membrane. As previously reported, marine spermatozoa can reorganize the lipid bilayer under homeostasis damages (*e.g.* osmotic stress) (Cosson, 2004). Nanomaterials can interact with membrane compounds such as ion channels, modifying ion homeostasis and leading to physiological disruptions (Yin *et al.*, 2019). For instance, a perturbation of the transmembrane Na⁺/K⁺-ATPases and Ca²⁺-ATPases was suggested as cause of the loss of motility in bovine spermatozoa displaying gold nanoparticle attachments (Taylor *et al.* 2014). As phosphatidylserine participates actively in the stability of essential transmembrane components such as Na⁺/K⁺-ATPases (Wheeler and Whittam, 1970), we hypothesize that the increase in the phosphatidylserine relative level could reflect a cellular response to handle effect of 50-NH₂ beads on ion channels such as conformational changes. Nevertheless, it seemed that this membrane reorganization was not sufficient to maintain cell motility. Therefore, further functional work is needed to establish effects of 50-NH₂ beads in the membrane organization and understand precisely the origin and the consequences of the relative increase of phosphatidylserine.

Aggregates of 50-COOH beads could trap oyster spermatozoa. Interestingly, the 50-COOH suspension displayed a higher aggregate size (2.3 times) in the experimental medium compared with seawater. The contribution of the high cations content (Na⁺, Ca²⁺, and K⁺) occurring naturally in the seminal fluid (Boulais *et al.* 2018) may have promoted interactions with the negative surface charge of the 50-COOH beads, thereby increasing homo-aggregation processes. Under these conditions, spermatozoa interact with microscale aggregates that could affect their movement. Indeed, confocal microscopy and SEM showed a myriad of spermatozoa trapped within large microscale aggregates formed by the 50-COOH beads in the experimental medium. This result suggests a physical entrapment of spermatozoa by 50-COOH aggregates and could explain the decrease in spermatozoa motility at the highest dose. This mechanism was previously suggested for coral spermatozoa when exposed to colloidal matter (suspended sediments) that formed aggregates (Humanes *et al.* 2017). The lack of an observed effect on D-larval yield may be an effect arising from dilution of the spermatozoa at the end of the exposure in 1.5 L to evaluate the reproductive success. This step could have liberated spermatozoa trapped in aggregates, thereby allowing a normal fertilization rate and thus no effect on the D-

larval yield. This also suggests there was no cell damage from the exposure, which is consistent with the other results, *i.e.*, no effects on membrane (composition, integrity) and on the ROS production. However, we cannot exclude a total absence of sperm health impairments as the high spermatozoa-to-oocyte ratio (100:1) can interfere in the result of the D-larval yield. Overall, millions of spermatozoa are produced/released but at the end, one is enough to deliver genetic material in oocyte. Therefore, the fraction of unaffected spermatozoa by 50-COOH aggregates could be sufficient to ensure fertilization.

No translocation of nano-PS beads into oyster spermatozoa. Conventionally, the risk of translocation into cells depends on the size of particles; *e.g.*, translocation efficiency of nano-PS beads into human cell lines is up to 7.8% for 50 nm beads and 0.8% for 100 nm beads (Walczak *et al.* 2015). Here, no evidence of translocation was highlighted in oyster spermatozoa by either confocal and electron microscopy for either 50-NH₂ or 50-COOH beads at 25 µg.mL⁻¹. This result could be explained by specific properties of the germ cells, *i.e.*, spermatozoa do not have endocytic processes, which are important pathways for internalizing nanoparticles (Taylor *et al.* 2015). Nevertheless, because (i) other studies have reported translocation of engineered nanoparticles into spermatozoa, *e.g.* 40 nm silver nanoparticles in mouse spermatozoa (Yoisingnurn *et al.* 2015), (ii) nano-PS beads have low contrast for TEM analyses (also noted by Jeong *et al.* (2018)), a relevant follow-up study would be to expose oyster spermatozoa to ¹⁴C-radiolabeled or metal-doped nano-PS which may greatly enhance nanoparticles detection and fate in marine cells and organisms (Al-Sid-Cheikh *et al.* 2018, Mitrano *et al.* 2019).

