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## A novel pectic polysaccharide-based hydrogel derived from okra (*Abelmoschus esculentus* L. Moench) for chronic diabetic wound healing

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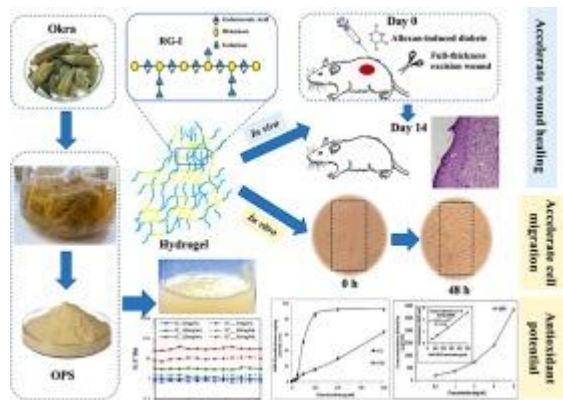
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### Abstract :

Hydrogels based on natural polysaccharides represent a growing group of suitable biomaterials for the elaboration of effective wound healing dressings, especially for the treatment of chronic wounds. This work was intended to prepare a polysaccharide-based hydrogel for diabetic wound healing which would help maintain the well-being of diabetes and improve their quality of life. For this purpose, a pectic polysaccharide (OPS) was extracted and purified, for the first time, from Tunisian okra pods and its physicochemical and rheological features, antioxidant and in vivo and in vitro wound healing activities were investigated. OPS, an acidic polysaccharide with a molecular weight of  $3.28 \times 10^6$  Da and a polydispersity index of 1.03, was mainly composed of galactose (24.45%), galacturonic acid (24.6%) and rhamnose (18.25%). Combined with FT-IR and NMR analyses, it consisted of a pectic rhamnogalacturonan I (RG-I) structure with galactan side chains. The OPS demonstrated antioxidant potential, gelling ability, cytocompatibility properties, non-cytotoxicity and cell migration and proliferation promoting activities, which met the requirements for wound dressings. Then, the in vivo cutaneous wound healing effect of OPS-based hydrogel was investigated using an alloxan-induced diabetic rat model, and results showed that it significantly accelerated the wound healing process by acting in the acceleration of the recovery of the dermis and inducing more blood vessels formation and tissue granulation.

Overall, these results provide new insights into the development of a promising and effective okra pectin-based hydrogel for the treatment of chronic diabetic wounds.

## Graphical abstract



## Highlights

► RG-I pectic polysaccharide (OPS) was extracted and purified from okra pods. ► Physico-chemical features of OPS were elucidated by FTIR, GC, SEC-MALLS and NMR. ► OPS solution exhibited a gel-like behavior. ► OPS demonstrated notable antioxidant capacity and stimulated cell migration and proliferation. ► OPS hydrogel effectively accelerated wound healing in alloxan-induced diabetic rats.

**Keywords** : Okra polysaccharide, Pectin, Physicochemical features, rheological properties, Antioxidant potential, hydrogel, Wound healing, Scratch assay, Cell migration, Diabetic wound.

## 1 **1. Introduction**

2 Okra (*Abelmoschus esculentus L.*), also commonly known as gumbo or lady's finger, is an  
3 annual vegetable belonging to the *Malvaceae* family. Native of Africa, this flowering plant is  
4 mainly planted in tropical, subtropical and warm temperate regions of Southern Europe, Middle  
5 East, America and Asia [1]. Apart supplying common nutrient like vitamins and minerals, Okra  
6 fruit is also a rich source of natural food ingredients displaying promising nutritional, functional  
7 and biological characteristics, among which polysaccharides (PSs) (~10.38–16.89% by weight)  
8 have drawn a great deal of attentions [2-5]. Indeed, PSs derived from Okra have shown a broad  
9 spectrum of biological activities, including antioxidant, immunomodulating, analgesic, anti-  
10 hyperglycemic, anti-inflammatory, antihyperlipidemic, anti-fatigue and intestinal functions [6-  
11 10]. Among water-soluble polysaccharides, okra pods mainly contain pectin and, to a lower  
12 extent, xyloglucan and glucuronoxylan [11]. Its complex structure together with its location  
13 between cellulose microfilaments makes pectin extraction from the cell wall difficult [12].  
14 Pectin is an acidic complex macromolecule which consists of linear and ramified regions. The  
15 linear homogalacturonan (HG) region consists of (1→4)-linked  $\alpha$ -d-galacturonic acid (GalAp)-  
16 units, while the ramified region is represented by different hetero-polysaccharides, type I  
17 rhamnogalacturonan (RG-I) and type II rhamnogalacturonan (RG-II) [13]. Pectins have  
18 attracted a great attention as natural ingredients in the food industry for their thickening, gelling  
19 and stabilizing properties [14,15]. Indeed, the application of pectins in food is mainly based on  
20 their hydrocolloid properties. Their ability to form gels, the gelation mechanism and rheological  
21 properties of pectin gels depend on the structural features of these polysaccharides.  
22 Furthermore, some of them gain more and more interest as possible health promoting  
23 polysaccharides [16,17]. Recently, owing to its advantages of cytocompatibility, nontoxicity,  
24 moisturizing property and biodegradability, commercial pectin has been investigated for using  
25 in several wound healing applications, including composite wound dressing, scaffolding and

26 skin protection [18-20]. Natural and modified pectins demonstrated several advantages to be  
27 used for wound healing, including sufficient water uptake, acidic character of pectin  
28 macromolecules, and high affinity to cationic growth factors [18]. Such properties enhance  
29 removal of exudates, repulsion of bacteria and thus improve the healing process [21]. On the  
30 other hand, with the advances of technology, hydrogels dressings have shown great potentials  
31 for the treatment of both acute and chronic wounds [22]. Particularly, polysaccharide-based  
32 hydrogels which are made up with natural biomaterials that are biodegradable and  
33 biocompatible present unique features as wound dressings and are widely applicable in clinical  
34 practices. They share not only common characteristics of hydrogels such as excellent tissue  
35 adhesion, swelling, water absorption, etc., but also other properties, including antioxidant,  
36 antimicrobial and anti-inflammatory activities, to accelerate wound re-epithelialization, mimic  
37 skin structure and induce skin regeneration [23-26].

38 As far as we know, no studies were focused on pectic polysaccharide extracted from Tunisian  
39 okra and its healing activity on diabetic wounds. Indeed, diabetic ulcers which are a common  
40 type of chronic non-healing wound, are the major complication of trauma, thus affecting  
41 patients' quality of life and frequently causing the need of amputation [27]. The difficult-to-  
42 heal fact of diabetic wounds has been reported to be associated with some issues, including  
43 bacterial infection leading to uncontrollable inflammation, high oxidative stress mainly induced  
44 by excessive reactive oxygen species (ROS), impaired angiogenesis, vascular damage leading  
45 to nutrient and oxygen supply disorders to the wound site, etc. [28]. Therefore, besides strictly  
46 controlling blood glucose level, it is also very important to improve the diabetic wound local  
47 microenvironment and provide favorable conditions for wound healing. To achieve this goal,  
48 advanced research is carrying out to develop multifunctional wound dressing materials able to  
49 address simultaneously different aspects of the wound healing process. Novel wound dressings

50 are thus constantly being investigated. Their efficiency to achieve fast healing at reasonable  
51 cost together with minimal inconvenience to the patients is a common targeted feature [29-32].  
52 Therefore, the development of novel wound dressing materials capable to decrease both risks  
53 of bacterial infections and oxidative stress, and accelerate the tissue regeneration process is  
54 utmost important to prevent long-term healing problems. For this purpose, in this study we  
55 purified a pectic polysaccharide from okra, assessed its physicochemical features and  
56 rheological properties, and evaluated, through *in vivo* and *in vitro* studies, its bioactivity and  
57 capacity, in form of hydrogel, for promoting wound healing in alloxan-induced diabetic rat  
58 model.

## 59 **2. Material and methods**

### 60 **2.1. Plant material, chemicals and animals**

61 Mature okra (*Abelmoschus esculentus L.*) pods (5–10 cm in length) were collected during the  
62 month of October 2020 from a local market of Sfax city, Tunisia. Collected samples were  
63 carefully assorted, cleaned, washed and further air-dried for 2 days before being packaged and  
64 stored at – 20° C until use. The used standards and reagents are: MTT (3-(4,5-dimethylthiazol-  
65 2-yl)-2,5-diphenyltetrazoliumbromide), DPPH (2,2-phenyl-1-picrylhydrazyl), trichloroacetic  
66 acid (TCA), dimethyl sulfoxide (DMSO) and Gallic Acid (GA) were purchased from Sigma-  
67 Aldrich company (St. Louis, MO, USA). Neutrase P1236 (from *Bacillus amyloliquefaciens*)  
68 and Purafect 2000<sup>E</sup> proteases were bought from Sigma-Aldrich (St. Louis, MO, USA) and  
69 Genencor International (Etats-Unis), respectively. Penicillin-streptomycin, fetal bovine serum  
70 (FBS), and Dulbecco's modified eagle medium (DMEM) were purchased from Life  
71 Technologies (Gibco, Paisley, UK). All other chemicals and solvent used were of analytical  
72 grade. B16 4A5 cells (RBRC-RCB0557) showing fibroblast-like characteristics, and Human  
73 embryonic kidney cells (HEK-293) (ATCC, CRL-1573, American Type Culture Collection,  
74 Manassan, VA) used in this study were prepared as described by Maalej et al. [33,34].

75 For the *in vivo* experiments, alloxan (CAS Number: 2244-11-3) was purchased from Sigma-  
76 Aldrich. « *Cytol Centella cream* ® » is a synthetic oil in water emulsion drug based on a natural  
77 titrated extract of *Centella asiatica* used as a therapeutic agent in wound healing. A total of 18  
78 male Wistar rats, weighting  $250 \pm 40$  g, were obtained from the Faculty of Sciences of Gabes-  
79 Tunisia (FSG). All animals were housed in an environmentally-controlled room, at constant  
80 temperature ( $22 \pm 1$  °C) with a 12/12 h light/dark cycle. All rats had free access to water and  
81 standard laboratory food. Procedures and animal comfort were controlled by the International  
82 Guidelines for Animal Care.

## 83 **2.2. Purification of okra polysaccharide (OPS)**

### 84 **2.2.1. Extraction of the crude OPS**

85 The okra pods (including seeds) were sliced and then mixed 10 times with distilled water, and  
86 were extracted at 70 °C, three times, for 2 h each time, under the same conditions. After  
87 centrifugation (6000 rpm, 15 min) to remove the debris of water insoluble materials, the  
88 recovered supernatants were pooled and concentrated to one third of the original liquid under a  
89 vacuum at 50 °C. The concentrate was precipitated twice with absolute ethanol (4 °C,  
90 overnight), followed by centrifugation at 6000 rpm for 15 min at 4 °C. Thereafter, the  
91 precipitate was dissolved in distilled water and then was lyophilized to obtain the crude okra  
92 polysaccharide.

### 93 **2.2.2. Enzymatic deproteinization**

94 The dried crude okra polysaccharide was dissolved in distilled water (0.2%, w/v) and the  
95 dispersion was treated for 2 h, successively, using two commercial proteases, Neutrase P1236  
96 and Purafect 2000<sup>E</sup> at the same enzyme to substrate ratio (E/S) of 5000 U/g of protein [35]. The  
97 pH was adjusted to the optimal level for each enzyme with 2.5 N NaOH prior to protease  
98 addition. The enzymatic hydrolyses were conducted at 50 °C for 2 h with pH values of 7.0 and  
99 10.0, for Neutrase and Purafect, respectively. The optimal hydrolysis conditions were:

100 Neutrase, 50 °C, pH 7.0; Purafect, 50 °C, pH 10.0. Reactions were terminated by heating at 100  
101 °C for 5 min. Then, the pH was adjusted to 7.0 and the mixture was precipitated with 95%  
102 ethanol and centrifuged (6000 rpm, 15 min). The obtained precipitate was finally lyophilized  
103 and regarded as enzyme-deproteinized okra polysaccharide.

