
Diatom artificial medium (DAM): a new artificial medium for the diatom *Haslea ostrearia* and other marine microalgae

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Abstract:

Artificial media are used in physiological studies of microalgae to maintain consistent conditions from one experiment to another and these media must be adapted to the needs of the organism studied. The artificial medium, in this case named diatom artificial medium (DAM), was designed to maintain long-term cultures of *Haslea ostrearia* and 19 other planktonic microalgae, and to allow physiological studies related to metal metabolism. The biomass and biochemical composition of *H. ostrearia* grown in the DAM and in a modified Provasoli medium were compared to assess the suitability of this new artificial medium for the culture of this diatom. The DAM provided sufficient nutrients to allow *H. ostrearia* to grow as efficiently as in the enriched seawater medium, without negative impact on metabolism. The DAM was tested with 19 other microalgae in order to widen its potential use, and 18 of the 19 showed a good adaptation to this medium. The chemical speciation of metals (Cd, Cu, Pb, Zn) was assessed using a speciation mathematical model. The presence of EDTA resulted in the total complexation of the trace metals implying that they were present in a sole chemical species in the DAM.

Keywords: Artificial culture medium - Microalgae - Diatom - Biochemical parameters - Metal speciation

33 **Introduction**

34 Natural seawater enriched with nutrients, trace metals and vitamins is commonly used
35 to maintain microalgal strains. The composition of seawater may be different for several
36 sampling sites and periods, thereby leading to less reproducible and reliable
37 experiments. Since 1890, media composed of seawater or seawater-like artificial
38 solutions have been developed to keep axenic microalgal cultures under controlled
39 laboratory conditions (Provasoli et al., 1957). Some artificial media were developed for
40 the study of marine microalgal physiology (Berland et al., 1973; Carvalho and Malcata,
41 2000; Lane and Morel, 2000; Reinfelder et al., 2000). Moreover, the bioaccumulation of
42 trace metals by microalgae was studied by using specific artificial media (Jensen et al.,
43 1982; Price and Morel, 1990; Ahner et al., 1994; Yee and Morel, 1996; Yan and Pan,
44 2002). The advantages of such media are an accurate control of the concentration and
45 speciation of metals in experimental conditions, and the opportunity to reproduce
46 further experiments under identical conditions.

47 Many studies about microalgae have focused on species largely used in aquaculture.
48 The diatom *Haslea ostrearia* (Simonsen) is involved in the oyster greening, which
49 consists in acquiring a blue green color on the gills and the labial palps (Ranson, 1937),
50 increasing value of oyster products. Robert (1983) obtained long term productive
51 cultures with this diatom after modifying the enriched seawater medium ES (Provasoli,
52 1968). In order to study some physiological aspects of this diatom, cultures were grown
53 using artificial media. The artificial medium ESAW (Harrison et al., 1980; Berges et al.,
54 2001) is based on an artificial seawater medium similar to that of Kester et al. (1967)
55 enriched with Provasoli's ES solution. Cultures of *H. ostrearia* in ESAW showed cell
56 deformation after several transfers (Mouget, pers. comm.). Moreau (1996) used the
57 artificial Aquil medium (Morel et al., 1979; Price et al., 1988) and the f/50 medium
58 derived from the enriched seawater f/2 medium (Guillard, 1982) for experiments
59 involving cultures of *H. ostrearia*. Cultures appeared to be more productive in the f/50
60 seawater medium than in the artificial medium, which implies specific nutritional
61 requirements for elements found in seawater but not in the artificial medium. Actually,
62 the medium Aquil designed was tailored to study metal metabolism in microalgae,
63 rather than a phytoplankton culturing medium in the first place (Morel et al., 1979).
64 Neuville and Daste (1971) used a synthetic medium for *H. ostrearia* in order to observe

