
Influence of typology and management practices on water pCO₂ and atmospheric CO₂ fluxes over two temperate shelf – estuary – marsh water continuums

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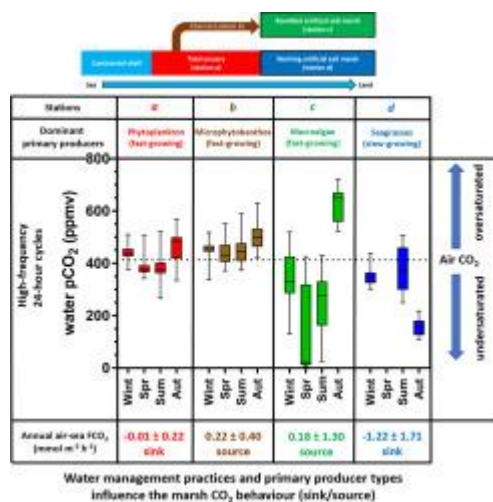
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Abstract :

Within the coastal zone, salt marshes often behave as atmospheric CO₂ sinks, allowing for blue carbon (C) sequestration associated with intense autotrophic metabolism. However, C dynamics over salt marshes are complex since various biogeochemical processes and fluxes occur at different terrestrial – aquatic – atmospheric exchange interfaces and spatiotemporal scales. This study focuses on seasonal, tidal and diurnal variations of water pCO₂, estimated water-air CO₂ fluxes and controlling factors along two temperate shelf – estuary – marsh continuums. The latter include typical coastal systems with artificial salt marshes that have contrasting water management practices and primary producer types. Our high-frequency biogeochemical measurements (seasonal 24-hour cycles) highlighted a strong control of ecosystem typology on inorganic C dynamics with lower water pCO₂ values in the artificial salt marshes, due to stronger biological activity and longer water residence times, than in the tidal estuary. In the marine-dominated estuary, water pCO₂ variations (267 - 569 ppmv) were strongly controlled by tidal effects and phytoplankton activity particularly in spring/summer. On the contrary, the greatest amplitudes in water pCO₂ were recorded in the artificial salt marshes (6 - 721 ppmv) due to intense macrophyte activity. In the rewilded marsh, nutrient inputs favoured spring/summer fast-growing macroalgae produced, in turn, strong fall atmospheric CO₂ outgassing from degraded algae waters and thus a net annual source of CO₂ to the atmosphere (17.5 g C m⁻² yr⁻¹). Conversely, specific management practices at the working marsh for salt-farming activity favoured rather slow-growing macrophytes (i.e. seagrasses) which greatly contribute to the yearly observed atmospheric CO₂ sink (-97.7 g C m⁻² yr⁻¹). In this work, we suggest that salt marsh management can be used to control the contribution of primary producers to marsh C budget as atmospheric CO₂ (sink and/or source).

Graphical abstract



Highlights

► Water pCO₂ variations exist along sea – land continuums according to station typology. ► Large pCO₂ amplitudes were recorded in artificial marshes at the diurnal/tidal scale. ► Water management practices in artificial salt marshes modulate their CO₂ behaviour. ► Fast-growing macroalgae in salt marshes produce a net atmospheric CO₂ degassing.

Keywords : shelf – estuary – marsh continuums, water pCO₂, air-water CO₂ fluxes, diurnal, tidal, seasonal scales, marsh management practices, macrophytes

1. Introduction

1 Marine coastal environments, which only account for 7% of the global ocean, perform
2 major ecological functions such as primary production, bacterial mineralization and organic
3 matter burial (Gattuso et al., 1998). The coastal zone presents a wide diversity of
4 geomorphological types and ecosystems (shelves, estuaries, bays, wetlands) shaping the
5 biogeochemical cycle coupling between land, ocean and atmosphere (Aufdenkampe et al.,
6 2011; Bauer et al., 2013). These dynamics and heterogeneous ecosystems vertically exchanges
7 large and variable quantities of carbon (C) with the atmosphere (Cole et al., 2007; Polsenaere
8 et al., 2012). At the global scale, continental shelves behave as atmospheric CO₂ sinks and
9 absorb 0.25 ± 0.05 Pg C yr⁻¹ (Bauer et al., 2013; Dai et al., 2022) due to phytoplankton primary
10 production (Cloern et al., 2014). On the contrary, CO₂ supersaturated estuarine waters emit 0.25
11 ± 0.05 Pg C yr⁻¹ to the atmosphere (Bauer et al., 2013) due to a high mineralization of organic
12 matter from the land (Frankignoulle et al., 1998; Borges and Abril, 2011). These atmospheric
13 C exchanges within the coastal zone are heterogeneous (Borges et al., 2005) and need to be
14 better taken into account in regional and global C budgets (Najjar et al., 2018). For instance,
15 coastal wetlands, including salt marshes located along inner shelf – estuary – marsh
16 continuums, absorb 0.55 ± 0.05 Pg C yr⁻¹ from the atmosphere (Bauer et al., 2013) and may
17 play a major role in atmospheric CO₂ uptake and associated organic C burial on Earth (Cai,
18 2011; Mcleod et al., 2011).

19 In salt marshes, inorganic C dynamics and water pCO₂ are influenced by several
20 physicochemical and biological processes within and between each ecosystem compartment
21 such as tidal exchanges, calcium carbonate precipitation/dissolution, benthic-pelagic coupling,
22 air-water exchanges and photosynthesis/respiration balance (Cai, 2011; Bauer et al., 2013;

23 Macreadie et al., 2017). Due to high photoautotrophy rates of both aquatic (phytoplankton and
24 macrophytes) and terrestrial (vascular plants) primary producers (Tobias and Neubauer, 2019),
25 these highly productive ecosystems mostly behave as net atmospheric C sinks (Schäfer et al.,
26 2014; Artigas et al., 2015; Forbrich and Giblin, 2015). A refractory part of organic C produced
27 through photosynthesis in these vegetated coastal ecosystems can then be sequestered in
28 sediments (Chmura et al., 2003) and stored as blue C, and greatly contribute to the
29 regional/global C cycle in comparison with terrestrial ecosystems (Mcleod et al., 2011). Salt
30 marshes also produce and horizontally export significant quantities of C through tidal water
31 advection (Najjar et al., 2018) which could, in turn, strongly influence the C balance of the
32 system itself as well as the estuary and shelf systems (Cai, 2011). The “marsh CO₂ pump”
33 hypothesis proposes that atmospheric CO₂ fixation by plants and phytoplankton in marshes and
34 the export of part of the associated C may be one of the major mechanisms making adjacent
35 coastal waters sources of CO₂ to the atmosphere (Wang and Cai, 2004). For instance, in a tidal
36 marsh area (USA; 12300 km²), Wang et al. (2016) estimated that 56% (39% inorganic and 17%
37 organic C forms) of its total net CO₂ fixation was exported to the coastal ocean. Nevertheless,
38 despite these major ecological potentials (storm protection, nursery areas, long-term C storage),
39 these interface areas are the most threatened in the world by land-use changes, climate changes
40 and sea level rise (Gu et al., 2018). Moreover, coastal eutrophication causes the loss of salt
41 marshes by decreasing the below-ground biomass of plant roots through microbial degradation
42 thereby producing a decrease in the geomorphic stability of marshes (Deegan et al., 2012).
43 Since the 1800s, salt marshes have lost about 25% of their global area with negative effects on
44 the atmospheric CO₂ sink and the associated C sequestration (Mcleod et al., 2011). Their
45 importance as ecosystem service reservoirs has made it possible to implement protection and

1 46 restoration policies that contribute to their better management and to the development of their
2 47 ecological and economic potentials (Gu et al., 2018; Adam, 2019).
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5 48 The high heterogeneity in biogeochemical processes within coastal systems at spatial and
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7 49 temporal scales (Cai, 2011; Bauer et al., 2013) requires more integrative C process and
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9 50 exchange measurements at the various terrestrial – aquatic – atmospheric interfaces over
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11 51 different time scales (tidal, diurnal and seasonal) to better understand the ecological functioning
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13 52 of these ecosystems facing global changes. Some studies in coastal wetlands such as salt
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15 53 marshes or tidal estuaries have taken water pCO₂ measurements at different temporal scales
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17 54 allowing the study of in situ CO₂ dynamics in relation to other biotic and abiotic processes. For
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19 55 instance, in an intertidal mangrove (Gaderu Creek, India), Borges (2003) showed a strong
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21 56 control of diurnal pCO₂ variations by tides and biological activity (primary production and
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23 57 respiration). However, still too few studies have taken high-frequency water pCO₂
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25 58 measurements in salt marshes at the diurnal and tidal scales. These temporal variations in water
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27 59 pCO₂ strongly affect associated air-water CO₂ fluxes that can, in turn, be estimated from the
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29 60 CO₂ gas transfer velocity, CO₂ solubility in the water and air-water CO₂ gradient (Borges, 2003;
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31 61 Crosswell et al., 2017). The atmospheric Eddy Covariance technique represents an alternative
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33 62 way to directly measure atmospheric CO₂ fluxes at the ecosystem scale (Baldocchi et al., 1988;
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35 63 Schäfer et al., 2014). This non-intrusive micrometeorological technique allows to study the
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37 64 metabolism of coastal ecosystems (sink or source) under real field conditions and to integrate
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39 65 them into regional C budgets (Polsenaere et al., 2012; Van Dam et al., 2021).
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50 66 The purpose of this study was to better understand CO₂ dynamics at different temporal
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52 67 scales and locations over two aquatic sea – land continuums along the Atlantic French coast on
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54 68 Ré Island. These continuums include coastal systems (shelf, estuary, marsh) such as those
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69 studied elsewhere by Cai (2011) and Bauer et al. (2013) with regards to horizontal and vertical
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2 70 C exchanges in the coastal ocean. Unlike tidal salt marshes, which are more generally discussed
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5 71 in the literature (Wang et al., 2016), here we studied two artificial salt marshes (i.e. salt ponds)
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7 72 in which water exchanges are controlled by dykes and locks for human uses (biodiversity
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10 73 protection or anthropogenic activities). Through in situ high-frequency measurements of
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12 74 biogeochemical parameters in waters and estimations of atmospheric CO₂ fluxes from 2018 to
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14 75 2020, we sought to (1) identify biophysical controlling factors of water pCO₂ by establishing
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16 76 biogeochemical relationships both at the seasonal and diurnal/tidal scales, (2) highlight the
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19 77 influence of continuum typologies on measured biogeochemical parameters and (3) identify
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22 78 role of station typologies and salt marsh management practices on temporal pCO₂ dynamics
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24 79 and associated CO₂ budgets. The results allowed us to contextualize the associated continuum
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27 80 metabolism among other studied systems from a C dynamic and budget point of view.
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32 82 **2. Materials and methods**

35 83 **2.1. Study sites**

38 84 **2.1.1. Tidal estuary (station *a*) and channel (station *b*)**

41 85 The Fier d'Ars estuary is a semi-closed maritime area of 750 ha on Ré Island within the
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43 86 French Atlantic Ocean and connected to the Breton Sound continental shelf (Fig. 1). It
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46 87 corresponds to a type II temperate tidal estuary according to Dürr et al.'s (2011) coastal system
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48 88 typology. At low tide (LT), its subtidal zone (in light blue; Fig. 1) is composed of mudflats
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51 89 (slikke) and tidal salt marshes (schorre) traversed by numerous channels of different sizes. At
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53 90 high tide (HT), the subtidal zone is flooded by coastal waters from the shelf up to the dykes
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56 91 (Fig. 1), managed to control water exchanges between the estuary and upstream artificial salt

92 marshes. Station *a*, with a maximum water height of 6.5 m, is located within the subtidal zone
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2 93 of the estuary along the main channel connected to the slikke (Fig. 1). Station *b* is a secondary
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4 94 tidal channel associated to the schorre and located at the back of the estuary before the dyke.
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7 95 With a maximum water height of 5.3 m, it is connected to the station *a* (distance of 1.6 km)
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9 96 enabling the supply of coastal waters to artificial salt marshes upstream from the dyke (Fig. 1).
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12 97 Artificial salt marshes are old tidal salt marshes divided into multiple ponds by dykes
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14 98 mainly located along European coasts for which water residence times (from a few hours to
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16 99 fifteen days according to the management practices; Bel Hassen, 2001) were originally
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18 100 controlled for salt-farming through locks (Tortajada et al., 2011).
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2.1.2. Rewilded artificial salt marsh (station *c*)

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29 103 Station *c* is a rewilded artificial salt marsh upstream from the dyke (surface area of
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31 104 40100 m², depth of 60 cm), protected and managed inside a National Natural Reserve (NNR).
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34 105 During HT periods, this rewilded marsh is supplied indirectly with coastal waters from the
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36 106 estuary by the station *b* channel (distance of 500 m between stations *b* and *c*; Fig. 1) through a
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38 107 lock management practice to promote biodiversity protection (former salt farm that has been
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40 108 rewilded; Fig. A.1). From November until March (winter period), the marsh lock is open only
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42 109 during the highest tidal amplitudes in order to have the best compromise between salt- and
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44 110 fresh-mixing waters (salinity around 30) to allow aquatic fauna passing from the continental
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46 111 shelf to the marsh. From April to October (summer period) with lower tidal amplitudes, the
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48 112 lock is permanently open to avoid large salinity fluctuations and favour the development of
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50 113 *Ruppia spp.* seagrass beds in the marsh (salinity between 30 and 45). Lately, this rewilded marsh
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114 is characterized by significant macroalgae development from early spring to late summer each
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2 115 year thereby preventing seagrass development (Champion et al., 2012).
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8 117 **2.1.3. Working artificial salt marsh (station *d*)**

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11 118 Station *d* is a working artificial salt marsh upstream from the dyke (surface area of 8500 m²,
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13 119 depth of 75 cm) directly communicating (no channel in between) with coastal waters from the
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15 120 estuary through a lock (distance of 2 km between stations *a* and *d*; Fig. 1). This working marsh
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17 121 was chosen for its specific management practice of the lock related to a salt-farming activity.
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19 122 In spring and summer, the lock of is regularly open to store salt waters and then allow to supply
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21 123 upstream ponds succession during the neap tides for the salt production through the evaporation
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23 124 (Fig. A.1). Moreover, the use of this working marsh requires a drying up and a cleaning once a
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25 125 year in early spring before the start of the salt production period to remove seagrass, macroalgae
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27 126 and organic matter in the marsh (Poitevin, personal communication).
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37 128 **2.1.4. Additional station: continental shelf (station *F*)**

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39 129 The Breton Sound corresponds to a coastal maritime area located on the French continental
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41 130 shelf, characterized by a surface area of 425 km² (Fig. 1). The Breton Sound continental shelf
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43 131 exchanges salt waters with the Atlantic Ocean to the west at each semi-diurnal tidal cycle and
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45 132 receives continental inputs through the Aiguillon Bay discharges to the east depending on
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47 133 hydrodynamic and meteorological conditions (Stanisiere et al., 2006; Soletchnik et al., 2015).
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49 134 The highest and lowest river water flows were recorded in winter and summer, respectively,
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51 135 influencing salinity of the shelf waters differently. Station *F* in the centre of the Breton Sound
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136 (Fig. 1) is located in a predominantly marine environment with a low freshwater contribution
137 (Stanisiere et al., 2006; Soletchnik et al., 2015). At each HT, the continental shelf supplies our
138 studied stations (*a*, *b*, *c* and *d*) with various water masses based on the tidal amplitudes and
139 seasonal periods along two aquatic sea – land continuums: (1) continental shelf (station *F*) –
140 estuary (station *a*) – channel (station *b*) – rewilded salt marsh (station *c*) and (2) continental
141 shelf (station *F*) – estuary (station *a*) – working salt marsh (station *d*). Conversely, at each LT,
142 different water masses from salt marshes are exported (indirectly through the station *b* channel
143 for station *c* or directly for station *d*) to the estuary and then to continental shelf (Fig. 1).

