#### **Chemical Geology**

January 2023, Volume 615 Pages 121202 (10p.) <a href="https://doi.org/10.1016/j.chemgeo.2022.121202">https://doi.org/10.1016/j.chemgeo.2022.121202</a> <a href="https://archimer.ifremer.fr/doc/00804/91580/">https://archimer.ifremer.fr/doc/00804/91580/</a>



# Calculation of cerium and lanthanum anomalies in geological and environmental samples

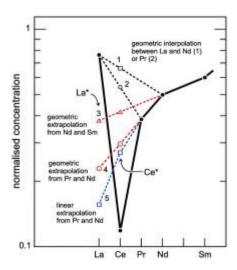
Barrat Jean-Alix 1, 2, \*, Bayon Germain 3, Lalonde Stefan 4

- <sup>1</sup> Univ Brest, CNRS, LEMAR, Institut Universitaire Européen de la Mer (IUEM), Place Nicolas Copernic, 29280 Plouzané, France
- <sup>2</sup> Institut Universitaire de France, France
- <sup>3</sup> Univ Brest, CNRS, Ifremer, Geo-Ocean, F-29280 Plouzané, France
- <sup>4</sup> Univ Brest, CNRS, Ifremer, Geo-Ocean, F-29280 Plouzané, France
- \* Corresponding author : Jean-Alix Barrat, email address : barrat@univ-brest.fr

#### Abstract:

The determination of La and Ce anomalies in natural waters, biological samples and biogenic sedimentary rocks can provide unique information on biogeochemical processes in Earth surface environments. Over the last decades, several approaches have been used for calculating La/La\* and Ce/Ce\*, based on the comparison between measured and theoretical abundances (La\* and Ce\*) extrapolated from neighboring rare earth element concentrations normalized to chondritic or shale reference values. These extrapolations can be achieved either linearly or semi-logarithmically ("geometrically"), both methods being used in the literature in the absence of any consensus. We show here, using a database of rocks exhibiting no La and Ce anomaly, that the linear extrapolation of La and Ce abundances can result in markedly different results depending on whether chondritic or shale values are used for normalization. The geometric extrapolation allows consistent calculation of La and Ce anomalies for the entire compositional range tested in this study, regardless of whether data are normalized to chondritic or shale reference values. The differences between linear and geometric extrapolations are illustrated by a few selected examples from the literature, including various carbonate rock and seawater samples, further demonstrating that linear extrapolation can result in erroneous estimates of La and Ce anomalies. We thus propose that La/La\* and Ce/Ce\* ratios in all geological and environmental samples should be determined using the geometric extrapolation only.

### **Graphical abstract**



### **Highlights**

► La/La\* and Ce/Ce\* ratios calculated with linear and geometric extrapolations are compared. ► Linear extrapolation can result in aberrant estimates of La and Ce anomalies. ► La/La\* and Ce/Ce\* ratios should be determined using the geometric extrapolation only.

Keywords: Rare earth elements, La anomaly, Ce anomaly, Linear extrapolation, Geometric extrapolation

## 1/ Introduction

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

58

59

60

61

62

63

64

65

66

67

68

69

70

71

72

73

Over the last sixty years, rare earth elements (REE) have become one of the most studied groups of elements in Earth sciences. These elements have the particularity of having an extremely coherent geochemical behavior that can be classically described with the "Masuda-Coryell plots" (Masuda, 1962; Coryell et al., 1963), known today as "rare earth patterns". The principle of these diagrams is simple. Rare earth elements are ordered by increasing atomic number, and their abundances in any given sample are normalized to a set of reference values that generally correspond to average concentrations for chondrites (e.g., Anders and Grevesse, 1989; Pourmand et al., 2012; Barrat et al., 2012; Palme et al., 2014) or shales (e.g., Nance and Taylor, 1976; Pourmand et al., 2012; Bau et al., 2018). The first advantage of these diagrams is that the Oddo-Harkins effect (i.e. chemical elements with even atomic numbers are more abundant than adjacent odd atomic number elements) disappears with normalization. Smooth REE patterns are generally obtained for most geological and environmental samples, mostly because all REE are trivalent and not significantly decoupled from each other under relevant physico-chemical conditions. There are notable exceptions, however. For instance, europium (Eu<sup>2+</sup> and Eu<sup>3+</sup>) and cerium (Ce<sup>3+</sup> and Ce<sup>4+</sup>) can be found in two valence states in geological samples. During particular magmatic or environmental processes, these two elements can be significantly decoupled from their neighboring REE, thereby producing specific elemental depletion or enrichment in normalized REE patterns. Additionally, significant anomalies in La, Sm, Gd and Tm can be also found in terrestrial rocks or natural waters, unrelated to any effect of valence. The occurrence of Tm anomalies in terrestrial rocks are thought to be inherited from the building blocks that formed our planet (Dauphas and Pourmand, 2015; Barrat et al., 2016). The origin of other anomalies is not always well understood, except when they are the result of environmental pollution (e.g., Bau and Dulski, 1996b; Kulaksiz and Bau, 2007, 2013; Merschel and Bau, 2015; Ma et al., 2019; Le Goff et al., 2019, Valdés-Vilchis, 2021).

Among these anomalies, Ce anomalies have received considerable interest. The distinctive behavior of Ce in the marine environment was discovered more than 50 years ago (e.g., Goldberg et al., 1963). The first reliable REE analyses of seawater, various authigenic phases, and ichtyoliths showed very early on that aqueous phases, marine carbonates and other seawater archives could display pronounced negative Ce anomalies (e.g., Piper, 1974; Elderfield and Pagett, 1986 and references therein). The role of Fe and Mn oxides in the development of these anomalies was subsequently proposed because Mn-rich nodules and

crusts displayed complementary positive anomalies in Ce (e.g., Goldberg et al., 1963; Piper, 1974). Today, it is generally well accepted that the decoupling of Ce from other REEs in the oceans, or more generally in aqueous environments, mainly results from oxidative scavenging of Ce by Fe and Mn hydroxides (e.g., Bau and Koschinsky, 2009). The occurrence of Ce anomalies in natural waters and in the biogenic or authigenic phases that precipitate from them is used as a proxy for oxidative conditions in Earth surface environments (e.g., German and Elderfield, 1990; German et al., 1991; Bau et al., 1997; Wallace et al., 2017; Bellefroid et al., 2018). Cerium anomalies are also of interest for magma-related studies, although their application to basalts and other igneous rocks remains limited. Lavas from subduction zones and oceanic islands occasionally show negative Ce anomalies, interpreted as reflecting a recycled sedimentary component in their mantle sources (e.g., Shimizu et al., 1992; Class and Le Roex, 2008). In contrast, zircons frequently show excess Ce (Ce<sup>4+</sup> having the same ionic radius as Zr<sup>4+</sup>), indicating preferential selective incorporation of Ce relative to neighboring REE during crystal growth. The resulting positive Ce anomalies could reflect the Ce<sup>4+</sup>/Ce<sup>3+</sup> ratios of their parental magmas, and therefore provide constraints on oxygen fugacity (e.g., Burnham and Berry, 2012; Trail et al., 2012; Smythe and Brenan, 2016).

74

75

76

77

78

79

80

81

82

83

84

85

86

87

88

89

90

91

92

93

94

95

96

97

98

99

100

101

102

103

104

105

106

Unlike Ce, the La anomaly in geological and environmental samples has received much less attention. Lanthanum anomalies represent a common feature in seawater and marine authigenic phases (e.g., Elderfield, 1988; Tostevin et al., 2016), but remains to date undocumented in igneous rocks. The cause of these anomalies is still poorly understood but could relate to the higher stability of La in solution relative to the other light REE (De Baar et al., 1985, Byrne and Kim, 1990, Byrne et al., 1996). Previous studies have suggested that marine barite could play a role in the development of La anomalies in seawater (Grenier et al., 2018). Indeed, Hein et al. (2007) have reported positive La anomalies in this phase. It is likely that the La excesses they measured, however, are analytical artefacts generated in the plasma, notably isobaric interferences from Ba (e.g., <sup>138</sup>BaH<sup>+</sup> on <sup>139</sup>La<sup>+</sup>), and need to be confirmed. Recently, a number of pioneering studies have demonstrated that biological activity can also fractionate light-REE (e.g., Pol et al., 2014; Semrau et al., 2018; Bayon et al., 2020a). Wang et al. (2020) reported positive La anomalies in methanotrophic mussels at submarine methane seeps, interpreted as resulting from microbial enzymatic activity related to aerobic methane oxidation. These results have suggested that the La anomaly could serve as a diagnostic tool for tracing past biological activity related to aerobic methanotrophy. Moreover, La also represents an emerging pollutant in modern environments due to its widespread industrial use

in magnetic alloys and catalysts for gasoline engines (Kulaksiz and Bau, 2013; Merschel and Bau, 2015). The emergence of anthropogenic La issues in Earth surface environments calls for a better understanding of the mechanisms that control the decoupling of La from neighboring REE in aquatic environments.