### 104 **2.2.3. Cetyl trimethyl ammonium bromide (CTAB) precipitation**

105 The deproteinized okra polysaccharide solution (0.5%) was precipitated by adding CTAB (100  
106 mM) under continuous stirring. After incubation at room temperature for 1 h, the formed  
107 precipitate was collected by centrifugation at 6000 rpm for 15 min and then re-suspended in  
108 100 mM NaCl solution. Then, the polysaccharide solution was precipitated by absolute ethanol  
109 (V/V), dialyzed against Milli-Q water and finally lyophilized. The obtained purified okra  
110 polysaccharide (OPS) was crushed in a pestle and mortar, weighted and stored at -20 °C until  
111 further use. The extraction yield was expressed as the % ratio of yield in grams of dried OPS to  
112 okra raw material.

## 113 **2.3. Structural characterization of OPS**

### 114 **2.3.1. Chemical analyses**

115 The total sugar content was determined by the phenol-sulfuric acid colorimetric method [36],  
116 using D-glucose as the standard. Total uronic acid content was estimated calorimetrically by the  
117 carbazole-sulfuric acid method [37], using D-glucuronic acid as standard. Sulfate content was  
118 determined according to the traditional method of barium chloride–gelatin [38]. The moisture  
119 and ash contents were determined according to the AOAC (Association of Official Analytical  
120 Chemists) standard procedures 930.15 and 942.05, respectively [39]. Soxhlet extraction with  
121 hexane was used to estimate the fat content (AOAC 920.85). Protein content ( $N \times 6.25$ ) was  
122 estimated using the Kjeldahl method [40] for the analysis of nitrogen. Elemental analyses were  
123 performed by the Central Micro-analysis Department of the CNRS at Gif/Yvette (France) for  
124 carbon (C), hydrogen (H), nitrogen (N) and sulfur (S) contents.



### 125 **2.3.2. Monosaccharide components analysis by Gas Chromatography (GC)**

126 Monosaccharide content was determined by GC analysis of trimethylsilyl derivatives (TMS)  
127 after acidic methanolysis [41]. GC analysis of the TMS methyl glycosides was performed on  
128 an Agilent 6850 Series GC System, equipped with a HP-5MS column (30 m x 0.25 mm)  
129 (Agilent Technologies, Palo Alto, CA, USA) and using He as carrier gas and a Flam Ionization  
130 Detector.

### 131 **2.3.3. Molecular weight determination**

132 The average molecular weight of OPS was determined by high-performance size-exclusion  
133 chromatography (HPSEC) coupled with a multiangle light scattering (MALLS, Dawn Heleos-  
134 II, Wyatt Technology Sc) and a differential refractive index (RI) (Optilab Wyatt Technology  
135 Sc) detectors. HPSEC system was composed of an HPLC system Prominence Shimadzu, a PL  
136 aquagel-OH mixed, 8  $\mu\text{m}$  (Agilent) guard column ( $U$  7.5mm  $\times$   $L$  50 mm), and a PL aquagel-  
137 OH mixed (Agilent) separation column. Samples were eluted with 0.1 M ammonium acetate.  
138 The molecular weight was calculated using  $dn/dc$  value of 0.145 ml/g. Elution was performed  
139 at 1 ml/min with 0.1 M ammonium acetate containing 0.03%  $\text{NaN}_3$ , filtrated on 0.1  $\mu\text{m}$   
140 membrane (Durapore Membrane, PVDF, Hydrophilic type VVLP, Millipore). Samples at 2  
141 mg/ml were filtrated on 0.45  $\mu\text{m}$  syringe filter prior to injection (100  $\mu\text{l}$ ). Data were computed  
142 with Astra 6.1 software for absolute molar mass determinations.

### 143 **2.3.4. Infra-Red and NMR spectroscopic analyses**

144 The Fourier Transform Infra-Red (FTIR) spectrum of OPS was obtained using a Perkin Elmer  
145 type FTIR 1000 spectrometer at room temperature and KBr pellets. The sample pellets were  
146 prepared at a pressure of 5 tons for 2 min. Pellets were scanned at room temperature (25  $^{\circ}\text{C}$ ) in  
147 the 400-4000  $\text{cm}^{-1}$  spectral range. Solid-state  $^{13}\text{C}$  NMR spectroscopy was carried out using a  
148 Bruker W300 spectrometer. 25 mg of sample was suspended in 1 ml  $\text{D}_2\text{O}$  at a high level of  
149 deuteration (99.997 %) to avoid the presence of relatively high water. The spectrum was



150 registered at a temperature of 25 °C and with a frequency of 75.5 MHz, 50 ms acquisition time,  
151 8 ms contact time and 5 s repetition time. <sup>13</sup>C chemical shifts (d, ppm) are quoted with respect  
152 to external sodium 4, 4-dimethyl-4-silapentane-1-sulfonate (0.0 ppm).

## 153 **2.4. Rheological measurements**

### 154 *Viscosity and viscoelastic properties*

155 OPS samples at different concentrations of 0.5, 1 and 1.5% were prepared in 0.9% NaCl and  
156 left overnight under continuous stirring to ensure complete solubilization.

157 The shear rheological measurements were made on DHR3 rheometer (TA Instruments). A  
158 plate-plate cell with a diameter of 25 mm was used. Its surface was rough to prevent slippage  
159 to the wall. The temperature was 25 °C maintained by Peltier heating system. Dynamic mode  
160 tests under small strains were performed by measuring firstly storage modulus G' and loss  
161 modulus G'' vs. strain to determine the linear behavior domain. Then, frequency sweeps at a  
162 fixed strain within the linear viscoelastic domain were performed. For viscosity measurements,  
163 the effect of shear rate (0.01–10 s<sup>-1</sup>) on OPS solutions viscosity was also examined.

164 All rheological measurements were carried out on freshly prepared samples and data were  
165 analyzed with the software that supported the rheometer.

### 166 *Zeta Potential measurement*

167 The Zeta potential was measured at 25 °C using a laser Doppler electrophoresis apparatus  
168 (Malvern Nano-Zetasizer ZS, UK). The concentration of OPS was around 0.5 mg/ml and the  
169 pH was adjusted by NaOH or HCl solution (0.001 M). The measurements were performed three  
170 times for each sample.

## 171 **2.5. Bioactivities of OPS**

### 172 **2.5.1. Determination of antioxidant activities *in vitro***

#### 173 **2.5.1.1. DPPH free radical scavenging activity assay**

174 The DPPH radical scavenging activity was carried out based on the method described by Brand-  
175 Williams et al. [42] with slight modifications. Briefly, in a 96 well plate, 10  $\mu$ l of OPS solution  
176 at different concentrations (12.5-400  $\mu$ g/ml) were mixed with 10  $\mu$ l of DPPH ethanolic solution  
177 (124  $\mu$ g/ml). Gallic acid (GA), the control (containing ethanol and DPPH solution) and the  
178 blank of each sample were prepared with the same method and under the same conditions.  
179 All solutions obtained were then incubated for 30 min at room temperature in the dark and the  
180 absorbance (A) was recorded at 517 nm. The percentage of inhibition of samples was calculated  
181 from obtained absorbance by the equation:

$$182 \quad \% \text{ Inhibition} = [(A_{\text{control}} - A_{\text{blank of control}}) - (A_{\text{sample}} - A_{\text{blank of sample}})] / (A_{\text{control}} - A_{\text{blank of control}}) \times 100$$

183 The equation of the obtained curve was allowed to calculate the  $IC_{50}$  corresponding to the  
184 sample concentration that reduced the initial DPPH $\cdot$  absorbance of 50%.

#### 185 **2.5.1.2. Ferric reducing antioxidant power (FRAP) assay**

186 The reducing power of OPS was evaluated using the ferric reducing ability of plasma (FRAP)  
187 assay according to the method described by Benzie & Strain [43] with slight modifications.  
188 Briefly, in a 96 well plate, the FRAP reagent ( $Fe^{3+}$ - TPTZ (2, 4, 6- tripyridyl triazine) complex)  
189 was mixed with OPS sample (0.5 to 8 mg/ml) or standard solutions.

190 Standard solutions consisted of  $FeSO_4 \cdot 7H_2O$  in different concentrations ranging from 0 to 0.5  
191 mg/ml. The blank of each sample and the control were also prepared under same conditions.

192 The standard calibration curve was then plotted using ferrous sulfate ( $FeSO_4 \cdot 7H_2O$ ) ranged  
193 from 0.062 to 0.5 mg/ml. The absorbance was read at 593 nm after 10 min incubation at 37  $^{\circ}C$ .

194 The reducing potential of iron, expressed as mmol  $EF_{FeSO_4/g}$  of OPS, was calculated using the  
195 linear regression curve of ferrous sulfate standard.

#### 196 **2.5.2. Cytocompatibility (MTT test)**

197 The MTT assay was performed to evaluate the cytotoxicity effect of the purified OPS on HEK-  
198 293 cells [33]. Different sample concentrations (0–1000  $\mu$ g/ml) were applied to the cells (1 x

199  $10^5$  live cells/ml) for 48 h. The absorbance (A) was measured at 570 nm using a microplate  
200 reader (Thermo Scientific Varioskan Flash). The cells treated with medium alone were  
201 considered 100% viable (control). The blank of each sample and the control was also prepared  
202 under same conditions. The cell viability (%) was measured as follows:

$$203 \quad \% \text{ Cell viability} = (A_{\text{sample}} - A_{\text{blank of sample}} / A_{\text{control}} - A_{\text{blank of control}}) \times 100$$

### 204 **2.5.3. Wound healing potential of OPS**

#### 205 **2.5.3.1. Scratch wound assay *in vitro***

206 B16 cells, showing fibroblast-like characteristics, were seeded in a 24-well plate at a cell  
207 density of  $10^5$  cells/ml until completely confluent cell monolayer was obtained. Then, the cell  
208 monolayer was scratched in a straight line using a sterile micropipette tip, which could mimic  
209 an incision wound in the literature [44]. After that, gentle washes with PBS were used to remove  
210 floating cell debris. Then, PBS was replaced with OPS solutions (0, 10 and 400  $\mu\text{g/ml}$ ),  
211 scratched areas were photographed ( $t=0$  h), and cells were incubated for an additional 24 and  
212 48 h at 37 °C prior to acquiring images of the scratched areas. The scratch area was measured  
213 using the Image-J software. Digital photographs were obtained using an inverted microscope  
214 (Olympus, Japan).

#### 215 **2.5.3.2. *In vivo* wound healing in diabetic rats' model**

##### 216 *Preparation of OPS hydrogel*

217 The hydrogel was prepared by dissolving the lyophilized OPS, at room temperature, in a 0.9%  
218 sodium chloride sterile saline solution, to give a final concentration of 5 mg/ml. The mixture  
219 was kept under agitation until a hydrogel was formed.

##### 220 *Diabetes Mellitus induction, wound creation and treatment*

221 The animals, allowed to acclimatize for one week, were intraperitoneally injected with freshly  
222 prepared solution of alloxan monohydrate dissolved in physiological saline solution, at a dose  
223 of about 130 mg/kg body weight. The injected rats immediately received 20% glucose solution

224 for 6 h to prevent fatal hypoglycemia that often follows alloxan treatment. For the next 24 h,  
225 the rats were given a 5% glucose solution as beverage. Three days post alloxan injection, the  
226 Type 1 diabetes onset was assessed via measuring, through the tail vein, blood glucose level  
227 using a digital glucometer (On Call Vivid). Rats that had blood glucose levels above 200 mg/dl  
228 were enrolled in the experiment. The animal's glycemia was assessed every three days to check  
229 the maintenance of high blood glucose levels. Fifteen days after confirmation of diabetic  
230 induction, wounds were done as previously described by Maalej et al. [45]. In brief, rats were  
231 anesthetized using intraperitoneal injection of 50 mg/kg of ketamine. A metal punch was used  
232 to demarcate an area of skin for removal. Then, full-thickness oval wounds (2×1 cm) were  
233 created on the shaved rat's dorsal interscapular region by removing a patch of skin with a pair  
234 of surgical scissors.