144 In this study, meteorological parameters (air temperature, rain, wind speed) corresponding
145 to our measurement cycles were used from the Eddy Covariance station (Campbell Scientific)
146 deployed on a nearby tidal salt marsh (station *e*; Fig. 1).

2.2. Measurement strategy and biogeochemical measurements

149 In the subsurface water (~30 cm depth), partial pressure of CO₂ (pCO₂), temperature,
150 salinity, turbidity, dissolved oxygen concentration (DO) and pH were autonomously measured
151 with in situ probes at a frequency of 1 min. during fifteen 24-h cycles at stations *a*, *b*, *c* and *d*
152 during each season (Table A.1). These high frequency measurements allowed to record relevant
153 temporal (diurnal, tidal and seasonal) and spatial (continuums) variations of water pCO₂ and
154 associated physicochemical parameters. The diurnal scale corresponds to fluctuations occurring
155 between daytime and night-time whereas, the tidal scale corresponds to fluctuations between
156 LT and HT. The seasonal measurement cycles were performed in 2018 at stations *a* and *b* and
157 in 2019/2020 at stations *c* and *d*. Measurements could not be taken at station *d* in spring 2020

158 due to the Covid pandemic. At station *F*, the same biogeochemical measurements were taken
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159 biweekly over the year 2018 (Table A.1) by Coignot et al. (2020).

160 A pCO₂ underwater probe (C-Sense™; PME/Turner Designs), a multiparameter probe
161 (EXO2; YSI) and a submersible fluorometer (C3™; Turner Designs) were deployed to measure
162 water pCO₂, physicochemical parameters and fluorescence, respectively. The measurement
163 range of the C-Sense probe is 0-2000 ppmv with an absolute accuracy of 60 ppmv (3% of the
164 full scale; Turner Designs). The C-Sense probe was calibrated by the manufacturer before the
165 study. The EXO2 probe was used to measure temperature (± 0.01 °C), salinity (± 0.5 salinity
166 unit), turbidity (± 0.3 NTU), DO concentration (± 3.1 $\mu\text{mol L}^{-1}$), DO saturation level ($\pm 1\%$)
167 and pH (± 0.01 pH unit). The pH was calibrated before and after each 24-h cycle using three
168 YSI buffer solutions (pH 4.01, pH 7.00 and pH 10.01) as outlined by Aminot and K  rouel
169 (2004). It was not possible to measure pH at stations *a* and *b*. The C3-fluorometer was used to
170 estimate the sub-surface Chl *a* values from the 10-min. fluorescence data. This latter was
171 deployed only at station *c* and *d* in summer 2019 and winter 2020.

172 Water pCO₂ measured by the C-Sense probe are influenced by the total dissolved gas
173 pressure (TDGP) which corresponds to the total pressure exhibited by all gases within the water
174 column. When this pressure greatly exceeded the pressure at which the C-Sense probe was
175 calibrated, the output needed to be corrected. Then, a pCO₂ correction was applied taking both
176 TDGP, atmospheric pressure during sensor calibration (1009 hPa) and the measured pCO₂ by
177 the C-Sense probe into account, as per equation $(\text{pCO}_{2\text{measured}} \times 1009) / \text{TDGP}$ (Turner Designs).
178 Over all 24-h cycles, the corrected pCO₂ values with TDGP were $2.6 \pm 0.9\%$ lower than the
179 measured pCO₂ values. Total alkalinity (TA) and dissolved inorganic carbon (DIC) were
180 estimated from salinity, temperature, pH and water pCO₂ using the carbonic acid constant from
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181 Mehrbach et al. (1973) as modified by Dickson and Millero (1987), the K_{HSO_4} constant from
 182 Dickson (1990) and the borate acidity constant from Lee et al. (2010). The CO_2 system
 183 calculation program (version 2.1.) performed these calculations (Lewis and Wallace, 1998).

2.3. Temperature and non-temperature effects on pCO_2 variations

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 186 To distinguish between the temperature and non-temperature effects on in situ pCO_2
 187 variations at the seasonal and diurnal scales, TpCO_2 (pCO_2 variations related to temperature
 188 physical effects, in ppmv) and NpCO_2 (pCO_2 variations related to non-temperature effects, in
 189 ppmv) were calculated respectively, following (Eq. 1) and (Eq. 2) from Takahashi et al. (2002):

$$190 \quad \text{TpCO}_2 = \text{pCO}_{2\text{mean}} \times \exp[0.0423 \times (T_{\text{obs}} - T_{\text{mean}})] \quad (1)$$

$$191 \quad \text{NpCO}_2 = \text{pCO}_{2\text{obs}} \times \exp[0.0423 \times (T_{\text{mean}} - T_{\text{obs}})] \quad (2)$$

192 where T_{obs} and $\text{pCO}_{2\text{obs}}$ are the temperature and pCO_2 values measured by the probes at each
 193 time step (1 min.), respectively. T_{mean} and $\text{pCO}_{2\text{mean}}$ are the temperature and pCO_2 averaged
 194 either at the seasonal (annual mean) or diurnal scale (means per 24-h cycle). TpCO_2 is only
 195 associated with the physical pump whereas, NpCO_2 is associated with biological processes,
 196 tidal advection and benthic-pelagic coupling (Cotovicz Jr. et al., 2015; Polsenaere et al., 2022).

2.4. Calculations of air-water CO_2 fluxes

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 199 For all 24-h measurement cycles, the gas transfer velocity (k_{600}) and hourly CO_2 fluxes
 200 (FCO_2) at the air-water interface were estimated following Ribas-Ribas et al. (2011) and
 201 Polsenaere et al. (2022) in coastal environments. At stations *a* and *b*, only air-water FCO_2 during

202 HT periods (four hours around each HT) were calculated whereas at stations *c* and *d*, all hourly

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2 203 FCO₂ were calculated using the following formula (Eq. 3):
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$$5 \quad 204 \quad \text{FCO}_2 = \alpha \times k \times \Delta\text{pCO}_2 \quad (3)$$

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8 205 where FCO₂ (mmol m⁻² h⁻¹) is the estimated air-water CO₂ fluxes, α (mol kg⁻¹ atm⁻¹) is the CO₂
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10 206 solubility coefficient in saltwater, k (cm h⁻¹) is the gas transfer velocity of CO₂ and ΔpCO_2

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13 207 (ppmv) is the gradient between mean water and air pCO₂. Water pCO₂ were measured by the
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16 208 C-Sense probe. Atmospheric CO₂ concentrations were measured by (1) the Eddy Covariance
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18 209 (station *e*; Fig. 1) for summer 2019 and winter 2020 and (2) the National Oceanic and
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21 210 Atmospheric Administration (NOAA) at the Mauna Loa Observatory for all other periods (see
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23 211 values caption Table 3). The α coefficient depends on water temperature and salinity and was
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25 212 calculated according to Weiss (1974). The k coefficient also significantly controls air-water
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28 213 FCO₂ since it directly takes turbulence processes at the air-water exchange interface into
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30 214 account (Polsenaere et al., 2013). In this study, k (or k_{660}) was calculated according to both
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33 215 Raymond and Cole (2001) (RC01; Eq. 4) and Wanninkhof et al. (2022) (W22; Eq. 5)
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35 216 corresponding to closed environments and more open coastal environments, respectively. These
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38 217 two parametrization methods for the k exchange coefficient were applied in order to compare
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40 218 the results.
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43 219 For closed freshwater environments (Raymond and Cole, 2001):
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$$45 \quad 46 \quad 220 \quad k_{600} = 1.91 \times \exp[0.35 \times U_{10}] \quad (4)$$

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49 221 For more open coastal environments (Wanninkhof et al., 2022):
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$$51 \quad 52 \quad 53 \quad 222 \quad k_{600} = 0.31 \times (U_{10})^2 \quad (5)$$

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223 The gas transfer coefficients normalized to a Schmidt number of 600 (k_{600}) obtained with the
224 two parametrization were then converted to the gas transfer velocity of CO₂ at the in situ
225 temperature and salinity (k_{660}) according to Jähne et al. (1987) as per the equation (6):

$$k_{660} = k_{600} / (660/Sc)^{0.5} \quad (6)$$

227 where k_{660} (cm h⁻¹) is the gas transfer velocity of CO₂ at the in situ temperature and salinity
228 according to the parametrizations of RC01 or W22, U_{10} (m s⁻¹) is the wind speed normalized to
229 10 m (Amorocho and DeVries, 1980) and Sc is the Schmidt number which describes both the
230 water viscosity and the molecular diffusion of the subsurface layer (Bade, 2009). In summer
231 2019 and winter 2020, wind speeds were measured by the Eddy Covariance (station *e*; Fig. 1)
232 at a height of 3.15 m; for all other periods, wind data were obtained from the “Infoclimat”
233 station (Fig. 1) measured at a height of 10 m (distances of 6.20, 4.85, 4.30 and 8.40 km from
234 stations *a*, *b*, *c* and *d*, respectively).

2.5. Chl *a* concentrations and fluorometer data calibration

237 In situ Chl *a* concentrations (µg L⁻¹) were measured from sub-surface water samples
238 collected only at stations *c* and *d*. Water samples (50-100 mL) were filtered through GF/F filters
239 (Whatman® Nuclepore™, porosity of 0.7 µm) and stored at -20°C until analysis. In the dark,
240 Chl *a* was extracted in 90% acetone with a glass rod. After 12 h of stirring at 4 °C to continue
241 the extraction, Chl *a* was analysed by monochromatic spectrophotometry at 665 nm (Aminot
242 and K  rouel, 2004).

243 For the fluorometer data, the calibration procedure was applied to derive Chl *a* values from
244 our water fluorescence measurements (Aminot & K  rouel, 2004). Chl *a* could be calculated

245 only at station *d* through the significant linear regressions ($p < 0.05$) between the fluorometer
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2 246 values and the in situ Chl *a* values sampled simultaneously in the marsh waters.
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8 248 **2.6. Statistical tools and analysis**

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11 249 For all measured variables, the high-frequency data (i.e. 1 min) did not respect a normal
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13 250 distribution (Shapiro-Wilk tests, $p < 0.05$). Non-parametric comparison tests such as the Mann-
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15 251 Whitney and Kruskal-Wallis tests were carried out with 0.05 level of significance. A Dunn test
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17 252 was used to perform a post-hoc multiple comparison of the Kruskal-Wallis test to detect
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19 253 significant differences among groups. The statistical tests as well as temporal graphs, linear
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21 254 regressions, boxplot and barplot were performed with the GraphPad Prism 7 software. The R-
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23 255 studio software was used to perform the principal component analysis (PCA) with the
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25 256 FactoMineR package (Lê et al., 2008) and the correlation matrices with the corrplot package
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27 257 (Wei and Simko, 2017). The PCA allows to study the distribution of seasonal data along the
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29 258 studied continuums (Fig. 2). It is based on the seasonal means of the temperature, salinity,
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31 259 turbidity, DO and pCO₂ measured (i) once every two weeks at station *F* and (ii) once every
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33 260 minute over 24-h cycles at stations *a*, *b*, *c* and *d*. Stepwise multilinear regression analysis were
34
35 261 performed to test the contribution of measured physicochemical variables (salinity,
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37 262 temperature, turbidity and oxygen) on water pCO₂ variations through the percentage of
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39 263 explained variance (adjusted R²; Harrell, 2015). Within each measurement cycle, the selected
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41 264 multilinear model ($p < 0.001$, $n = 1441$) had the highest adjusted R² with all variables explaining
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43 265 at least 5% of the pCO₂ variation. Analysis were performed with Statgraphics Centurion 19
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45 266 software.
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268 3. Results

269 3.1. Biogeochemical overview of the aquatic continuums

270 Over our measurement periods, thermal conditions for the years 2018 and 2019 were similar
271 following a classical seasonal trend. However, July 2018, July 2019 and February 2020 were
272 warmer than the 1990-2020 reference period (Table 1). Annual cumulative precipitations in
273 2018 and 2019 were higher than the historical data with March 2018 and October 2019 as the
274 rainiest months (Table 1). Salinity at station *F* as the water source flowing into the two
275 continuums did not vary significantly between the years 2018, 2019, 2020 and the 2000-2017
276 reference period (Kruskal-Wallis test, $p = 0.77$; Fig. A.2).

277 At station *F*, over the biweekly measurements in 2018, water temperature varied from 7.5
278 (winter) to 21.7 °C (summer) whereas at station *a* over our seasonal 24-h cycles, values varied
279 from 9.1 (winter) to 26.9 °C (summer). Along the aquatic continuum, the water temperature
280 significantly increased from station *a* to stations *b* and *c* (Mann-Whitney tests, $p < 0.05$).
281 Salinity ranged from 28.9 (winter) to 35.4 (autumn) at station *F*, whereas values varied from
282 31.4 (winter) to 35.7 (autumn) at station *a*, from 27.5 (winter) to 36.9 (autumn) at station *b*,
283 from 27.0 (winter) to 42.6 (summer) at station *c* and from 21.3 (winter) to 38.4 (autumn) at
284 station *d* with large salinity gradients at stations *c* and *d* (Table 2). In average over the year, the
285 station *a* waters were slightly oversaturated in oxygen compared to the atmosphere with DO
286 saturation levels ranging between 70 (LT during dawn) and 150% (LT during dusk) during the
287 summer cycle (Fig. 3). The station *b* waters were close to the saturation value with the
288 atmosphere with a lower maximum value (120%) during the summer cycle (Fig. 4). Larger
289 amplitudes of DO saturation level were recorded in the artificial salt marshes with values
290 ranging from 36 to 176% at station *c* (summer; Fig. 5) and from 49 to 150% at station *d*

291 (summer; Fig. 6). The annual levels of water CO₂ undersaturation with respect to the
 292 atmosphere were 48%, 16%, 65% and 86% at stations *a*, *b*, *c* and *d*, respectively, with a strong
 293 annual CO₂ oversaturation at station *b* (Fig. 4). The greatest amplitude in water pCO₂ was
 294 recorded at station *c* with values varying from 6 (spring) to 721 ppmv (autumn; Fig. 5).