Over the last few decades, different methods have been proposed for calculating La and Ce anomalies in geological and environmental samples, yet no consensus exists among geochemists on best practices for these calculations. In the literature, historical conventions or habits prevail. These anomalies are calculated from normalized concentrations and by interpolating or extrapolating La and Ce concentrations (La\* and Ce\*) assuming smooth REE patterns with linear or logarithmic scales. This results in anomaly values that can be very different from one study to another and cannot be compared. Here we show that some commonly used approaches for calculating La and Ce anomalies can lead to aberrant results, hence our call for a standardization of the calculation.

# 2/ The different ways to calculate La and Ce anomalies

### 2.1/ Normalization values

The patterns of average CI chondrites and average shales do not show significant anomalies in La and Ce when normalized to each other. Therefore, one would expect to calculate similar La/La\* or Ce/Ce\* values in any given sample following normalization to either chondritic or shale reference values. We will see below that this is not necessarily the case. In this work, we used the average of the Orgueil chondrite concentrations measured by Barrat et al. (2012), and the Post Archean Australian Shale (PAAS) average obtained by Pourmand et al. (2012), which was recalculated relative to our standard values to correct for a slight calibration bias (Barrat et al., 2020). These preferred normalization values are given in Table 1. In the following, X<sub>CI</sub> and X<sub>SN</sub> corresponds to the element X concentrations normalized to chondritic or shales values, respectively.

### 2.2/ The calculation of Ce and La anomalies

By definition, an anomaly visualized in a normalized-REE diagram for an element (X) can be quantified by dividing its measured abundance by its theoretical concentration in the

absence of any anomaly  $(X^*)$ . This latter can be calculated by interpolation or extrapolation using the normalized abundances of neighboring elements, assuming a smooth REE pattern. The measured/theoretical elemental ratio  $(X/X^*)$  thus makes it possible to quantitatively measure a positive  $(X/X^*>1)$  or a negative  $(X/X^*<1)$  anomaly for this element.

Many approaches to calculating Ce\* and La\* have been proposed in the last few decades. In the case of magmatic rocks or zircons, which are devoid of La anomalies, Ce\* is given by the geometric mean of the normalized concentrations of La and Pr, or interpolated "semi-logarithmically" between La and Nd when Pr abundances are not determined:

145 
$$Ce/Ce^* = Ce_{CI}/(La_{CI} \times Pr_{CI})^{1/2}$$
 (1)

146 or

147 
$$\text{Ce/Ce}^* = \text{Ce}_{\text{Cl}}/(\text{La}_{\text{Cl}}^{2/3} \times \text{Nd}_{\text{Cl}}^{1/3})$$
 (2)

The choices of a geometric mean or the semi-log interpolation are justified here by the fact that the REE patterns are plotted in semi-log diagrams, where linear abscissa correspond to the number of protons (Z) of REE and ordinates refer to normalized concentrations conventionally displayed in logarithmic scale.

During the 1970s, new developments in sedimentary geochemistry were accompanied by increasing use of reference shale values for normalizing measured REE abundances (e.g., Piper, 1974). In these early studies, Ce\* was interpolated as described above (e.g., Piper (1974) with Equation 2) but also linearly, considering that shale-normalized REE patterns were sometimes plotted using linear scales for both ordinates and abscissae. Additionally, prior to the 1990s and the advent of ICP-MS, REE abundances were mostly measured by neutron activation and Pr concentrations were rarely determined, meaning that Ce\* was generally interpolated between La and Nd (e.g., Elderfield and Greaves, 1981), and occasionally between La and Sm (Toyoda et al., 1990):

162 
$$\text{Ce/Ce*} = 3 \text{ Ce}_{\text{SN}}/(2 \text{ La}_{\text{SN}} + \text{Nd}_{\text{SN}})$$
 (3)

163 
$$Ce/Ce^* = 5 Ce_{SN}/(4 La_{SN} + Sm_{SN})$$
 (4)

It was only with the development of plasma source mass spectrometry in the mid-1980's that the simultaneous determination of the concentrations of all the REE's in a given sample became routine, and the number of analyses that also included Pr increased exponentially. The calculation of the Ce anomaly with Ce\* linearly interpolated between La and Pr<sub>7</sub> is given by the following equation (e.g., De Baar et al., 1983):

169 
$$Ce/Ce^* = 2 Ce_{SN}/(La_{SN} + Pr_{SN})$$
 (5)

However, many natural waters, biogenic or authigenic precipitates, and biogenic sediments display La anomalies, thereby biasing the determination of the Ce anomaly using the above formula. With this approach, the calculation of Ce\* calculated with La abundances using either linear or geometrical interpolations (Fig. 1) resulted in biased results, generating for instance Ce/Ce\* ratios < 1 even in the case of samples devoid of any Ce anomaly. For this reason, Bau and Dulski (1996a) developed a Ce/Ce\* vs. Pr/Pr\* diagram (Fig. 2, with Ce\* linearly interpolated between La and Pr (equation 5), and Pr\* linearly interpolated between Ce and Nd,  $Pr*_{SN}=(Ce_{SN}+Nd_{SN})/2$ ). This plot can be used to identify whether La anomalies are present or not, and whether Ce/Ce\* ratios < 1 correspond to true negative Ce anomalies or not. This diagram, which was also popularized by Webb and Kamber (2000), is frequently used today. We will come back to it later.

Since one cannot properly interpolate Ce\* using La with Pr or Nd abundances, one can instead extrapolate Ce\* and La\* with Pr and Nd abundances (Fig. 1). To avoid ambiguity, we use here the symbols  $X^{*g}$  and  $X^{*l}$  for the geometric (semi-log) and linear extrapolations, respectively, of the theoretical concentrations of X. The following equations give  $La_{SN}^{*g}$ ,  $Ce_{SN}^{*g}$ ,  $La_{SN}^{*l}$ , and  $Ce_{SN}^{*l}$  [e.g., Bolhar et al. (2004) for the linear interpolations and Lawrence et al. (2006) for the geometric interpolations], and of course similar equations can be written for the chondritic normalization:

188 
$$La_{SN}^{*g} = Pr_{SN}^{3}/Nd_{SN}^{2}$$
 (6)

189 
$$Ce_{SN}^{*g} = Pr_{SN}^2/Nd_{SN}$$
 (7)

190 
$$La_{SN}^{*1} = 3 Pr_{SN} - 2 Nd_{SN}$$
 (8)

191 
$$Ce_{SN}^{*l} = 2 Pr_{SN} - Nd_{SN}$$
 (9)

These are the equations used by most teams working today on sedimentary rocks or natural waters, without consensus on whether geometric or linear interpolations should be used or not. Questions arise about whether linear or geometric extrapolation allows for a better estimation of the anomalies. Additionally, do the normalization values (chondrite or shale) lead to different estimates of La and Ce anomalies?

198

199

206

207

208

209

210

211

212

213

214

215

216

217

218

219

220

221

222

223

224

225

196

197

# 2.3/ Which type of extrapolation to select?

- An ideal extrapolation should meet the following criteria: 200
- To allow for the best estimation of La\* and Ce\* concentrations; 201
- 202 - To be universal, i.e. to correctly estimate La\* and Ce\* over the entire compositional range
- encountered in Earth systems; 203
- To be independent of normalization values; in other words, the type of extrapolation must be 204 able to calculate consistent La/La\* or Ce/Ce\* ratios even if the data are normalized to 205 chondritic or shale values.

We have built a database including a total of 286 magmatic rocks, covering a large range of light-REE depletion or enrichment, in order to evaluate the ability of both linear and geometric extrapolations to meet these requirements (Hamelin et al., 2009, 2010; Cordier et al., 2010; Daoud et al., 2010; Pelleter et al., 2014; Barrat et al., 2016; Caroff et al., 2021, Pelleter et al., 2014). The chosen suite of igneous rocks ranges from highly-depleted MORBs to highly-enriched alkaline rocks through lamprophyres, and some evolved rocks [(La/Sm)<sub>CI</sub> = 0.24 - 13.4]. All of these rocks were analyzed using the same procedure and calibration to avoid any potential analytical bias (e.g., Barrat et al., 2012). Based on replicate analyses of standard reference materials and samples (e.g., Barrat et al., 2012, 2016, 2020; Charles et al., 2021), the precision for abundances is typically much better than 5 %, and 3 % for element ratios (one relative standard deviation - RSD), including La/La\* and Ce/Ce\*. Finally, and obviously most importantly, these rocks do not have Ce, nor La anomalies: their "correct" La/La\* or Ce/Ce\* ratios are hence ~1.