Conclusion

This study demonstrated interactions between nano-PS beads and oyster spermatozoa with spermotoxicity as a function of the surface functionalization. Several cellular disruptions and possibly translocation were expected, but only a toxicity related to the adhesion of nano-PS beads on the spermatozoa was observed. This finding calls for further analyses on transmembrane components, such as recognition proteins or ion channels, to establish the toxicity pathway.

Although this study highlighted effects of nano-PS particles on oyster spermatozoa, we remain cautious about any extrapolation to the natural environment. Firstly, we used model nanoparticles with spherical shape and surface functionalization available commercially,

whereas environmental nanoplastics could be more complex in terms of both shape (*e.g.*, irregular) and surface chemistry (*e.g.*, presence of pollutants, eco-corona) (Paul-Pont et al., 2018). This complexity must be considered in future studies using oyster spermatozoa as a sensitive model. From a toxicological point of view it would be interesting in future works to test other nanomaterials possessing $-NH_2$ or $-COOH$ functionalizations to verify if the toxicity is only due to the surface properties without implications of the core material. Secondly, we used dose-response experiments to determine the effect concentration of nano-PS beads because the concentration of nanoplastics in ocean is currently unavailable due to methodological gaps. Toxic effects were observed at extremely high concentrations, suggesting a very limited risk to oyster spermatozoa in the natural environment. Due to the lack of environmental data on such small particles, our results will have to be confronted in the future with data obtained in the field for the nanoplastic compartment. Therefore, the characterization and quantification of environmental nanoplastics, as well as the production of model particles that better mimic natural nanoplastics, are required to provide accurate risks scenarios and to determine the extent to which these nanoplastic contaminants constitute a threat for marine ecosystems.

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The authors report no conflict of interest.

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Table

Table 1. Relative size (FSC; A.U.), relative complexity (SSC; A.U.), percentage of live cells (%), ROS production (green fluorescence, A.U.), and acrosomal integrity (red fluorescence, A.U.) of oyster spermatozoa after 1 h of exposure to 50-NH₂ and 50-COOH beads at five concentrations: 0 (Control), 0.1, 1, 10, and 25 µg.mL⁻¹. Data are given as means ± SD (n = 6 except for the acrosomal integrity, where n = 4). ANOVA was used to compare treatments with Tukey HSD for pairwise comparisons at the 5% level; homogeneous groups share the same letter. NA = Not applicable as the acrosomal integrity was only assessed on control and treatments that affected the reproductive success was affected.

Treatments	Relative Size (FSC; A.U.)	Relative Complexity (SSC; A.U.)	Live cells (%)	ROS Production (A.U.)	Acrosomal Integrity (A.U.)
Control	64.1 ± 1.3 ^a	41.3 ± 1.7 ^a	94.2 ± 1.9 ^a	24.6 ± 11.4 ^a	259.3 ± 33.2 ^a
50-NH ₂ 0.1	64.1 ± 1.4 ^a	40.8 ± 1.5 ^a	95.5 ± 2.5 ^a	22.6 ± 8.6 ^a	NA

	1	64.5 ± 1.3 ^a	42.5 ± 1.6 ^a	95.2 ± 1.8 ^a	25.4 ± 10.8 ^a	NA
	10	65.1 ± 1.7 ^a	44.9 ± 1.4 ^a	93.1 ± 8.5 ^a	22.6 ± 8.7 ^a	250.8 ± 53.9 ^a
	25	64.6 ± 0.7 ^a	40.6 ± 3.1 ^a	97.4 ± 1.6 ^a	19.3 ± 8.2 ^a	297.8 ± 14.9 ^a
50-COOH	0.1	64.5 ± 1.2 ^a	41.6 ± 2.2 ^a	95.1 ± 2.1 ^a	20.0 ± 4.3 ^a	NA
	1	63.9 ± 1.1 ^a	39.9 ± 6.6 ^a	95.4 ± 2.5 ^a	22.7 ± 5.7 ^a	NA
	10	64.7 ± 0.9 ^a	49.5 ± 1.8 ^b	95.7 ± 2.4 ^a	20.4 ± 4.8 ^a	NA
	25	64.4 ± 0.9 ^a	55.1 ± 2.4 ^b	96.4 ± 2.2 ^a	17.3 ± 4.7 ^a	NA