295 At station *F*, over the year 2018, Chl *a* values increased from winter ($0.7 \pm 0.1 \mu\text{g L}^{-1}$) to
 296 spring-summer (2.5 ± 1.6 and $1.6 \pm 1.0 \mu\text{g L}^{-1}$, respectively), before decreasing in autumn (0.8
 297 $\pm 0.5 \mu\text{g L}^{-1}$). At station *c*, the highest and lowest Chl *a* values were recorded in autumn 2019
 298 ($8.1 \pm 0.4 \mu\text{g L}^{-1}$) and winter 2020 ($1.3 \pm 0.3 \mu\text{g L}^{-1}$), respectively whereas at station *d*, the
 299 highest and lowest values were recorded in winter 2020 ($3.4 \pm 0.4 \mu\text{g L}^{-1}$) and summer 2019
 300 ($1.9 \pm 0.3 \mu\text{g L}^{-1}$), respectively. Moreover, at station *c*, from spring to autumn 2019, a free
 301 floating macroalgae development (*Ulva spp.*) was observed in the subsurface waters and on
 302 sediments (Fig. A.3). On the contrary, at station *d*, no macroalgae development occurred,
 303 allowing for the seagrass growth (*Ruppia spp.*) in the marsh (Fig. A.3).

3.2. Seasonal variations and controls along the aquatic continuums

306 Seasonally, the PCA reveals that stations along the aquatic continuums were vertically
 307 distinguished according to pCO₂, turbidity and DO saturation within PC2 explaining 35.5% of
 308 the total variance (Fig. 2). Within this axis, our results confirmed that water pCO₂ were
 309 seasonally negatively correlated with DO saturation ($r_{\text{Pearson}} = -0.56$; $n = 19$; $p < 0.05$) and
 310 positively correlated with turbidity ($r_{\text{Pearson}} = 0.54$; $n = 19$; $p < 0.05$) (Fig. 2). Station *b* recorded
 311 the highest water pCO₂ values and the lowest DO saturation levels compared to the three other
 312 studied stations (except in autumn; Table 2 and Fig. 2). Station *b* was also characterized by the
 313 highest turbidity values along the aquatic continuums (from 1.6 to 41 NTU). The PCA also

314 shows seasonal data were horizontally distinguished according to salinity and temperature
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2 315 within PC1 explaining 42.3% of the total variance (Fig. 2). Generally, the highest and lowest
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4 316 salinity values were recorded in summer and winter, respectively (Fig. 2), except at station *d*
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7 317 where the highest salinity were recorded in autumn (Table 2). At all studied stations,
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9 318 temperature and salinity values significantly varied between each seasonal 24-h cycles
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12 319 (Kruskall-Wallis tests, $p < 0.0001$).

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15 320 Along the aquatic continuums, the PCA reveals contrasted seasonal variations of water
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17 321 pCO₂, particularly in artificial salt marshes (Fig. 2). At station *F*, in 2018, no significant
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19 322 difference in water pCO₂ were recorded at the seasonal scale (Kruskall-Wallis test, $p = 0.13$),
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22 323 although the highest and lowest seasonal means were recorded in winter and spring,
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25 324 respectively (Table 2 and Fig. 7). At station *a*, in 2018, water pCO₂ showed the same seasonal
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27 325 pattern decreasing from winter to summer before increasing in autumn, whereas station *b*
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29 326 showed lower seasonal variations over the same measurement periods (Table 2 and Fig. 7). In
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31
32 327 contrast, stations *c* and *d* showed larger seasonal pCO₂ variations (Fig. 7). Station *c* waters were
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34 328 undersaturated in CO₂ in spring 2019, summer 2019 and winter 2020 but oversaturated in
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36 329 autumn 2019 (622 ± 57 ppmv). At the same time, station *d* waters were undersaturated in CO₂
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39 330 in summer, autumn and winter with the largest water CO₂ undersaturation recorded in autumn
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42 331 (155 ± 30 ppmv) in contrast to station *c* (Fig. 7). At all studied stations, water pCO₂ significantly
43
44 332 differed between seasons (Kruskall-Wallis tests, $p < 0.05$), except for station *a* between spring
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46 333 and summer (Dunn's post-test, $p > 0.99$).

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49 334 The same seasonal NpCO₂ variations were observed at stations *a* and *b* in 2018, with a
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52 335 decrease from winter (595 and 624 ppmv, respectively) to summer (296 and 347 ppmv,
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54
55 336 respectively) and an increase towards autumn (420 and 439 ppmv, respectively; Fig. 7). At
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337 station *c*, the seasonal mean NpCO_2 value increased sharply from summer 2019 (193 ppmv) to
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2 338 autumn 2019 (630 ppmv) and then decreased towards winter 2020 (441 ppmv) whereas at
3
4 339 station *d*, values slightly decreased from summer (286 ppmv) to autumn 2019 (160 ppmv)
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7 340 before increasing towards winter 2020 (453 ppmv; Fig. 7). Regarding temperature effects on
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9 341 water pCO_2 , the highest and lowest seasonal TpCO_2 values were measured in summer and
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11 342 winter, respectively, with seasonal TpCO_2 values followed systematically by seasonal water
12
13 343 temperature variations (Fig. 7). At station *a*, ΔTpCO_2 offset recorded from winter to summer
14
15 344 2018 ($\Delta\text{TpCO}_2 = 240$ ppmv, from 310 to 550 ppmv) concomitantly to the water temperature
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17 345 increase of 13.4 °C partly compensated non-thermal effects on water pCO_2 during this period
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19 346 ($\Delta\text{NpCO}_2 = 299$ ppmv, from 595 to 296 ppmv).
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348 3.3. Diurnal/tidal variations and controls along the aquatic continuums

349 At stations *a* and *b* in winter, salinity varied at the tidal scale with the lowest values at LT
350 and the highest values at HT whereas from spring to autumn, the opposite pattern was recorded
351 with salinity decreases at each incoming tide from the shelf (Figs. 3 and 4). At stations *c* and *d*,
352 even stronger salinity gradients were recorded, especially in summer with decreases of 9 and 5
353 salinity units, respectively (Figs. 5 and 6). At station *c*, coastal water inflows led to an increase
354 in salinity only in winter (Fig. 5). At station *d*, salinity and turbidity did not vary neither in
355 autumn or in winter (Fig. 6).
356

357 The largest diurnal/tidal variations in the water pCO_2 and DO concentrations occurred
358 during summer with pCO_2 ranges of 255, 216, 405 and 258 ppmv at stations *a*, *b*, *c* and *d*,
359 respectively, and DO ranges of 153.2, 152.2, 262.8 and 205.6 $\mu\text{mol L}^{-1}$ at stations *a*, *b*, *c* and *d*,
respectively. At stations *a* and *b*, the low tide periods during the day (LT/D) occurring at dawn

360 showed higher water pCO₂ values and lower DO saturation levels than the low tide periods
361 during the night (LT/N) occurring at dusk under similar salinity ranges, particularly in summer
362 (Figs. 3 and 4). In general, our diurnal cycles showed a pCO₂ decrease that was negatively
363 correlated to a DO increase during daytime (except at station *c* in spring; Fig. 5) and an opposite
364 pattern during night-time (except at station *c* in summer; Fig. 5). At station *c*, in winter during
365 the LT periods (no salinity variation), water pCO₂ decreased of 390 ppmv during the day (from
366 09:00 to 17:00) and increased of 230 ppmv during the night (from 20:00 to 05:00), while DO
367 increased of 76.6 μmol L⁻¹ and decreased of 29.0 μmol L⁻¹, respectively (Fig. 5). At station *d*,
368 the same diurnal water pCO₂ and DO patterns were recorded at each 24-h cycle (Fig. 6).
369 However, these strong diurnal pCO₂ and DO variations were significantly disrupted once
370 coastal water advection and marsh management practices occurred.

371 Strong tidal variations in water pCO₂ were recorded during all seasonal cycles except at
372 station *d* both in autumn and winter (Fig. 6). At stations *a* and *b*, incoming tides from the shelf
373 during the day produced rapid water pCO₂ decreases from an oversaturation to a slight water
374 undersaturation, particularly in spring (-121 and -167 ppmv, respectively) and summer (-139
375 and -115 ppmv, respectively; Figs. 3 and 4). Only at station *a*, ebbing tides during the day
376 generated an additional pCO₂ decrease to reach the lowest values (Fig. 3). At station *a*, in
377 summer and autumn and at station *b* over the four seasons, incoming tides during the night
378 produced pCO₂ increases leading to water oversaturation periods (Figs. 3 and 4). Along the
379 continuum, at station *c* during the night, higher water pCO₂ values were recorded at HT than at
380 LT, especially in spring (363 ± 85 and 16 ± 5 ppmv at HT/N and LT/N, respectively) and in
381 winter (431 ± 6 and 323 ± 53 ppmv at HT/N and LT/N, respectively; Fig. 8). The same tidal
382 pCO₂ pattern was also recorded at station *c* in summer during the day (323 ± 88 and 197 ± 141
383 ppmv at HT/D and LT/D, respectively; Fig. 8). In spring, the station *c* marsh recorded lowest

384 water pCO₂ values both the day and the night during the marsh confinement but coastal water
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 2 385 inflows from the station *b* channel instantly produced a large and rapid increase in water pCO₂
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 5 386 (+395 ppmv) mostly within a two-hour period (Fig. 8).
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 8 387 For all 24-h cycles, strong and significant correlations between pCO₂ and NpCO₂ were
 9
 10 388 computed (Figs. 3-6). Similarly, water pCO₂ values were negatively correlated with DO
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 12 389 saturation levels at 24-h cycles ($n = 1441$, $p < 0.05$), with r_{Spearman} ranging from -0.67 (winter)
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 15 390 to -0.97 (autumn) at station *a*, from -0.63 (summer) to -0.87 (autumn) at station *b*, from -0.54
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 17 391 (winter) to -0.86 (autumn) at station *c* and from -0.59 (winter) to -0.80 (autumn) at station *d*. At
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 19 392 station *d*, negative correlations were obtained between measured pCO₂ and estimated Chl *a* in
 20
 21 393 summer ($r_{\text{Spearman}} = -0.44$; $n = 105$; $p < 0.05$) and in winter ($r_{\text{Spearman}} = -0.60$; $n = 144$; $p < 0.05$)
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 23 394 (Fig. 6). At station *a*, in winter, the multilinear regression analyses highlighted that water pCO₂
 24
 25 395 were controlled by DO, temperature and salinity whereas over other seasons, pCO₂ were
 26
 27 396 strongly controlled only by DO with the highest R² values (Table 4). At station *a*, in spring and
 28
 29 397 summer, estimated TA values were weakly correlated with measured pCO₂ and pH values (Fig.
 30
 31 398 A.4) whereas in autumn, stronger correlations TA versus pCO₂ were recorded ($R^2 = 0.89$ and
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 33 399 $R^2 = 0.71$, respectively; $n = 1441$; $p < 0.05$). At station *c*, water pCO₂ were mainly controlled
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 35 400 by both salinity and DO in spring, by salinity in summer and by DO in autumn (Table 4).
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 37 401 Finally, at station *d*, pCO₂ were mostly explained by salinity and DO in summer (salt farming
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 39 402 period) and by DO and temperature in autumn/winter (marsh confinement periods) (Table 4).
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50 404 3.4. Air-water CO₂ flux variations

51 405 Mean air-water CO₂ fluxes (FCO₂) according to the W22 parametrization were
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 53 406 estimated to be -0.01 ± 0.22 , 0.22 ± 0.40 , 0.18 ± 1.37 and -1.22 ± 1.71 mmol m⁻² h⁻¹ at stations
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407 *a* (sink), *b* (source), *c* (source) and *d* (sink), respectively, whereas station *F* waters behaved as
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2 408 a CO₂ source ($0.30 \pm 1.04 \text{ mmol m}^{-2} \text{ h}^{-1}$). Large seasonal and diurnal variations were observed
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4 409 at the studied stations (Fig. 9). On average, station *a* showed positive FCO₂ values in both
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7 410 winter and autumn (slight CO₂ source) but negative means in spring and summer (slight CO₂
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9
10 411 sink; Table 3 and Fig. 9). At station *b*, positive FCO₂ values were estimated, with maximum
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12 412 and minimum FCO₂ mean values occurring in winter and summer, respectively (Table 3 and
13
14 413 Fig. 9). Station *c* behaved as a CO₂ sink in spring, summer and winter and as a strong CO₂
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16
17 414 source in autumn (Table 3). At this marsh station, FCO₂ varied between -3.00 and 0.03 mmol
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19 415 m⁻² h⁻¹ in spring and between 0.61 and 4.61 mmol m⁻² h⁻¹ in autumn (Fig. 9). Station *d* behaved
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21
22 416 as a CO₂ sink in summer, autumn and winter with the largest atmospheric CO₂ uptake in autumn
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24 417 (Table 3) where FCO₂ varied between -6.03 and -1.79 mmol m⁻² h⁻¹ (Fig. 9).
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30 419 4. Discussion

31 32 33 420 4.1. Biogeochemical parameter relationships and pCO₂ controls along the shelf - 34 35 421 estuary - marsh continuums 36 37