65 production by this diatom of a specific blue-green pigment, called “marennine”, with
66 short term cultures. This medium was very rich in nutrients and trace elements
67 compared to other media (enriched natural water or artificial seawater).
68 The aim of this study was to develop an artificial medium suitable to maintain long term
69 productive cultures of *H. ostrearia*, and allowing physiological studies related to trace
70 metal metabolism. For this reason, the artificial medium DAM (Diatom Artificial
71 Medium) was developed after the Aquil model. Modifications were introduced
72 according to literature data about *H. ostrearia* and artificial media, and assays based on
73 the variation of the abundance of several elements (presence/absence, ratio). The ability
74 of the DAM to sustain *H. ostrearia* was tested with a culture that lasted for two and a
75 half months. For each growth cycle, the biomass and the total carbohydrate and protein
76 contents were compared with those obtained with a strain grown in enriched seawater
77 medium of ES type (Provasoli, 1968; Robert, 1983) usually used for *H. ostrearia*
78 cultures (Robert, 1983; Lebeau et al., 2000; Tremblin et al., 2000; Robert et al., 2002).
79 The DAM was also tested using nineteen different species of microalgae, essentially
80 diatoms, through growth cycles, in order to widen its potential use.
81 To understand metal/algal interactions, knowledge of speciation in the medium is
82 essential. The chemical speciation of essential (copper and zinc) and non essential
83 (cadmium and lead) metals was assessed using a speciation mathematical model.

84

85 **Methods**

86

87 *Composition of the media, DAM and ES1/3*

88 The composition of nutrient stock solutions and the DAM preparation protocol are
89 given in Table 1. The trace element solution (III) was acidified in order to keep metals
90 in solution and to prevent adsorption onto the flask. EDTA was added to the trace
91 element solution before FeCl₃ incorporation to avoid precipitation of ferric hydroxide
92 (Morel et al., 1979). The pH of the medium was adjusted at 7.8 using a 1M hydrochloric
93 acid solution. The filtered vitamin solution (IV) was added to the medium after being
94 sterilized by filtration. This vitamin solution can be prepared in advance, fractioned and
95 stored at -18°C.

Table 1

96 The composition of the modified enriched seawater medium ES1/3, derived from the ES
97 medium (Provasoli, 1968; Robert, 1983), is presented in Table 2 and is compared to the
98 final concentration of elements in DAM and in the ES medium described by Provasoli
99 in 1968. The ES1/3 enrichment solution results in the third of the ES enrichment for a
100 main part of elements and the vitamin solution is that described by Guillard and Ryther
101 (1962). The optimal biomass attained in ES and ES1/3 was within the same range; but
102 ES1/3 appeared to be more suitable for *H. ostrearia*. Whereas Robert (1978) showed a
103 depletion of the culture maintained in ES medium with a decrease of the biomass and an
104 increase of the number of deformed cells, this author did not mention any depletion in
105 cultures of *H. ostrearia* performed in ES1/3.

106

107 *Culture conditions*

108 All the algal strains used in these experiments were provided by the Nantes Culture
109 Collection (WDCM 856, ISOMer, Université de Nantes, France). All the experiments
110 were carried out at 17°C under 80 $\mu\text{mol photons}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ irradiance with a 14:10 hour
111 photoperiod. The cultures were grown either in 250 mL flasks with 150 mL of sterilised
112 medium (*Haslea crucigera* (NCC 32), *Haslea ostrearia* (NCC 143), *Odontella aurita*
113 (NCC 87), *Phaeodactylum tricornutum* (NCC 45), *Rhizosolenia setigera* (NCC 81),
114 *Skeletonema costatum* (NCC 52), *Thalassionema sp.* (NCC 69) and *Tetraselmis suecica*
115 (NCC 62)) or in 100 mL tubes with 25 mL of DAM (all the other species tested in this
116 study). Each flask/tube was inoculated with the parent culture maintained in enriched
117 seawater at an initial concentration of 5000 $\text{cells}\cdot\text{mL}^{-1}$ (cycle n). When the cultures
118 reached the stationary phase, the cells were transferred into a new flask containing fresh
119 medium at 5000 $\text{cells}\cdot\text{mL}^{-1}$ (cycle n+1).

120 In order to test the capacity of the new medium to maintain the diatom *H. ostrearia*, a
121 strain was grown in the DAM and in the ES1/3 during five consecutive cycles. For each
122 culture condition, five replicates were realised.

123 The DAM was developed to grow *H. ostrearia* but its suitability for growing other
124 species of microalgae was tested to widen its potential use for physiological
125 experiments. Cultures in tubes or in flasks were run in a single example through 10 or 6
126 transfers depending on species. The experiment duration (4 months) allowed 10 cycles
127 to take place for cultures carried out in tubes while only 6 cycles were achieved by

128 cultures carried out in flasks. The experiments included several repeated microscopic
129 observations to identify shape modifications and to estimate number of cells per mL.