Table 2. Total lipid content (mg.10⁹ spermatozoa⁻¹) and lipid class composition of oyster spermatozoa after 1 h exposure to various treatments: Control; 50-COOH beads (25 µg.mL⁻¹); 50-NH₂ beads (10 µg.mL⁻¹); 50-NH₂ beads (25 µg.mL⁻¹). Lipid composition was expressed as the mass percentage of each class to the total lipid content (n = 4; means ± SD). ANOVA method was used to compare treatments with Tukey HSD for pairwise comparisons at the 5% level; *: p < 0.01. PE: phosphatidylethanolamine, PI: phosphatidylinositol, PSer: phosphatidylserine, CL: cardiolipin; CAEP: ceramide amino-ethylphosphonate, PC: phosphatidylcholine.

	Treatments			
	Control	50-COOH (25 $\mu\text{g.mL}^{-1}$)	50-NH ₂ (10 $\mu\text{g.mL}^{-1}$)	50-NH ₂ (25 $\mu\text{g.mL}^{-1}$)
% Triglycerides	1.59 \pm 0.76	1.66 \pm 0.49	1.56 \pm 0.65	1.67 \pm 0.63
% Sterols	7.04 \pm 0.60	6.79 \pm 0.48	6.93 \pm 0.90	6.37 \pm 0.30
% Free fatty acids	0.18 \pm 0.09	0.14 \pm 0.01	0.16 \pm 0.02	0.11 \pm 0.01
% PE	23.13 \pm 0.76	23.27 \pm 0.50	23.24 \pm 0.84	23.03 \pm 0.70
% PI+CAEP	18.90 \pm 1.24	19.19 \pm 1.20	19.15 \pm 1.51	18.99 \pm 0.79
% PSer	6.61 \pm 0.28	6.50 \pm 0.17	6.85 \pm 0.20	7.52 \pm 0.17*
% CL	5.65 \pm 0.40	5.97 \pm 0.23	5.70 \pm 0.03	5.66 \pm 0.16
% PC	36.90 \pm 0.55	36.49 \pm 1.21	36.40 \pm 0.42	36.64 \pm 0.77
% Membrane	98.23 \pm 0.03	98.20 \pm 0.05	98.28 \pm 0.07	98.22 \pm 0.02
% Storage	1.77 \pm 0.72	1.80 \pm 0.50	1.72 \pm 0.63	1.78 \pm 0.62
PC/PE	1.60 \pm 0.07	1.57 \pm 0.06	1.57 \pm 0.04	1.59 \pm 0.07
Total lipid content (mg.10 ⁹ spermatozoa ⁻¹)	4.03 \pm 0.24	3.79 \pm 0.50	4.20 \pm 0.72	3.97 \pm 0.20