We calculated the Ce/Ce\* ratio with Ce\* interpolated geometrically between La and Pr (equation 1) for the rocks in our database. This ratio varies from 0.92 to 1.11 only, with most of the samples exhibiting Ce/Ce\* between 0.95 and 1.05, confirming the lack of significant Ce anomaly (Fig. 2a). The curvature of some of the patterns is largely responsible for this range. In the Ce<sub>SN</sub>/Ce<sub>SN</sub>\* vs. Pr<sub>SN</sub>/Pr<sub>SN</sub>\* plot of Bau and Dulski (1996), where Ce\* and Pr\* are linearly interpolated, only half of studied rock samples are located in the panel

attributed to samples having no Ce and La anomalies; the other half being scattered in an area corresponding to positive anomalies in Ce with either positive or negative La anomaly. This discrepancy is somewhat surprising, as one would have expected the vast majority of studied rocks to fall within the area corresponding to samples having no anomalies. This is not related to the accuracy of the analyses used (< 2% (1 RSD) for the Ce<sub>SN</sub>/Ce<sub>SN</sub>\* and Pr<sub>SN</sub>/Pr<sub>SN</sub>\* – the reader is referred to Barrat et al. (2016, 2020) and Charles et al. (2021) for examples of results obtained on various types of standards). This can be explained, among other things, by the fact that the rocks used in our database display a much greater LREE compositional range relative to those used by Bau and Dulski (1996), and by the poor approximations of Ce\* and Pr\* by linear interpolations when the patterns are highly fractionated (here, the samples that deviate most from the expected area, are the most light-REE-depleted ones).

Next, we use Ce/Nd vs. Pr/Nd plots to compare linear or semi log extrapolations for Ce, normalizing ratios to either chondritic (Fig. 3a) or PAAS values (Fig. 3b). Indeed, the Ce\*g/Nd and Ce\*l/Nd ratios define parabolas and straight lines in these plots, respectively:

$$Ce_{CI}^{*g}/Nd_{CI} = (Pr_{CI}/Nd_{CI})^2$$
 and  $Ce_{SN}^{*g}/Nd_{SN} = (Pr_{SN}/Nd_{SN})^2$  (10)

241 
$$Ce_{CI}^{*l}/Nd_{CI} = 2 (Pr_{Cl}/Nd_{CI}) - 1 \text{ and } Ce_{SN}^{*l}/Nd_{SN} = 2 (Pr_{SN}/Nd_{SN}) - 1$$
 (11)

The rocks of our database having no Ce anomaly, they can be directly compared to these curves or lines. Figure 3 shows directly that the parabolas calculated using a geometric extrapolation reproduce very satisfactorily the Ce/Nd ratios of the rocks over the whole range of Pr/Nd ratios considered in this study. On the other hand, the lines corresponding to the linear extrapolation allow for an acceptable approximation of Ce\* in a given range of values only. Note that in each diagram the parabola and the line are tangent to the point (1,1). If we consider that both interpolations give acceptable results when Ce\*g/Ce\*l is between 1 and 1.05, we can easily calculate that we can then use indifferently one or the other extrapolation only when (Pr/Nd)<sub>CI</sub> or (Pr/Nd)<sub>SN</sub> are between 0.82 and 1.28. Whenever (Pr/Nd)<sub>CI</sub> or (Pr/Nd)<sub>SN</sub> plot outside this range of values, the Ce/Ce\*1 and Ce/Ce\*g ratios diverge, due to a clear underestimation of Ce\*1. The choice of normalization values is not without consequence if linear extrapolation is used. The use of shales instead of chondritic reference values results in a shift towards the left of the diagram [because (Pr/Nd)<sub>SN</sub><(Pr/Nd)<sub>CI</sub>], and, as a consequence, the Ce<sub>Cl</sub>/Ce<sub>Cl</sub><sup>\*1</sup> and Ce<sub>SN</sub>/Ce<sub>SN</sub><sup>\*1</sup> ratios may be very different depending on the Pr/Nd ratios. In this study, this is strikingly illustrated by the fact that the Ce<sub>Cl</sub>/Ce<sub>Cl</sub>\*<sup>1</sup> ratio varies from 0.91 to 1.24 only, while Ce<sub>SN</sub>/Ce<sub>SN</sub>\*1 varies from 0.96 to 5.11. On the other hand,

the  $Ce_{SN}/Ce_{SN}^{*g}$  and  $Ce_{CI}/Ce_{CI}^{*g}$  ratios are perfectly proportional, with  $(Ce_{SN}/Ce_{SN}^{*g})/(Ce_{CI}/Ce_{CI}^{*g})$  being equal to the PAAS  $Ce_{CI}/Ce_{CI}^{*g}$  ratio.

In Fig. 4, we show the Ce/Ce\*g and Ce/Ce\*l ratios obtained after normalization to either chondritic or PAAS values. The diagrams not only indicate that the ranges of values obtained are different, but that the correlations are poor: Ce anomalies calculated in different ways are not comparable.

We followed the same approach for La anomalies. In La/Nd vs. Pr/Nd plots, the La\*g/Nd and La\*l/Nd ratios define cubic curves and straight lines, respectively:

266 
$$La_{CI}^{*g}/Nd_{CI} = (Pr_{CI}/Nd_{CI})^3$$
 and  $La_{SN}^{*g}/Nd_{SN} = (Pr_{SN}/Nd_{SN})^3$  (12)

260

261

262

263

264

265

268

269

270

271

272

273

274

275

276

277

278

279

280

281

282

283

284

285

286

287

288

$$La_{CI}^{*1}/Nd_{CI} = 3 (Pr_{CI}/Nd_{CI}) - 2 \text{ and } La_{SN}^{*1}/Nd_{SN} = 3 (Pr_{SN}/Nd_{SN}) - 2$$
 (13)

As shown above, the curves corresponding to the geometric extrapolation superimpose well on the correlation trends displayed by the rock data used in this study (Fig. 5). The straight lines corresponding to the linear extrapolation only allow a good estimation of the La/Nd ratios for the patterns exhibit little REE decoupling (i.e., when (Pr/Nd)<sub>CI</sub> or (Pr/Nd)<sub>SN</sub> close to 1). Calculation of La anomalies using the linear extrapolation even appears to be inconsistent for many samples. When the data are normalized to chondritic reference values, the linear extrapolation underestimates considerably La\* values, leading to anomalously high La/La\*1 ratios. The situation is much more problematic when data are normalized to PAAS values, especially for the most light-REE depleted samples. Not only does Equation 8 underestimate La<sub>SN</sub>\*, but when  $(Pr/Nd)_{SN} \le 2/3$ , La<sub>SN</sub>\*<sup>1</sup>  $\le 0$  and the La<sub>SN</sub>/La<sub>SN</sub>\*<sup>1</sup> ratio becomes negative or tends toward -∞ when (Pr/Nd)<sub>SN</sub> is just below 2/3. The La<sub>SN</sub>/La<sub>SN</sub>\*1 ratio also tends toward  $+\infty$  when  $(Pr/Nd)_{SN}$  is just above 2/3. These cases are not uncommon on Earth: the La<sub>SN</sub>/La<sub>SN</sub>\*1 ratios calculated with our database range from -216.3 to 103.5, while much smaller ranges around 1 are obtained with other La/La\* calculations (Fig. 6). As for the Ce anomaly, the ratios La<sub>SN</sub>/La<sub>SN</sub>\*g and La<sub>CI</sub>/La<sub>CI</sub>\*g are perfectly proportional, and the La/La\*g and La/La\*1 ratios are not strongly correlated, even when the latter ratios are not aberrant. The range of (Pr/Nd)<sub>SN</sub> ratios for which La anomaly calculations are similar for linear or geometric interpolations (La\*g/La\*1<1.05) is narrow and only between 0.9 and 1.16.

These results demonstrate that La and Ce anomalies calculated using linear extrapolation can lead to biased or even aberrant values in many cases. Moreover, the calculated anomalies are very dependent on the type of normalization used. On the other

hand, the results obtained with our database indicate that the use of geometric extrapolation results in more reliable estimates of La and Ce anomalies, regardless of whether chondrite or shale are used for normalization, and this applies for the whole range of compositions tested in this study.

## 3/ Some examples

The theoretical ground discussed above is illustrated below for a few examples taken from the literature (Table 1). It is not our goal here to discuss the differences obtained between the anomalies calculated for each type of rocks or waters, normalizations and extrapolations. We have chosen to discuss in more detail specific cases of seawater and biogenic carbonates in order to show how the choice of extrapolation can affect data interpretation.