130

131 *Culture monitoring*

132 In order to compare the strain grown in the DAM with those grown in the modified ES
133 medium (ES1/3), growth rate and biochemical composition were determined for *H.*
134 *ostrearia* cultures. The cells were counted daily from the beginning of the exponential
135 phase to the stationary phase using a Nageotte hematocytometer. At this step, each culture
136 was filtered on Whatman GF-F membrane for analysis of a few biochemical parameters.
137 The physiological status of the microalgae was estimated using carbohydrate, pigment
138 (carotenoids and chlorophyll *a*) and protein concentrations determined by the
139 spectrophotometric methods of Dubois et al. (1956), Lorenzen (1967) and Lowry et al.
140 (1951) respectively.

141

142 *Metal speciation in the DAM*

143 A speciation mathematical model (MOCO) developed by IFREMER was used to
144 calculate essential metals (Cu and Zn) speciation in the DAM. Development and
145 applications of this speciation model has already been described (Gonzalez et al., 2001a;
146 2001b; Laurier et al., 2003). The complexation constants used for the calculation of the
147 chemical species taken into account were compiled by Morel and Hering (1993).
148 Calculations were also performed with the addition of non-essential metals (Cd and Pb)
149 to determine their speciation in the DAM. Simulations were run with the initial element
150 concentrations at pH 7.8, either with or without EDTA. Speciation of EDTA was
151 determined and the competition of metals with Ca^{2+} and Mg^{2+} was taken into account.

152

153 *Statistical analysis*

154 The normality and the homogeneity of variances being checked, the biomass and
155 biochemical parameters obtained for cultures grown in DAM and in enriched seawater
156 ES1/3 were compared using Student's t-test and one way analysis of variance
157 (ANOVA) with a significance level at $\alpha < 0.05$ using Sigma Stat 2.0 (Jandel Scientific)
158 software. *A posteriori* tests (Tukey tests) were run with ANOVA data.

159

160 **Results**

161

162 *Growth experiments*

163 Fig. 1 shows the growth curves of *H. ostrearia* cultures carried out in the DAM and in
164 the ES1/3. For the five cycles presented in Fig. 1, the strain seems to grow better in the
165 DAM than in the modified ES medium, but the statistical analysis did not put forward
166 any significant difference with the exception of the third ($p < 0.001$) and the fifth
167 transfers ($p < 0.001$). The total carbohydrate and protein contents of *H. ostrearia* (Fig.
168 2) were stable with time for each tested medium (DAM and ES1/3). No significant
169 difference was shown between these contents at each sampling date. The carotenoid and
170 chlorophyll *a* cell concentrations (Fig. 3) were significantly higher ($p < 0.05$) in cells
171 grown in the DAM than those in cells grown in the ES1/3 since the second transfer. The
172 repeated microscopic observations did not reveal a significant proportion of deformed
173 cells in DAM compared to ES1/3.

Fig. 1

Fig. 2

174

175 *Growth tests using various microalgae*

176 The results of the cultures carried out with 19 other microalgal species in DAM are
177 summarised in Table 3, which lists the species' names, the NCC strain codes and the
178 number of successful cycles.

Fig. 3

Table 3

179 All species kept in tubes grew successfully over ten cycles. Among the strains kept in
180 flasks, only *Rhizosolenia setigera* showed difficulties to grow in the DAM. Neither
181 shape modification nor growth decrease were noticed for the five other species kept in
182 flasks after 6 cycles achieved. Microscopic observations did not put forward any
183 modification of algal cells.

184

185 *Chemical speciation of Cd, Cu, Pb and Zn*

186 Calculations of Cd, Cu, Pb and Zn speciation are presented in Table 4. Without EDTA,
187 each considered metal had specific affinities for inorganic ligands. The major amount of
188 copper was associated to carbonates (97%) whereas zinc was mainly associated to
189 phosphates (90%). The cadmium was mainly (90%) associated to chlorides. Lead was
190 present as two major species, phosphates and chlorides for 50% and 40 % of the total
191 lead respectively.