38 422 At the tidal estuary (*a*) and its associated channel (*b*), seasonal non-thermal effects
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41 423 (biological and tidal affects) inducing heterotrophy in winter and autotrophy in summer were
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43 424 offset by thermal effects and resulted in low seasonal variations of water pCO₂ (Fig. 7). Similar
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46 425 observations were reported in two marine-dominated estuaries (Jiang et al., 2008) and in the
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48 426 Arcachon coastal lagoon (Polsenaere et al., 2022). In the estuarine waters here, thermal effects
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51 427 decreased and increased in situ pCO₂ values of 30 and 40% in winter and summer, respectively.
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53 428 On the contrary, at the rewilded marsh (*c*), seasonal water pCO₂ were strongly controlled by
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56 429 non-thermal effects promoting autotrophy both in spring and summer and heterotrophy in
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430 autumn, mostly due to macroalgae activity whereas at the working marsh (*d*), thermal effects
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2 431 in summer and biological effects in autumn strongly controlled seasonal water pCO₂ (Fig. 7).
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5 432 At the diurnal scale, the biological influence on continuum water pCO₂ dynamics through
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8 433 autotrophic and heterotrophic processes was significant at each station as endorsed by strong
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10 434 linear relationships between pCO₂ and DO, especially in autumn (Table 4). At the tidal estuary
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12 435 (*a*), from spring to autumn, diurnal pCO₂ variations were mostly controlled by the
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14 436 photosynthesis versus respiration balance of planktonic communities; indeed, more than 80%
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16 437 of the pCO₂ variance was modelled with DO only (Table 4). Dai et al. (2009) highlighted that
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19 438 CO₂ biogeochemical processes in coastal environments such as our estuary are generally
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21 439 controlled by non-thermal effects, like biological activity, compared to more open systems.
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23 440 Several studies have shown a major biological control on diurnal pCO₂ variations in coastal
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25 441 systems such as the temperate Bay of Brest (France; Bozec et al., 2011), the temperate Arcachon
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27 442 lagoon (France; Polsenaere et al., 2022), the shallow subtropical estuary in Tampa Bay (USA;
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29 443 Yates et al., 2007) and the tropical coastal embayment at Guanabara Bay (Brazil; Cotovicz Jr.
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31 444 et al., 2015). At the rewilded marsh (*c*), while in spring and summer, the high primary
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33 445 production of macroalgae induced large periods of water CO₂ undersaturation with respect to
34
35 446 the atmosphere, winter pCO₂ variations were rather induced by planktonic community activity
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37 447 (ciliates > 2 10⁴ cell L⁻¹; unpubl. result). At the working marsh (*d*), the negative correlations
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39 448 between pCO₂ and Chl *a* associated with strong non-thermal contributions ($\Delta\text{NpCO}_2 = 318$
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41 449 ppmv in summer for instance) showed a major biological influence on diurnal pCO₂ variations
42
43 450 as well. By comparison, in a *Zostera marina* meadow (South Bay, USA), Berg et al. (2019)
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45 451 measured similar diurnal fluctuations of water pCO₂ that were directly controlled by seagrass
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47 452 metabolism with diurnal ranges of 528 and 603 ppmv in spring and summer, respectively.
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453 Tidal advection between continental shelf and salt marshes also significantly controlled
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2 454 water pCO₂ dynamics and associated station biogeochemical status. This control was supported
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4 455 by the linear relationships between water pCO₂ and salinity as observed at each station from
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7 456 winter to summer whereas, relationships with turbidity were rather related to the hydrodynamic
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10 457 forcing on water pCO₂ (Table 4). At the tidal estuary (*a*) and its associated channel (*b*), daytime
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12 458 incoming tides created a significant decrease in water pCO₂ since the advected shelf waters
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14 459 were CO₂ undersaturated with respect to the atmosphere contrary to estuarine waters. In the
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17 460 Arcachon lagoon, seasonal measurement cycles also showed a strong tidal control on inorganic
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19 461 C parameters with lower pCO₂ values measured at high tide than at low tide irrespective of day
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22 462 or night status (Polsenaere et al., 2022). Even stronger tidal influences on water pCO₂ (from
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24 463 1380 to 4770 ppmv) were observed during a summer cycle in the Gaderu Creek mangrove
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26 464 (Borges, 2003). In summer, the large pCO₂ decrease in the studied estuarine waters at ebbing
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29 465 tide the day, associated with strong DO saturation level increase was probably due to CO₂
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31 466 undersaturated water exports from the productive salt marshes upstream. Indeed, in spring and
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34 467 summer, the rewilded marsh waters (*c*) were CO₂ depleted due to strong autotrophy activity.
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36 468 Conversely, more CO₂-enriched coastal water inflows from the shelf and the estuary into the
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39 469 marsh instantly produced significant salinity decreases and pCO₂ increases (Table 4).
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41 470 Therefore, at each semi-diurnal tidal cycle, horizontal advection had significant effects on water
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44 471 pCO₂ dynamics (except during marsh confinement) but variations strongly depended on the
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46 472 biogeochemical state of advected waters from upstream/downstream and the ecosystem
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48 473 typology (estuary, marsh, channel).
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476 4.2. Continuum typologies revealed from measured biogeochemical parameters

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3 477 In the coastal ocean, a strong influence of ecosystem typology (continental shelf, estuary,
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5 478 marsh) on biogeochemistry is generally observed and particularly, on inorganic C (Bauer et al.,
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8 479 2013). In our study, in 2018, watershed-influenced shelf waters (*F*) were characterized by lower
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10 480 salinity values than the Atlantic Ocean (35.6; Vandermeirsch 2012), confirmed over the 2000-
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12 481 2017 reference period (Belin et al., 2021). However, shelf waters showed a rather weak
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15 482 influence of terrestrial inputs on water pCO₂ dynamics (annual non-significant salinity and
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17 483 pCO₂ relationship, $p = 0.88$). At this shelf station, phytoplankton bloom with centric diatoms
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20 484 generally occurs in spring and summer (Guarini et al., 2004) and can induce water CO₂
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22 485 undersaturation as also observed on the Belgian continental shelf (Borges and Frankignoulle,
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24
25 486 1999). Along the continuum, the tidal estuary (*a*) influenced by buffered shelf waters was CO₂
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27 487 undersaturated at 4% in winter and 82% in spring/summer that could be attributed to coastal
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30 488 water autotrophic activity at this period. Previous study carried out in the same estuarine waters
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32 489 measured Chl *a* concentrations from 0.2 (winter) to 3.5 µg L⁻¹ (spring/summer) and a Chl *a*
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34 490 export suggesting a net primary production within this tidal estuary (Bel Hassen, 2001).
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37 491 Additionally, Savelli et al. (2019) observed in a nearby intertidal zone that microphytobenthos
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39 492 (MPB) may also contribute to estuarine water CO₂ undersaturation and to the overall water
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42 493 column Chl *a* concentration through tidal resuspension. Due to the small insular catchement
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44 494 area (1200 ha) consisting only of salt marshes (no terrestrial water input), the CO₂ dynamics in
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47 495 the tidal estuary (*a*) is different from other estuaries worldwide (Borges and Abril, 2011).
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49 496 Similarly, the marine-dominated estuary of Sapelo Sound (USA) was also characterized by
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51 497 lower water pCO₂ values than river-dominated ones (Borges and Abril, 2011) but bacterial
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54 498 remineralization of organic carbon produced by *Spartina* in nearby salt marshes strongly
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56 499 increased water pCO₂ in summer (Jiang et al., 2008) contrarily to our studied estuary.

500 Overall, channel (*b*) waters between the estuary and salt marshes showed the longest periods
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2 501 of CO₂ oversaturation with respect to the atmosphere. At this channel, strong hydrodynamic
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4 502 forcings during incoming and ebbing tides produced more turbid waters due to organic matter
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7 503 resuspension from muds (Guarini et al., 2008). It probably limited the primary production
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10 504 (phytoplankton, MPB) by low light availability in water column (Cloern, 1987) and, on the
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12 505 contrary, favoured heterotrophic processes (Polsenaere et al., 2022). In channel waters, we
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14 506 recorded lower DO concentrations (-10%) and higher pCO₂ values (+10%) than in estuarine
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17 507 waters under similar salinity ranges. In channel waters at the same location, Tortajada (2011)
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19 508 measured POC/Chl *a* > 200 mg mg⁻¹ throughout the year and even POC/Chl *a* > 600 mg mg⁻¹
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22 509 in autumn. This detrital/heterotrophic material may indicate microbial mineralization processes
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24 510 from MPB and confirm the water CO₂ oversaturation periods. However, channel (*b*) waters
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27 511 showed lower pCO₂ values than those from other coastal channel systems probably due to low
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29 512 terrestrial water inputs upstream/downstream over the estuary. The Sancti Petri Channel waters
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31 513 and its nearby salt marshes between the Atlantic Ocean and the Cadiz Bay (Spain) were also
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34 514 mainly CO₂ oversaturated (281 - 862 ppmv), due to DIC inputs from diagenetic processes of
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36 515 organic matter in mudflats that constitute a CO₂ source to water column (Burgos et al., 2018).
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39 516 Indeed, within the Duplin River salt marsh-estuary coastal system (USA), higher summer pCO₂
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41 517 and DIC values were recorded at low tide in channel waters (12000 ppmv and 4300 μmol L⁻¹,
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43 518 respectively) than at high tide in marsh waters (1600 ppmv and 2200 μmol L⁻¹, respectively;
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46 519 Wang et al., 2018).
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49 520 Contrary to estuarine and channel waters, artificial salt marshes (*c* and *d*) waters were
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51 521 characterized by the lowest turbidity and highest salinity values due to longer water residence
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54 522 times. These lower hydrodynamic conditions promoted the development of primary producers
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57 523 and as a result, biological CO₂ uptake associated with the highest DO saturation levels. In turn,
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524 the salt marshes showed lower water pCO₂ values and longer water CO₂ undersaturation periods
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2 525 mainly due to a strong macrophyte activity (macroalgae at rewilded marsh and seagrasses at
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4 526 working marsh) than the tidal estuary and elsewhere similar wetland typologies (Borges, 2003;
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7 527 Burgos et al., 2018; Wang et al., 2018; Berg et al., 2019). Unlike seagrasses which are known
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10 528 to be important blue C systems (Mcleod et al., 2011), macroalgae developing in coastal
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12 529 wetlands have a limited capacity to store C over the long-term. However, studies have shown
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14 530 their potential contribution to coastal blue C by (i) storing large organic matter quantities in
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16 531 their living biomass through their high primary production (Raven, 2018) and (ii) transferring
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19 532 it to adjacent systems through tides and storage in coastal sediments (Duarte and Cebrián, 1996;
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22 533 Hill et al., 2015; Krause-Jensen and Duarte, 2016). Our pCO₂ observations are in accordance
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24 534 with these reports with the role of C storage by macrophytes in these shallow salt marshes.
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30 536 **4.3. Temporal carbon modulation by management practices**

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33 537 Management practices at the artificial salt marshes correspond to specific water lock actions
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36 538 linked to anthropogenic activities. They can strongly modulate water fluxes from the estuary
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38 539 and thereby influence marsh pCO₂ dynamics. At the rewilded marsh (c), the specific
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41 540 management practice by the NNR produced favourable conditions for free floating macroalgae
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43 541 development from early spring to late summer under low water marsh hydrodynamic and high
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45 542 air and water temperature conditions (Newton and Thornber, 2013). These macroalgae indicate
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48 543 a degraded-eutrophic status of the marsh with excess nutrient inputs, as described for other
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51 544 coastal ecosystems (Teichberg et al., 2010; Le Fur et al., 2018). In our study, nearby marsh
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53 545 aquafarming activities occurring upstream from the estuary can communicate with the rewilded
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55 546 marsh (c) through channels and result in high nutrient conditions which could explain observed
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547 macroalgae blooms (Tortajada, 2011). Indeed, at the station *b* channel in September 2018, high
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2 548 DIN concentrations were reported ($60.0 \mu\text{mol L}^{-1}$; unpublished results). Moreover, shelf waters
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4 549 influenced by terrestrial inputs in winter could also lead to nutrient inputs at the rewilded marsh.
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7 550 Indeed, at the station *F* shelf in winter 2019, NO_3^- ranged between 29 and $107 \mu\text{mol L}^{-1}$ (Belin
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9 551 et al., 2021). Consequently, these fast-growing macroalgae probably prevented the growth of
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11 552 phytoplankton and seagrasses by nutrient and oxygen competition and light limitation in the
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13 553 water column (Sand-Jensen and Borum, 1991; Le Fur et al., 2018). Simultaneously, in spring
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15 554 and summer, the large water CO_2 undersaturation periods due to the macroalgae autotrophy
16
17 555 were maintained through occasional inflows of CO_2 oversaturated channel waters under weak
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19 556 tidal amplitudes. This result is confirmed by significantly higher salinity values in the rewilded
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21 557 marsh waters than in the channel waters during the spring and summer sampling periods (Table
22
23 558 2). On the contrary, macroalgae degradation in autumn probably by microbial remineralization
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25 559 processes (Hill et al., 2015) produced in turn the highest pCO_2 values and the longest
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27 560 oversaturation periods recorded in marsh waters. These heterotrophic processes were confirmed
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29 561 by high NH_4^+ levels ($62 \mu\text{mol L}^{-1}$; unpublished results) and low DO saturation levels recorded
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31 562 at this period and as described by Newton and Thornber (2013).
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39 563 Contrarily to the rewilded marsh (*c*), the working marsh (*d*) is managed for salt production
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41 564 in the upstream ponds along the continuum and is directly connected to the estuary (*a*) with no
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43 565 channel in between (Fig. 10). Salt production requires a subtle lock hydraulic management of
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45 566 the marsh depending on the frequency of the coastal water supplies that are mainly controlled
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47 567 by the salt manufacturer and meteorological conditions (rainfall, sunshine and wind) to favour
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49 568 the evaporation process (Paticat, 2007). Therefore, contrary to rewilded marsh, coastal water
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51 569 inflows to the working marsh were performed sparingly with small daily volumes to limit these
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53 570 water mixing effects (i.e. rapid accumulation of large water volumes through rainfall events or
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1 571 spring tides stop the increase in temperature and salinity in marsh waters; Paticat, 2007). At the
2 572 working marsh in summer, water pCO₂ values were significantly higher than those measured
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4 573 at the same period at the rewilded marsh but reflected those from the estuary (Table 2). This
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7 574 could also be linked to a lower activity of the primary producers during the summer period
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9 575 dedicated to salt production as confirmed by higher thermal than non-thermal effects on water
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12 576 pCO₂ (Fig. 7). On the other hand, in autumn and winter, lower hydrodynamic conditions due to
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14 577 lock closure (standstill salt farming activity) led to lower water turbidity (< 2 NTU) and nutrient
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16 578 input into the marsh (DIN < 2 μmol L⁻¹; unpublished results) and the growth of seagrasses and
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19 579 phytoplankton produced, in turn, the lowest water pCO₂ values. Overall, in Mediterranean poly-
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21 580 euhaline lagoons, Le Fur et al. (2018) confirmed that nutrient pollution influence the
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24 581 contribution of primary producers from perennial seagrasses in oligotrophic waters to fast-
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26 582 growing macroalgae in eutrophic waters. Similarly, other studies have suggested that the coastal
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29 583 ecosystem management by reducing anthropogenic nutrients could favour blue C ecosystems
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31 584 such as seagrasses and salt marshes (Macreadie et al., 2017, Palacios et al., 2021).