Figure captions

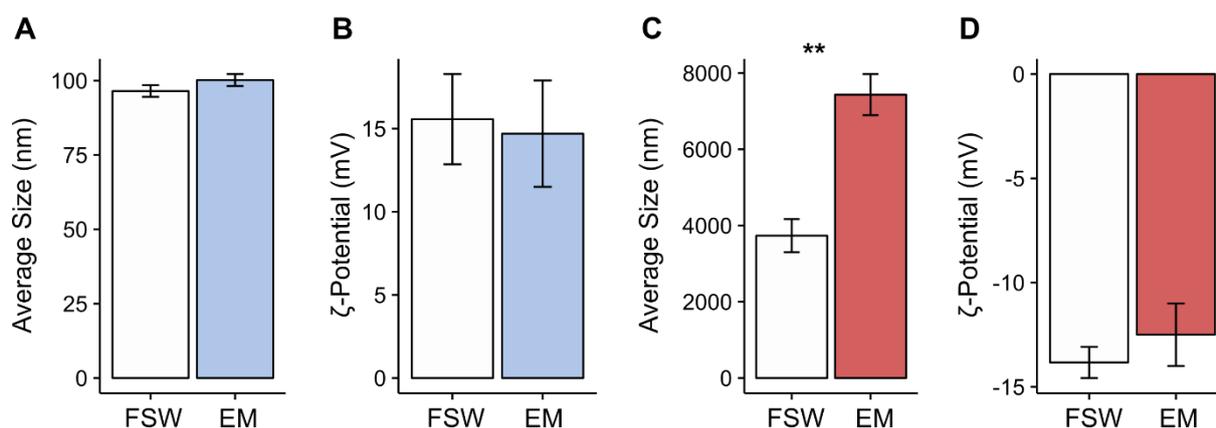


Figure 1. Size (nm; A and C) and ζ -potential (mV; B and D) in filtered seawater (FSW, from Tallec *et al.*, 2018) and the experimental medium (EM, present study) of two 50 nm nano-PS: 50-NH₂ beads (A and B); 50-COOH beads (C and D). Data are given as means \pm SD (n=3). The *t*-test method was used to compare data between FSW and EM (** = p-value < 0.01).