Table 4

192 The addition of EDTA resulted in a total complexation (100%) of the considered metals
193 with this molecule. This complexation was assessed with concentration values from 10^{-8}
194 mol.L⁻¹ to 10^{-6} mol.L⁻¹ of the different metals.

195

196 **Discussion**

197 The Diatom Artificial Medium DAM allows long term and productive culturing *H.*
198 *ostrearia* in controlled conditions. The cell shape and the global biochemical
199 composition appeared to be maintained after several transfers, which implies that DAM
200 contains the various elements in sufficient amounts for the optimal development of this
201 diatom. Requirement in silicon and copper appeared to be a main factor in the
202 development of the diatom. The slight increase in cell density observed from a transfer
203 to another could be explained by an acclimation of the strain to the medium supplies,
204 since carbohydrate and protein contents are similar for all transfers. Cultures of *H.*
205 *ostrearia* in larger volumes (10 L, 25 L and 200 L) have been attempted at our
206 laboratory and the biomass achieved was as high as that obtained in enriched seawater
207 medium (cultures realised in ES1/3 and F/2 media, unpublished data). Consequently,
208 DAM has to be considered as a well-adapted artificial medium for *H. ostrearia* culture.
209 Furthermore, DAM allowed the culture of 18 species of microalgae through several
210 transfers, other than *H. ostrearia*, even though the artificial medium did not seem to be
211 convenient for the strain of *Rhizosolenia setigera* used in this experiment. It is
212 noteworthy that this diatom did not grow either in Aquil beyond 3 transfers (Morel et
213 al., 1979).

214 The speciation model showed that, in the absence of EDTA in DAM, the speciation of
215 the considered metals was similar to that reported for seawater (Florence, 1982;
216 Fernando, 1995, Stumm and Morgan, 1996). The presence of EDTA resulted in the
217 complexation of metals from low concentrations (encountered in natural ecosystems) to
218 higher concentrations (used in experimental studies). A low bioavailability of metals
219 decreases their direct toxicity. Using an adapted ratio of EDTA/metals, it is possible to
220 control the amount of metals bioavailable to microalgae and to mimic the low
221 concentrations found in natural environments. Moreover, when EDTA was not added to
222 DAM, the four tested metals were presented as various chemical species, whereas, in
223 EDTA-added DAM, these metals were essentially in a sole species. The interaction

224 metals/microalgae is, thus, dependant on the dissociation constants of the complexes
225 EDTA-metals.

226 In conclusion, the medium DAM can be used for growing various microalgae in
227 medium term and for studying metal impact on the physiology of these microalgae.

228 Using this artificial medium would allow to study, in controlled conditions, the potential
229 bioaccumulation of metals in *H. ostrearia* (absorption and adsorption of metals,
230 kinetics) and to evaluate their impact on the growth and the culture quality.

231

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237

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329 **Figure legends**

330

331 Fig. 1 Growth curves of *Haslea ostrearia* maintained for two and a half months
332 in the Diatom Artificial Medium (◆) and in the enriched seawater
333 medium ES1/3 (×). Mean ± standard deviation.

334

335 Fig. 2 Concentrations in carbohydrates and proteins per cell for culture of
336 *Haslea ostrearia* maintained for two and a half months in the Diatom
337 Artificial Medium and in the enriched seawater medium ES1/3. Mean ±
338 standard deviation.

339

340 Fig. 3 Concentrations of carotenoids and chlorophyll *a* per cell for cultures of
341 *Haslea ostrearia* maintained for two and a half months in the Diatom
342 Artificial Medium and in the enriched seawater medium ES1/3. Mean ±
343 standard deviation.