#### 3.1. Carbonates

We selected 6 series of carbonate samples corresponding to microbialites or stromatolites of different ages: the 3.45-Ga-old Strelley Pool stromatolites (Van Kranendonk et al., 2003); the 2.84-Ga-old Mushandike stromatolites (Kamber et al, 2004); the 2.7-Ga-old Fortescue stromatolites (Bohlar and Van Kranendonk, 2007); the 2.52-Ga-old Campbellrand stromatolites (Kamber and Webb, 2001); the late Devonian reefal carbonates from the Lennard Shelf (Nothdurft et al., 2004); and the Holocene microbialites from the Heron Reef (Webb and Kamber, 2000). All data used here were obtained in the same laboratory (ACQUIRE, Brisbane) following similar analytical procedures, and are of excellent quality. These series of samples do not of course cover the full compositional range existing for these types of carbonates, but nevertheless display important variations in various REE signatures.

We plotted these analyses in the Ce/Ce\* vs. Pr/Pr\* diagram of Bau and Dulski (1996a), which can be used, as discussed above, to identify the presence of Ce and La anomalies (Fig. 7a). In this diagram, Ce\* and Pr\* are linearly interpolated between La and Pr and Ce and Nd respectively. All but 3 samples have a Ce/Ce\* ratio<1 when calculated in this way, but only those with a Pr/Pr\* ratio >1 have a true negative Ce anomaly, due to the bias introduced by the La anomalies. We calculated the Ce and La anomalies using both geometrical and linear extrapolation from PAAS-normalized concentrations. The obtained

Ce<sub>SN</sub>/Ce<sub>SN</sub>\*g and Ce<sub>SN</sub>/Ce<sub>SN</sub>\*l ratios are very similar (Fig. 8a) for most samples. Only 4 samples from Strelley Pool deviate significantly from the trend, their Ce<sub>SN</sub>\*l certainly being underestimated. The La<sub>SN</sub>/La<sub>SN</sub>\*g and La<sub>SN</sub>/La<sub>SN</sub>\*l ratios are for most samples reasonably well correlated, but the Holocene samples markedly deviate from the trend with La<sub>SN</sub>/La<sub>SN</sub>\*l ratios greater than La<sup>SN</sup>/La<sub>SN</sub>\*g, and 4 Strelley Pool samples display outlier La<sub>SN</sub>/La<sub>SN</sub>\*l values (=-27.7 to -4.6). These calculations show that in many cases, one can use either linear or geometric extrapolations to estimate La or Ce anomalies, without detecting an anomaly. This is particularly the case when the patterns are little or not fractionated, but linear extrapolation can nevertheless generate artifacts or even aberrant values. To avoid the latter, we recommend using only the geometrical extrapolation to calculate La or Ce anomalies.

Although the Ce/Ce\* vs. Pr/Pr\* diagram proposed by Bau and Dulski (1996a) can bring useful insights for discussing REE patterns and the origin of La and Ce anomalies, it also suffers from inherent drawbacks that are briefly described below:

- The Ce/Ce\* ratio used in this diagram depends on a Ce\* value linearly interpolated between La and Pr. This ratio does not allow a correct quantification of the anomaly, because in addition to the problems related to the linear interpolation, it can be largely biased for the samples with an anomaly in La, as already reported by these authors
- the Pr/Pr\* ratio allows for the detection of samples with positive or negative anomalies in La, but does not allow the quantification of the latter.

For all the above-mentioned reasons, we propose using instead the Ce/Ce\*g vs. La/La\*g diagram, which is better suited for illustrating whether any given sample displays La or Ce anomalies, and which also provides direct quantification of these anomalies. Additionally, the use of the geometric extrapolation ensures that calculated anomalies in this diagram are independent of the type of reference used for normalization, and above all it avoids the calculation of erroneous Ce/Ce\* or La/La\* ratios due to artifacts linked to linear extrapolation. For the carbonate samples selected here, the Ce/Ce\*g vs. La/La\*g diagram shows that these carbonates exhibit a wide range of positive La anomalies, but also allows one to identify a clear distinction between those Archean samples characterized by the absence of marked negative Ce anomalies (Ce/Ce\*g  $\geq$  1), in contrast with the Devonian or Holocene carbonates.

#### 3.2. Seawater

We employ here a previously published REE database for seawater samples (n=1649; Bayon et al., 2020b). We normalized the concentrations with PAAS and examined (Ce/Nd)<sub>SN</sub> and (La/Nd)<sub>SN</sub> vs. (Pr/Nd) <sub>SN</sub> systematics (Fig. 9). The (Pr/Nd)<sub>SN</sub> ratios range from 0.49 to 1.19: 62% of the analyses have (Pr/Nd)<sub>SN</sub> ratios <0.82, and thus have Ce/Ce\*<sup>1</sup> and Ce/Ce\*<sup>2</sup> ratios that differ by more than 5%; 97% of the analyses have (Pr/Nd)<sub>SN</sub> ratios<0.9, and thus have La/La\*<sup>1</sup> and La/La\*<sup>2</sup> ratios that differ by more than 5%. The position of the points with respect to the calculated curves and lines shows unambiguously that the choice of the extrapolation method is critical here, as the La\*<sup>1</sup> or Ce\*<sup>1</sup> concentrations are most often underestimated by the calculation, or even aberrant (La\*<sup>1</sup><0 for many samples).

In order to illustrate the pitfalls of using the linear extrapolation for the case of seawater samples, we chose two hydrographic stations from the China Sea (Alibo and Nozaki, 2000) and the Kerguelen Plateau (Grenier et al., 2018), and examined the vertical profiles of La anomalies calculated by normalizing the data with respect to both CI-chondrite and PAAS, extrapolating La\* linearly or geometrically (Fig. 10). For both stations, La<sub>SN</sub>/La<sub>SN</sub>\*1 ratios are always much larger than those estimated geometrically by normalizing with PAAS or with CI-chondrite: for the first station, La<sub>SN</sub>/La<sub>SN</sub>\*1 ratios are 1.30 to 1.42 times larger than La<sub>SN</sub>/La<sub>SN</sub>\*g ratios, and for the second station they are 1.54 to 3.71 times larger. The La anomalies estimated with the La<sub>SN</sub>/La<sub>SN</sub>\*1 ratios are obviously strongly exaggerated, and these calculations must be rejected. Note that the Laci/Laci\*1 ratios are very close to the Laci/Laci\*g or La<sub>SN</sub>/La<sub>SN</sub>\*g ratios, and similar or identical profiles are obtained with these three ratios. These results adds further support that the use of geometric extrapolation is best suited for calculating La and Ce anomalies relative to the linear extrapolation. The fact that the Lacı/Lacı\*1 ratios are correct here is fortuitous, and is easily explained. The (Pr/Nd)CI ratios are higher than the (Pr/Nd)<sub>SN</sub> ratios, and are then in the range of values for which geometric and linear extrapolations give equivalent results.

## 4/ Conclusion

An extended REE database for a suite of igneous rocks devoid of La and Ce anomalies, was used to investigate the effects of linear and geometric extrapolations for calculating La/La\* and Ce/Ce\* ratios in geological and environmental samples, based on Pr and Nd concentrations. We show that the linear extrapolation only provides reliable estimates

of La\* and Ce\* for a limited range of REE compositions. These calculations can lead in many cases to biased La/La\* and Ce/Ce\* values, which are also critically dependent on the type of normalization used (i.e. chondritic versus shale reference values). The artifacts generated when using the linear extrapolation are illustrated with examples from the literature for seawater and biogenic carbonates, demonstrating that it can lead to misleading interpretations regarding the presence and/or significance of La and Ce anomalies. Finally, we show that the use of geometric extrapolation ensures reliable quantitative calculation of Ce and La anomalies in all samples, which remain unaffected by the type of normalization used. We propose that linear extrapolations be discontinued, and instead recommend the exclusive use of geometric extrapolations to quantify La and Ce anomalies.

# Acknowledgements.

- We thank Karen Johannesson for the editorial handling, Robert Bolhar and an anonymous
- 396 reviewer for their constructive reviews.

# References

399

398

- 400 Alibo, D.S., Nozaki, Y. (2000) Dissolved rare earth elements in the South China Sea: geochemical characterization of the water masses. *J. Geophys. Res.* 105, C12, 28771-28783.
- 402 Anders E. and Grevesse N. (1989) Abundances of the elements: meteoritic and solar. *Geochim. Cosmochim.* 403 *Acta* 53, 197–214.
- Barrat J.A., Zanda B., Moynier F., Bollinger C., Liorzou C., and Bayon G. (2012) Geochemistry of CI chondrites: Major and trace elements, and Cu and Zn isotopes. *Geochim. Cosmochim. Acta* **83**, 79-92.
- Barrat J.A., Dauphas N., Gillet P., Bollinger C., Etoubleau J., Bischoff A., Yamaguchi A. (2016) Evidence from Tm anomalies for non-CI refractory lithophile element proportions in terrestrial planets and achondrites. *Geochim. Cosmochim. Acta*, **176**, 1-17.