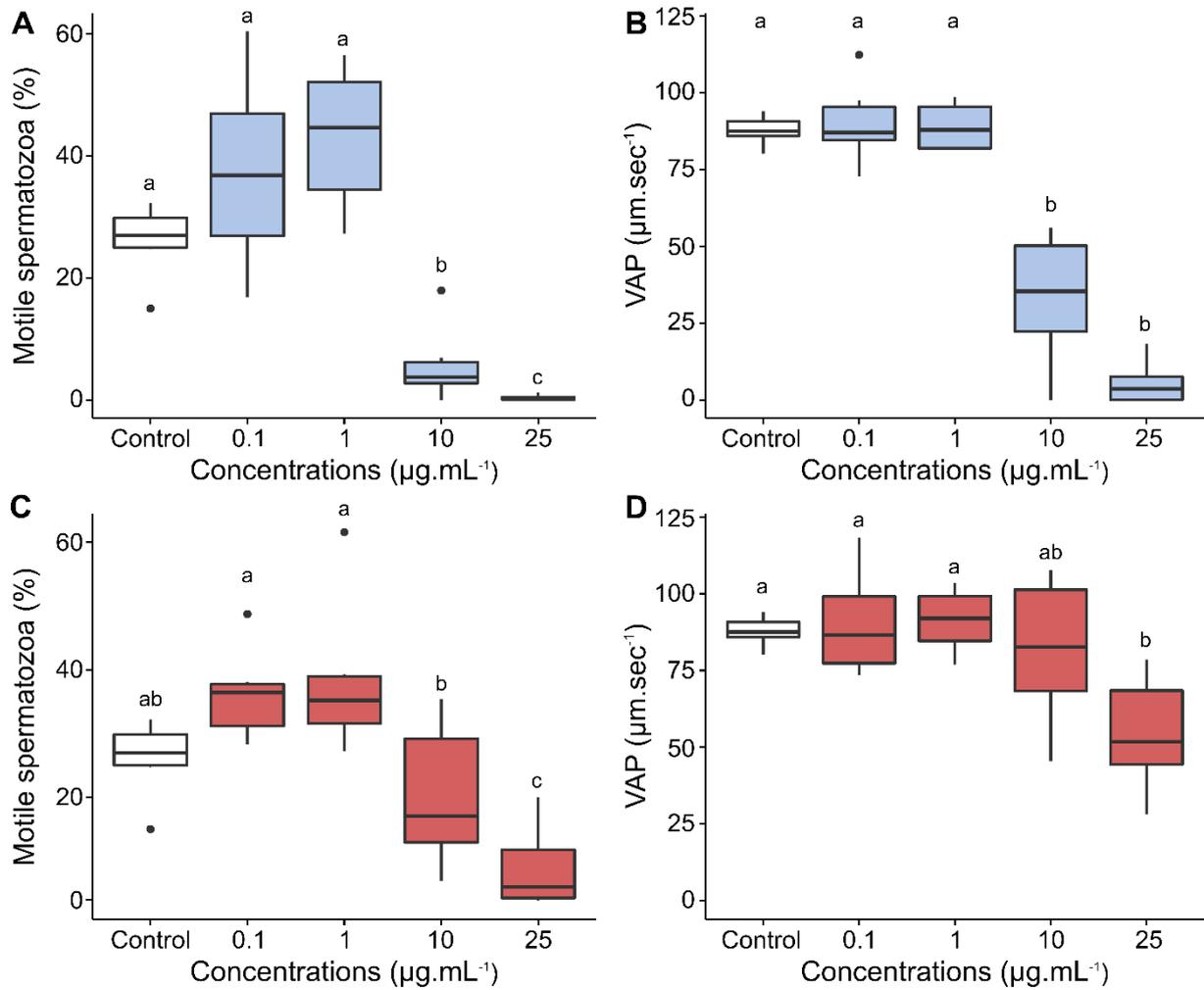


Figure 2. Percentage of motile spermatozoa (%; A and C) and velocity of the average path (VAP; $\mu\text{m.sec}^{-1}$; B and D) of oyster spermatozoa after 1 h of exposure to 50-NH₂ (blue, A and B) and 50-COOH beads (red, C and D) at five concentrations: 0 (Control), 0.1, 1, 10, and 25 $\mu\text{g.mL}^{-1}$. N = 6. ANOVA (A, C and D) or Kruskal-Wallis tests (B) with pairwise comparisons were made between treatments at the 5% level; homogeneous groups share the same letter.

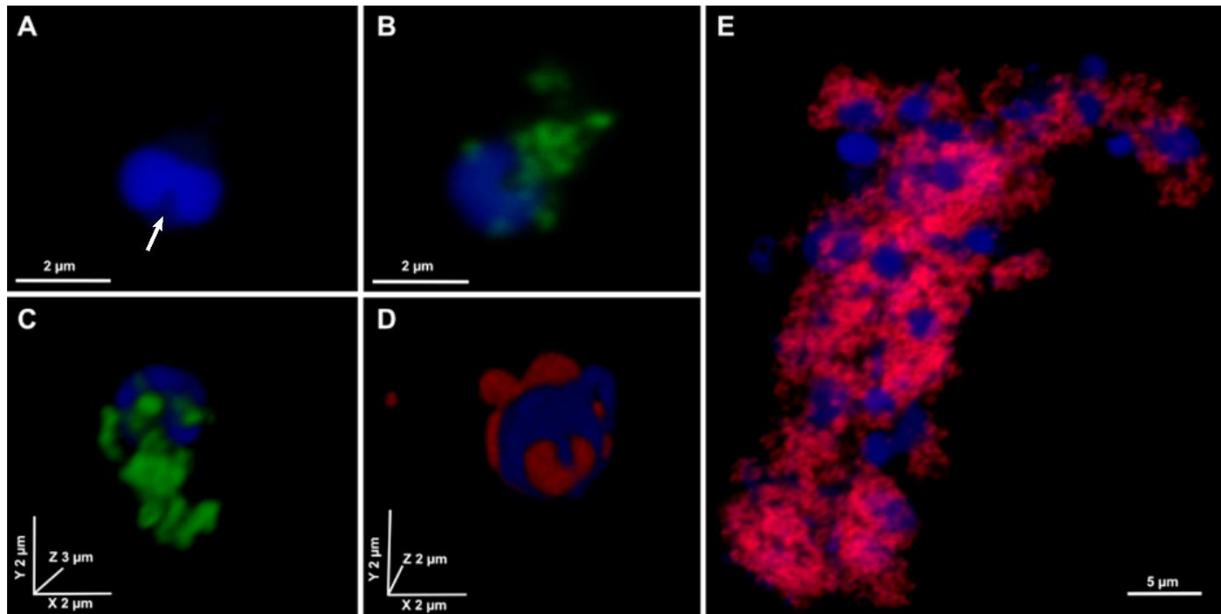


Figure 3. Confocal laser scanning microscopy of oyster spermatozoa (blue) after 1 h of exposure to nano-PS beads: (A) Control treatment; (B and C) Adhesion of 50-NH₂ beads (green; 25 µg.mL⁻¹) on the spermatozoa head; (D) Adhesion of 50-COOH beads (red; 25 µg.mL⁻¹) on the spermatozoa head and (E) Observation of spermatozoa embedded in microscale aggregates of 50-COOH beads. The arrow shows the location of the acrosome. Size is represented by the scale bar.