235 All animals were randomly divided into three groups each of six rats. Group I was untreated  
236 and served as the control (just cleaning the diabetic wounds with a physiological saline  
237 solution). Group II was treated with a synthetic reference drug « *Cytol Centella cream* ® » and  
238 served as a reference standard (positive control), while Group III was treated with the prepared  
239 OPS hydrogel and served as the test group.

240 After rinsing wounds with the physiological saline solution, the test sample (OPS hydrogel)  
241 and the reference drug (*Cytol Centella cream*) were applied, in a fine layer covering the surface  
242 of the wound, every 2 days starting from the wound induction (day 1), until the first group was  
243 completely healed (day 14). Once treated or just cleaned with the physiological saline solution,  
244 the wounds were covered with a compression dressing.

245 All the rats were anaesthetized with ether, sacrificed on the 14<sup>th</sup> post-wounding day and the  
246 granulation tissue was excised from the sacrificed animals. A part of wet tissue was preserved  
247 for hydroxyproline estimation and another one was fixed in formalin 10% (v/v), embedded in  
248 paraffin and processed for histological observation.

249 *Wound healing evaluation parameters*

250 To evaluate the healing process of the 3 studied groups, we relied on two clinical macroscopic  
251 criteria, including qualitative (color of the wound) and quantitative (wound closure rates)  
252 criteria and one microscopic criterion (histological evaluation).

253 *Chromatic study*

254 The chromatic evaluation of the healing process was done through the photography, every 2  
255 days, of wounds. This study consisted of attributing a chromatic code to the wound of each rat  
256 as follows: bright red=blood covering the wound; dark red=coagulation of blood in the  
257 epidermis; red=granulation tissue and pink=epithelialization phase [46].

258 *Wound closure rate and epithelization time*

259 The rate of wound closure of individual animal (n=6) from control and treated groups was used  
260 as an indicator of wound healing. Progressive decrease in the wound size was monitored  
261 periodically every 2 days interval using transparent graph sheet and a marker. The shapes of  
262 the wounds were scanned and uploaded to the computer. The wound surface areas were  
263 measured using Autodesk AutoCAD 2015 software application and then converted into percent  
264 values taking the size of the wound at the time of wounding as 100%. Wound closure, which  
265 indicates the formation of new epithelial tissue to cover the wound, was expressed as reduction  
266 in percentage of the original wound size using the following expression:

$$267 \quad \text{Wound closure rate (\%)} = (A_1 - A_n)/A_1 \times 100$$

268 where  $A_1$  and  $A_n$  are the initial wound area (day 1) and wound area on day n, respectively.

269 Falling of scab (dead-tissue remnants) without any residual raw wound was taken as end point  
270 of complete epithelialization and the days required for this were taken as period of  
271 epithelization [47].

272 *Histological analyses*

273 Tissue specimen samples from wound site of all studied groups were explanted and fixed in 4%  
274 paraformaldehyde solution, embedded in paraffin wax, and cut and stained with hematoxylin-  
275 eosin (HE) stain [48]. The sections were observed under a light microscope regarding fibroblast  
276 proliferation, collagen formation, angiogenesis, and epithelialization.

## 277 **2.6. Statistical analysis**

278 Data from all tests are presented as the mean  $\pm$  standard deviation (SD). Student's t-test was  
279 applied to ascertain significant differences between groups. Differences were considered to be  
280 statistically significant at  $p < 0.05$ . All analyses were carried out in triplicate.

## 281 **3. Results and Discussion**

### 282 **3.1. Biochemical and physicochemical characterization of OPS**

#### 283 **3.1.1. Basic composition and content determination**

284  
285 The water-soluble polysaccharide OPS was obtained by hot water extraction from dried  
286 Tunisian okra (*Abelmoschus esculentus*) pods, enzymatic deproteinization, ethanol and CTAB  
287 precipitation. The physicochemical parameters characterizing OPS are summarized in Table 1.  
288 Under the reported conditions, the OPS extraction yield was  $26 \pm 2.5\%$  of dried material.  
289 Although the extraction yield in this study is slightly lower than the reports of Nagpal et al. [49]  
290 (31.52%) and Li et al. [50] (29.4%), it is much higher than that of the previous studies of  
291 Elkhailifa et al. [51] (2.66%), Zhang et al [52] (7.9%), Wang et al. [5] (10.35%), Kpodo et al.  
292 [53] (14.6%) and Samavati [54] (16.9%). Such variations in the percentage yield of okra  
293 polysaccharide could be due to several factors, including the physical (hydrated or dried) and  
294 maturation states of pods, the cultivation region and the extraction method used [50]. As shown  
295 in Table 1, the chemical analysis results showed that the contents of total sugar, uronic acid and  
296 protein in OPS were  $70.61 \pm 3.55\%$ ,  $27 \pm 1.5\%$  and  $6.54 \pm 0.12\%$ , respectively.

### 297 **3.1.2. Monosaccharide composition and molar ratios**

298 GC analysis of the monosaccharide composition showed that OPS is a hetero - polysaccharide  
299 and consisted of four types of monosaccharides, namely rhamnose (Rha), galactose (Gal) and  
300 galacturonic acid (GalA), followed by minor amount of glucose (Glc) (Fig. 1). According to  
301 Table 2, the Gal and GalA contents (related to dry weight) in OPS were about 24%, followed  
302 by Rha (18%), while Glc has the lowest level of 1.25%, suggesting the pectic nature of the  
303 polysaccharide extracted from the okra pods. Low Glc content indicates that the present  
304 extraction protocol results in pectin fraction with minor amounts of co-extracted cellulose,  
305 hemicelluloses or starch [45]. Similarly, polysaccharides obtained from the pods [55], flowers  
306 [56] and leaves [57] of okra plants have been reported to contain pectic polysaccharides.  
307 According to the literature, among plant cell wall polysaccharides, pectins are characterized by  
308 high versatility and complexity of their structure. Indeed, in terms of its monosaccharide profile,  
309 pectin is a complex of several polysaccharides consisting mainly of galacturonic acid and  
310 rhamnose units, as well as a variety of neutral sugars including arabinose, galactose, and lesser  
311 amounts of other sugars [58]. Furthermore, pectic polysaccharides consisted of different  
312 domains, the most important of which are HG and RG-I often described, respectively, as the  
313 “smooth” and “hairy” regions [59]. RG-I regions contain arabinan or galactan side chains  
314 depending on pectin origin. RG-II is a much less common but strongly conserved branched  
315 region, composed by the repetition of four branches containing neutral and acid residues  
316 departing from a backbone of HG [60].

317 Molar ratios of pectin’s monosaccharides can help to reveal more important structural  
318 information through the determination of the % of key pectin regions HG and RG-I. According  
319 to Denman and Morris [61], several sugar ratios are often used as an expression for the  
320 occurrence and properties of pectin. The ratio of pectin backbone sugar GalA to the neutral  
321 sugars involved in side chains ( $R_1$ ) and which is useful tool to estimate the linearity of pectin.



322 Ratio 2 ( $R_2$ ) is defined as the proportion of Rha to GalA, thus reflecting the contribution of  
323 RG-I segments to the entire pectin population. While  $R_3$  value is indicative of the length of  
324 galactose side chains of RG. As shown in Table 2, OPS exhibited high  $R_3$  value (1.3) which  
325 indicates high branching of the RG-I segment. Furthermore, the RG-I molar content of OPS  
326 was found to be greater (94%), as compared to HG proportion (4.5%), thus indicating the  
327 abundance of “hairy” regions in the extracted pectin from Tunisian okra pods. As previously  
328 mentioned, the type of pectin is highly dependent on plant sources as well as extraction  
329 methods. Kpodo et al. [53], while conducting a comparative study on pectin characterization  
330 from six okra genotypes, showed that it is possible to tailor the structure of the extracted pectin  
331 by selecting the appropriate genotype, thus creating pectins with an extensive spectrum of  
332 functionalities. For the extraction of okra polysaccharides, it was demonstrated that most of the  
333 water extracted polysaccharides belong to RG-I type [6,62]. In summary, okra pods extraction  
334 resulted in an RG-I enriched OPS with short galactose-containing side chains.

### 335 **3.1.3. Homogeneity and molecular weight determination**

336 Molecular size characteristics of the isolated OPS were evaluated by size exclusion  
337 chromatography (SEC) coupled to multiangle laser light scattering and weight average (Mw),  
338 number average (Mn) molecular weights and polydispersity index (Mw/Mn) were determined.  
339 The obtained elution profile (Fig. 2) revealed narrow Mw distribution representing one polymer  
340 population of high molecular weight as indicated by the presence of one peak. The Mw and Mn  
341 values of OPS RG-I were  $3.127 \times 10^3$  and  $3.083 \times 10^3$  kDa, respectively. The Mw of OPS was  
342 comparable to okra RG-I polysaccharide showing a Mw larger than  $2.99 \times 10^3$  kDa [10], but  
343 higher than those previously reported for pectin samples isolated from six okra genotypes which  
344 were of Mw ranging from 0.7 to  $1.7 \times 10^3$  kDa [41]. Chen et al. [63] reported higher average  
345 Mw of pectin from *Abelmoschus esculentus* of  $5.94 \times 10^3$  kDa. Ma et al. [64] reported, when  
346 studying the Mw of three polysaccharides extracted from okra under different conditions, that

347 macromolecules were mainly concentrated in the fraction extracted at 60 °C, and the  
348 polysaccharide might be degraded at higher temperature or under acidic pH extraction. The  
349 polydispersity index value of OPS was 1.03, indicating its narrow molecular distribution and  
350 uniform dispersity in an aqueous solution.

#### 351 **3.1.4. FTIR and $^{13}\text{C}$ NMR spectra of OPS**

352 The FTIR spectrum of OPS, showing typical signals of a polysaccharide, was depicted in Fig.  
353 3A. The broad and strong area of absorption between 3100 and 3500  $\text{cm}^{-1}$  due to O–H  
354 stretching absorption is attributed to the vibrational modes of inter- and intramolecular  
355 hydrogen bonds [65]. The band around 2950  $\text{cm}^{-1}$  refers to C–H absorption. The absorptions  
356 around 1601 and 1418  $\text{cm}^{-1}$  were due to the stretching vibrations of C=O and –COOH,  
357 respectively, confirming the presence of GalA residues in the sample [63]. Two strong  
358 absorbance peaks between 1074 and 1039  $\text{cm}^{-1}$  are characteristics of glycosidic linkages  
359 between sugar units.

360 The structural features of OPS were further elucidated by  $^{13}\text{C}$  NMR spectral analysis. In a  $^{13}\text{C}$   
361 spectrum, the signals derived from  $\alpha$ -anomeric carbons usually appear in the 95-101 ppm region  
362 while most of the  $\beta$ -anomeric carbons will appear in the range 101-105 ppm [66]. The  $^{13}\text{C}$  NMR  
363 spectrum of OPS, displayed in Fig. 3B, showed peaks around 177.99 ppm indicative of the  
364 carbonyl group (C=O) of (GalA) [67]. The two signals at 106.78 and 101.07 ppm were assigned  
365 to the anomeric C-1 of (Gal) and (GalA) residues, respectively. Peaks in the range of 18-20  
366 ppm can be assigned to the methyl groups of the rhamnose residue. The  $^{13}\text{C}$  NMR spectrum of  
367 OPS was consistent with the previously reported spectra of RG-I regions extracted from six  
368 okra genotypes [53]. Moreover, such  $^{13}\text{C}$  NMR values confirm the results obtained by the GC  
369 analysis showing the presence of some monosaccharides, such as GalA, Gal and Rha which are  
370 the principal OPS sugars.