344

1 Table 1 Preparation of the Diatom Artificial Medium (DAM).

2

Composition of stock solutions	
Solution I: Ion solution	1 liter
H ₃ BO ₃	1.50 g
KBr	5.00 g
KCl	35.00 g
NaF	0.15 g
NaHCO ₃	10.00 g
SrCl ₂ -6H ₂ O	0.85 g
Solution II :Nutrients (Three different solutions are prepared and stored at 6°C)	1 liter of each solution
NaH ₂ PO ₄ -H ₂ O	1.38 g
NaNO ₃	25.50 g
Na ₂ SiO ₃ -9H ₂ O	28.40 g
Solution III :Trace Elements (in HNO₃ 0.01M)	1 liter
CoCl ₂ -6H ₂ O	1mL of a 10 g.L ⁻¹ solution
CuSO ₄ -5H ₂ O	1mL of a 9.80 g.L ⁻¹ solution
MnCl ₂ .4H ₂ O	1mL of a 180 g.L ⁻¹ solution
Na ₂ MoO ₄ .2H ₂ O	1mL of a 6.30 g.L ⁻¹ solution
Na ₂ SeO ₃ -5H ₂ O	1mL of a 0.85 g.L ⁻¹ solution
NiCl ₂ -6H ₂ O	1mL of a 0.74 g.L ⁻¹ solution
ZnSO ₄ -7H ₂ O	1mL of a 22 g.L ⁻¹ solution
FeCl ₃ -6H ₂ O	3.15 g
Na ₂ EDTA-2H ₂ O	4.36 g
Solution IV: Vitamin solution (After sterilization)	1 liter
Biotin	10mL of a 0.1g.L ⁻¹ solution
Cyanocobalamin	1mL of a 1 g.L ⁻¹ solution
Thiamine-HCl	0,20 g
Composition of the DAM	
Salts	
NaCl	20.570 g
Na ₂ SO ₄	3.067 g
CaCl ₂ -2H ₂ O	1.150 g
MgCl ₂ -6H ₂ O	11.100 g
Solution I: Ion solution	15 mL
Solution II: Nutrients	
NaH ₂ PO ₄ -H ₂ O	2 mL
NaNO ₃	1 mL
Na ₂ SiO ₃ -9H ₂ O	2 mL
Solution III: Trace Elements	2 mL
Solution IV: Vitamin solution (After sterilization)	1 mL
Deionised water	<i>Fill up to 1 liter</i>

3

1 Table 2 Final concentrations of elements in DAM, ES and ES1/3 media.

2

	DAM Final concentration (mol.L ⁻¹)	ES (Provasoli, 1968) Final concentration (mol.L ⁻¹)	ES1/3 (Robert, 1983) Final concentration (mol.L ⁻¹)
Salts			
NaCl	3.52×10 ⁻¹		
Na ₂ SO ₄	2.16×10 ⁻²		
CaCl ₂ -2H ₂ O	7.82×10 ⁻³		
MgCl ₂ -6H ₂ O	5.46×10 ⁻²		
Solution I :Ion solution			
H ₃ BO ₃	3.64×10 ⁻⁴	9.22×10 ⁻⁵	3.07×10 ⁻⁵
KBr	6.30×10 ⁻⁴		
KCl	7.04×10 ⁻³		
NaF	5.36×10 ⁻⁵		
NaHCO ₃	1.79×10 ⁻³	9.52×10 ⁻⁴	9.52×10 ⁻⁴
SrCl ₂ -6H ₂ O	4.61×10 ⁻⁵		
Solution II :Nutrients			
NaH ₂ PO ₄ -H ₂ O	2.00×10 ⁻⁵		
Na ₂ Glycerophosphate-6H ₂ O		4.63×10 ⁻⁵	1.54×10 ⁻⁵
NaNO ₃	3.00×10 ⁻⁴	8.24×10 ⁻⁴	2.75×10 ⁻⁴
Na ₂ SiO ₃ -9H ₂ O	2.00×10 ⁻⁴	5.00×10 ⁻⁵	5.00×10 ⁻⁵
Solution III :Trace			
CoCl ₂ -6H ₂ O	8.41×10 ⁻⁸		
CoSO ₄ -5H ₂ O		8.54×10 ⁻⁸	2.85×10 ⁻⁴
CuSO ₄ -5H ₂ O	7.85×10 ⁻⁸		
FeCl ₃ -6H ₂ O	2.33×10 ⁻⁵	9.06×10 ⁻⁷	3.02×10 ⁻⁷
Fe(NH ₄) ₂ (SO ₄) ₂ -6H ₂ O		8.95×10 ⁻⁶	2.98×10 ⁻⁶
MnCl ₂ -4H ₂ O	1.82×10 ⁻⁶		
MnSO ₄ -4H ₂ O		3.26×10 ⁻⁶	1.09×10 ⁻⁶
Na ₂ MoO ₄ -2H ₂ O	5.21×10 ⁻⁸		
Na ₂ SeO ₃ -5H ₂ O	6.46×10 ⁻⁹		
NiCl ₂ -6H ₂ O	6.30×10 ⁻⁹		
ZnSO ₄ -7H ₂ O	1.53×10 ⁻⁷	3.83×10 ⁻⁷	1.28×10 ⁻⁷
Na ₂ EDTA-2H ₂ O	2.34×10 ⁻⁵	2.23×10 ⁻⁴	7.43×10 ⁻⁵
Solution IV: Vitamin			
Biotin	4.09×10 ⁻⁹	4.09×10 ⁻⁹	8.19×10 ⁻⁸
Cyanocobalamin	7.38×10 ⁻¹⁰	1.48×10 ⁻⁹	2.95×10 ⁻⁹
Thiamine-HCl	5.93×10 ⁻⁷	2.96×10 ⁻⁷	2.96×10 ⁻⁶
Optimal Biomass Achieved (10 ³ Cells.mL ⁻¹)	Present study	230-250 at long term, decrease of the dividion rate of the culture and increase of the deformed cell ratio	200-250 (Robert, 1983, Present study)