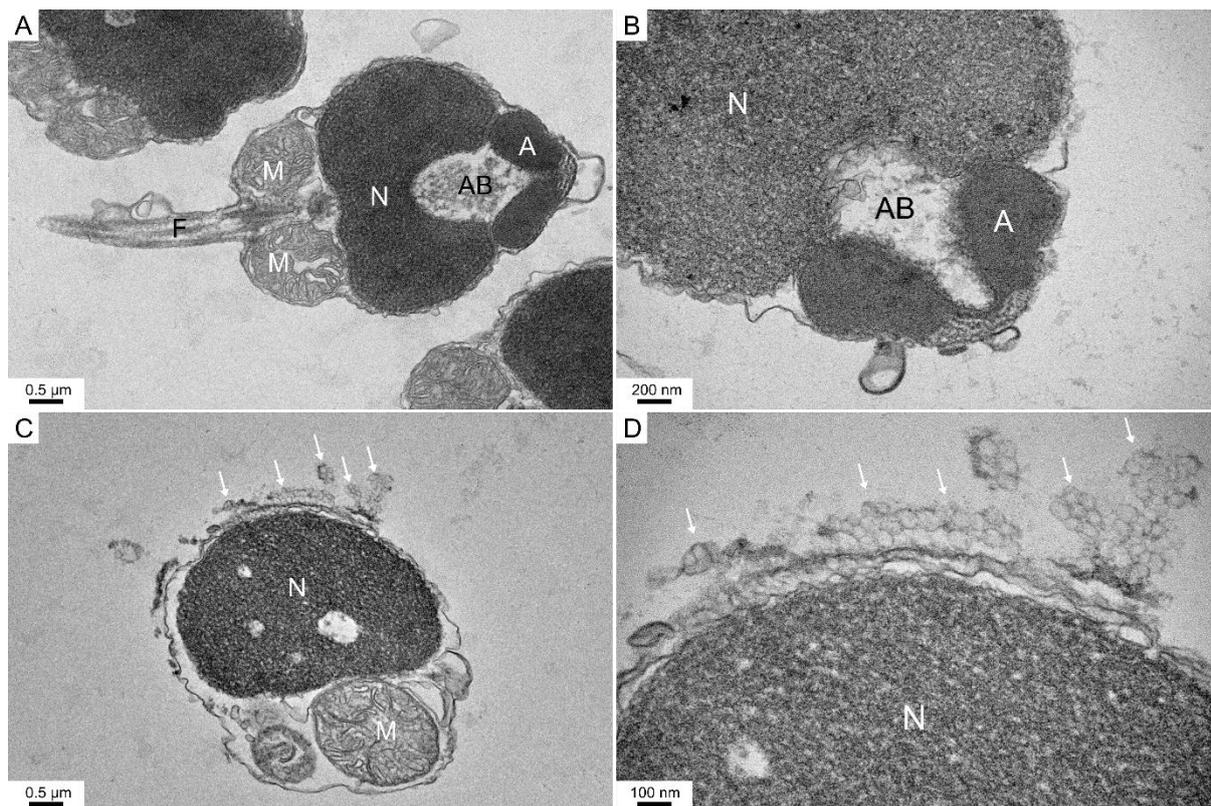


Figure 4. Transmission electron microscopy observations of oyster spermatozoa after 1 h of exposure to nano-PS beads. (A and B) Longitudinal sections of spermatozoa from the control treatment; (C and D) Longitudinal sections of spermatozoa exposed to $25 \mu\text{g.mL}^{-1}$ of 50-NH₂ beads. White arrows indicate adhesion of 50-NH₂ to the spermatozoa membrane. A: Acrosome; AB: Axial body; N: Nucleus; M: Mitochondria; F: Flagellum. Size is represented by the scale bar.