409 410

Barrat J.A., Bayon G., Wang X., Le Goff S., Rouget M.L., Gueguen B., Ben Salem B. (2020) A new chemical separation procedure for the determination of rare earth elements and yttrium abundances in carbonates by ICP-MS. *Talanta* **219**, 121244.

412 413

411

Bau, M., Dulski, P. (1996a) Distribution of yttrium and rare-earth elements in the Penge and Kuruman iron formations, Transvaal Supergroup, South Africa. *Precambrian Res.* **79**, 37–55.

416

Bau, M., Dulski, P. (1996b) Anthropogenic origin of positive gadolinium anomalies in river waters. *Earth Planet. Sci. Lett.* **143**, 245-255.

419

Bau M., Romer R.L., Lüders V., Dulski P. (2003) Tracing element sources of hydrothermal mineral deposits:
REE and Y distribution and Sr-Nd-Pb isotopes in fluorite from MVT deposits in the Pennine Orefield,
England. *Mineralium Deposita* 38, 992-1008.

423 424

425

Bau M., Koschinsky A. (2009) Oxidative scavenging of cerium on hydrous Fe oxide: Evidence from the distribution of rare earth elements and yttrium between Fe oxides and Mn oxides in hydrogenetic ferromanganese crusts. *Geochem. J.* **43**, 37–47.

426 427

Bau M., Schmidt K., Pack A., Bendel B., Kraemer D. (2018) The European Shale: an improved data set for normalization of rare earth elements and yttrium concentrations in environmental and biological samples from Europe. *Applied Geochem.* **90**, 142-149.

430 431

Bayon G., Lambert T., Vigier N., De Dekker P., Freslon N., Jang K., Larkin C.S., Piotrovski A.M., Tachikawa K., Thollon M., Tipper E.D. (2020a) Rare earth element and neodymium isotope tracing of sedimentary rock weathering. *Chem. Geol.* **553**, 119794.

435

Bayon G., Lemaitre N., Barrat J.A., Wang X., Feng D., Duperron S. (2020b) Microbial utilization of rare earth elements at cold seeps related to aerobic methane oxidation. *Chem. Geol.* **555**, 119832.

438

Bellefroid E.J., Hood A.V.S., Hoffman P.F., Thomas M.D., Reinhard C.T., Planavsky N.J. (2018) Constraints
 on Paleoproterozoic atmospheric oxygen levels. *PNAS* 115, 8104-8109.

441

Bolhar R., Van Kranendonk M.J. (2007) A non-marine depositional setting for the northern Fortescue Group,
Pilbara Craton, inferred from trace element geochemistry of stromatolitic carbonates. *Precambrian Res.*155, 229-250.

445

Bolhar R., Kamber B.S., Moorbath S., Fedo C.M., and Whitehouse, M.J. (2004). Characterisation of early Archaean chemical sediments by trace element signatures. *Earth Planet. Sci. Lett.* **222**, 43-60.

448

Burnham A.D., Berry A.J. (2012) An experimental study of trace element partitioning between zircon and melt as a function of oxygen fugacity. *Geochim. Cosmochim. Acta* **95**, 196-212.

Byrne R.H., Kim K.H. (1990) Rare earth element scavenging in seawater. *Geochim. Cosmochim. Acta* **54**, 2645-453 2656.

454

457

466

484

493

497

- Byrne R.H., Liu X., and Schijf J. (1996). The influence of phosphate coprecipitation on rare earth distributions in natural waters. *Geochim. Cosmochim. Acta* **60**, 3341–3346.
- Caroff M., Barrat J.A., Le Gall B. (2021) Kersantites and associated intrusives from the type locality (Kersanton), Variscan Belt of Western Armorica. *Gondwana Res.* **98**, 46-62.
- Cordier C., Benoit M., Hémond C., Dyment J., Le Gall B., Briais A., Kitazawa M. (2010), Time scales of melt
   extraction revealed by distribution of lava composition across a ridge axis. *Geochem. Geophys. Geosyst.* 11, O0AC06.doi:10.1029/2010GC003074.
- Charles C., Barrat J.A., Pelleter E. (2021) Trace element determinations in Fe–Mn oxides by high resolution ICP-MS after Tm addition. *Talanta* **233**, 122446.
- Class C., Le Roex A.P. (2008) Ce anomalies in Gough Island lavas Trace element characteristics of a recycled
   sediment component. *Earth Planet. Sci. Lett.* 265, 475-486.
- Coryell C.D., Chase J. W., and Winchester J.W. (1963) A procedure for geochemical interpretation of terrestrial
   rare-earth abundance patterns. *J. Geophys. Res.* 68, 559-566
- Cotruvo, jr., J.A. (2019) The chemistry of lanthanides in biology: recent discoveries, emerging principles, and technological applications. *ACS Cent. Sci.* 5, 1496–1506.
- Daoud M.A., Maury R.C., Barrat J.A., Taylor R.N., Le Gall B., Guillou H., Cotten J., Rolet J. (2010) A LREEdepleted component in the Afar plume: further evidence from Quaternary Djibouti basalts. *Lithos* **114**, 327-336.
- Dauphas N, Pourmand A. (2015) Thulium anomalies and rare earth element patterns in meteorites and Earth:
  Nebular fractionation and the nugget effect. *Geochim. Cosmochim. Acta* **163**, 234-261.
- De Baar H.J.W., Bacon M.P., Brewer P.G. (1983) Rare-earth distributions with a positive Ce anomaly in the Western North Atlantic Ocean. *Nature* **301**, 324-327.
- De Baar H.J.W., Bacon M.P., Brewer P.G., and Bruland, K.W. (1985) Rare earth elements in the Pacific and Atlantic oceans. *Geochim. Cosmochim. Acta* **49**, 1943–1959.
- Elderfield H. (1988) The oceanic chemistry of the rare-earth elements. *Philosophical Transaction of the Royal Society of London. Series A, Mathematical and Physical Sciences* **325**, 105-126.
- Elderfield H., and Greaves M.J. (1981) Negative cerium anomalies in the rare earth element patterns of oceanic ferromanganese nodules. *Earth Planet. Sci. Lett.* **55**, 163-170.
- Elderfield H., Pagett R. (1986) Rare Earth elements in Ichthyoliths: variations with redox conditions and depositional environment. In: Riley, J.P. (Ed.), Science of the Total Environment. Elsevier, Amsterdam, pp. 175–197.
- 498 German C.R., Elderfield H. (1990) Application of the Ce anomaly as a paleoredox indicator: The ground rules.
   499 *Paleoceanography* 5, 823-833.
- German C.R., Holliday B.P., Elderfield H. (1991) Redox cycling of rare earth elements in the suboxic zone of the Black Sea. *Geochim. Cosmochim. Acta* **55**, 3553-3558.
- 502 Goldberg E.D., Koide M., Schmitt R.A. (1963) Rare earth distributions in the marine environment. *J. Geophys.* 503 *Res.* **68**, 4209-4217.
- Grenier M, Garcia-Solsona E., Lemaitre N., Trull T.W., Bouvier V., Nonnotte P., Van Beek P., Souhaut M.,
  Lacan F., Jeandel C. (2018) Differentiating lithogenic supplies, water mass transport, and biological
  processes on and off the Kerguelen Plateau using rare earth element concentrations and neodymium
  isotopic compositions. *Frontiers in Marine Science* 5, 426.

510 Hamelin C., Seitz H.M., Barrat J.A., Dosso L., Maury R.C., Chaussidon M. (2009) A low δ<sup>7</sup>Li lower crustal 511 component: evidence from an alkalic intraplate volcanic series (Chaîne des Puys, French Massif Central). 512 Chem. Geol. 266, 214-226.

513 514 Hamelin C., Dosso L., Hanan B., Barrat J.A., Klingelhöfer F., Ondréas H. (2010) Sr-Nd-Hf isotopes along the Pacific Antarctic Ridge from 41 to 53°S. Geophys Res. Lett. 37, L10303, doi:10.1029/2010GL042979.

515 516

Hein J. R., Zierenberg R. A., Maynard J. B., and Hannington, M. D. (2007). Multifarious barite-forming environments along a rifted continental margin, Southern California Borderland. Deep-Sea Res. II 54, 1327-1349.

518 519

517

520 Kamber B.S. and Webb G.E. (2001) The geochemistry of late Archaean microbial carbonate: Implications for 521 ocean chemistry and continental erosion history. Geochim. Cosmochim. Acta 65, 2509–2525.