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36 37 38 586 **4.4. Metabolism assessment of the sea-land continuums**

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41 587 The tidal estuary (*a*) behaved on average as a yearly CO₂ sink close to the atmospheric
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43 588 equilibrium (Fig. 10), although a significant sink was measured in the spring/summer due to
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45 589 autotrophic activity of phytoplankton in coastal waters. Conversely, over the same
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48 590 meteorological periods, the channel (*b*) was a net annual source from its turbid waters to the
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51 591 atmosphere due to several water CO₂ oversaturation periods (Fig. 10), particularly in winter,
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53 592 characterized by high gas transfer velocities (Table 3). In the Arcachon lagoon, estimated
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55 593 atmospheric CO₂ sources were higher with seasonal means ranging from 0.06 ± 0.04 (winter)

594 to $0.62 \pm 0.66 \text{ mmol m}^{-2} \text{ h}^{-1}$ (summer) with significant diurnal fluctuations (Polsenaere et al.,
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2 595 2022). In this study, eutrophic waters of the rewilded marsh (*c*) behaved as a yearly source of
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5 596 atmospheric CO_2 (Fig. 10), when macroalgae degradation produced strong atmospheric CO_2
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7 597 effluxes. On the contrary, oligotrophic waters of the working marsh (*d*) behaved as a large
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10 598 yearly CO_2 sink (Fig. 10), favoured by low tidal advection in the absence of salt-farming
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12 599 activities. Within the Duplin River salt marsh-estuary system, both channel and marsh waters
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14 600 degassed CO_2 to the atmosphere and, unlike our stations, the highest and lowest sources were
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16 601 recorded in summer (5.50 and $3.90 \text{ mmol m}^{-2} \text{ h}^{-1}$ from channel and marsh waters, respectively)
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18 602 and in winter (0.70 and $0.60 \text{ mmol m}^{-2} \text{ h}^{-1}$ from channel and marsh waters, respectively),
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20 603 respectively (Wang et al., 2018). Overall, the Duplin system emits more atmospheric CO_2 than
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22 604 the Fier d’Ars system, probably due to its more intense estuarine heterotrophic metabolism.
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27 605 In autumn, the lack of variations in wind speeds between stations *a* and *b* in 2018 and
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29 606 between stations *c* and *d* in 2019, whereas atmospheric CO_2 exchanges significantly changed,
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31 607 highlighted the predominance of air-water CO_2 gradients in the control of flux directions either
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33 608 as a sink or a source (Table 3). However, at the seasonal scale, turbulence processes measured
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35 609 at the air-water interface played an important role in CO_2 flux variability and magnitude. For
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37 610 instance, at station *a* between spring and summer and at station *b* between winter and summer,
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39 611 wind speed variability produced significant FCO_2 variations although no significant air-water
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41 612 CO_2 gradients were measured (Table 3). Atmospheric exchanges in salt marshes are therefore
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43 613 dependent on the CO_2 saturation state of the water column considering that the wind only acts
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45 614 as a driver of the flux (Polsenaere et al., 2022). Moreover, the methodological calculations and
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47 615 associated differences chosen for the exchange coefficient parameterizations (higher fluxes
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49 616 with RC01 than with W22; Table 3) may produce even more contrasts in the estimated air-
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51 617 water FCO_2 (Cotovicz Jr. et al., 2015; Polsenaere et al., 2022).
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618 By scaling-up and considering stations *a* and *b* together along the continuum, estuarine and
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2 619 channel waters behaved as an annual atmospheric CO₂ source of 7.3 g C m⁻² yr⁻¹. Whereas, the
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4 620 rewilded marsh emitted 17.5 g C m⁻² yr⁻¹ to the atmosphere, the working marsh absorbed 97.7
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6 621 g C m⁻² yr⁻¹ from the atmosphere. A larger scale study along three shelf – estuary – tidal wetland
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8 622 continuums on the Atlantic coast of the United States also showed strong spatial variations in
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10 623 atmospheric CO₂ fluxes with uptake to wetland and shelf waters (523.2 ± 148.1 and 10.5 ± 1.8
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12 624 g C m⁻² yr⁻¹, respectively) and a source from estuarine waters (110.0 ± 44.5 g C m⁻² yr⁻¹; Najjar
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14 625 et al., 2018). During our study, contrasting coastal stations were sampled via seasonal 24-h
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16 626 cycles to estimate the air-water CO₂ exchanges. However, longer seasonal measurement periods
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18 627 would be more representative of the strong temporal variability in *k*₆₆₀, water pCO₂ and other
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20 628 biogeochemical parameters. At the nearby tidal salt marsh (*e*), emerged for 75% of time during
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22 629 low tides and neap tides, another flux methodology using the atmospheric Eddy Covariance
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24 630 technique was deployed to continuously measure in situ CO₂ fluxes at the ecosystem scale
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26 631 coming from all habitats (aquatic and terrestrial vegetations, mudflats, channels). Over the year
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28 632 2020, a net uptake of 483 g C m⁻² yr⁻¹ from the atmosphere was measured, indicating a stronger
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30 633 CO₂ sink in tidal marshes than artificial marshes due to higher halophytic plant photosynthesis
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32 634 activity. However, it is also important to study the whole marsh metabolism taking terrestrial
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34 635 and aquatic compartments into account and distinguishing their respective contributions to
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36 636 atmospheric fluxes and the regional C budgets of the associated marshes (Mayen et al., in prep.).
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50 638 5. Conclusion

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53 639 Along the continuums, estuarine and channel waters were slightly oversaturated in CO₂
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55 640 characterized by seasonal compensations of thermal and non-thermal effects whereas, upstream
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641 marsh waters were mostly undersaturated in CO₂ due to stronger biological activity and longer
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2 642 water residence times. At the diurnal/tidal scale, our high-resolution analyses highlighted large
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5 643 water pCO₂ variations in salt marshes, controlled by production and respiration of macrophytes
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7 644 and coastal water inflows. However, anthropogenic management in salt marshes could strongly
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10 645 influence the contribution and turnover of macrophytes and, consequently, the marsh CO₂
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12 646 sink/source behaviour. Due to eutrophication in the rewilded marsh, development of the fast-
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14 647 growing macroalgae produced an overall net annual atmospheric CO₂ source through their
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17 648 degradation. Our results suggest a winter marsh confinement follow by drying up to limit
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19 649 nutrient inputs and macroalgae development and on the contrary, favour rather slow-growing
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22 650 macrophytes (i.e. seagrasses) which could ultimately contribute to blue C sequestration.
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25 651

28 652 **Acknowledgements**

31 653 We would like to sincerely thank the oyster farmers for their help with taking samples at station
32
33
34 654 *a*, Julien Gernigon from the Lilleau des Niges NNR (LPO) and Brice Collonier from the Loix
35
36 655 Ecomuseum for their help and the information given at stations *b*, *c*, *d* and *e*. We are grateful to
37
38
39 656 our colleagues Jean-Michel Chabirand, James Grizon and Philippe Geairon for their help with
40
41 657 field sensor deployments and QGIS work. We also thank Quentin Ternon, Gabriel Devique and
42
43 658 Jonathan Deborde for their contribution in the field. This paper is a contribution to the Master
44
45
46 659 and Ph.D. thesis of Jérémy Mayen (Ifremer funding), the ANR-PAMPAS project (Agence
47
48 660 Nationale de la Recherche « Evolution de l'identité patrimoniale des marais des Pertuis
49
50
51 661 Charentais en réponse à l'aléa de submersion marine », ANR-18-CE32-0006) and the CNRS-
52
53 662 INSU LEFE-DYCIDEMAIM project (DYnamique du Carbone aux Interfaces D'Échange des
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Fig. 1. The Fier d' Ars estuary (Ré Island, French Atlantic coast) and locations of the four studied stations along aquatic continuums: tidal estuary at station *a*, channel at station *b*, rewilded artificial salt marsh at station *c* (in green) and working artificial salt marsh at station *d* (in blue). The dyke (in red) delimits terrestrial and maritime areas. The locks in the two studied artificial marshes are represented within the two map expansions. An atmospheric Eddy Covariance station was deployed at station *e* on the tidal salt marsh downstream from the dyke. Station *F* is located in the centre of the Breton Sound continental shelf; station *a* is located at the entry of the estuary; stations *b*, *c* and *e* are within the National Natural Reserve to the west of the estuary; station *d* to the east of the estuary is within a salt-farm.

Fig. 2. Principal Component Analysis (PCA) of the biogeochemical parameters measured at each season along the studied aquatic continuums (stations *a*, *b*, *c* and *d*). The PCA is based on temperature (Temp), salinity, turbidity, dissolved oxygen concentration (DO), dissolved oxygen saturation level (DO-sat.) and pCO₂ mean values for each 24-h cycle. Stations *a*, *b*, *c* and *d* are represented in red, brown, green and blue, respectively. The additional station *F* is represented in light blue. Win: Winter; Spr: Spring; Sum: Summer; Aut: Autumn.

Fig. 3. Temporal variations at station *a* (tidal estuary) of water temperature (°C), salinity, DO saturation level (DO-sat., %), turbidity (NTU), pCO₂, NpCO₂ (pCO₂ variations related to non-temperature effects, ppmv) and TpCO₂ (pCO₂ variations related to temperature physical effects, ppmv) during each 24-h cycle from winter 2018 to autumn 2018. Parameters were autonomously measured once per minute by in situ probes. Water heights (H, m) were retrieved from the SHOM station (9 km away; Fig. 1). Grey areas correspond to night-time periods. Vertical dotted lines correspond to high tides. Horizontal dotted lines correspond to the CO₂ atmospheric concentration (411 ppm; NOAA 2018). Each graduation of the x-axis corresponds to one hour.

Fig. 4. Temporal variations at station *b* (channel) of water temperature (°C), salinity, DO saturation level (DO-sat., %), turbidity (NTU), pCO₂, NpCO₂ and TpCO₂ (ppmv) during each 24-h cycle from winter 2018 to autumn 2018. See the Fig. 3 caption for more details.

Fig. 5. Temporal variations at station *c* (rewilded artificial salt marsh) of water temperature (°C), salinity, DO saturation level (DO-sat., %), turbidity (NTU), pCO₂, NpCO₂ and TpCO₂ (ppmv) and *in situ* Chl *a* (µg L⁻¹) during each 24-h cycle from spring 2019 to winter 2020. Vertical dotted lines correspond to coastal water inflows to the marsh during incoming tide. Horizontal dotted lines correspond to the atmospheric CO₂ concentration simultaneously measured (i) by the Eddy Covariance (station *e*) during the summer and winter cycles and (ii) by NOAA during the spring and autumn cycles. *In situ* Chl *a* values are represented by black crosses (Chl *a*_{measured}); no water samples could be taken in spring 2019. See the Fig. 3 caption for more details.

Fig. 6. Temporal variations at station *d* (working artificial salt marsh) of water temperature (°C), salinity, DO saturation level (DO-sat., %), turbidity (NTU), pCO₂, NpCO₂ and TpCO₂ (ppmv) and *in situ* Chl *a* (µg L⁻¹) during each 24-h cycle from summer 2019 to winter 2020. Chl *a* values in green were derived from the C3-fluorometer every 10 min. (Chl *a*_{estimated}). See the Fig. 3 caption for more details.

Fig. 7. Derived temperature-normalized pCO₂ (seasonal NpCO₂, blue curves with empty blue dots) and thermally forced pCO₂ (seasonal TpCO₂, pink curves with empty pink dots) at the seasonal scale at stations *F*, *a*, *b*, *c* and *d*. Seasonal means of water temperature (in red dotted lines) and pCO₂ (red curves with filled red dots) are also represented. Horizontal dotted lines correspond to CO₂ atmospheric concentration (411 ppm; NOAA 2018).

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Fig. 8. Diurnal/tidal correlation plots of temperature vs. salinity and water pCO₂ vs. salinity at stations *b* and *c* for each season. Only significant R² (slopes significantly different from zero; n = 1441; p < 0.05) are showed. HT/D: high tide day; LT/D: low tide day; HT/N: high tide night; LT/N: low tide night. At station *c*, HT periods correspond to coastal water inflows to the marsh. Note that the temperature and salinity ranges across the seasons are not the same. Horizontal dotted lines correspond to the atmospheric CO₂ concentration.

Fig. 9. Seasonal and spatial variations in estimated CO₂ fluxes (FCO₂, in mmol m⁻² h⁻¹) at the water-atmosphere interface at stations *F*, *a*, *b*, *c* and *d*. The means and associated standard deviations over each 24-h cycle are shown. *k*₆₆₀ and FCO₂ estimations were calculated according to the R22 parametrization. FCO₂ values at stations *a* and *b* are only given for high tide periods.

Fig. 10. CO₂ budget over the two aquatic sea - land continuums: (1) continental shelf - estuary - channel - rewilded artificial salt marsh and (2) continental shelf - estuary - working artificial salt marsh. Annual means (± SD) and ranges (min - max) of water pCO₂ (ppmv) and air-water FCO₂ (mmol m⁻² h⁻¹) are showed. The picture of station *c* in spring 2019 (© P. Polsenaere) allows to visualize the macroalgae bloom.

Fig. 1.