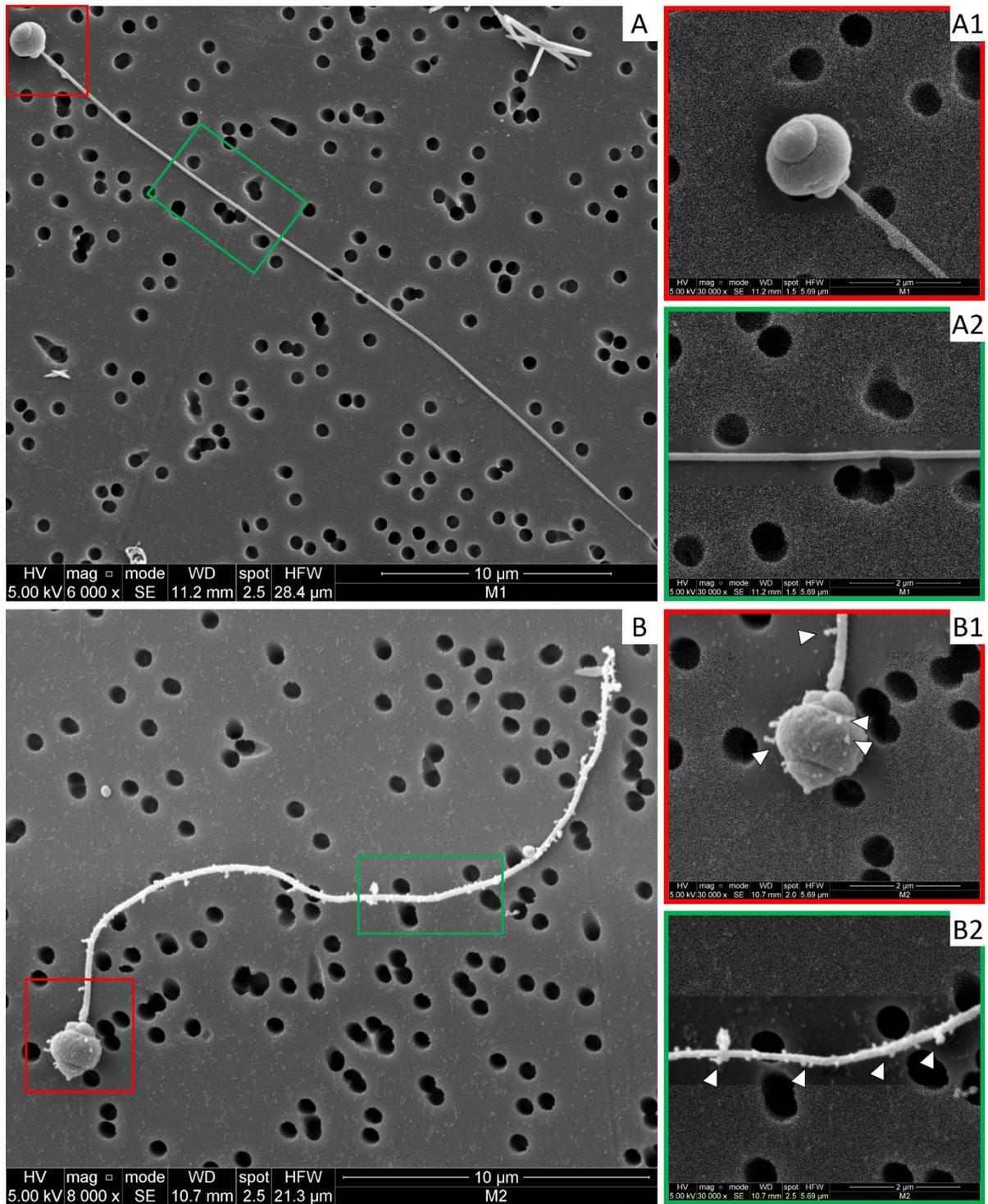


Figure 5. Scanning electron microscopy observations of oyster spermatozoa after 1 h of exposure to nano-PS beads. (A, A1 and A2) Spermatozoa from the control treatment; (B, B1 and B2) Spermatozoa exposed to 25 $\mu\text{g.mL}^{-1}$ of 50-NH₂ beads. White arrows indicate adhesion of 50-NH₂ beads to the spermatozoa head and flagellum. Size is represented by the scale bar.