522 523

Kamber B.S., Bolhar R., Webb G.E. (2004). Geochemistry of late Archaean stromatolites from Zimbabwe: evidence for microbial life in restricted epicontinental seas. Precambrian Res. 132, 379-399.

524 525

526

529

530

Kulaksiz S., Bau M. (2007) Contrasting behaviour of anthropogenic gadolinium and natural rare earth elements in estuaries and the gadolinium input into the North Sea. Earth Planet. Sci. Lett. 260, 361-371.

527 528

Kulaksiz S., Bau M. (2013) Anthropogenic dissolved and colloid/nanoparticle-bound samarium, lanthanum and gadolinium in the Rhine River and the impending destruction of the natural rare earth element distribution in rivers. Earth Planet. Sci. Lett. 362, 43-50.

531 532 533

Lawrence M.G., Greig A., Collerson K.D., Kamber B.S. (2006) Rare Earth Element and Yttrium variability in South East QueenslandWaterways. *Aquat. Geochem.* **12**, 39–72.

534 535 536

Le Goff S., Barrat J.A., Chauvaud L., Paulet Y.M., Gueguen B., Ben Salem D. (2019) Compound-specific recording of gadolinium pollution in coastal waters by great scallops. Scientific Reports 9, 8015.

537 538 539

Ma L., Dang D.H., Wang W., Evans R.D., Wang W.X. (2019) Rare earth elements in the Pearl River Delta of China: Potential impacts of the REE industry on water, suspended particles and oysters. Environmental Pollution 244, 190-201

541 542 543

540

Masuda A. (1962) Regularities in variation of relative abundances of Lanthanide elements and an attempt to analyse separation index patterns of some minerals. J. Earth Sci. Nagoya Univ. 10, 173-187.

544 545 546

Merschel G., Bau M. (2015) Rare earth elements in the aragonitic shell of freshwater mussel Corbicula fluminea and the bioavailability of anthropogenic lanthanum, samarium and gadolinium in river water. Sci. Total Environment 533, 91-101.

547 548 549

550 Nance W.B., Taylor S.R., (1976) Rare earth element patterns and crustal evolution-I. Australian post-Archean 551 sedimentary rocks. Geochim. Cosmochim. Acta 40, 1539–1551.

552

553 Northdurft L.D., Webb G.E., Kamber B.S. (2004) Rare earth element geochemistry of Late Devonian reefal 554 carbonates, Canning Basin, Western Australia: confirmation of a seawater REE proxy in ancient limestones Geochim. Cosmochim. Acta 68, 263-283.

555

556 Palme H., Lodders K. and Jones A. (2014) Solar System Abundances of the Elements. In: Holland H.D. and 557 Turekian K.K. (eds.) Treatise on Geochemistry, Second Edition, vol. 2, pp. 15-36. Oxford: Elsevier.

558

559 Piper D.Z. (1974a) Rare earth elements in the sedimentary cycle: a summary. Chem. Geol. 14, 285--304.

560

561 Piper D.Z. (1974b) Rare earth elements in ferromanganese nodules and other marine phases. Geochim. Cosmochim. Acta 38, 1007–1022. 562

563

564 Pelleter A.A., Caroff M., Cordier C., Bachelery P., Nehlig P., Debeuf D., Arnaud N. (2014) Melilite-bearing 565 lavas in Mayotte (France): An insight into the mantle source below the Comores. Lithos 208-209, 281-297.

Pol, A. *et al.* (2014) Rare earth metals are essential for methanotrophic life in volcanic mudpots. *Environmental Microbiology* **16**, 255–264.

- Pourmand, A., Dauphas, N., Ireland, T.J. (2012) A novel extraction chromatography and MC-ICP-MS technique
   for rapid analysis of REE, Sc and Y: Revising CI-chondrite and Post-Archean Australian Shale (PAAS)
   abundances. *Chem. Geol.* 291, 38-54.
- 574 Semrau, J.D., DiSpirito, A.A., Gu, W., Yoon, S. (2018) Metals and Methanotrophy. *Appl Environ Microbiol* **84**, e02289-17.
- 577 Shimizu H., Sawatari H., Kawata Y., Dunkley P.N., Masuda A. (1992) Ce and Nd isotope geochemistry on island arc volcanic rocks with negative Ce anomaly: existence of sources with concave REE patterns in the mantle beneath the Solomon and Bonin island arcs. *Contrib. Mineral. Petrol.* **110**, 242-252.
- 581 Smythe D.J., Brenan J.M. (2016) Magmatic oxygen fugacity estimated using zircon-melt partitioning of cerium. *Earth Planet. Sci. Lett.* **453**, 260-266.
  - Toyoda K., Nakamura Y., Masuda A., 1990. Rare earth elements of Pacific pelagic sediments. *Geochim. Cosmochim. Acta* **54**, 1093–1103.
  - Tostevin R., Graham A., Shields G.A., Tarbuck G.M., Tianchen He, Clarkson M.O., Wood R.A. (2016) Effective use of cerium anomalies as a redox proxy in carbonate-dominated marine settings. *Chem. Geol.* **438**, 146-162
  - Trail D., Watson E.B., Tailby N.D. (2012) Ce and Eu anomalies in zircon as proxies for the oxidation state of magmas. *Geochim. Cosmochim. Acta* **97**, 70-87.
  - Valdés-Vilchis S., Sànchez-Beristain F., Bernal J.P., Juàrez-Aguilar E.A. (2021) Rare Earth Elements and Yttrium (REE+Y) patterns in recent *Anadara brasiliana* shells from Playa Norte, Barra de Cazones (Veracruz, Mexico): Evidence of anthropogenic contamination linked to river output? *Journal of South American Earth Sciences* 110, 103368.
  - Van Kranendonk M.J., Webb G.E., Kamber B.S. (2003) New geological and trace element evidence from 3.45 Ga stromatolitic carbonates in the Pilbara Craton: support of a marine, biogenic origin and for a reducing Archaean ocean. *Geobiology* **1**, 91-108.
  - Wallace M.W., Hood A.S., Shuster A., Greig A., Planavsky N.J., Reed C.P. (2017) Oxygenation history of the Neoproterozoic to early Phanerozoic and the rise of land plants. *Earth Planet. Sci. Lett.* **466**, 12-19.
  - Wang X., Barrat J.A., Bayon G., Chauvaud L., Feng D. (2020) Lanthanum anomalies as fingerprints of methanotrophy. *Geochem. Persp. Let.* **14**, 26–30.
- Webb G.E., Kamber B.S. (2000) Rare earth elements in Holocene reefal microbialites: a new shallow seawater proxy. *Geochim. Cosmochim. Acta* **64**, 1557-1565.

	CI-chondrite	CI-chondrite	PAAS	PAAS	MORB	fluorite	Mn-nodule	stromatolite	BIF
ref.	1	1	2	2	3	4	5	6	6
#					PI 18-06	CT2a	GSMC-1	2-9-11a	IF-G
unit	μg/g	μmol/kg	μg/g	μmol/kg	μg/g	μg/g	μg/g	ng/g	ng/g
Y	1.56	17.55	32.2	362		35.6	259	1014.2	9135
La	0.235	1.692	44.75	322.2	0.504	0.54	326	56.3	2706
Ce	0.600	4.28	87.29	623.0	2.13	0.89	1246	85.9	3902
Pr	0.091	0.646	10.1	71.68	0.464	0.23	68.74	14	430
Nd	0.464	3.22	36.98	256.4	3.00	1.42	283	77.7	1731
Sm	0.153	1.018	6.908	45.94	1.30	0.57	58.4	45.9	399
Eu	0.0586	0.386	1.188	7.818	0.582	0.18	14.36	27.1	362
Gd	0.206	1.31	5.958	37.89	2.18	1.33	61.68	89.6	667
Tb	0.0375	0.236	0.894	5.625	0.432	0.23	9.53	13.6	112
Dy	0.254	1.563	5.272	32.44	3.16	1.71	56.48	70	791
Но	0.0566	0.343	1.078	6.536	0.743	0.4	11.58	17.7	207
Er	0.166	0.992	3.094	18.50	2.23	1.15	31.93	54.8	619
Tm	0.0262	0.155	0.468	2.770		0.13			
Yb	0.168	0.971	3.028	17.50	2.28	0.61	29.2	48	580
Lu	0.0246	0.141	0.438	2.503	0.34	0.07	4.26	8.6	90.4
La <sub>CI</sub> /La <sub>CI</sub> *g	1	1	0.88	0.88	0.68	1.33	1.20	1.84	1.52
Ceci/Ceci* g	1	1	0.94	0.94	0.88	0.71	2.22	1.01	1.09
La <sub>CI</sub> /La <sub>CI</sub> * <sup>1</sup>	1	1	1.10	1.10	0.91	1.57	1.33	1.89	1.71
Ceci/Ceci*1	1	1	1.02	1.02	0.95	0.74	2.31	1.02	1.14
La <sub>SN</sub> /La <sub>SN</sub> * g	1.13	1.13	1	1	0.76	1.51	1.35	2.09	1.72
Ce <sub>SN</sub> /Ce <sub>SN</sub> * g	1.06	1.06	1	1	0.94	0.75	2.36	1.08	1.15
La <sub>SN</sub> /La <sub>SN</sub> * 1	2.71	2.71	1	1	-0.46	-1.42	1.42	-28.69	1.77
Cesn/Cesn*1	1.26	1.26	1	1	2.27	1.43	2.40	1.47	1.17