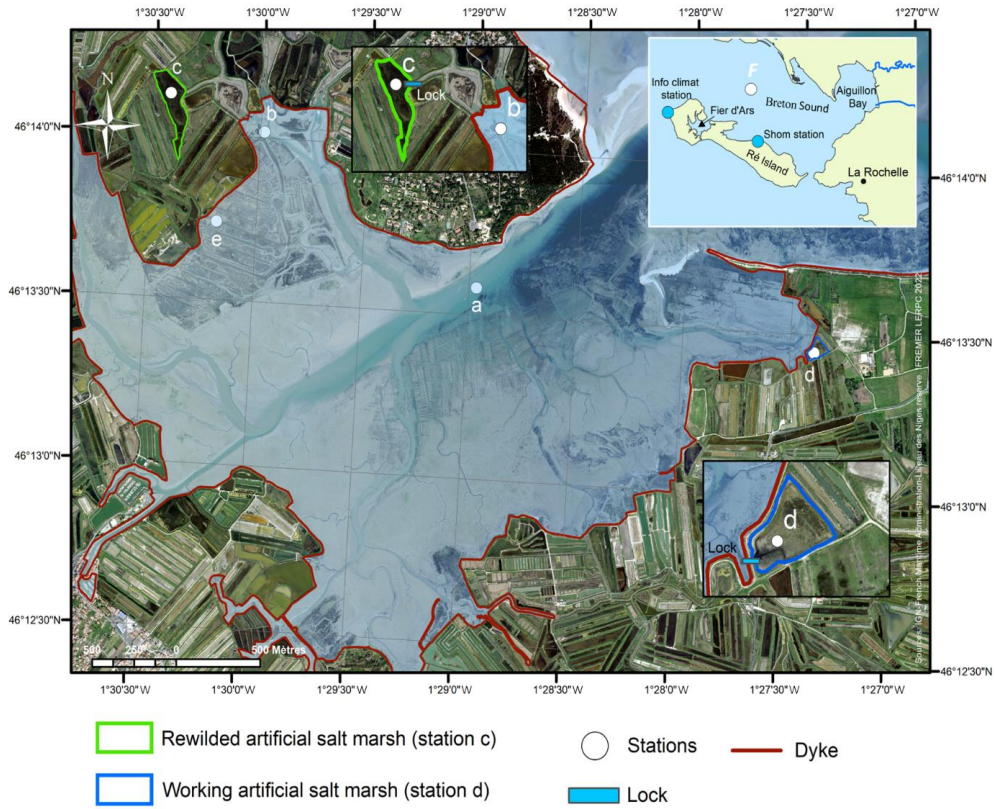
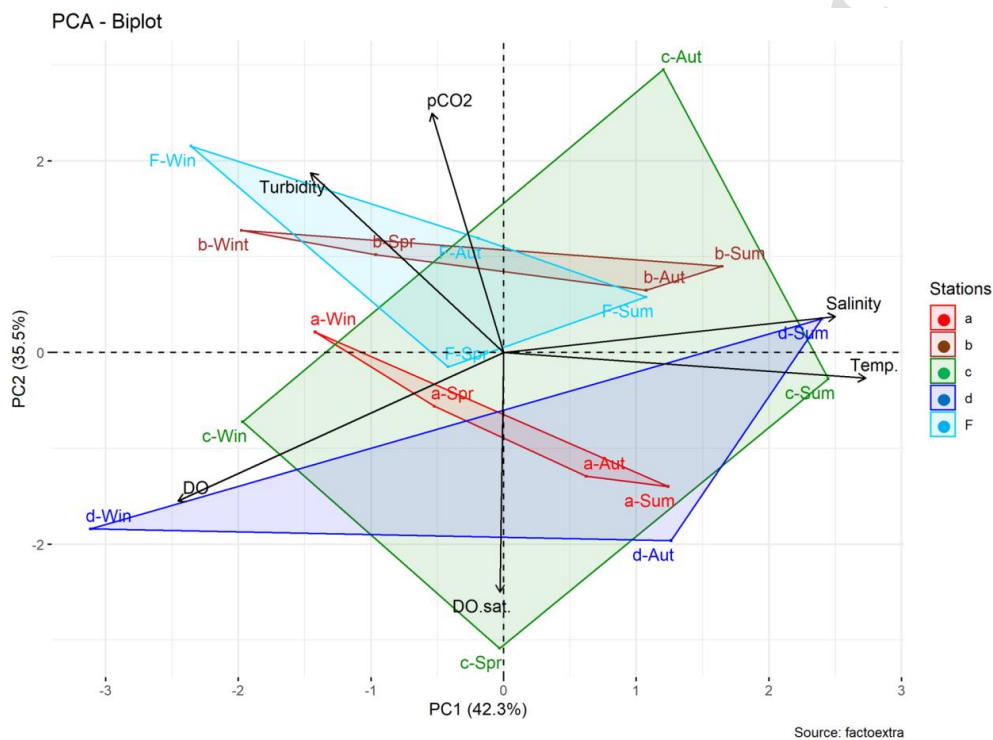


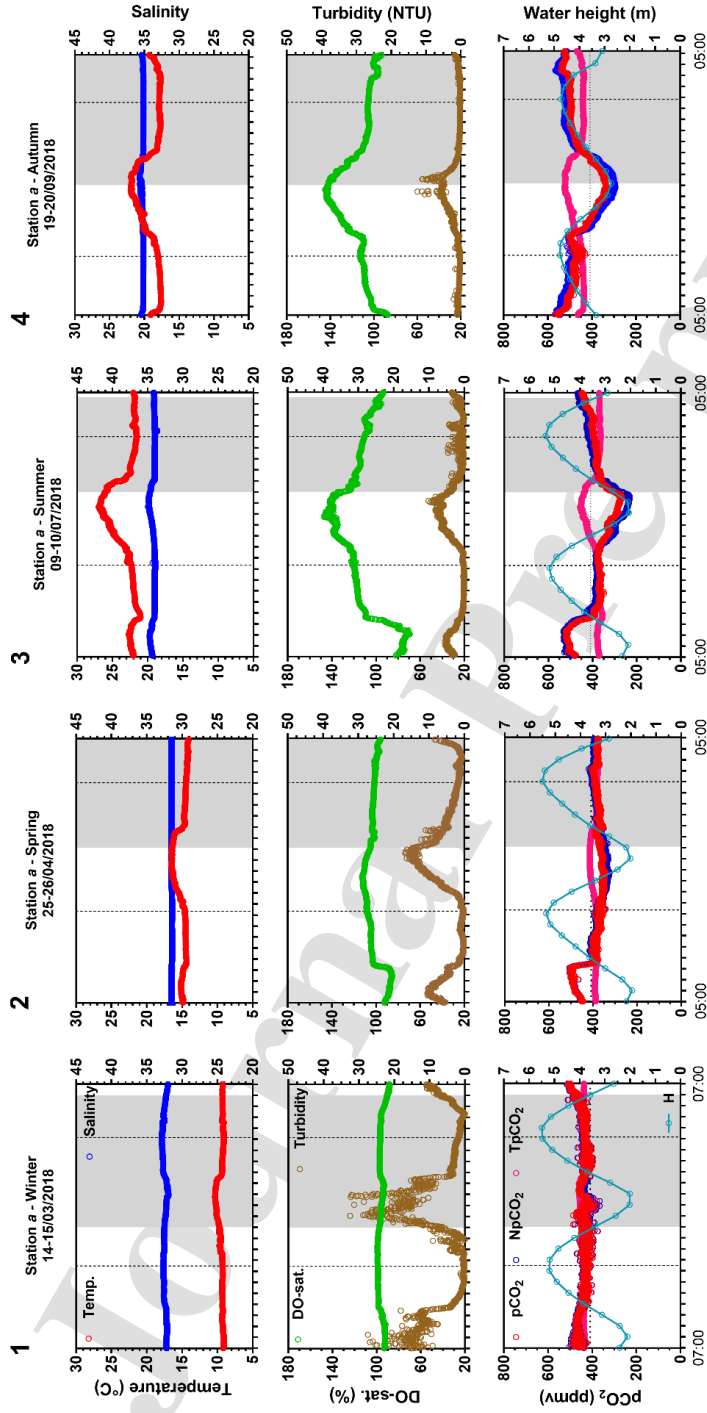
Fig. 2.



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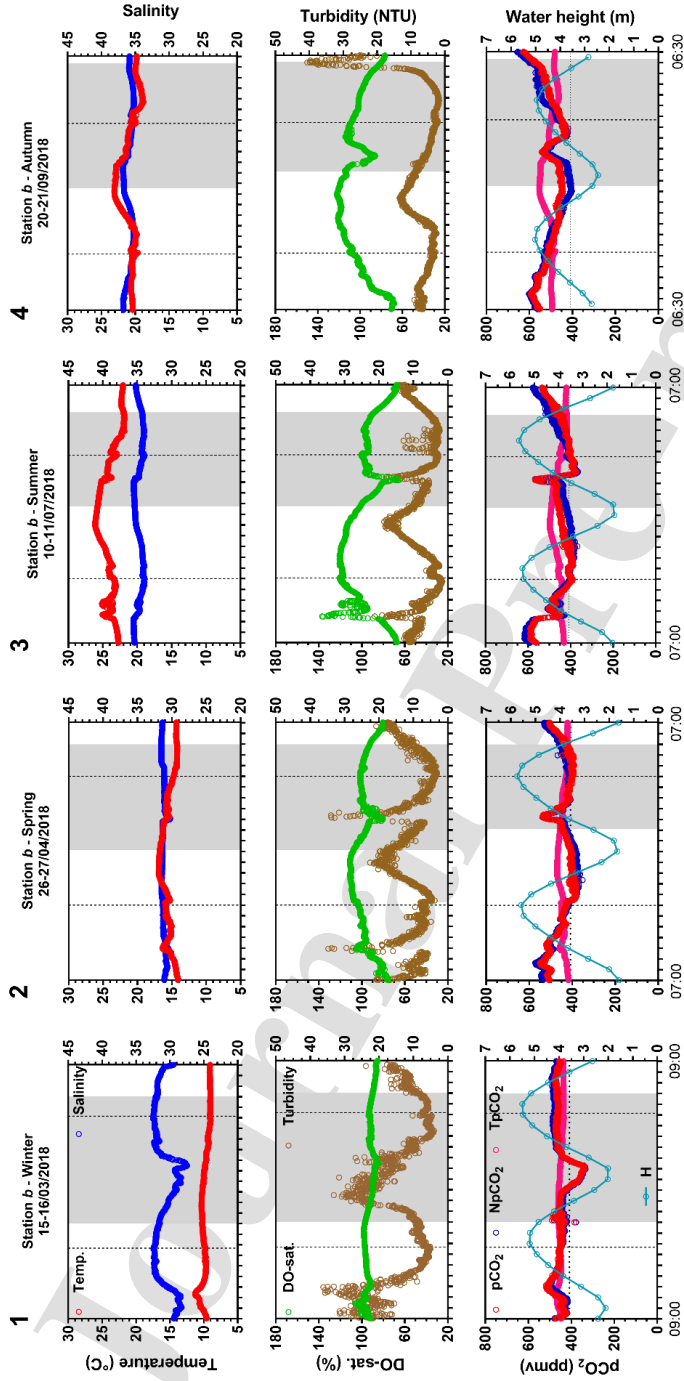
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Fig. 3.



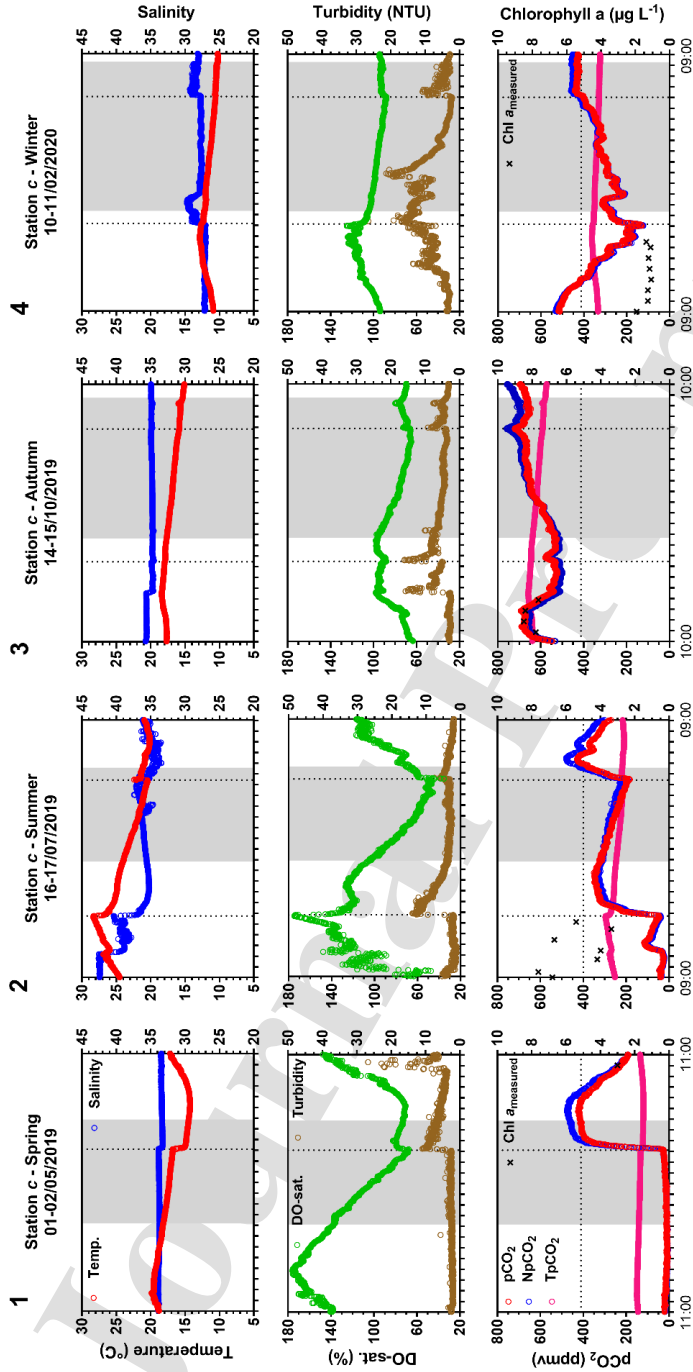
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Fig. 4.



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Fig. 5.