#### 371 **3.2. Rheological characterization**

372 The rheological behavior of OPS solutions was investigated by oscillatory sweep measurements  
373 at 25 °C, of the storage modulus ( $G'$ ), loss modulus ( $G''$ ) as a function of frequency ( $f$ ) in the  
374 linear domain and the steady-state viscosity ( $\eta$ ) vs shear rate (Fig. 4). Measurements were  
375 performed at constant pH and ionic strength to avoid any fluctuation induced by these  
376 parameters.

377 In Fig. 4A, in the domain of OPS concentration investigated, both  $G'$  and  $G''$  were nearly  
378 independent of frequency with the  $G'$  values almost larger than  $G''$  over the applied frequency,  
379 which is consistent with a gel-like behavior and good storage stability of the OPS solution, even  
380 at low concentration (0.5%). Indeed, Mezger [68] reported that materials having highest values  
381 of  $G'$  modulus show highest stability over time.

382 Such gelling effect is the consequence of the properties OPS macromolecular chain to generate  
383 in aqueous solution an interconnected three-dimensional network, due to the high molecular  
384 weight of OPS (exceeding  $10^6$  Da) combined with its polyelectrolyte character induced by the  
385 presence of ionizable carboxylic groups [69]. At pH 7, the carboxylic groups were mostly in  
386 the ionic form ( $\text{COO}^-$ ), promoting the polymer chains expansion in the coiled form and increase  
387 in their hydrodynamic volume, which favors the entanglement of the polymer chains. An  
388 increase in the magnitude of  $G'$  with the increasing concentration of OPS is observed, which is  
389 expected, since the increase in the polymer concentration will increase the number of chains  
390 junction and entanglement density. This led to a higher cohesion in the polymer network formed  
391 by coiled pectin chains thanks to the set-up of intermolecular interaction through hydrogen  
392 bonding between pectin chains. It was established that gel formation in RG-I rich in galactan  
393 side chains results from interactions between galactan chains. The 2D-correlation FTIR  
394 spectroscopy [70] together with molecular modelling [71] showed that in such gels, galactan  
395 adopts a regular helical structure and chains interact in a side-by-side antiparallel fashion.

396 The steady-state viscosity ( $\eta$ ) as a function of shear rate (Fig. 4B) displayed a typical shear-  
397 thinning behavior of pectin gel which is attributed to the disentanglement and alignment of  
398 pectin along the shear direction under the low shear force, causing a gradual decrease in  
399 viscosity, which agrees with literature data [70]. Such shear-thinning (pseudo plastic) behaviour  
400 has been widely observed for many food gums [71]. At higher shear rates, the breakdown of  
401 intermolecular junctions became faster than the rate of network reformation and caused the  
402 observed decrease in apparent viscosity [60, 70].

403 The zeta potential of 0.05% okra pectin in aqueous solutions at different pH values is presented  
404 in Fig. 4C. In the pH range from 2.0-12.0, the zeta potential remained negative, with an  
405 increment in the absolute value as pH is going up. This indicated that OPS chains bear negative  
406 ionic groups over this pH range, and their ionic degree is more pronounced with increasing pH.  
407 This evolution is presumably due the carboxylic groups of OPS that turned more dissociated as  
408 the pH is increasing [72].

409 It has been reported that rheological properties of pectic polysaccharides depend on both  
410 intrinsic and extrinsic factors, including the number and distribution pattern of free carboxyl  
411 groups, the molecular weight and types of pectin, pectin concentration, ion concentration, pH,  
412 temperature and ionic strength [71]. Other factors like extraction conditions, contribution of  
413 proteins, drying method and extraction method may also influence the rheological properties of  
414 these polysaccharides [73].

### 415 **3.3. Antioxidant potential *in vitro***

416 In an attempt to accelerate the wound healing process, many investigations have been carried  
417 out using an antioxidant strategy, as the excessive production of reactive oxygen species (ROS)  
418 promotes the imbalance between antioxidants and oxidants, thus leading to a slow tissue  
419 regeneration and healing process. Particularly, besides the impaired vessel formation, damage

420 from ROS are also among many factors that can hinder diabetic wounds healing, and thus  
421 results in severe complication of diabetes [74].

422 For wound management, biopolymers-based dressings with antioxidant properties have  
423 appealed increasing attention to minimize the damage caused by reactive chemical species,  
424 especially in the case of diabetic chronic wounds.

425 DPPH (2,2-diphenyl-1-picrylhydrazyl) radicals are relatively stable free radicals that are often  
426 used to evaluate the antioxidant radical scavenging activity of various compounds. As  
427 illustrated in Fig. 5A, the DPPH radical scavenging activity of OPS increased gradually as the  
428 concentration increased, presenting a dose-dependent relationship. It exhibited a value of  $63.5$   
429  $\pm 1.98$  % at a concentration of  $0.4$  mg/ml. The  $IC_{50}$  values of OPS and gallic acid (GA) were  
430  $0.32$  and  $0.048$  mg/ml, respectively. The results showed that OPS owned a stronger antioxidant  
431 activity than those of previously reported okra polysaccharides [5,55,64].

432 The FRAP (ferric reducing antioxidant power) assay takes advantage of electron-transfer  
433 reaction and is considered a common way to evaluate the antioxidant potential. The reducing  
434 power of OPS was assessed based on its ability to reduce  $Fe^{3+}$  to  $Fe^{2+}$  and the results in ferrous  
435 sulfate equivalent (EFeSO<sub>4</sub>) is presented in Fig. 5B. OPS displayed a reducing potential of the  
436 ferric ion of  $1.28 \pm 0.018$  mmol EFeSO<sub>4</sub>·7H<sub>2</sub>O/g. Chen et al. [63] revealed that okra  
437 polysaccharides extracted by using ultrasound at different frequencies exhibited FRAP values  
438 lower than  $21$   $\mu$ mol  $Fe^{2+}$ /g sample.

439 According to both antioxidant assays, OPS exhibited a potent antioxidant activity *in vitro* which  
440 may be likely due to the presence of the RG-I pectic domain, as previously reported. Indeed,  
441 according to Zhang et al. [75], uronic acid content might be an important molecular feature for  
442 the antioxidant properties of polysaccharides. Li et al. [76] demonstrated that acidic  
443 polysaccharide fractions with the highest uronic acid content showed stronger free radical  
444 scavenging activities than the neutral polysaccharide fraction. Similarly, Wu et al. [77] showed

445 that acidic polysaccharides, which contained amounts of uronic acid, exhibited stronger  
446 antioxidant activity. Besides, Rao et al. [78] found that three types of uronic acid exhibited  
447 strong antioxidant effect in the order of polygalacturonic acid > glucuronic acid > galacturonic  
448 acid, suggesting degree of polymerization may impart the activity. According to the previous  
449 report of Ma et al. [64], the pectin of OPW-60 containing the highest branching rate and the  
450 lowest linearity and thus which may contain more hydrogen bonding sites, exhibited the  
451 strongest antioxidant capacity.

452 Interestingly, OPS demonstrated the potential to act as a natural antioxidant which would  
453 broaden the development and utilization of okra resources. Moreover, since antioxidants assist  
454 in controlling oxidative stress at the chronic wound site, thereby accelerating the healing  
455 process, OPS-based hydrogel could contribute to the conception of an appropriate environment  
456 for wound healing through the protection of wound tissue from oxidative damage.

#### 457 **3.4. Biocompatibility analysis**

458 The safety of OPS was evaluated on normal HEK-293 cells cultivated with different  
459 concentrations of okra polysaccharide ranging from 10 to 1000 µg/ml, and the MTT assay was  
460 employed to determine cell viability. The results in Fig. 6 show that, compared to the cell  
461 viability of the control group, the co-culture with OPS for 48 h did not lead up to a remarkable  
462 decrease of the cell viability which remains higher than 91%. The results clearly illustrated that  
463 OPS possessed good cytocompatibility and non-toxicity to human normal cells at  
464 concentrations up to 1 mg/ml. It is well-established in the literature that pectins may be  
465 considered as a biocompatible and non-toxic biopolymer. According to da Costa Amaral et al.  
466 [78], two pectin fractions from gabiropa pulp exhibited a cytotoxic effect, even in low  
467 concentrations, against human glioblastoma cancer cells, while no cytotoxicity was observed  
468 in normal fibroblast cells.

#### 469 **3.5. Wound healing activity**

470 In this study, the potential wound healing property of the polysaccharide (OPS) extracted from  
471 the Tunisian okra pods was investigated by analyzing its effects, *in vitro* through the cell  
472 migration assay, which play a key role in the wound healing process, and *in vivo* through the  
473 assessment its efficiency to accelerate wound healing using alloxan-induced diabetic rat model.

#### 474 **3.5.1. Scratch wound assay *in vitro***

475 The scratch assay was utilized to evaluate the *in vitro* potential of OPS in cell migration and  
476 proliferation, into the wounded area (Fig. 7). Our findings indicated that the polysaccharide  
477 stimulated cell proliferation and migration compared to the control, and thereby decreased the  
478 cell free gap (distance) in a concentration dependent manner (Fig. 7A). Indeed, cells at the  
479 wound edges remove toward the surface traversing layers and lead to the re-epithelialization of  
480 the wound surface. The highest wound closure rate was 99.09% compared to control (23.96%),  
481 after 48 h of incubation (Fig. 7B).

482 According to the efficiency of OPS to promote the cell proliferation, a key phenomenon in the  
483 re-epithelialization process during wound healing [79], together with its non-toxicity and  
484 antioxidant capacity we have found, it was of interest to study, *in vivo*, the healing capacity of  
485 this polysaccharide on treating chronic wounds.

#### 486 **3.5.2. Effect of OPS hydrogel on diabetic wound rats**

487 Alloxan is one of the most known diabetogenic agents often used to produce a model of type I  
488 diabetes mellitus in experimental studies involving diabetes [79]. In the present study, the  
489 average blood glucose of rats suffered from alloxan injection reached above 200 mg/dl, along  
490 with polyuria and visible body weight decrease. Such symptoms indicated the successful  
491 establishment of the diabetic model. In our wound healing rat model, a single full thickness  
492 oval skin wound (about 160 mm<sup>2</sup>) was made on the shaved dorsal interscapular region of each  
493 diabetic rat. The wound healing potential of the OPS hydrogel was evaluated on the basis of



494 some parameters, including the rate of wound closure, wound color, state of inflammation and  
495 epithelialization time along with the histological evaluation of the healed tissues.

496 The evaluation of the wound closure by monitoring the skin wound area is a parameter widely  
497 used, since it is easily accessible and handled and has clinical applicability. Fig. 8A showed the  
498 representative photos of wounds in different rats' groups at days 1, 3, 5, 10, 12 and 14. As  
499 shown, macroscopic follow-up of the hydrogel OPS-treated wound area, throughout the  
500 treatment period, did not show any signs of infection and inflammation, such as the presence  
501 of pus, secretion or a foul odor, nor a delay in the healing process. In contrast, in the untreated  
502 group (Group I), wound healing was significantly delayed because of the impaired blood  
503 circulation in diabetic rats, which still show, at the end of the experiment (day 14), an open  
504 wound with red rounding tissues.

505 After local application of OPS hydrogel and *Cytol Centella*® cream, a dark red coloration was  
506 observed on the third day for both treated groups (Groups II and III) which gives evidence of  
507 the initiation of the healing process through the formation of blood clots with cellular debris.  
508 On the 5<sup>th</sup> day, a healthy and brown in color wounds were observed in both treated groups due  
509 to plasma exudation with formation of superficial crusts at the site of the wound until the 10<sup>th</sup>  
510 day. Nevertheless, wounds that received the reference drug *Cytol Centella*® cream revealed a  
511 thicker, harder and darker crust around the wound area than those that received the OPS  
512 hydrogel, leading to a drier wound. It has been previously reported that crusts act as a physical  
513 barrier to protect the wound from external infections and germs. However, less dry the wound  
514 was, rapidly the epidermal cells migrate, and hence the epithelialization process can be  
515 accelerated [37,73]. On day 12, in OPS hydrogel-treated rats, the crust began to detach with  
516 growth of new skin tissue to let appear a pink blade coloration, and complete closure of the  
517 wounds was noticed on the 14<sup>th</sup> day. However, all untreated animals still show, at the end of  
518 the experiment, an open wound with red rounding tissues.