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1 Table 3 Transfers of different microalgae in DAM (Grown in 25 mL tubes,
 2 except species noted (*) grown in 250 mL flasks)

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Species	Clone	Successful Transfers
Bacillariophyceae		
<i>Amphora hyalina</i>	NCC 1 D-Am.hy.	10
<i>Bacillaria paradoxa</i>	NCC 6 D-Ba.pa.	10
<i>Chaetoceros sp.</i>	NCC 8 D-Ch.sp.	10
<i>Coscinodiscus granii</i>	NCC 11 D-Co.gr.	10
<i>Entomoneis alata</i>	NCC 17 D-En.al.	10
<i>Haslea crucigera*</i>	NCC 32 D-Ha.cr.	6
<i>Heliotheca thamesis</i>	NCC 59 D-He.th.	10
<i>Navicula ramosissima</i>	NCC 73 D-Na.ra.	10
<i>Nitzschia compressa</i>	NCC 38 D-Ni.co.	10
<i>Odontella aurita*</i>	NCC 87 D-Od.au.	6
<i>Phaeodactylum tricorutum*</i>	NCC 45 D-Ph.tr.	6
<i>Pleurosigma intermedium</i>	NCC 78 D-Ple.in.	10
<i>Rhizosolenia setigera*</i>	NCC 81 D-Rh.se.	2-†
<i>Skeletonema costatum*</i>	NCC 52 D-Sk.co.B2	6
<i>Thalassionema sp.*</i>	NCC 69 D-Thn.sp.	6
Other microalgae		
<i>Dunaliella sp.</i>	NCC 14 C-Du.sp.	10
<i>Isochrysis sp.</i>	NCC 24 Pry-Is.sp.	10
<i>Porphyridium cruentum</i>	NCC 49 R-Po.cr.	10
<i>Tetraselmis suecica*</i>	NCC 62 Pra-Te.Su.	10

4 *cultures performed in flasks.