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Fig. 6

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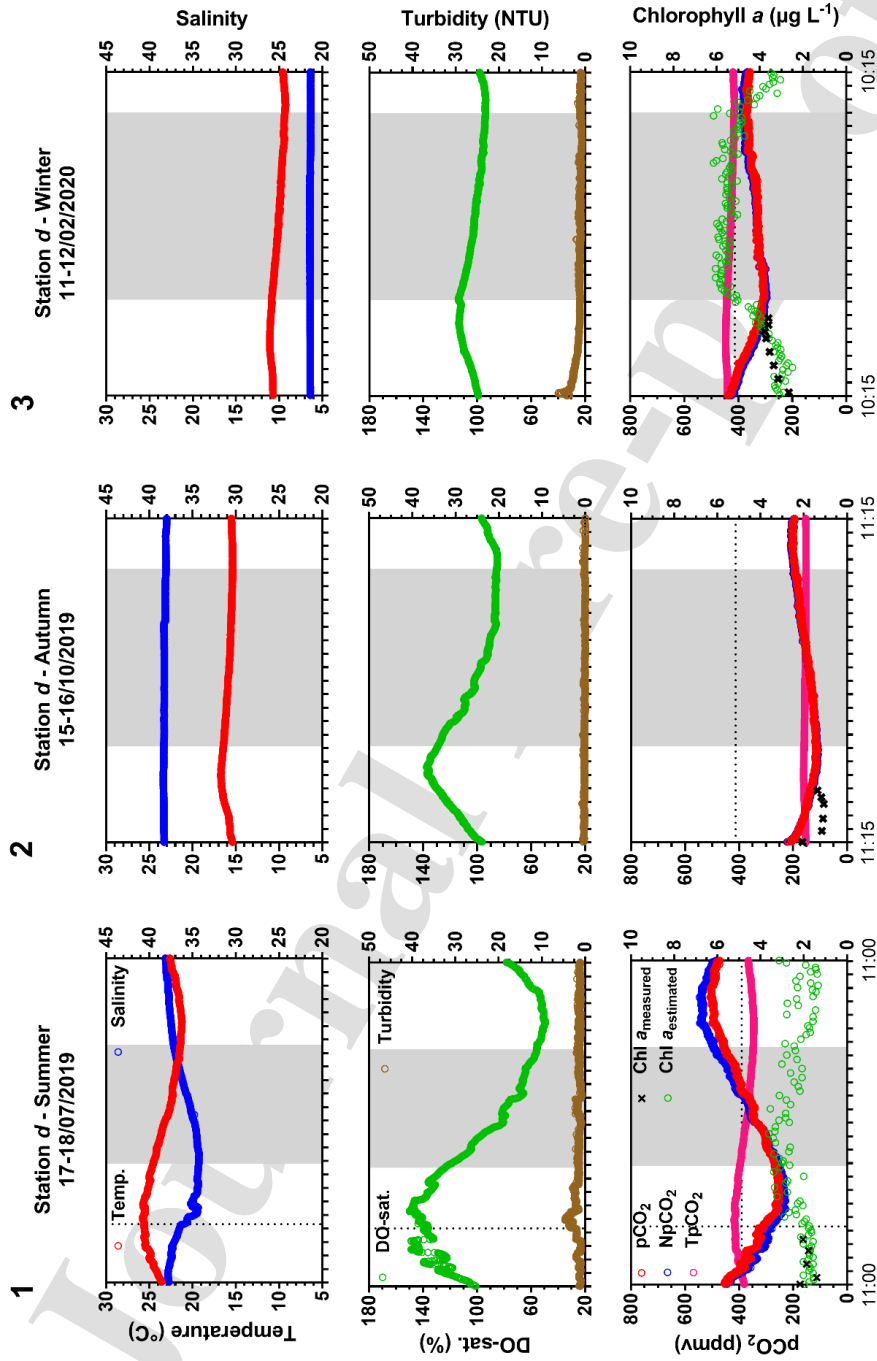


Fig. 7.

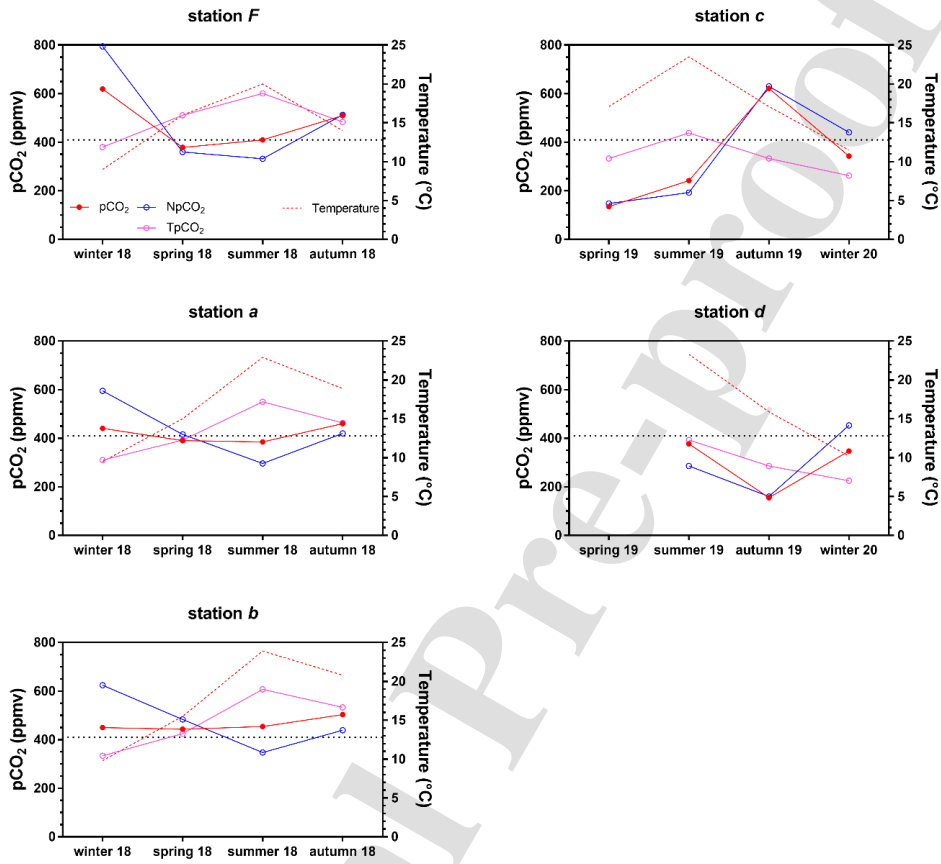


Fig. 8.

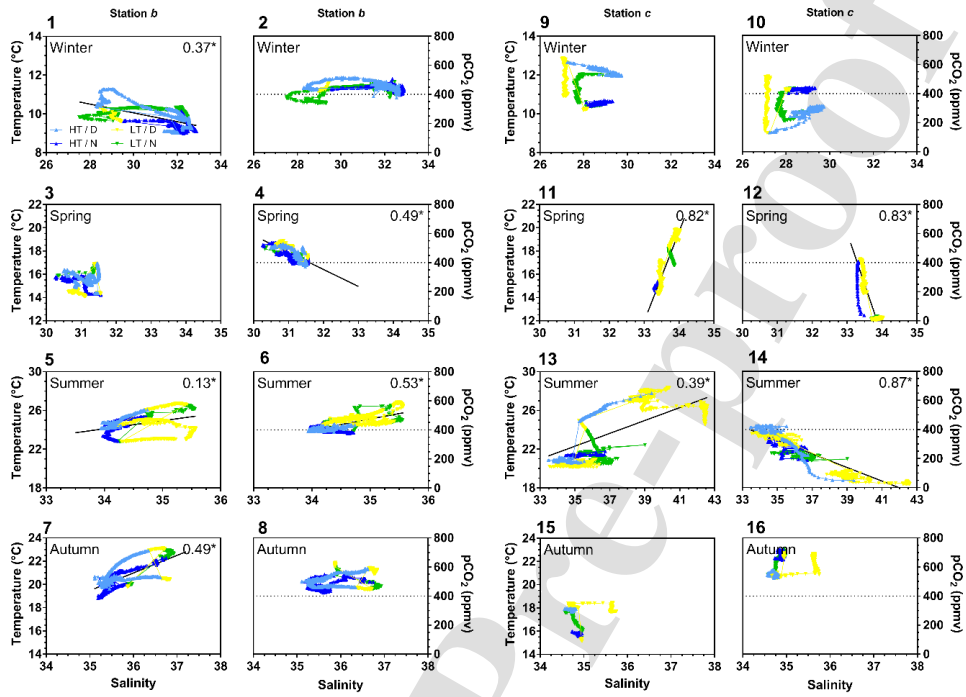
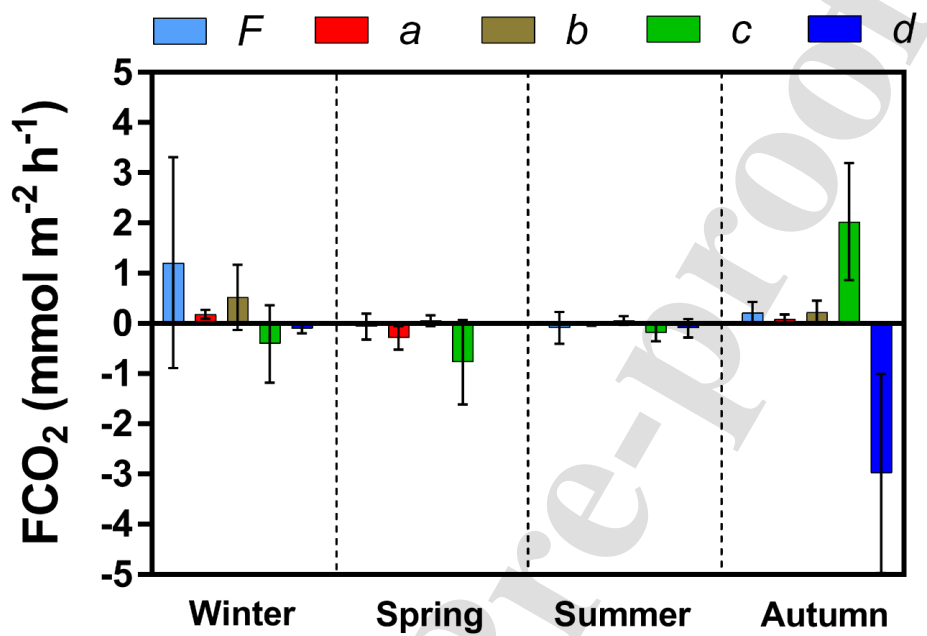
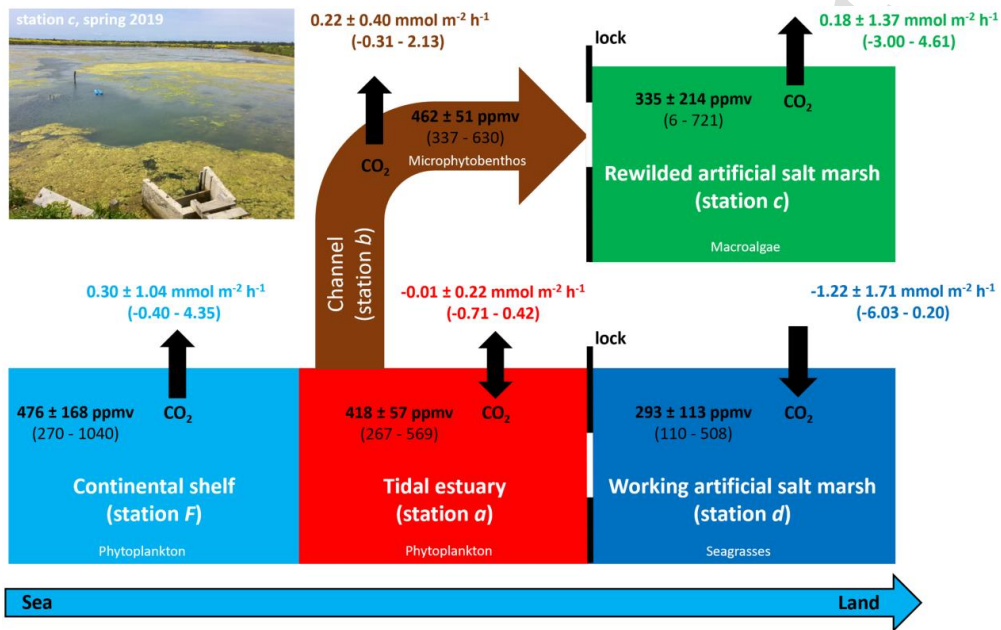


Fig. 9.



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Fig. 10.



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Table 1: Meteorological conditions (air temperature in °C and cumulative precipitation in mm) obtained from the “Infoclimat” station on Ré Island (Fig. 1; <https://www.infoclimat.fr>) at the monthly and annual scales over our measurement periods in 2018, 2019 and 2020 in bold compared to the reference period (1990-2020).

Year	Season	Month	Mean air temperature (°C)	Difference with the reference period (°C)	Cumulative precipitation (mm)	Difference with the reference period (mm)
2018	Winter	March	9.1	-0.7	127	+70
	Spring	April	13.4	+1.4	58	-3
	Summer	July	22.2	+1.7	59	+19
	Autumn	September	19.0	+1.0	9	-51
		Annual	14.3	+0.8	786	+32
2019	Spring	April	11.9	-0.1	57	-4
	Summer	July	22.5	+2.0	33	-7
	Autumn	October	15.8	+1.1	117	+33
		Annual	14.1	+0.6	827	+73
2020	Winter	February	10.4	+3.1	68	+12

Table 2: Seasonal means (\pm SD) and ranges (min - max) of temperature ($^{\circ}$ C), salinity, DO ($\mu\text{mol L}^{-1}$), pH (NBS scale) and pCO_2 (ppmv) values measured (i) once every two weeks in 2018 at station *F* (Coignot et al., 2020) and (ii) during each 24-h cycle from 2018 to 2020 at stations *a*, *b*, *c* and *d* in this study.