519 The quantified wound areas of each group in different time points were shown in Fig. 8B. As  
520 compared to the other counterparts, the application OPS hydrogel exhibited the optimal wound  
521 healing activity during the whole experimental period, which is in accordance with Fig. 8A.

522 Both the reference (*Cytol Centella*®) and OPS hydrogel-treated groups resulted in a faster  
523 reduction of the wound diameter than the diabetic control group (Group I), with approximately  
524 52 and 40% of wound contraction achieved on the 5<sup>th</sup> day, respectively. Moreover, OPS  
525 hydrogel was more efficient than the standard *Cytol Centella*® cream on healing contraction  
526 of diabetic rats ( $p < 0.05$ ). Indeed, 100% of wound healing was recorded at the end of experiment  
527 in OPS-treated group, against 89.08 and 80.3% for *Cytol Centella*-treated and untreated diabetic  
528 groups, respectively.

529 Images taken from tissue biopsies of wound sections of each group on the 14<sup>th</sup> post excision  
530 day, after hematoxylin-eosin (HE) staining, showed a rapid re-instatement of normal tissue  
531 structures following the application of OPS hydrogel, in comparison to the reference drug-  
532 treated group and the control diabetic group (Fig. 9). In wounds of diabetic animals, a delay in  
533 healing was evident as epidermal morphology was irregular and incomplete with localized  
534 tissue destruction (Fig. 9A). The dermis of the untreated diabetic wound revealed the presence  
535 of a moderate number of inflammatory cells (yellow arrow) which represent chronic stage of  
536 inflammation together with a pronounced hyperemia of capillary blood vessels (red arrows).  
537 Moreover, we note the presence of a foreign body reaction (star) which may be probably the  
538 result of the gauze fibers adherence to the healing tissues and then their penetration in the dermis  
539 of the wound's tissue after being enclosed by newly grown tissue. The granulation tissue  
540 demonstrated an aggregation of macrophages with moderate collagen fibers (blue arrows). The  
541 application of the reference drug and OPS hydrogel was able to restore the healing process in  
542 these diabetic rats, with better results observed following the treatment with OPS hydrogel.  
543 Indeed, a fully re-epithelization with a well-structured layer of epidermis, faster keratinization,

544 higher collagen density and newly formed blood vessels (black arrows) occupying the dermal  
545 could be observed.

546 All these features demonstrated that the wound healing effect of OPS hydrogel was better than  
547 the well-known healer *Cytol Centella*® cream, which is in accordance with the chromatic  
548 study.

549 Several factors have been reported to contribute in promoting the wound healing process,  
550 including the antioxidant property, mitogenic effect, moisturizing ability, etc. [74]. Antioxidant  
551 dressings have attracted widespread attention, particularly for chronic or hard-to-heal wounds  
552 treatment that often stagnate in the inflammatory phase due to of oxidative stress. Indeed,  
553 antioxidants have been demonstrated to improve the wound microenvironment by removing  
554 excessive reactive oxygen and thus reducing oxidative stress and accelerating healing [75].

555 Interestingly, the antioxidant character of OPS might be among the most efficient contributing  
556 factors for improved diabetic wound healing. Nonetheless, the lack of antimicrobial activity  
557 may be regarded as the main disadvantage of the application of the pectic OPS hydrogel as  
558 wound dressing. Indeed, besides oxidative stress, microbial infection is also considered among  
559 the factors delaying the wound-healing process and thus should receive careful attention in  
560 chronic wound treatment. Accordingly, antibacterial and biofilm-preventing activities in the  
561 area of wound healing is an interesting property that should be displayed by an efficient wound  
562 dressing to be a convincing candidate for the healing of chronic wounds. Bustamante-Torres et  
563 al. [80] reported that despite their versatile properties, including biodegradability, nontoxicity  
564 and biocompatibility, biopolymers usually have no antimicrobial properties by themselves,  
565 except for the chitosan [81] (Olmos & González-Benito, 2021). Therefore, antimicrobial agent  
566 incorporation into the natural polymeric matrix emerged as a possible alternative to overcome  
567 this drawback and to impart antimicrobial properties to the biocomposite, thus improving its  
568 wound healing performance [82]. Interestingly, combination of pectic OPS with other synthetic

569 or natural polymers, such as chitosan, may be explored to reduce the disadvantages of each  
570 individual polymer by itself, leading to composite dressings with increased antibacterial  
571 activities towards chronic wounds, in which each biopolymer can contribute to different stage  
572 of wound healing.

573 Besides, the gelling ability of some wound dressings play a major role in maintaining a  
574 favorable wound microenvironment. Polymers-based hydrogel dressings have demonstrated to  
575 possess unique features, including their capacity to maintain the wound-bed with ideal humidity  
576 [76]. This means that, thanks to their swelling property, hydrogels can absorb and retain a  
577 minimum amount of fluid from a heavily exuding wound and at the same time donate moisture  
578 to dry tissue thus promoting autolytic debridement. They have also been shown to provide a  
579 three-dimensional network structure for cell adhesion, migration and proliferation as well as  
580 transportation of cytokines, nutrients and metabolic waste, which make the wounds heal faster  
581 [77]. It should also be noted that the mechanical characteristics of hydrogels are critical in  
582 pharmacological and biological applications. In this study, the influence of frequency on  
583 viscoelastic characteristics of pectic OPS confirms its high gelling ability at a concentration of  
584 0.5%. Its shear-thinning behavior characterized by decreasing viscosity with increasing shear  
585 rate is an important feature in the use of a gelling skin care product [77x]. Indeed, it has been  
586 reported that products having high viscosity at low shear force show a firm, stable and well-  
587 bodied products, with good standup properties, in contrast, the viscosity decrease at higher  
588 shear force allowed the product when applied to be absorbed into the skin easily [77y]. In  
589 addition, the zero-shear viscosity of OPS solutions, showing a material behavior at a minimum  
590 stress and which is considered as an important parameter in assessing the capacity of use and  
591 storage of a material [77y], was found to be higher than 10 Pa.s, suggesting that pectic OPS gel  
592 might be considered strong enough to have good stability during its storage. These above-  
593 mentioned rheological properties highlight that pectic OPS-based hydrogel fulfill the

594 mechanical properties required for a product to be suited for applications as wound healing  
595 biomaterial.

596 To sum up, the development of multifunctional wound healing materials with biological  
597 properties and which meet the growing consumer interest in bio-based products is a major  
598 challenge for the pharmacological and dermato-cosmetic sectors of applications. Particularly,  
599 plant-based biopolymers for efficient application as dressing to wound healing are believed to  
600 offer effective, affordable, and accessible forms of treatment. Nowadays, they are viewed as an  
601 efficient alternative for wound healing thanks to their widespread availability besides their  
602 promising medicinal value.

603 Finally, based on the above findings reported, we should mention that the OPS hydrogel,  
604 developed in this study, meets the basic requirements of an efficient wound dressing, such as  
605 bioavailability, biocompatibility, non-toxicity, biodegradability, hydrophilicity and easy-  
606 handling, and more importantly good mechanical characteristics, antioxidant activity, cell  
607 migration and proliferation promoting properties, moisturizing ability and histocompatibility.  
608 As compared to other hydrogels, the main advantages of OPS are that it is very inexpensive  
609 being easily extracted from an available nutritive vegetable crop, easy to produce, and is easily  
610 applied to the wound thanks to its good mechanical properties.

611 Interestingly, the present study can provide a useful and efficient healing biomaterial,  
612 particularly a pectic polysaccharide based-hydrogel for diabetic chronic wound management,  
613 extracted for the first time from the Tunisian okra pods.

#### 614 **4. Conclusions**

615 In this study, a pectic polysaccharide was extracted and purified from okra, which was  
616 identified to contain RG-I domain. Its molecular weight was estimated to be  $3.127 \times 10^3$  kDa.  
617 Antioxidant potential of OPS, especially in relation to its ability to sequester DPPH radicals  
618 and to reduce the ferric ion, was also observed. In addition, after treatment with OPS hydrogel,

619 in alloxan-induced diabetic rats, the wound healing process was considerably accelerated by  
620 stimulating re-epithelialization, cell migration and proliferation and blood vessels formation.  
621 In summary, the biological activities presented by OPS highlight the potential of this okra-  
622 derived polysaccharide to be applied in pharmacological and dermato-cosmetic sectors, and  
623 suggest, particularly, its potential as a hydrogel for use in diabetic wound dressing sector whose  
624 market is continuously growing and promising. Furthermore, this study may offer a new route  
625 for utilizing Tunisian okra, an easily available nutritive vegetable crop, as a valuable source for  
626 the large-scale production of value-added biomaterials, which would certainly broaden its  
627 development and utilization.

#### 628 **CRedit authorship contribution statement**

629 **Hana Maalej:** Conceptualization; Investigation; Methodology; Writing original draft; Data  
630 curation. **Amina Maalej:** Investigation; Methodology; Formal analysis. **Asma Bayach:**  
631 Investigation. **Agata Zykwincka:** Investigation; Data curation; Methodology; Writing-review  
632 & editing. **Sylvia Collic-Jouault:** Supervision; Resources; Writing-review & editing.  
633 **Corinne Siquin:** Investigation; Methodology. **Laetitia Marchand:** Investigation;  
634 Methodology. **Naourez Ktari:** Investigation. **Sana Bardaa:** Investigation. **Riadh Ben Salah:**  
635 Resources. **Mohamed Chamkha:** Supervision; Resources. **Sami Boufi:** Investigation. **Moncef**  
636 **Nasri:** Supervision; Resources.

#### 637 **Declaration of competing interest**

638 The authors declare that they have no known competing financial interests or personal  
639 relationships that could have appeared to influence the work reported in this paper.

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**Table 1: Yield and chemical components of OPS**

	<b>OPS</b>
Yield (g/100 g dry okra powder)	26 ± 2.5
Moisture (%)	7.03 ± 0.04
Protein (Kjeldahl) (%)	6.54 ± 0.12
Fat (%)	3.33 ± 0.09
Carbohydrates (%)	70.61 ± 3.55
Sulfates (%)	1.51 ± 0.1
Uronic acid (%)	27 ± 1.5
Ash (%)	8.98 ± 0.03
% C	34.92 ± 1.54
% H	5.72 ± 0.12
% N	1.06 ± 0.01
% S	0.35 ± 0.05

The % is related to dry weight. Data are expressed as means ± standard deviation (SD; n = 3).

**Table 2: Monosaccharide composition of OPS**

	<b>OPS</b>
Rhamnose <sup>a</sup>	18.25% ± 0.35 (28.46)
Arabinose <sup>a</sup>	ND <sup>b</sup>
Galactose <sup>a</sup>	24.45% ± 0.49 (37.25)
Glucose <sup>a</sup>	1.25% ± 0.07 (1.30)
Galacturonic acid <sup>a</sup>	24.6% ± 0.28 (32.99)
Glucuronic acid <sup>a</sup>	ND <sup>b</sup>
R <sub>1</sub> <sup>c</sup>	0.5
R <sub>2</sub> <sup>c</sup>	0.9
R <sub>3</sub> <sup>c</sup>	1.3
R <sub>4</sub> <sup>c</sup>	1.3
HG <sup>c</sup>	4.5%
RG-I <sup>c</sup>	94%
HG/RG-I <sup>c</sup>	0.04

The % is related to dry weight

<sup>a</sup> Each value was expressed the mean ± standard deviation (n = 3).

In parentheses, data is presented as mol% of each sample.