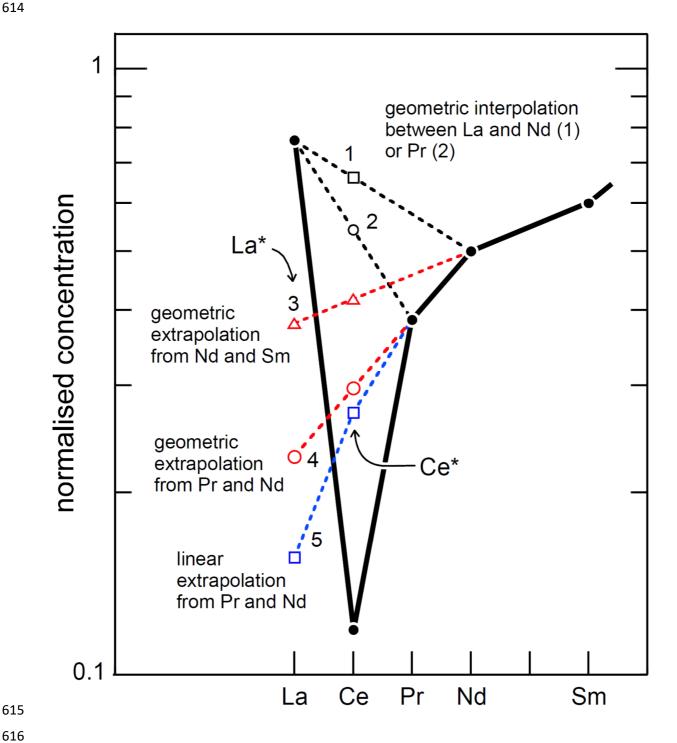


Figure 1. The different ways of calculating La\* and Ce\*.

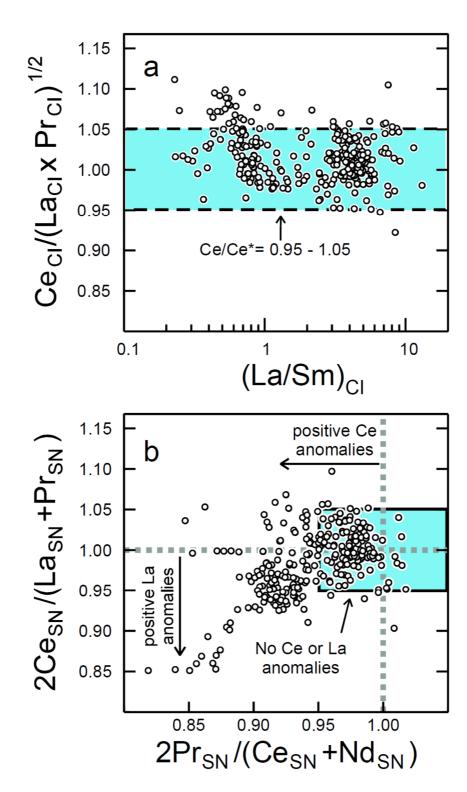


Figure 2. The 286 samples devoid of La and Ce anomalies of the database used to compare linear and geometric extrapolations for calculating La\* and Ce\*, are plotted in a Ce/Ce\* vs. La/Sm plot (a) and in the Ce/Ce\* vs. Pr/Pr\* plot of Bau and Dulski (1996a) (b). Ce\* is geometrically interpolated in (a) and linearly interpolated in (b).

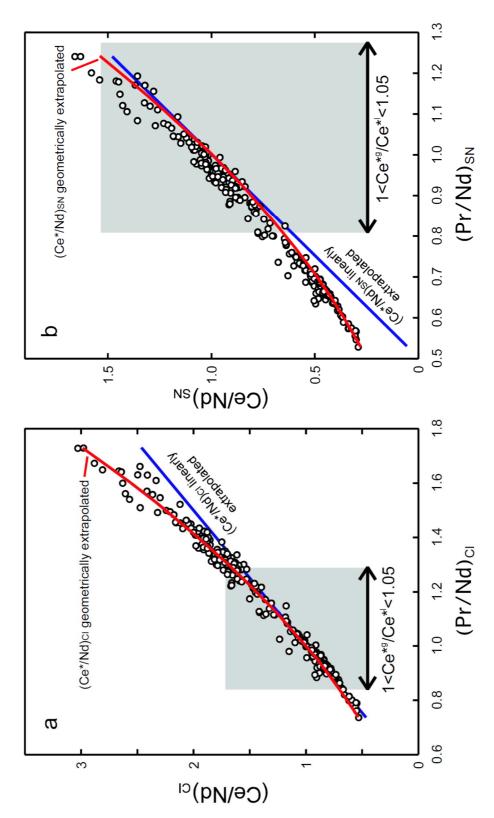


Figure 3. Ce/Nd vs. Pr/Nd plots for the samples devoid of La and Ce anomalies of our database, used here to compare linear (blue line) and geometric (red parabola) extrapolations for Ce\*. The data are normalized with CI chondrite (a) or with PAAS (b).

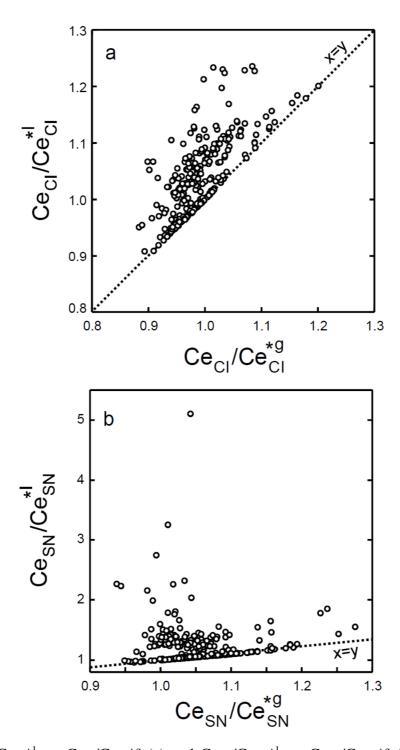


Figure 4.  $Ce_{CI}/Ce_{CI}^{*1}$  vs.  $Ce_{CI}/Ce_{CI}^{*g}$  (a) and  $Ce_{SN}/Ce_{SN}^{*1}$  vs.  $Ce_{SN}/Ce_{SN}^{*g}$  (b) plots for the samples devoid of Ce anomalies used to test the different extrapolations. Notice the ranges of values obtained.

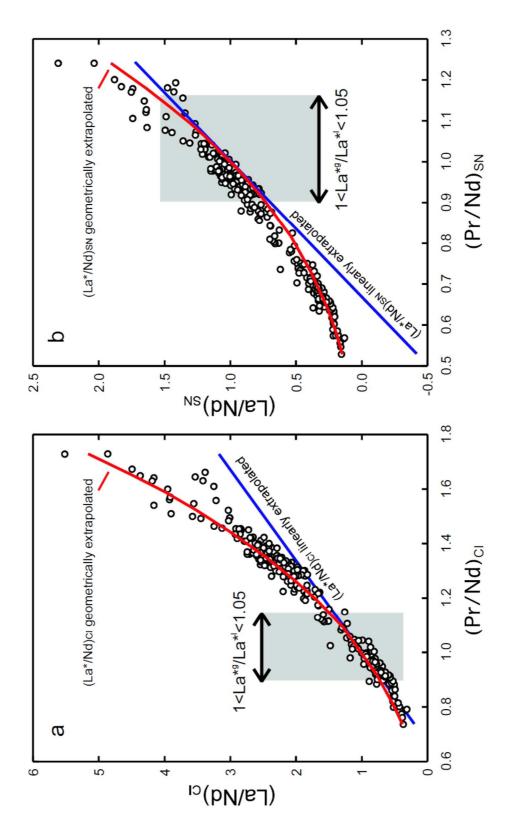


Figure 5. La/Nd vs. Pr/Nd plots for the samples devoid of La and Ce anomalies of our database used to compare linear (blue line) and geometric (red cubic) extrapolations for Ce\*. The data are normalized with CI chondrite (a) or with PAAS (b).