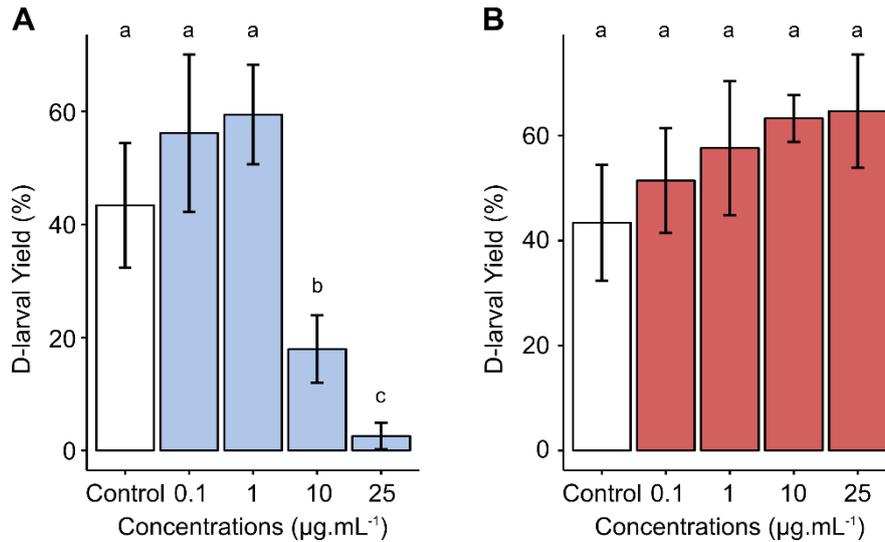


Figure 6. D-larval yield obtained following fertilization with spermatozoa exposed to five concentrations (0 – Control, 0.1, 1, 10, and 25 µg.mL⁻¹) of 50-NH₂ (A) and 50-COOH beads (B). Data are given as means ± SD (n = 6). ANOVA and pairwise comparisons were made between treatments using Tukey’s HSD at the 5% level; homogeneous groups share the same letter.

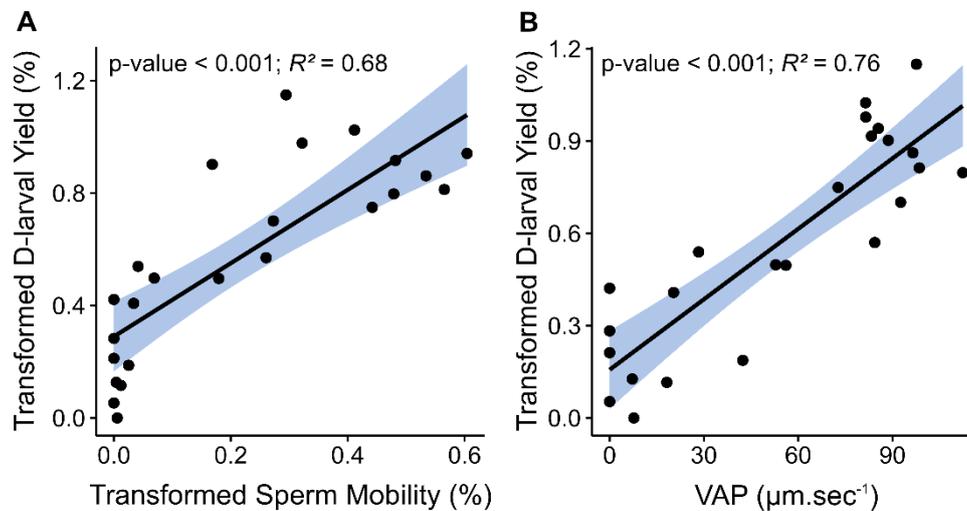


Figure 7. Correlations between spermatozoa motility (A) or VAP (B) and the D-larval yield after spermatozoa exposures to 50-NH₂ beads. Data used for the graphical representation were the data transformed as described in the materials and methods part. Statistical analyses were performed with the Pearson test at the 5% level (n = 6).