<sup>b</sup> ND: not detectable or lower than the limit of quantification.

<sup>c</sup> According to reference (Denman and Morris, 2015),

R<sub>1</sub> = GalA/(Rha + Ara + Gal), R<sub>2</sub> = Rha/GalA, R<sub>3</sub> = Gal/Rha,

HG = GalA – Rha, RG-I = 2Rha + Ara + Gal.



**Figure Captions**

**Figure 1:** GC analysis of monosaccharide components of okra polysaccharide.

**Figure 2:** HPSEC-MALS elution pattern of OPS.

**Figure 3:** Infrared spectrometry **(A)** and  $^{13}\text{C}$  NMR analysis **(B)** of OPS.

**Figure 4:** **(A)** Storage modulus ( $G'$ ) and loss modulus ( $G''$ ) vs. frequency at different concentrations, **(B)** viscosity vs shear rate in presence of NaCl (0.9%), and **(C)** Zeta-potential vs pH of OPS solution.

**Figure 5:** Antioxidant potential of OPS: **(A)**: Scavenging effect on DPPH radicals; **(B)**: Ferric reducing antioxidant power (FRAP).

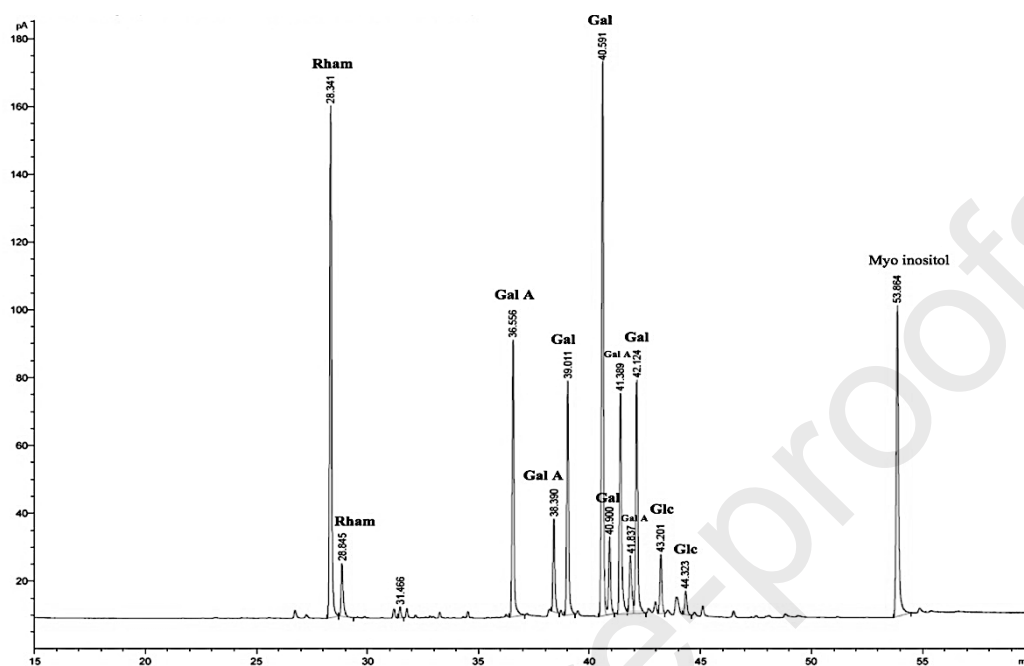
**Figure 6:** Effects of OPS on HEK-293 cells viability: **(A)**: MTT analysis; **(B)**: cellular density after 48 h incubation without (i) or with 1 mg/ml OPS (ii).

**Figure 7:** Scratch assay using B16 cells **(A)** and wound closure expressed as a percentage of wound size relative to the size of the initial wound **(B)**.

**Figure 8:** **(A)** Representative images of full-thickness skin defects and **(B)** the wound healing rate (%) for different group treatment on wounds at days 1, 3, 5, 10, 12 and 14.

**Figure 9:** Histological HE staining analysis of wounded skin tissue sections in the diabetic untreated **(A)**, *Cytol Centella*-treated **(B)** and OPS hydrogel-treated **(C)** groups on the 14<sup>th</sup> day post-wounding (100× magnification).

Fig.1



Rham: rhamnose; GalA: galacturonic acid; Gal: galactose; Glc: glucose

Fig.2

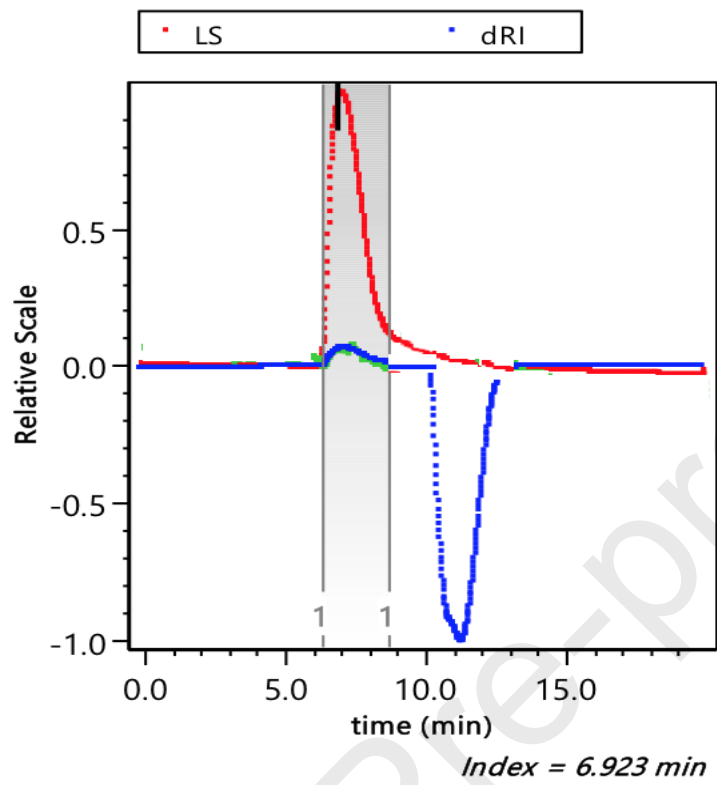
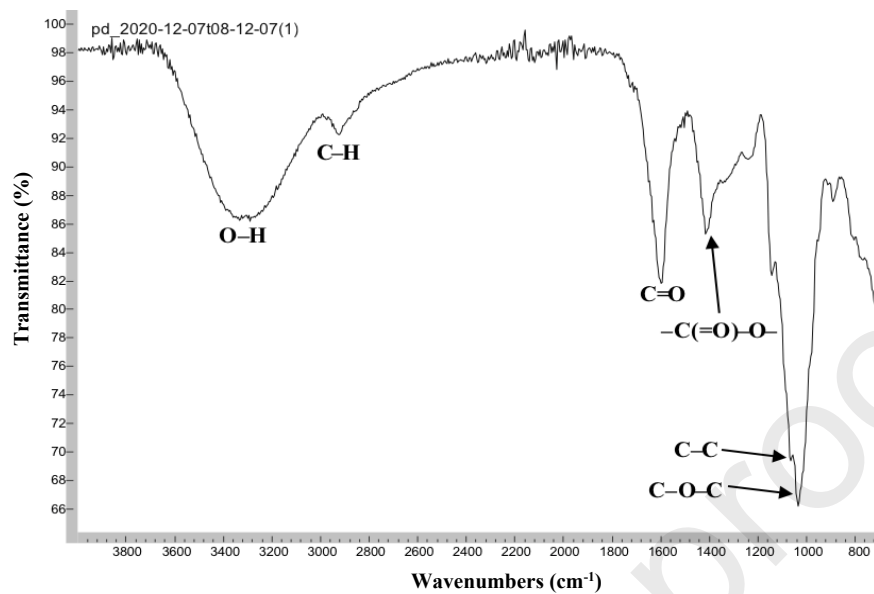
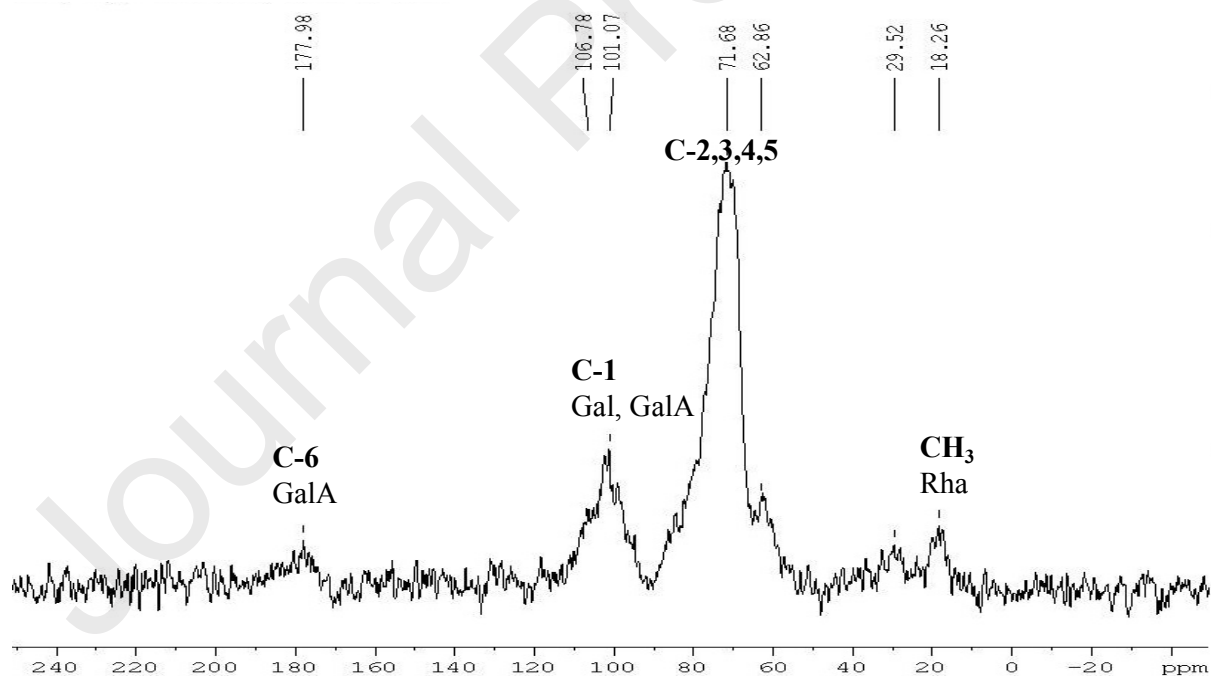