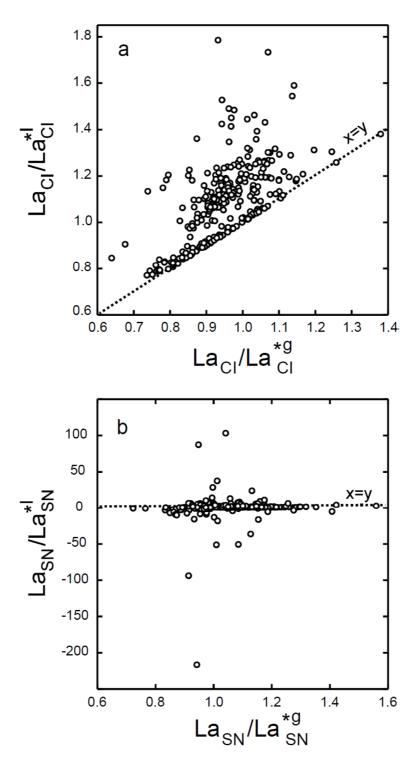


Figure 6.  $La_{CI}/La_{CI}^{*1}$  vs.  $La_{CI}/La_{CI}^{*g}$  (a) and  $La_{SN}/La_{SN}^{*1}$  vs.  $La_{SN}/La_{SN}^{*g}$  (b) plots for the samples devoid of La anomalies used to test the different extrapolations. Notice the ranges of values obtained.

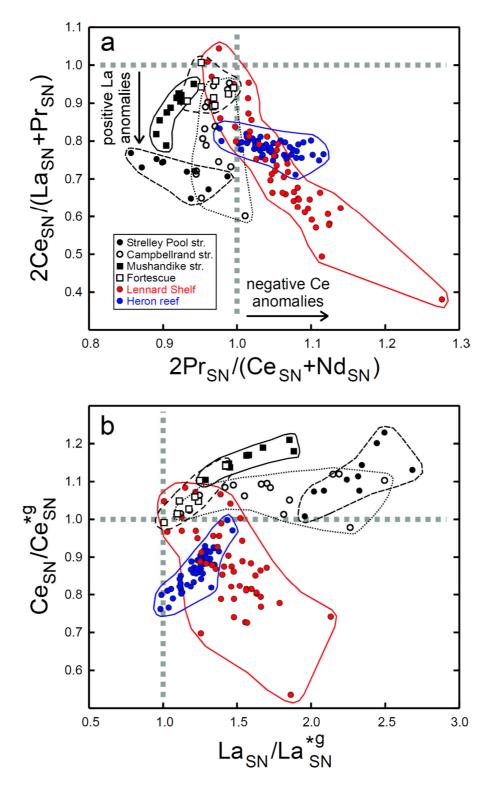


Figure 7. Selected carbonates are plotted in the Ce/Ce\* vs. Pr/Pr\* plot of Bau and Dulski (1996a) (a) where Ce\* and Pr\* are linearly interpolated, and in a Ce/Ce\* vs. La/La\* where Ce\* and La\* are geometrically extrapolated. Notice the different Ce/Ce\* ranges obtained. See text for more details.

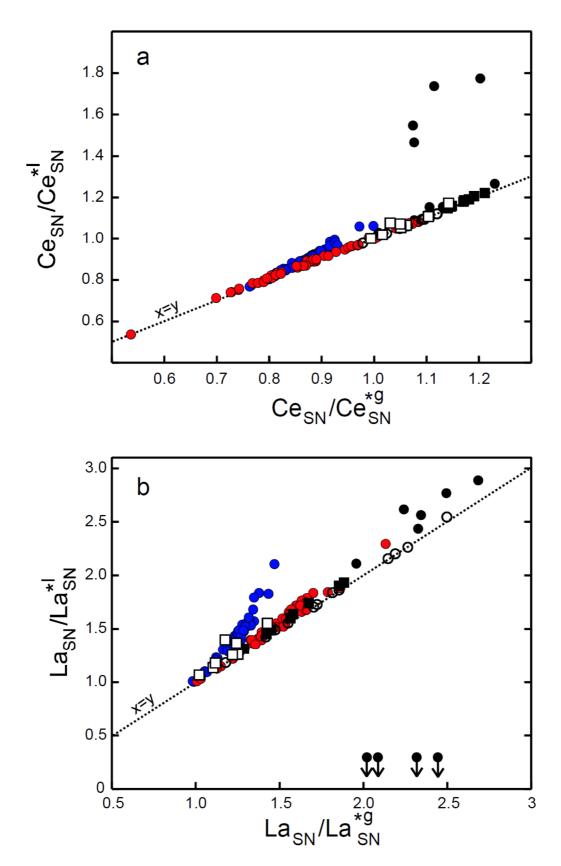


Figure 8.  $Ce_{SN}/Ce_{SN}^{*l}$  vs.  $Ce_{SN}/Ce_{SN}^{*g}$  (a) and  $La_{SN}/La_{SN}^{*l}$  vs.  $La_{SN}/La_{SN}^{*g}$  (b) plots for the selected carbonates used to test the different extrapolations (same caption as Fig. 7).

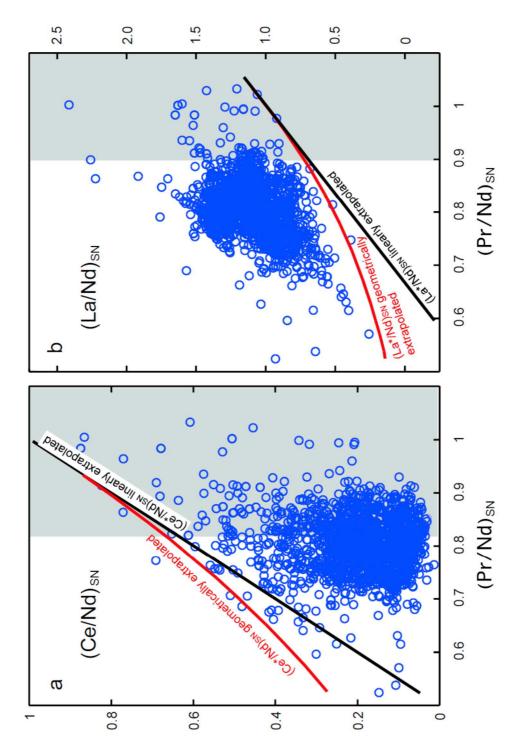


Figure 9.  $(Ce/Nd)_{SN}$  (a) and  $(La/Nd)_{SN}$  (b) vs.  $(Pr/Nd)_{SN}$  plots for seawater. The shaded areas correspond to the range in Pr/Nd ratios for which the linear and geometric extrapolations are similar (1<X\*I/X\*g<1.05). A large proportion of the samples are outside the ranges where the linearly extrapolated Ce\* or La\* are equivalent to the geometrically extrapolated ones.

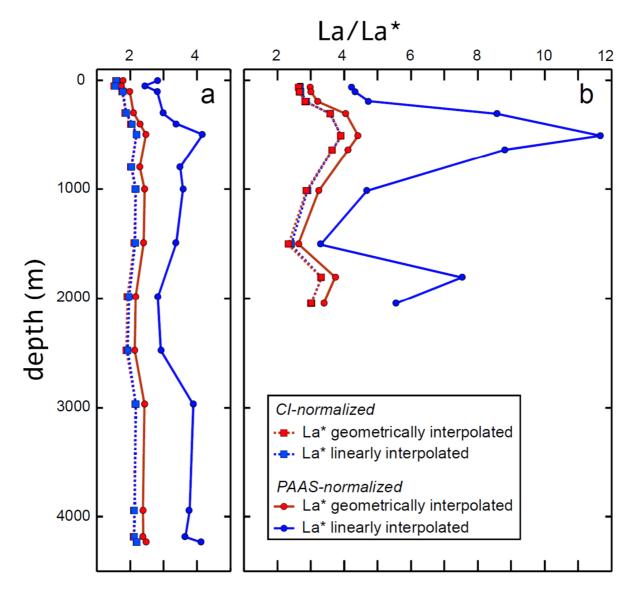


Figure 10. Vertical profiles of La anomaly (La/La\*) at (a) station PA-11, South China Sea (February 11 and 12, 1997;  $15^{\circ}22'N$ ,  $115^{\circ}17'E$ ; depth: 4240 m; Alibo and Nozaki, 2000) and at (b) the meander core station E1, Kerguelen Plateau (October 30, 2011;  $72.178^{\circ}E$ ,  $48.498^{\circ}S$ ; depth: 2058 m, Grenier et al., 2018). La/La\* was calculated linearly and geometrically with data normalized with chondrite or with PAAS. The La<sub>SN</sub>/La<sub>SN</sub>\*1 values are always much larger than the other La/La\* estimates, and is an artifact.