5 †no algal growth after transfer

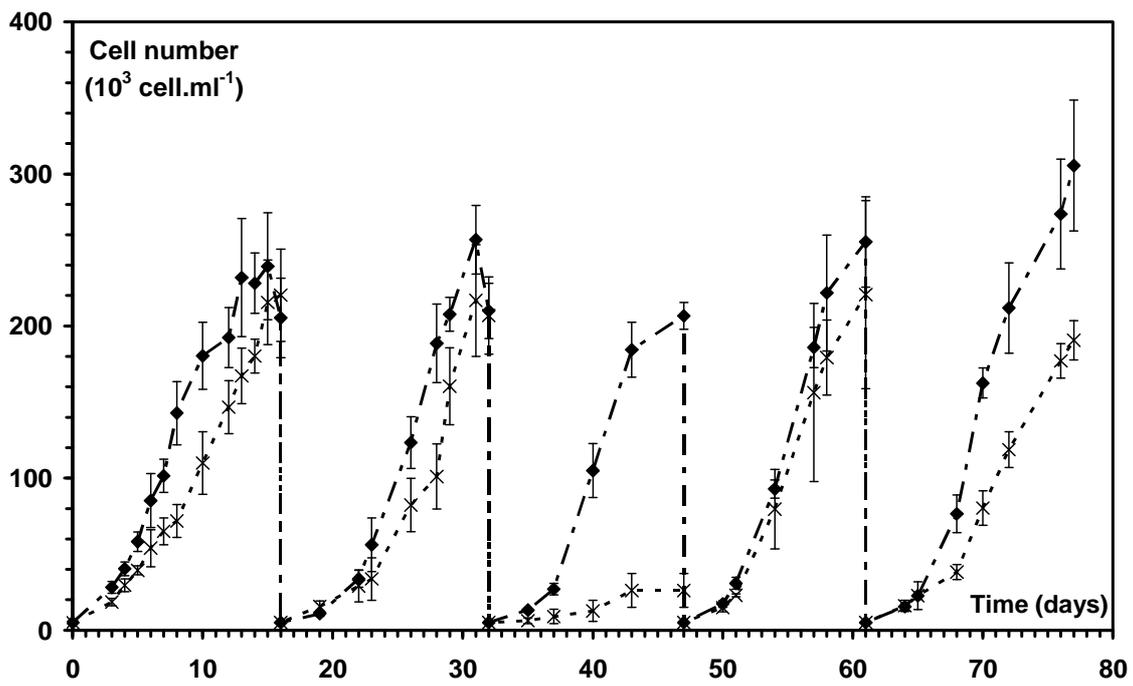
1 Table 4 Calculations of metal speciation (Cu, Zn, Cd, Pb) in the artificial
 2 medium DAM with and without presence of EDTA (in percentage of the
 3 total concentration; nc: non calculated)

Association metal-ligand (%)	Cu ^a	Zn ^a	Cd ^b	Pb ^b
<i>Without EDTA</i>				
Free metal (M ⁿ⁺)	0.030	6.412	3.723	5.515
Chloride (4.8×10 ⁻¹ mol.L ⁻¹)	0.011	2.340	95.233	39.177
Hydroxyde	0.014	0.210	0.006	2.367
Phosphate (2×10 ⁻⁵ mol.L ⁻¹)	2.643	89.889	n.c.	48.869
Carbonate (1.8×10 ⁻³ mol.L ⁻¹)	97.293	n.c.	n.c.	n.c.
Sulfate (2.2×10 ⁻² mol.L ⁻¹)	0.009	1.149	1.037	4.072
<i>With EDTA^c</i>				
EDTA	100	100	100	100
Other species	< 1×10 ⁻⁸	< 1×10 ⁻⁸	< 1×10 ⁻⁸	< 1×10 ⁻⁸

5 ^a with concentrations of the artificial medium DAM: [Cu] 7.85×10⁻⁸ mol.L⁻¹ and [Zn]
 6 1.53×10⁻⁷ mol.L⁻¹