Fig.3

A



B



GalA: galacturonic acid; Gal: galactose; Rha: rhamnose

Fig.4

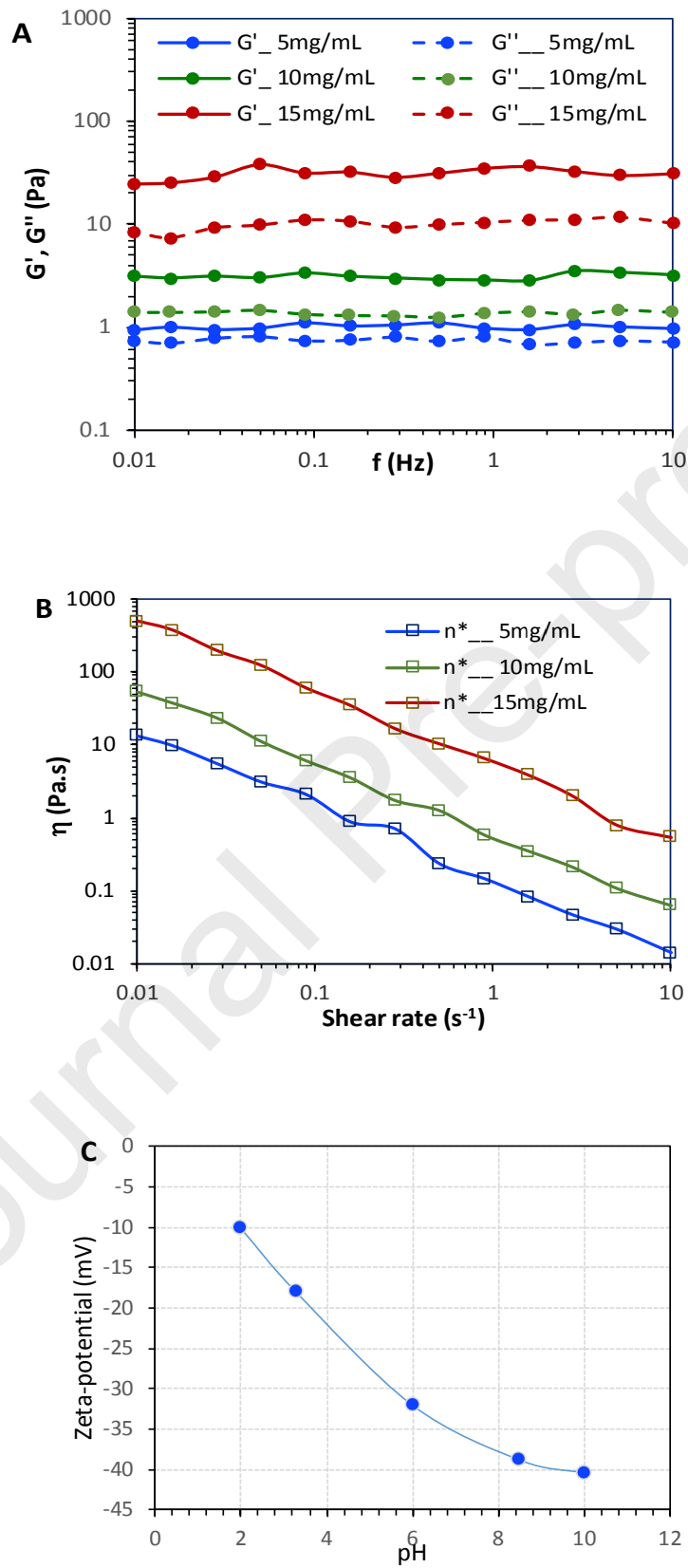
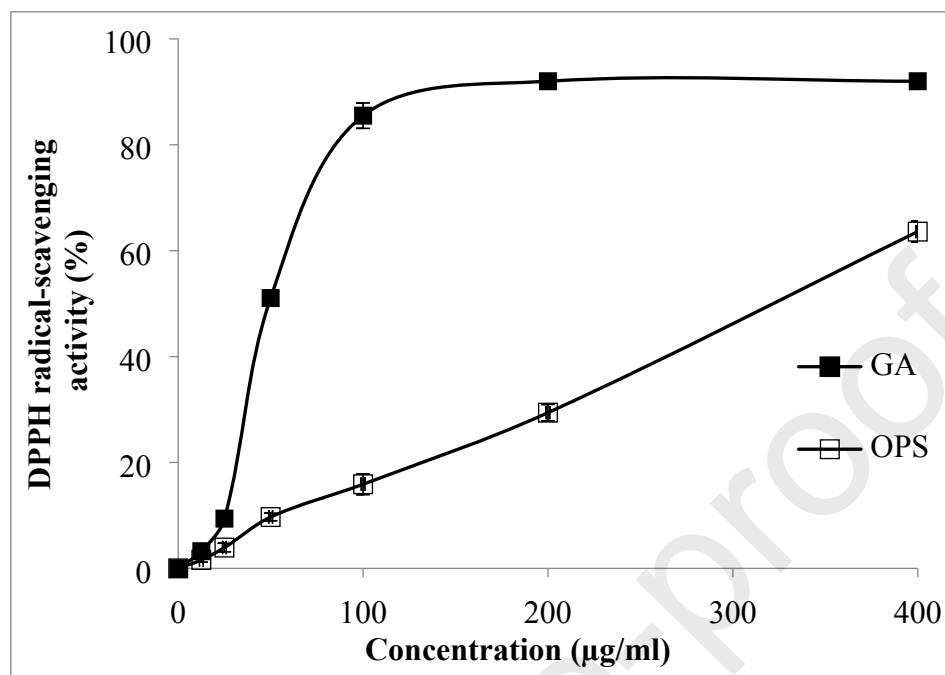
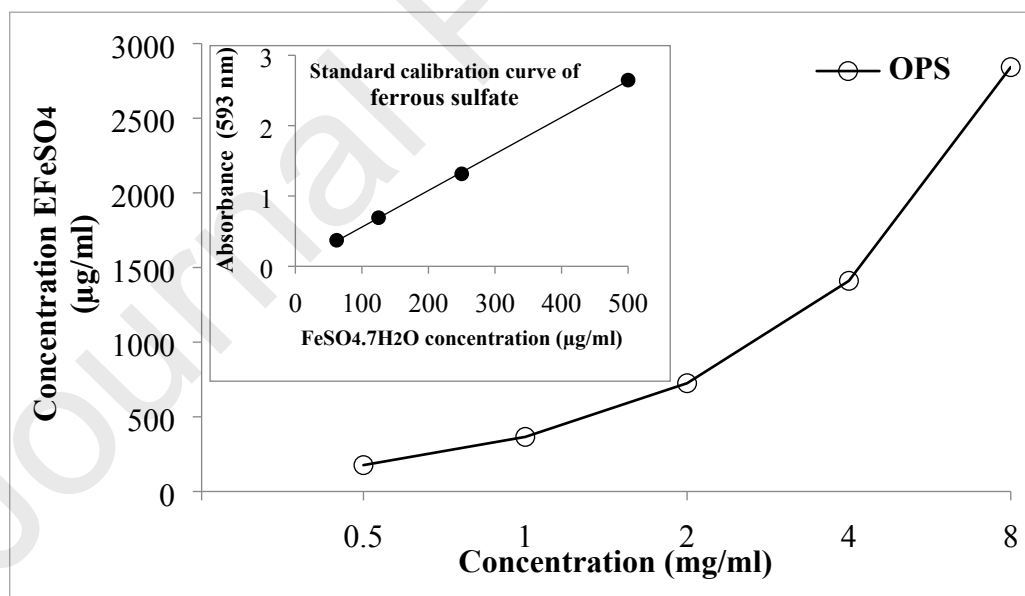


Fig.5

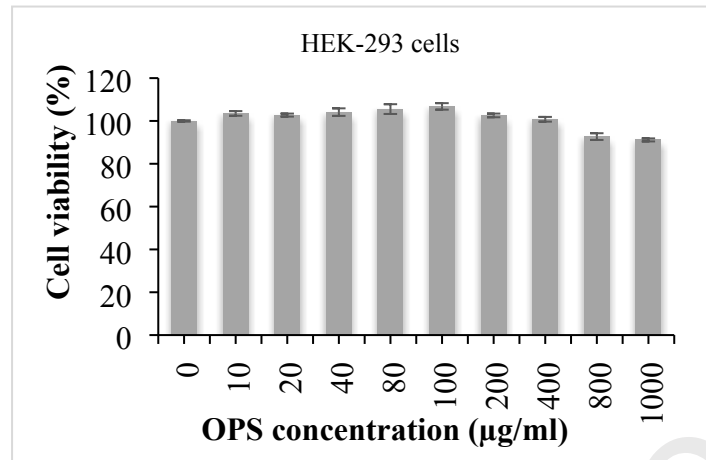
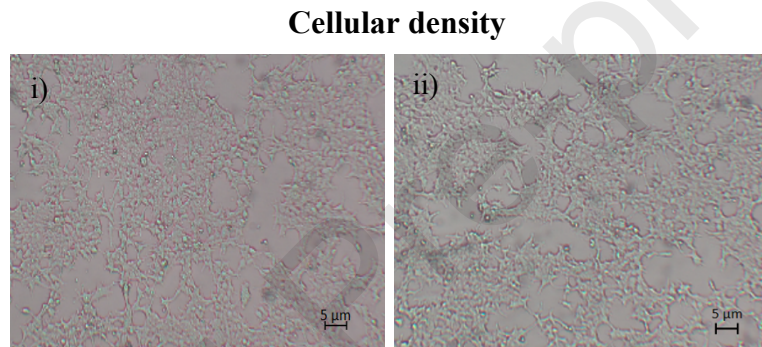
A



B



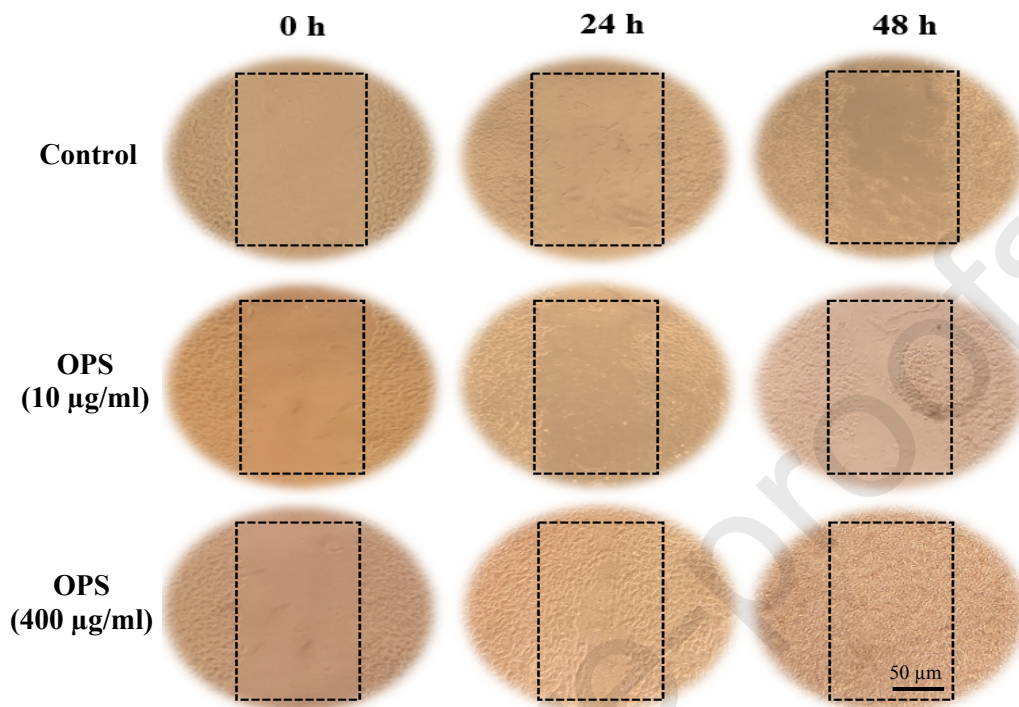
Gallic acid (GA) was used as positive control; EFeSO<sub>4</sub>: Equivalent FeSO<sub>4</sub>  
Values are means ± SD (n = 3)

**Fig.6****A****B**

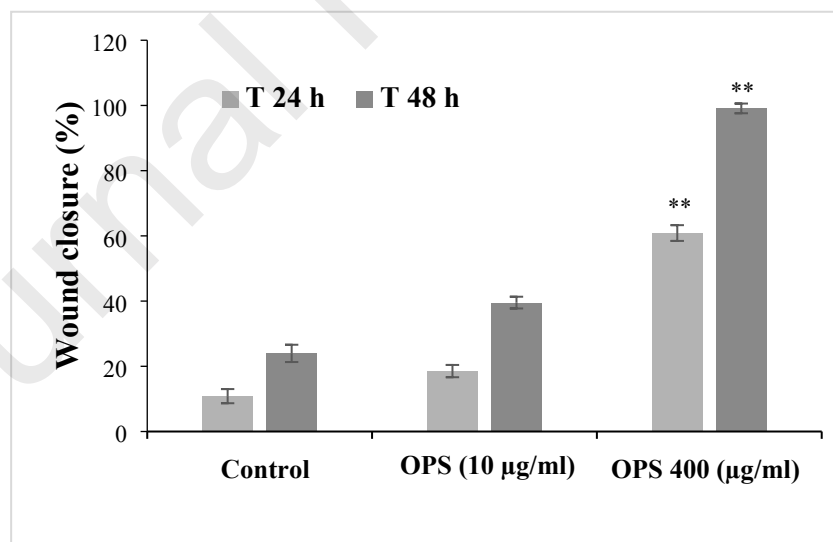
HEK293 cells ( $1 \times 10^5$  live cells/ml) were cultured in different concentrations of OPS (0–1000 µg/ml) for 48 h. The cells treated with medium alone were considered 100% viable (control). Cell morphology was observed under inverted phase contrast microscope (magnification 40x).