		Temperature ($^{\circ}$ C)	Salinity	DO ($\mu\text{mol L}^{-1}$)	pH (NBS)	pCO_2 (ppmv)
Winter 2018	<i>F</i>	9.1 \pm 1.3 (7.5 - 10.6)	31.4 \pm 1.8 (28.9 - 33.0)	269.6 \pm 29.7 (225.0 - 285.3)	8.04 \pm 0.17 (7.79 - 8.17)	619 \pm 285 (415 - 1040)
March 2018	<i>a</i>	9.5 \pm 0.4 (9.1 - 10.4)	32.5 \pm 0.2 (31.9 - 32.9)	278.7 \pm 7.7 (257.2 - 288.8)	-	441 \pm 21 (377 - 510)
March 2018	<i>b</i>	9.8 \pm 0.5 (9.0 - 11.3)	31.0 \pm 1.4 (27.5 - 32.5)	269.7 \pm 9.3 (251.6 - 307.5)	-	450 \pm 33 (337 - 518)
February 2020	<i>c</i>	11.5 \pm 0.7 (10.2 - 12.9)	27.8 \pm 0.7 (27.0 - 29.7)	287.5 \pm 22.0 (256.9 - 350.0)	8.20 \pm 0.14 (7.94 - 8.53)	343 \pm 87 (130 - 519)
February 2020	<i>d</i>	10.2 \pm 0.6 (9.2 - 11.1)	21.4 \pm 0.0 (21.3 - 21.5)	314.5 \pm 15.9 (293.1 - 343.8)	8.27 \pm 0.04 (8.16 - 8.32)	347 \pm 30 (302 - 438)
Spring 2018	<i>F</i>	16.0 \pm 2.5 (13.2 - 19.1)	32.4 \pm 1.5 (30.8 - 34.2)	270.0 \pm 28.5 (245.3 - 308.2)	8.23 \pm 0.09 (8.11 - 8.33)	379 \pm 89 (279 - 495)
April 2018	<i>a</i>	15.0 \pm 0.7 (14.1 - 16.5)	31.5 \pm 0.0 (31.4 - 31.5)	265.8 \pm 16.4 (221.9 - 285.3)	8.17 \pm 0.03 (8.09 - 8.21)	390 \pm 40 (342 - 505)
April 2018	<i>b</i>	15.5 \pm 0.9 (14.1 - 16.9)	31.2 \pm 0.3 (30.3 - 31.6)	252.2 \pm 20.1 (200.3 - 279.4)	8.05 \pm 0.02 (7.98 - 8.09)	443 \pm 44 (371 - 551)
May 2019	<i>c</i>	17.1 \pm 1.8 (14.3 - 19.9)	33.7 \pm 0.2 (33.3 - 34.0)	287.5 \pm 78.4 (168.4 - 415.0)	8.78 \pm 0.43 (8.12 - 9.23)	135 \pm 165 (6 - 425)
-	<i>d</i>	-	-	-	-	-
Summer 2018	<i>F</i>	19.9 \pm 1.5 (18.2 - 21.7)	34.6 \pm 0.6 (34.0 - 35.2)	235.6 \pm 22.3 (204.4 - 253.1)	8.20 \pm 0.12 (8.05 - 8.34)	410 \pm 130 (270 - 572)
July 2018	<i>a</i>	22.9 \pm 1.6 (21.0 - 26.9)	34.2 \pm 0.3 (33.6 - 34.9)	249.4 \pm 35.9 (153.4 - 306.6)	8.22 \pm 0.07 (8.12 - 8.39)	385 \pm 60 (267 - 522)
July 2018	<i>b</i>	23.9 \pm 1.3 (21.9 - 26.1)	34.7 \pm 0.5 (33.9 - 35.6)	211.8 \pm 33.9 (139.7 - 291.9)	8.02 \pm 0.05 (7.89 - 8.12)	454 \pm 55 (374 - 590)
July 2019	<i>c</i>	23.5 \pm 2.5 (20.1 - 28.4)	36.8 \pm 2.3 (33.5 - 42.6)	206.8 \pm 58.5 (76.9 - 339.7)	8.31 \pm 0.23 (8.01 - 8.94)	242 \pm 116 (25 - 430)
July 2019	<i>d</i>	23.3 \pm 1.6 (21.2 - 28.1)	35.8 \pm 1.4 (33.3 - 38.1)	202.4 \pm 70.4 (108.8 - 314.4)	7.97 \pm 0.09 (7.84 - 8.11)	377 \pm 85 (250 - 508)
Autumn 2018	<i>F</i>	14.7 \pm 3.2 (10.3 - 17.5)	34.3 \pm 2.1 (30.6 - 35.4)	249.1 \pm 23.5 (225.0 - 284.4)	8.09 \pm 0.06 (8.04 - 8.18)	510 \pm 70 (403 - 580)
September 2018	<i>a</i>	18.9 \pm 1.5 (17.6 - 22.0)	35.2 \pm 0.1 (35.1 - 35.7)	267.4 \pm 25.9 (204.7 - 323.4)	7.98 \pm 0.07 (7.90 - 8.16)	460 \pm 58 (334 - 569)
September 2018	<i>b</i>	20.8 \pm 1.2 (18.8 - 23.1)	35.9 \pm 0.5 (35.1 - 36.9)	232.1 \pm 30.3 (153.8 - 275.3)	7.84 \pm 0.05 (7.74 - 7.94)	503 \pm 46 (422 - 630)
October 2019	<i>c</i>	17.1 \pm 0.9 (15.1 - 18.5)	35.0 \pm 0.3 (34.6 - 35.7)	194.3 \pm 24.4 (152.2 - 236.9)	7.82 \pm 0.04 (7.74 - 7.91)	622 \pm 57 (522 - 721)
October 2019	<i>d</i>	15.9 \pm 0.4 (15.4 - 16.8)	38.2 \pm 0.1 (38.0 - 38.4)	255.8 \pm 39.9 (210.6 - 331.9)	8.17 \pm 0.07 (8.07 - 8.28)	155 \pm 30 (110 - 218)

Table 3: Seasonal means (\pm SD) and ranges (min - max) of wind speed (km h^{-1}), gas transfer velocity (k_{660} , cm h^{-1}) and estimated water-atmosphere CO_2 flux (FCO_2 , $\text{mmol m}^{-2} \text{h}^{-1}$) values measured (i) once every two weeks in 2018 at station *F* (Coignot et al., 2020) and (ii) during each 24-h cycle from 2018 to 2020 at stations *a*, *b*, *c* and *d* in the present study. Air CO_2 concentrations used for FCO_2 calculations are: 408 ppm (stations *a*, *b* and *F* in 2018), 411 ppm (station *c* in spring 2019), 413 ppm (stations *c* and *d* in autumn 2019), 400 ppm (stations *c* and *d* in summer 2019) and 403 ppm (stations *c* and *d* in winter 2020; see M&M sections).

		Wind speed (km h^{-1})	k_{660} (cm h^{-1})		FCO_2 ($\text{mmol m}^{-2} \text{h}^{-1}$)	
			W22	RC01	W22	RC01
Winter 2018	<i>F</i>	19 \pm 10 (7 - 29)	7.45 \pm 6.30 (0.91 - 14.24)	11.29 \pm 8.83 (2.79 - 21.62)	1.21 \pm 2.10 (0.01 - 4.35)	1.85 \pm 3.18 (0.02 - 6.60)
March 2018	<i>a</i>	29 \pm 4 (22 - 37)	14.93 \pm 4.04 (8.63 - 24.68)	24.75 \pm 9.92 (12.08 - 52.53)	0.18 \pm 0.09 (0.09 - 0.42)	0.28 \pm 0.14 (0.13 - 0.69)
March 2018	<i>b</i>	32 \pm 12 (13 - 54)	21.39 \pm 15.99 (3.02 - 54.68)	66.17 \pm 87.62 (5.06 - 285.40)	0.52 \pm 0.65 (-0.31 - 2.13)	1.91 \pm 3.45 (-0.46 - 11.11)
February 2020	<i>c</i>	27 \pm 5 (16 - 34)	23.17 \pm 10.28 (6.93 - 39.66)	30.36 \pm 13.28 (9.21 - 50.44)	-0.40 \pm 0.77 (-2.02 - 0.94)	-0.68 \pm 1.31 (-3.58 - 1.58)
February 2020	<i>d</i>	15 \pm 5 (4 - 24)	6.92 \pm 3.17 (2.15 - 14.23)	9.46 \pm 4.34 (2.96 - 19.78)	-0.10 \pm 0.09 (-0.31 - 0.09)	-0.15 \pm 0.13 (-0.41 - 0.13)
Spring 2018	<i>F</i>	15 \pm 20 (11 - 20)	5.35 \pm 3.44 (2.41 - 9.32)	7.78 \pm 3.91 (4.43 - 12.40)	-0.06 \pm 0.26 (-0.40 - 0.22)	-0.09 \pm 0.34 (-0.54 - 0.30)
April 2018	<i>a</i>	33 \pm 7 (24 - 43)	24.01 \pm 9.56 (11.85 - 38.49)	51.64 \pm 32.90 (16.94 - 108.70)	-0.29 \pm 0.23 (-0.71 - 0.11)	-0.66 \pm 0.65 (-1.81 - 0.18)
April 2018	<i>b</i>	14 \pm 5 (4 - 22)	4.62 \pm 2.96 (0.35 - 10.44)	7.23 \pm 3.46 (2.55 - 13.62)	0.05 \pm 0.11 (-0.06 - 0.33)	0.08 \pm 0.15 (-0.08 - 0.46)
May 2020	<i>c</i>	18 \pm 7 (6 - 31)	8.99 \pm 6.31 (0.86 - 22.99)	12.86 \pm 9.09 (3.17 - 37.67)	-0.77 \pm 0.84 (-3.00 - 0.03)	-1.21 \pm 1.30 (-5.09 - 0.04)
-	<i>d</i>	-	-	-	-	-
Summer 2018	<i>F</i>	20 \pm 12 (4 - 32)	12.85 \pm 10.75 (0.36 - 24.50)	20.41 \pm 17.10 (2.56 - 40.89)	-0.09 \pm 0.31 (-0.34 - 0.31)	-0.08 \pm 0.49 (-0.50 - 0.51)
July 2018	<i>a</i>	11 \pm 3 (6 - 17)	3.50 \pm 1.97 (0.89 - 7.21)	6.35 \pm 2.13 (3.55 - 10.40)	-0.02 \pm 0.03 (-0.08 - 0.08)	-0.03 \pm 0.06 (-0.14 - 0.16)
July 2018	<i>b</i>	18 \pm 7 (7 - 30)	9.69 \pm 6.37 (1.29 - 23.89)	14.66 \pm 9.28 (4.15 - 39.18)	0.06 \pm 0.08 (-0.07 - 0.22)	0.09 \pm 0.13 (-0.11 - 0.37)
July 2019	<i>c</i>	13 \pm 4 (4 - 19)	4.56 \pm 2.59 (0.43 - 9.46)	7.59 \pm 2.92 (3.16 - 13.32)	-0.19 \pm 0.17 (-0.70 - 0.01)	-0.33 \pm 0.24 (-1.01 - 0.02)
July 2019	<i>d</i>	15 \pm 6 (2 - 23)	7.12 \pm 4.60 (0.11 - 14.77)	10.77 \pm 5.72 (2.41 - 20.96)	-0.09 \pm 0.18 (-0.43 - 0.20)	-0.12 \pm 0.27 (-0.60 - 0.29)
Autumn 2018	<i>F</i>	17 \pm 5 (11 - 22)	6.37 \pm 4.09 (2.25 - 10.83)	9.01 \pm 4.98 (4.08 - 14.46)	0.21 \pm 0.22 (-0.02 - 0.56)	0.30 \pm 0.29 (-0.02 - 0.75)
September 2018	<i>a</i>	11 \pm 6 (4 - 20)	3.55 \pm 3.27 (0.36 - 8.98)	6.21 \pm 3.66 (2.65 - 12.53)	0.09 \pm 0.09 (0.01 - 0.30)	0.15 \pm 0.11 (0.01 - 0.41)
September 2018	<i>b</i>	17 \pm 6 (11 - 32)	7.32 \pm 5.46 (2.90 - 24.27)	9.46 \pm 7.91 (4.02 - 42.48)	0.22 \pm 0.23 (0.07 - 1.00)	0.28 \pm 0.33 (0.05 - 1.75)
October 2019	<i>c</i>	35 \pm 8 (20 - 48)	27.39 \pm 11.52 (9.01 - 48.79)	63.79 \pm 42.95 (12.58 - 179.80)	2.03 \pm 1.17 (0.61 - 4.61)	4.86 \pm 4.22 (0.85 - 16.98)
October 2019	<i>d</i>	42 \pm 7 (31 - 54)	38.22 \pm 12.55 (20.54 - 62.36)	122.80 \pm 84.13 (34.76 - 325.40)	-3.43 \pm 1.09 (-6.03 - -1.79)	-10.91 \pm 7.35 (-31.46 - -3.03)

Table 4. Stepwise multilinear regression analyses to test the contribution of physicochemical variables on water pCO₂ variations through the percentage of explained variance (adjusted R²). Each selected multilinear model (p < 0.001, n = 1441) had the highest adjusted R² value with all variables explaining at least 5% of the pCO₂ variation. In bold is indicated the parameter explaining at least 50% of the pCO₂ variation. Input variables: DO-sat. (dissolved oxygen saturation level), T (water temperature), S (salinity) and TU (turbidity). The statistic (F) and adjusted R² (adj. R²) are given.

			Equations	F	adj. R ²
Winter	March 2018	<i>a</i>	pCO ₂ = 2227.7 - 3.7 DO-sat. - 19.9 T - 38.2 S	1004	67.7%
	March 2018	<i>b</i>	pCO ₂ = -373.8 + 21.1 S + 0.8 TU + 15.9 T	377	43.9%
	February 2020	<i>c</i>	pCO ₂ = 1460.5 - 158.3 T + 7.0 DO-sat.	1180	62.1%
	February 2020	<i>d</i>	pCO ₂ = 446.6 - 10.8 DO-sat. + 99.1 T	4723	86.8%
Spring	April 2018	<i>a</i>	pCO ₂ = 949.8 - 5.5 DO-sat.	8255	85.2%
	April 2018	<i>b</i>	pCO ₂ = 2542.0 - 4.0 DO-sat. - 61.7 S + 14.2 T	1923	80.0%
	May 2019	<i>c</i>	pCO ₂ = 21777.7 - 640.8 S - 0.6 DO-sat.	3668	83.6%
	-	<i>d</i>	-	-	-
Summer	July 2018	<i>a</i>	pCO ₂ = 747.2 - 3.2 DO-sat.	30524	95.5%
	July 2018	<i>b</i>	pCO ₂ = -2440.0 + 100.8 S - 25.4 T	3330	82.7%
	July 2019	<i>c</i>	pCO ₂ = 1961.5 - 46.7 S	9401	86.8%
	July 2019	<i>d</i>	pCO ₂ = -961.8 + 40.1 S - 1.2 DO-sat.	47983	98.5%
Autumn	September 2018	<i>a</i>	pCO ₂ = 923.1 - 4.1 DO-sat.	34905	96.4%
	September 2018	<i>b</i>	pCO ₂ = 782.0 - 2.8 DO-sat. + 1.1 TU	3066	81.0%
	October 2019	<i>c</i>	pCO ₂ = 1009.3 - 4.9 DO-sat.	8831	86.0%
	October 2019	<i>d</i>	pCO ₂ = 1932.7 + 1.6 DO-sat. - 122.7 T	3253	81.9%

CRediT author statement

Jérémy Mayen: Methodology, Software, Validation, Formal analysis, Investigation, Writing - Original - Draft, Writing - Review & Editing. **Pierre Polsenaere:** Conceptualization, Methodology, Software, Validation, Formal analysis, Investigation, Ressources, Writing - Review & Editing, Supervision, Project Administration, Funding acquisition. **Aurore Regaudie De Gioux:** Writing - Review & Editing, Supervision, Funding acquisition. **Christine Dupuy:** Validation, Writing - Review & Editing. **Marie Vagner:** Writing - Review & Editing, Funding acquisition. **Jean-Christophe Lemesle:** Ressources. **Benoit Poitevin:** Ressources. **Philippe Souchu:** Validation, Writing - Review & Editing, Supervision, Funding acquisition.

Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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