7 ^b with an addition of 10⁻⁸ mol.L⁻¹ of total cadmium and lead to the DAM

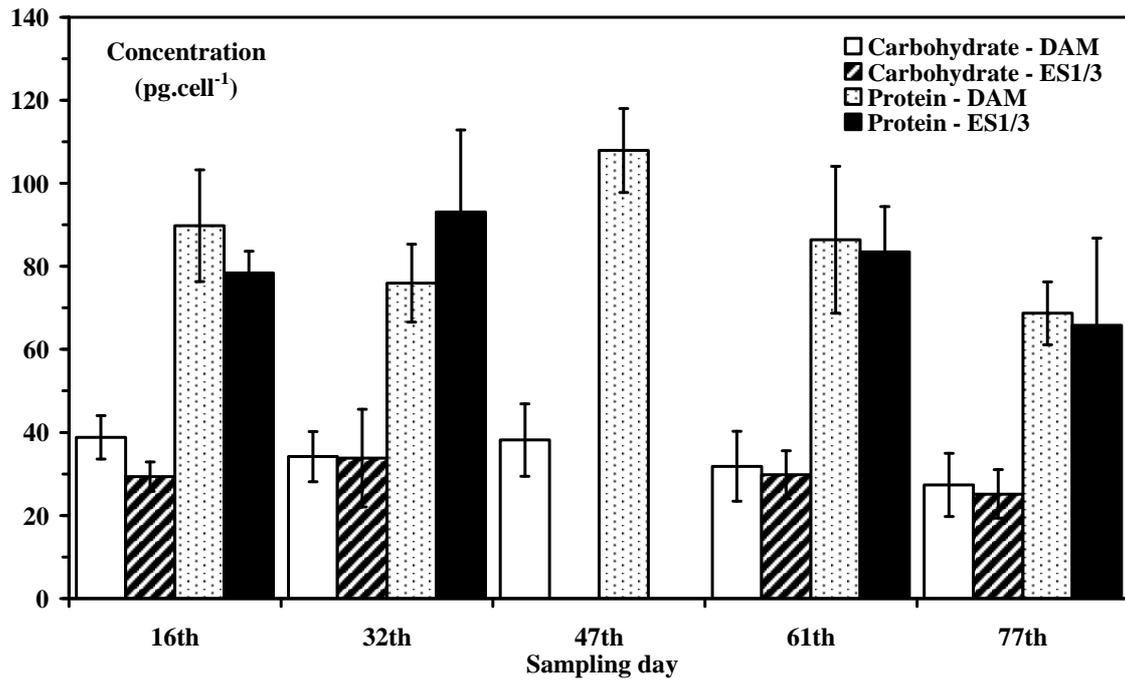
8 ^c major species of EDTA(Y): HY³⁻ (97.5%), H₂Y²⁻ (2.5%)



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Fig. 1 Gagneux-Moreaux, Moreau, Gonzalez, Cosson

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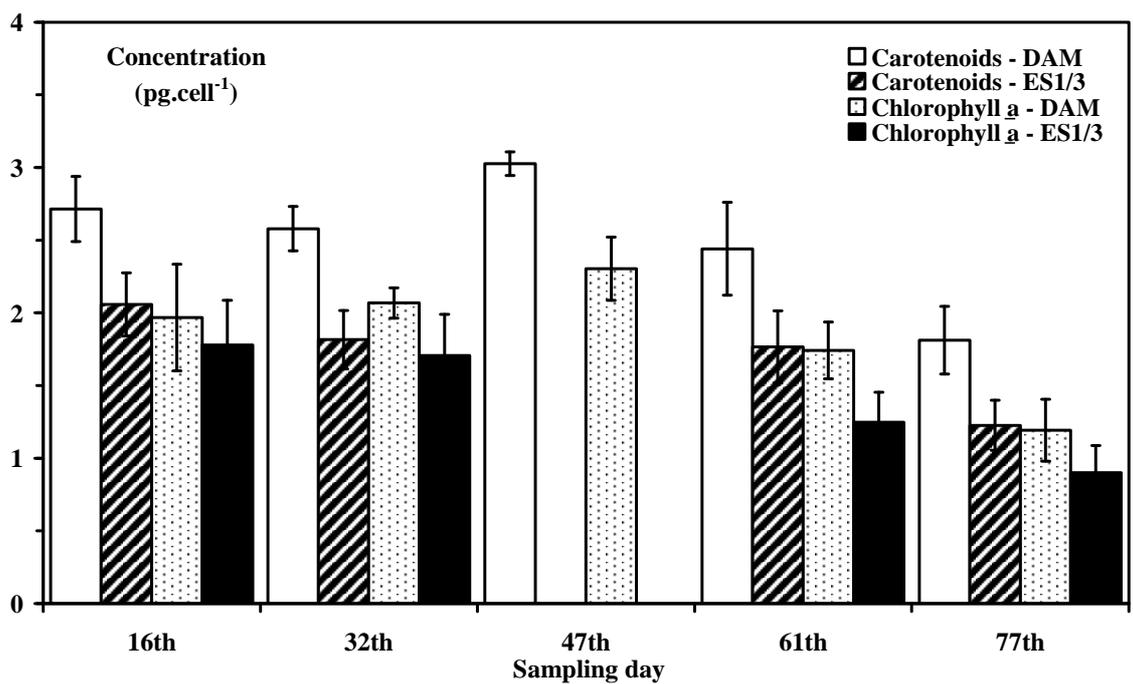


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Fig. 2 Gagneux-Moreaux, Moreau, Gonzalez, Cosson

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Fig. 3 Gagneux-Moreaux, Moreau, Gonzalez, Cosson