Fig.7

A



B

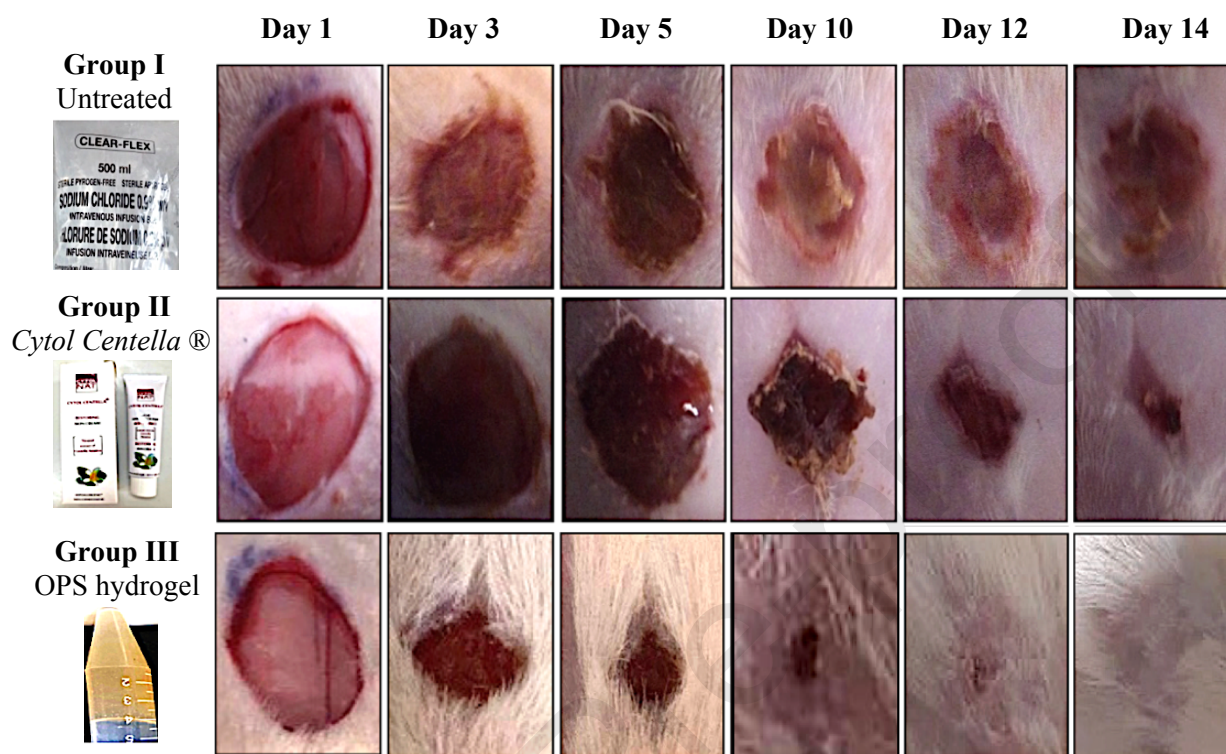


Cells were treated with 10 and 400 µg/ml of OPS for 24 and 48 h and untreated cells were used as control. Percentage of wound closure was measured and presented on a histogram using Image-J software. \*P < 0.05 and \*\*P < 0.005 vs untreated cells (Control). Percentage of wound healing was measured and presented on a histogram using Image-J software.

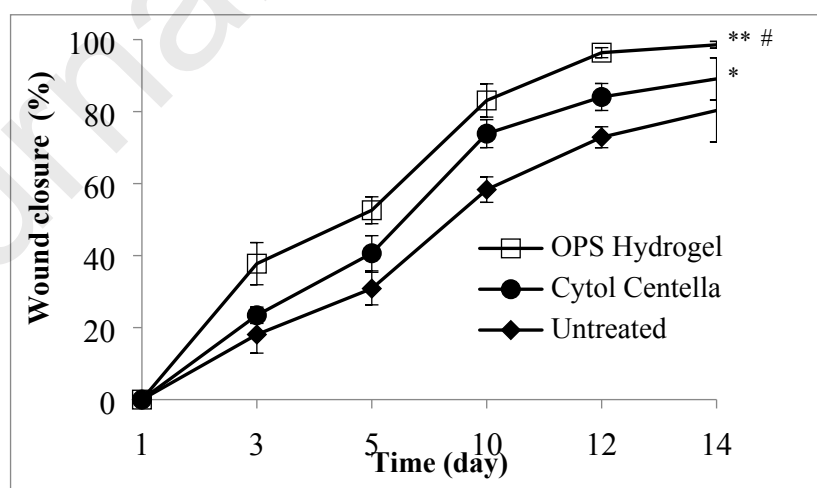


Fig.8

A



B



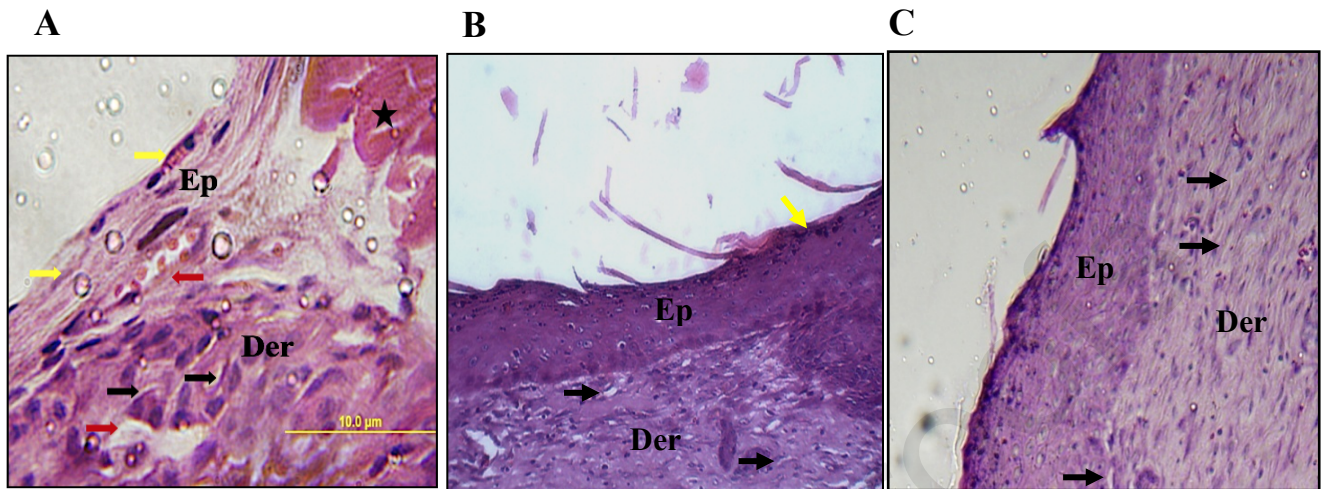
The photographs of the wounds are representative of six rats in each group;

The data represent the mean of six rats.

\* $P < 0.05$  and \*\* $P < 0.005$  vs untreated diabetic group (Group I)

# $P < 0.05$  vs the reference group (Group II)

Fig.9



A, B and C correspond to untreated, *Cytol Centella*-treated and OPS hydrogel-treated groups;

Ep: epidermis ; Der: dermis ;  $\rightarrow$  : inflammatory cells ;  $\rightarrow$  : blood vessels ;  $\rightarrow$  : vessel's congestion ;  $\star$  : inflammatory cells

### Conflict of Interest

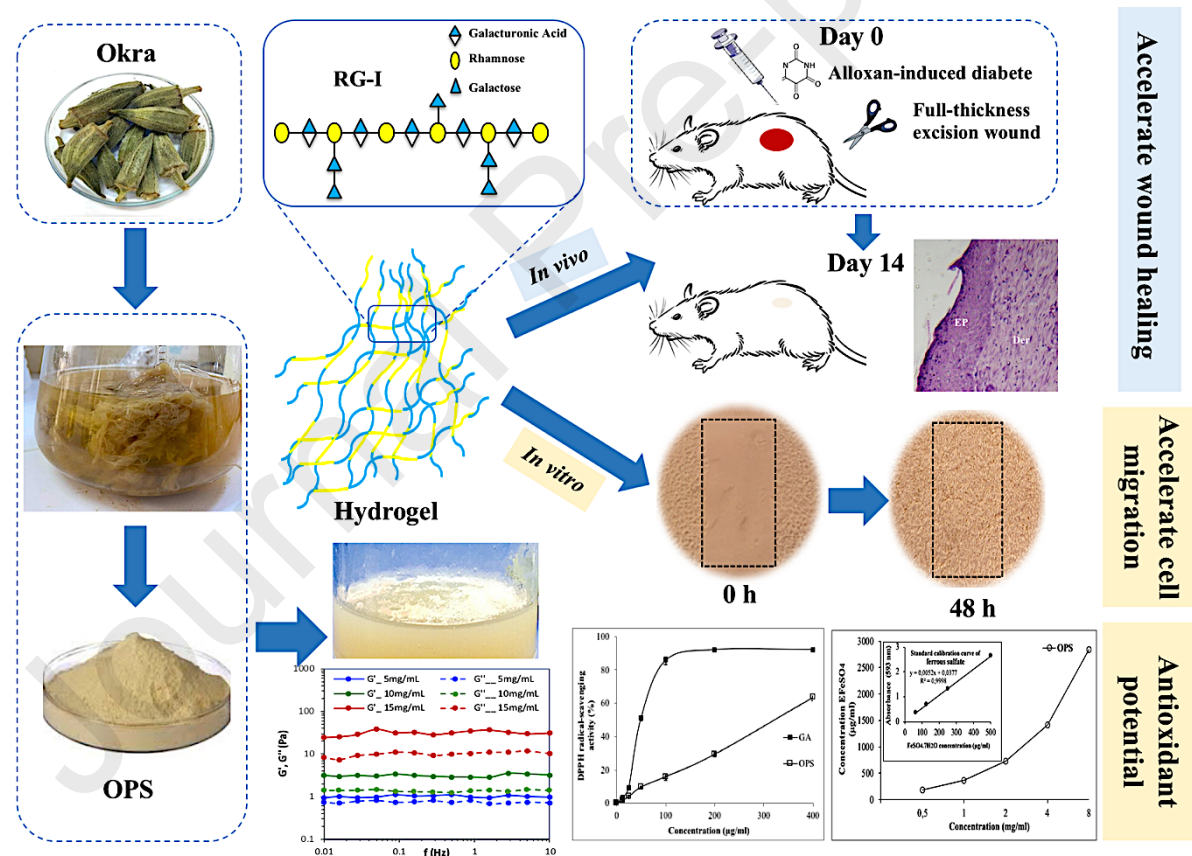
The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

## Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

## Graphical Abstract



## Highlights

- RG-I pectic polysaccharide (OPS) was extracted and purified from okra pods.
- Physico-chemical features of OPS were elucidated by FTIR, GC, SEC-MALLS and NMR.

- OPS solution exhibited a gel-like behavior.
- OPS demonstrated notable antioxidant capacity and stimulated cell migration and proliferation.
- OPS hydrogel effectively accelerated wound healing in alloxan-induced diabetic rats.

Journal Pre-proofs